

1 **Molecular dynamics simulations of nanosheets of polymeric carbon nitride and**
2 **comparison with experimental observations**

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15 **Abstract**

16 A computational study of the properties of polymeric carbon nitride using molecular
17 dynamics is presented. The analysis of ideal infinite-extent sheets permits to evaluate the
18 effect of temperature on the network of hydrogen bonds responsible for the linkage of the
19 material. The weakening of this binding mechanism at sufficiently high temperatures,
20 together with the inter-layer interactions characteristic of this type of 2D materials, is
21 shown to determine the conformation properties of polymeric carbon nitride at the
22 nanoscale. The results obtained from the simulation of finite samples in the canonical
23 ensemble at varying temperatures are consistent with those from the characterization of
24 our experimentally synthesized samples. Hydrogen bonding between adjacent polymer

25 ribbons leads this process and is the cause of the typical crumpled structure of this
26 material.

27 **Keywords:** carbon nitride; conformation; molecular dynamics; MOPAC; nanosheets

28 **1. INTRODUCTION**

29 Polymeric carbon nitride has attracted large interest as a precursor for the synthesis of
30 super-hard carbon nitride phases (1-10). This material finds many applications in the wide
31 field of sensors due to the combination of its chemical sensitivity with its optical and
32 semiconductor properties (11-13). Also, monolayer polymeric carbon nitride forms a new
33 family of wide-gap semiconducting materials, in which the gap can be tuned by small
34 variations in the stress applied to the layers or by the presence of adatoms (14).

35 The various existing synthesis routes provide predominantly amorphous samples
36 structured at the nanoscale level as graphite-like (15), nanosheets (16-19), nanotubes (20),
37 or nanopillars (21-22). They can also be structured in mesoporous conformations that lead
38 to new applications, like the generation of nanostructured metal nitrides (23).
39 Conformation as single sheets that can be exfoliated from graphitic carbon nitride samples
40 (24) or directly grown (25) permits to increase its efficiency as a catalyst. The detailed
41 arrangement of this material is of most relevance for the specific applications and strongly
42 depends on the ambient conditions of the synthesis (16-19) and on other parameters, such
43 as the precursor type (25). Understanding of the detailed supramolecular mechanisms that
44 rule the conformation of this material is thus essential for the optimization of the synthesis
45 procedures and the development of new related materials and applications.

46 In this work, we perform a study of the conformation properties of polymeric carbon
47 nitride sheets under different ambient conditions using atomistic molecular dynamics
48 (MD) simulations. The analyses address both the general intrinsic properties of ideal
49 infinite-extent 2D polymeric carbon nitride sheets and the evolution in the structural

50 organization of finite size samples; specifically, the routes towards the formation of
51 crumpled nanosheets.

52 MD (26) is a useful tool in the study of the conformational transitions of very large
53 molecular and supramolecular systems like, for instance, protein folding (27). It provides
54 not only the final minimum energy structures, but the MD trajectories also afford the
55 detailed pathways and a valuable insight into the underlying mechanisms determining the
56 evolution. Molecular dynamics simulations have been extensively used in the study of the
57 intrinsic properties of 2D materials, particularly graphene (28-31) and related materials
58 like graphane (32) or multilayer graphene (33). These systems are especially interesting
59 from a theoretical point of view since, according to the Mermin-Wagner theorem (34),
60 long-wavelength fluctuations destroy any crystalline order in pure 2D systems.
61 Nevertheless, 2D membranes can exist in a 3D space, where the displacements are not
62 constrained in two dimensions.

63 Graphene-like materials can be modelled as polymerized or tethered membranes,
64 which have been subject of intense theoretical investigations (35-37) and which are
65 characterized by the fact that bonds linking the membrane sites are unbreakable. Even
66 within the physical constraint of self-avoidance, these 2D membranes typically undergo a
67 transition to a crumpled state (or a tubular state in the presence of anisotropy (38)),
68 whereas a flat state, with strong height fluctuations, is also possible due to the coupling
69 between bending and stretching modes. On the other hand, polymeric carbon nitride sheets
70 are characterized by their binding through a network of relatively weak hydrogen bonds,
71 far from the ideal unbreakable condition, that links the melon linear arrays of the basic
72 heptazine building blocks. Using MD simulations, we find that the morphological
73 properties of the material under study are predominantly determined by the behaviour of
74 the hydrogen bonds under the varying ambient conditions and the inter-planar interactions

75 between the heptazine units. In spite of the limited size of our simulation cells when
76 compared with the actual spatial extent of the investigated features, the results of the
77 computations are consistent with the experimental results.

78 **2. METHODS**

79 **2.1. Experimental**

80 2.1.1. *X-ray diffraction measurements*

81 X-ray diffraction patterns were obtained by means of a powder diffractometer ENRAF-
82 NONIUS FR590, detector INEL CPS 120 (BRUKER AXS, The Netherlands), with
83 monochromator-quartz discriminator, Debye-Scherrer geometry, using Cu K α radiation,
84 and glass capillaries for sample mounting. The samples were ground in an agate mortar
85 and sifted. The measurements always lasted for 1 hour, and crystalline silicon was used
86 as a standard.

87 2.1.2. *TEM and SEM characterization*

88 Transmission electron microscopy (TEM) and scanning electron microscopy (SEM)
89 analyses were performed with a JEOL JEM-FS2200 HRP and a JEOL JSM-820 SEM
90 microscope (JEOL, Japan), respectively.

91 **2.2. Computational**

92 The MD simulations were performed using the large-scale atomic molecular massively
93 parallel simulator (LAMMPS) (39) using the DREIDING force field (40). This generic
94 model includes parameters for H and the non-metallic main-group elements, plus a few
95 metals. It specifically takes into account the effect of hydrogen bonds, which are
96 particularly relevant for the material under analysis.

97 The data used for the MD simulations were built from the 2D ground state geometry of
98 the PM6 Hamiltonian (41), calculated using MOPAC (42). The DREIDING force field
99 polymer layers used as initial conditions for the MD computations were relaxed using

100 LAMMPS, both as finite size and periodic infinite-extent sheets. The minimized
101 geometries yielded $a=19.240 \text{ \AA}$, $b=13.365 \text{ \AA}$ and $\gamma=90.000^\circ$. The values of b and γ were
102 in excellent agreement with those derived from semi-empirical quantum chemistry
103 calculations and previously reported in (19), with a somewhat larger deviation in the value
104 of a . MD optimized geometries displayed very good planarity, very similar to that of the
105 semi-empirical computations. Two nearly square sheets were generated with 7776 and
106 31104 atoms and spanning 13×9 and 24×18 unit cells of the 2D polymer, respectively.
107 Their approximate sizes were $159.8 \text{ \AA} \times 159.5 \text{ \AA}$ and $319.7 \text{ \AA} \times 319.0 \text{ \AA}$, respectively.

108 The MD simulations of infinite 2D sheets were performed in the isothermal-isobaric
109 (NPT) ensemble with periodic boundary conditions. The calculations for finite size
110 systems, were started with well-formed one, three or six nearly square polymer sheets,
111 and were performed in the NVT ensemble. The equations of motion used in LAMMPS
112 were those of Shinoda, Shiga, and Mikami (43) and the time integration schemes closely
113 followed those derived in (44). Thermostating and barostating were implemented using
114 Nosé-Hoover algorithm.

115 **3. RESULTS AND DISCUSSION**

116 **3.1. MD Simulations**

117 The properties of single sheets of polymeric carbon nitride are first analysed. These are
118 the basic building blocks of three-dimensional arrangements (15, 19) and also the desired
119 final products for many applications. These arrangements can potentially be obtained by
120 exfoliation of more complex materials or by direct growth. The study of the response of
121 infinite sheets to changes in temperature and pressure permit to quantify, in particular, the
122 behaviour of the hydrogen bond network as the temperature grows. The results obtained
123 are consistent with subsequent studies on the processes responsible for the morphological
124 organization of the material under treatment at high temperatures.

125 3.1.1. *Infinite sheets*

126 The intrinsic properties of ideal, infinite extent, 2D arrangements were addressed using
127 periodic boundary conditions in the NPT ensemble. Final trajectory equilibrium states for
128 simulations at zero pressure and $T=100$ K and $T=700$ K are displayed in Fig. 3. The results
129 show how a finite temperature induced height fluctuations in the sheet, similarly to the
130 case of graphene (32-33). The average absolute deviations from the surface level h are
131 shown in Fig. 2(a) for the range between 100 K and 1000 K. The results for the two sheets
132 of different size were very similar, with the same approximate linear dependence of h in
133 that range. As opposed to graphene, these height fluctuations have an important effect on
134 the structural integrity of the layer through their impact on the hydrogen bonds network,
135 as it can be appreciated by comparing subplots (a) and (b) with (c) and (d), respectively,
136 in Fig. 1. This can be clearly appraised in the dependence of the energy contributed by
137 hydrogen bonds normalized with the total number of heptazine units in the simulation
138 domain shown in Fig. 2(b). These results show some dependence with the size of the
139 sample under study. The contribution of the hydrogen bond network to lower the energy
140 of the system decreases as temperature is increased and part of these bonds are broken.
141 Further, the rate of change of this energy with T saturates at higher temperatures due to
142 the finite number of total hydrogen bonds.

143 The structural integrity of other non-hydrogen-bonded 2D materials is preserved under
144 very large in-plane uniaxial or biaxial stress, resulting in a net change of the potential
145 energy in the membrane (32). In the polymeric carbon nitride presented herein,
146 simulations conducted even at low temperature show that increasing the applied finite
147 uniaxial or biaxial in-plane stress eventually results in the collapse of the 2D polymer at a
148 finite time that depends on the applied pressure. The atom displacements in the onset of
149 the instabilities resemble those associated to the low frequency lattice modes of the 2D

150 polymer (19), which may play a key role in the nascent growth of the instabilities.

151 3.1.2. *Finite size samples*

152 The simulations performed on finite size samples show that, in general, the dynamic
153 trajectories leading to their final conformations are driven by two main effects: the
154 previously described weakening of the hydrogen bonds network as temperature increases,
155 plus the plane-to-plane interactions that take place after polymer sections are loosened or
156 released from their hydrogen bond restraints. These two combined effects produce the
157 folding and piling of the material layers in different manners, determining the final
158 arrangement of the material. The detailed evolution depends not only on the ambient
159 conditions, but also on the amount and the characteristics of the initially interacting
160 material.

161 The conformation dynamics of a finite single sheet of polymeric carbon nitride were
162 analysed from computations performed in the canonical ensemble. The final states of the
163 calculations of 500 psec evolutions of a layer with 13×9 unit cells in a volume of
164 $200 \times 200 \times 200 \text{ \AA}^3$ at different temperatures are shown in Fig. 3. In all cases, equilibration
165 of the structures took place within the first 100 psec.

166 For the lowest temperatures in the sequence, 500 K and 600 K, the sheet essentially
167 kept its integrity. At 500 K only a very small defect in the 2D structure was found,
168 localized in one of the corners of the sheet, and corresponded to the disordered two-layer
169 piling of a section spanning very few heptazine units. At 600 K, a defect (a bilayer stripe)
170 could be identified. The dynamics of its formation, starting with the folding of one of the
171 corners and the sliding of the feature along the 2D structure, revealed a very interesting
172 balance between the inter-heptazine interaction and its hindrance by the hydrogen bonds
173 at this temperature.

174 At 700 K, the repetition of the folding and propagation of defect stripes produced the
175 packing of several layers. This disordered and non-homogeneous piling, in turn, induced
176 some curvature in the formed multilayer structures. As the temperature was increased, the
177 folding of the structure and the packing of multiple layers was facilitated by the
178 debilitation of the network of hydrogen bonds. The route to the layer piling then included
179 the break-up of stripes of material and their layer-to-layer interaction. As the hydrogen
180 bond network weakened with the temperature growth, the prominence of inter-planar
181 interaction increased and pilings with larger numbers of layers were produced.

182 When similar computations were performed on a planar system spanning 26×18 unit
183 cells, this larger sheet seemed to be more robust against the folding process at 700 K, with
184 morphological modifications that were limited to a somewhat smaller fraction of the total
185 sheet surface than in the previous case. This can be explained by the fact that the initiation
186 of the dynamical evolution eventually leading to the stacking and crumpling of the sheet
187 would be facilitated by the irregularities in the system and the processes would become
188 less likely in the more homogeneous system. At higher temperatures (900 K), the initial
189 breakup of the hydrogen links between 1D chains and their inter-planar interaction led to
190 the formation of a thread of 1D chains stacked in the direction perpendicular to the
191 heptazine planes.

192 Fig. 4 shows the results obtained from the equilibration in the NVT ensemble of three
193 26×18 sheets, which were initially widely separated and with a random disposition. The
194 computations were performed at different temperatures and, in all cases, the duration of
195 the simulation is 500 psec. Equilibrium was reached within the first 200 psec. At the lower
196 temperatures, 500 K and 700 K, the evolution was dominated by the interaction between
197 the well-formed layers, which resulted in their piling in an essentially three-layer system.
198 When the temperature was increased to 900 K, the thermally-driven sheet undulations

199 broke parts of the hydrogen-bond networks, loosening portions of the initial sheets. The
200 subsequent plane-to-plane interactions of the fragmented material resulted in the
201 formation of an irregular flake with a curved profile. The number of layers at some
202 locations of this disordered cell was larger than three.

203 The results derived from the evolution in the canonical ensemble were very similar
204 when the initial conditions consisted of six identical, well separated and placed at random
205 sheets. The time required to reach an equilibrium state in this case was significantly larger.
206 As in the previous case of three sheets in the starting conditions, the evolution at the lower
207 temperatures produced the piling of the initially well-formed layers. The system
208 dynamics, in this regime, were dominated by the layer-to-layer interactions. At 900 K, the
209 thermally induced shaking and break-up of the initial sheets and the following layer-to-
210 layer interaction of the resulting fragments finally produced a curved and disordered flake
211 with a number of stacked layers that, at some places, was larger than the initial figure of
212 six. Fig. 5 shows the resulting flake after being annealed by cooling it at low temperature.
213 The top view in Fig. 5(a) displays a turbostratic arrangement of the layers of material
214 within the flake, consistent with the experimental observations. The side view of Fig. 5(b)
215 shows the curvature of the flake and the existence of stacked layers that can clearly exceed
216 the number of six even at the edge of the flake. There are portions of the flake interior that
217 display a larger numbers of layers.

218 Fig. 6 shows the radial distribution function (RDF) of the atoms in the flake processed
219 at 900 K. The RDF of a single sheet of material at very low temperature is also plotted for
220 comparison purposes. This second curve has been arbitrarily scaled to facilitate its
221 contrast. The main peaks due to the in-layer interatomic distances in the single sheet of
222 material are also clearly distinguishable in the sample that has been processed at high
223 temperature. Nevertheless, the disorder in the flake is evident from the broadening of these

224 peaks. In the figure, the peak at 3.21 Å in the signal corresponding to the sample heated
225 to 900 K has been highlighted. Even though there are nearby contributions from in-layer
226 distances, the height of this peak relative to that of the low temperature single-sheet curve
227 is much higher than in the neighbouring peaks. Therefore, this feature can be
228 predominantly attributed to the interlayer stacking. The separation obtained from the
229 calculations was in excellent agreement with the measured values at the corresponding
230 processing temperature.

231 **3.2. X-ray diffractometry (XRD), scanning (SEM), and transmission electron** 232 **microscopy (TEM) observations**

233 A marked corrugation of the polymeric carbon sheets, synthesized from melamine
234 cyanurate (discussed in (19)), was confirmed at temperatures above 600 K. Particles with
235 tubular shape, formed by crumpling of flake particles, were observed at synthesis
236 temperatures around 700 K (Fig. 7).

237 In fact, at 900 K, the crumpled particles, formed not only tubular particles but also
238 globular ones, as tubules were further shrunk up to form globules with holes (Fig. 7).

239 This characteristic should be linked to H-bonding, and -as simulations showed- borders
240 would be more instable and tend to crumple, transforming the flakes in either tubular or
241 globular particles.

242 It is noteworthy to point out that this folding effect is coupled to a tighter interlayer
243 distance upon synthesis temperature increase, indicating that -concomitant to the borders'
244 folding- the stacking order increases, as shown in Fig. 8.

245 Therefore, the increase in the number of stacked sheets and the larger interlayer
246 interactions would be linked to the folding effect that produces crumpled particles. In a
247 previous research, it was found that the layers behaved as independent ones (i.e., as a 2D
248 material) below a synthesis temperature of 800 K (around 650 °C). Above this

249 temperature, the folding effect increased the number of stacked layers as well as the order
250 along the *c* direction. In summary, the diminution of the interaction on the layer plane
251 implicates a stronger interplanar H-bond interaction.

252 This corrugation effect could be conveniently used to produce globular, spherical or
253 tubular particles. Moreover, this effect of folding seemed to affect THz absorption in a
254 considerable way (19), so that it could be used as a sensor of pressure in special
255 environments. Hydrogen bond in carbon nitride (obtained from urea) was experimentally
256 investigated by Hu et al. (X) and they concluded that melem oligomers were formed and
257 arranged in layers. In spite of that study we suggest that the polymeric structure is
258 compatible with the crumpled morphology. Hydrogen bonding weakening between
259 adjacent polymer ribbons is the main cause of the typical crumpled structure of this
260 material, especially observed at high synthesis temperatures (*ca.* 900 K). This process
261 makes the material more sensitive to the external environment through the possibility of
262 interacting by hydrogen bonding with molecules such as CO₂, etc.

263 **4. CONCLUSION**

264 A computational study of the morphological properties of a carbon nitride polymer was
265 performed at the nanoscale using molecular dynamics. In spite of the fact that this survey
266 was restrained by the limited size of the simulation domains, a good agreement with the
267 experimental results was found. The synthesized samples consisted of crumpled
268 nanosheets formed by a turbostratic arrangement of polymer layers, which is consistent
269 with the small-size molecular dynamics simulations at the corresponding processing
270 temperatures.

271 The analysis of the computer simulations results have provided a clear picture of the
272 mechanisms that would lead to such arrangement. One determining factor is the weakness
273 of the hydrogen bonding network that keeps together the 2D polymer and that determines

274 many of its physical and chemical properties(45). This fact had been previously noted, for
275 instance, in relation with the calculation of the electronic band-gap of this material. Due
276 to the very small interactions between heptazine units, the calculated band-gap for the 2D
277 system is very similar to that of the monomer (19). At sufficiently high temperatures,
278 molecular agitation can partly release this relatively weak bonding network and permit the
279 conformational change in the material. The second element is the effect of the layer-to-
280 layer interaction, typical of 2D carbon materials, that drives such change. Hydrogen
281 bonding weakening between adjacent polymer ribbons is the main cause of the typical
282 crumpled structure of this material, especially observed at synthesis temperatures above
283 ca. 900 K. However, this property makes this material more sensitive to the external
284 environment, and thus more suitable to be used as a catalyst and a sensor.

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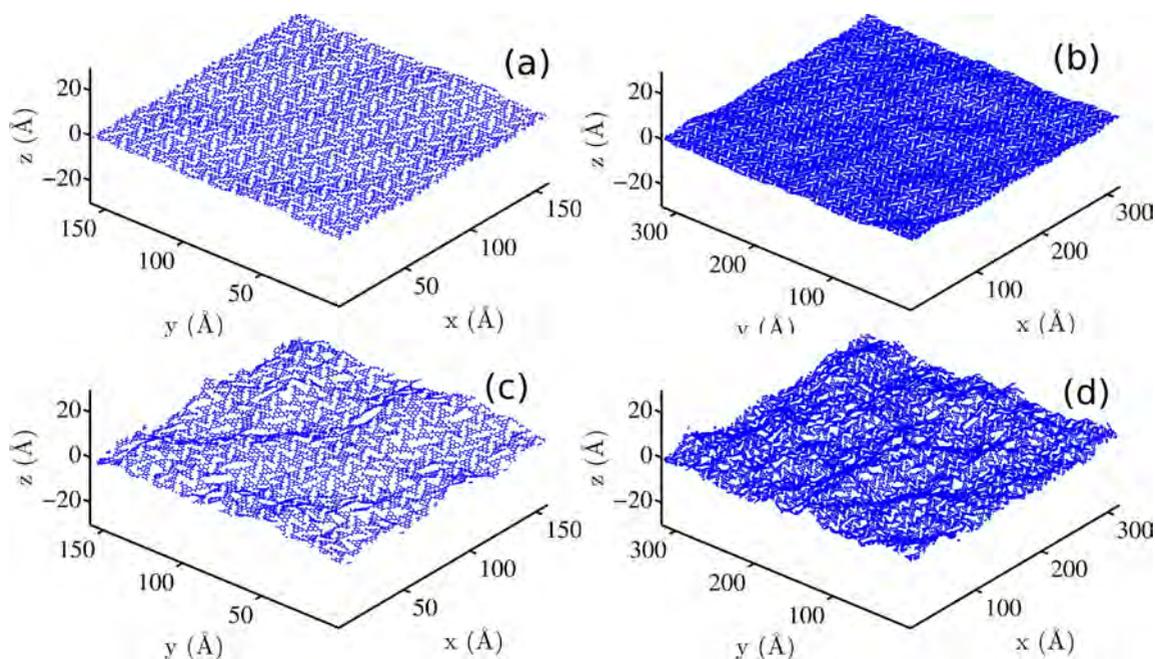
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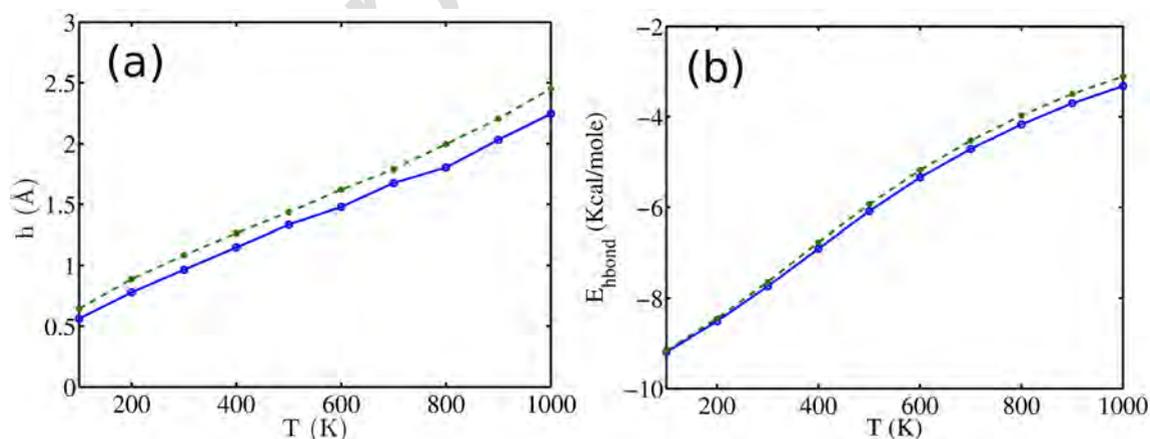
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418 Fig. 1. Equilibrium polymer sheets at zero pressure obtained using periodic boundary
 419 conditions. Left column results ((a) and (c)) correspond to a period of 13×9 unit cells;
 420 right column ones ((b) and (d)) correspond to a period of 26×18 unit cells. Top row results
 421 ((a) and (b)) are for a temperature of $T=100$ K and lower row ones ((c) and (d)) for $T=700$
 422 K.

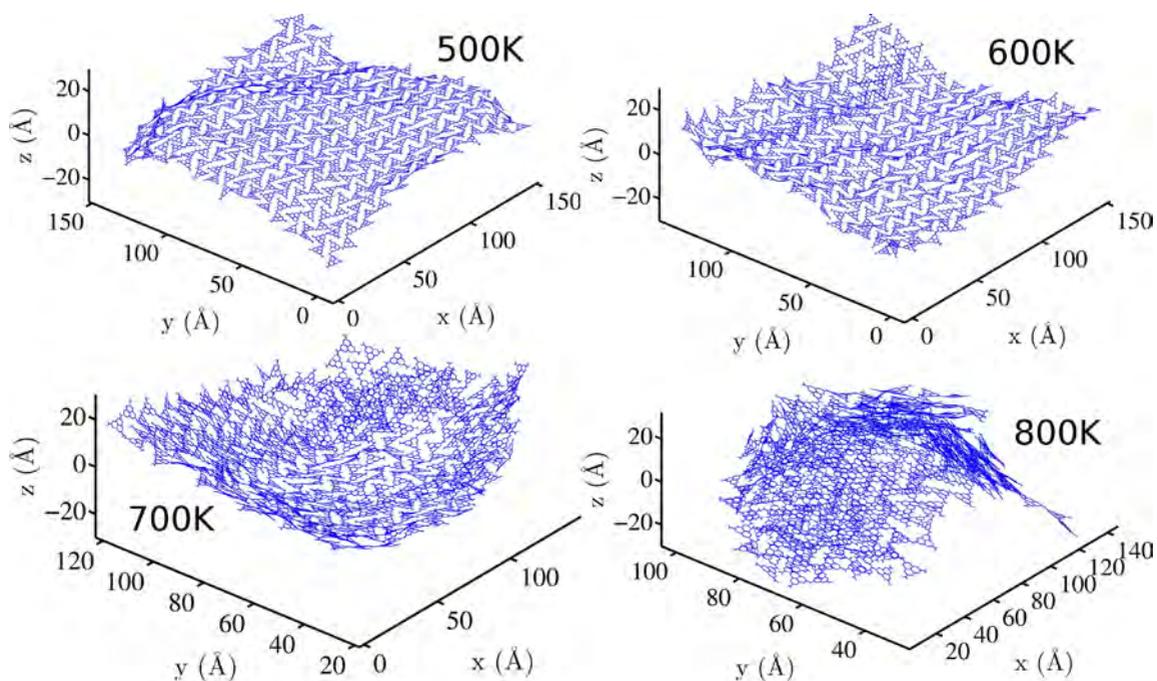
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425 Fig. 2. (a) Average height fluctuations and (b) hydrogen bond contribution to the system
 426 energy per heptazine unit averaged over a simulation run after equilibration.

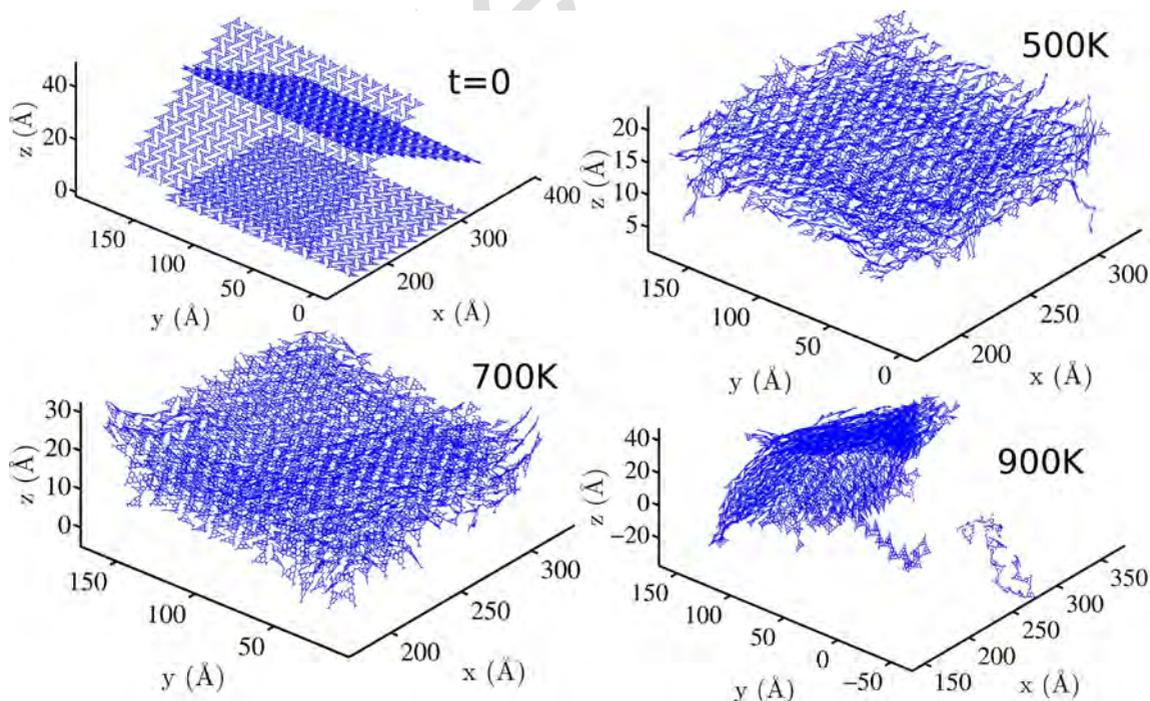
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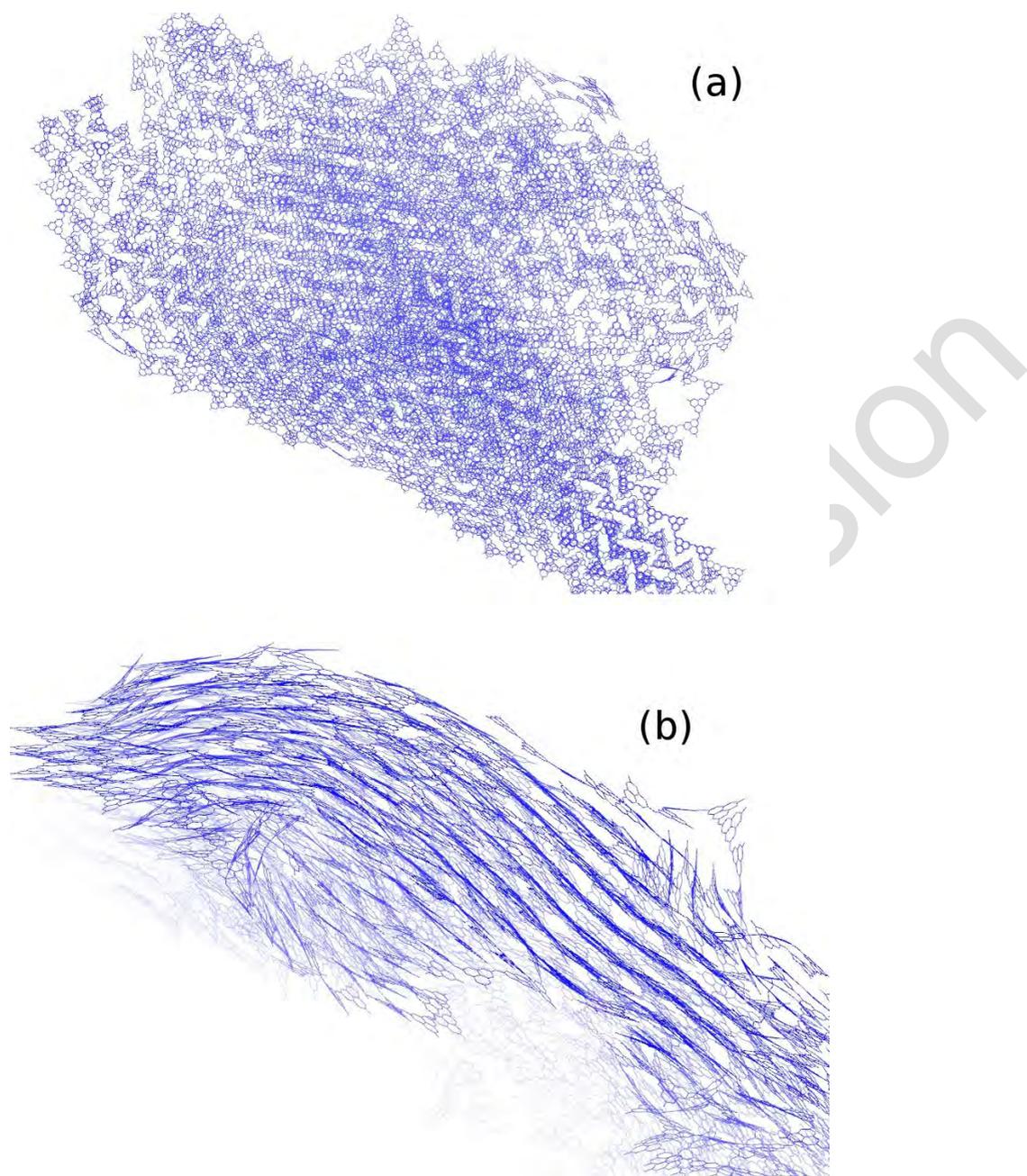
429 Fig. 3. Final trajectory states after 500 psec evolution in the NVT ensemble at different
 430 temperatures. The initial condition in all cases is a ground state, approximately square,
 431 sheet spanning over 13×9 unit cells of the 2D polymer.

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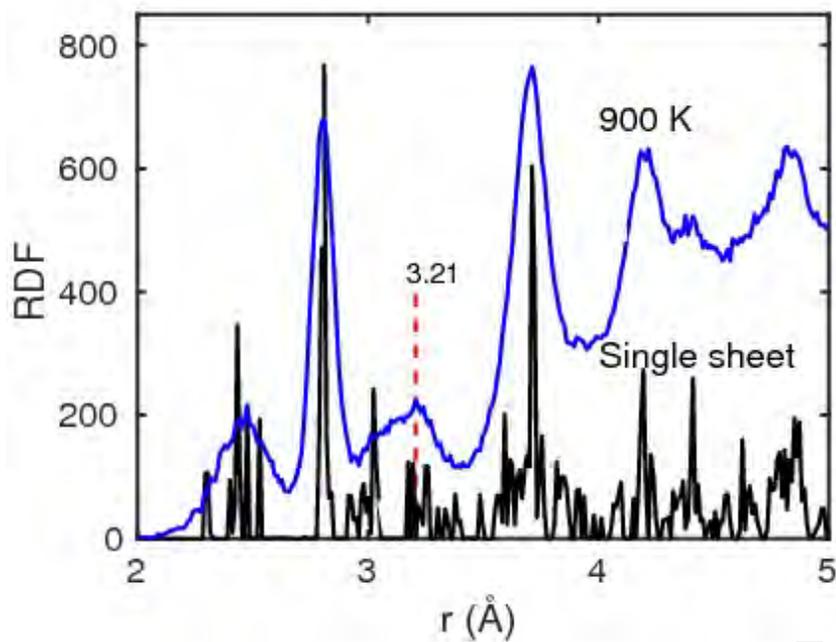
434 Fig. 4. Initial ($t=0$) and equilibrium final trajectory states at different temperatures after
 435 the evolution of three sheets in the canonical ensemble.



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438 Fig. 5. A flake obtained by annealing the final equilibrium geometry after evolution in the
439 canonical ensemble at 900 K starting from six sheets of polymeric carbon nitride. (a) top
440 view, (b) side view of the flake edge. Plots have been produced using VMD software (46).

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443 Fig. 6. Radial distribution function computed for flake depicted in Fig. 5 and for a single
444 sheet of material at low temperature.

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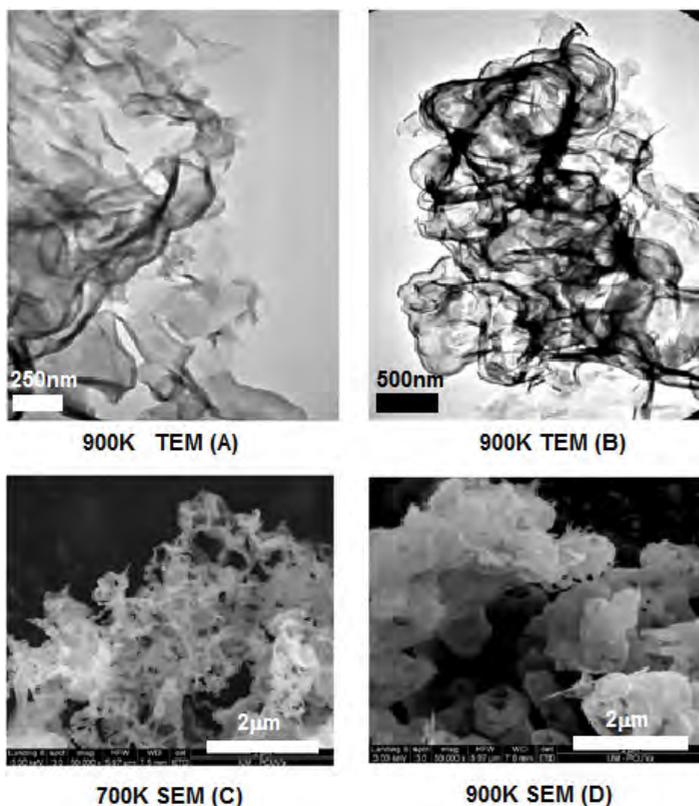
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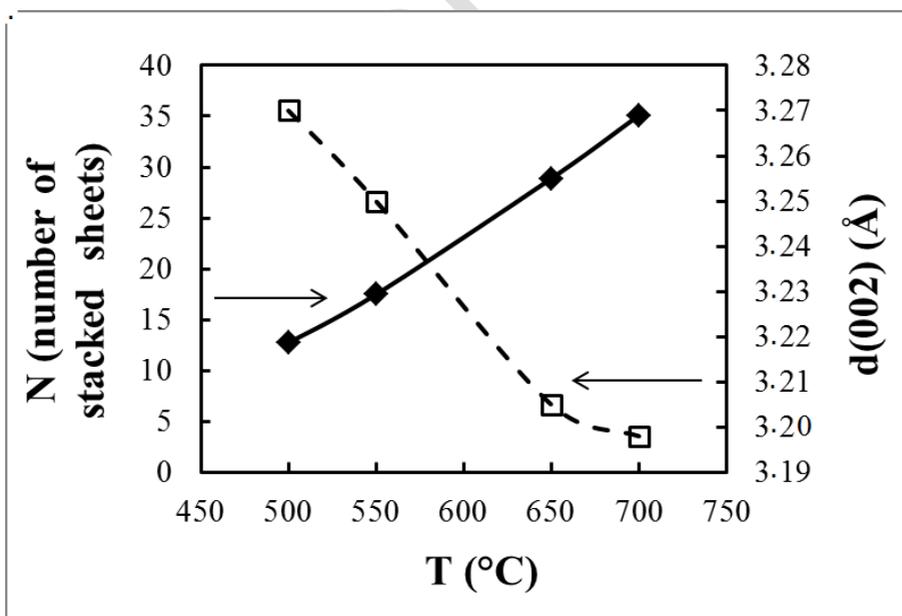
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458 Fig. 7. TEM and SEM images of polymeric carbon nitride obtained at 700 and 900 K from
 459 melamine cyanurate pyrolysis.

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462 Fig. 8. Variation of the interlayer distance and stacked sheets as a function of the synthesis
 463 temperature (19).