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Drastic enhancement of mechanical properties of Ca₃Co₄O₉ by B₄C addition

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ABSTRACT

 $Ca_3Co_4O_9$ with B_4C additions in different proportions (0, 0.10, 0.25, 0.5, and 0.75 wt.%) have been fabricated using the classical solid-state method. Powder XRD patterns have displayed that $Ca_3Co_4O_9$ phase is the major one in all samples. Microstructural observations showed that B_4C has been superficially oxidized, producing liquid B_2O_3 during sintering, and reacting with the $Ca_3Co_4O_9$ grains to produce bridges between them. In spite of the increase of porosity, these bridges led to an important raise (more than two times) of mechanical properties when compared to the pristine materials. On the other hand, while B_4C addition has not influenced S values, it has decreased electrical resistivity, thermal conductivity, and thermal expansion. Consequently, ZT values have been also increased, reaching 0.24 at 800 °C in 0.25 wt.% B_4C containing samples, which is very close to the best values reported in the literature, and two times higher than the obtained in pure materials in this work.

1. Introduction

Thermoelectric (TE) technology is considered as a very promising one to enhance energy transforming devices efficiency by harvesting wasted heat [1]. This characteristic is associated to the Seebeck effect, which is inherent to the TE materials [2]. On the other hand, these materials should display appropriate thermoelectric properties for practical applications, evaluated using the equation [3]:

$$ZT = \frac{S^2 T \sigma}{\kappa} \tag{1}$$

where ZT is the dimensionless Figure-of-Merit, and S, T, σ , and κ are Seebeck coefficient, absolute temperature, and electrical, and thermal conductivity, respectively.

TE materials are considered adequate for practical applications when their *ZT* values are higher than 1. Nowadays, only intermetallic materials display these values at low-medium temperatures [4,5], with the drawback of their usual chemical composition, comprising heavy elements. Moreover, they cannot be used at high temperatures due to oxidation processes, unless they are sealed to avoid oxidant environments. On the other hand, oxide materials possess high chemical and thermal stability, and can work at high temperatures. Moreover, they are composed of abundant, cheaper, and environmentally friendly elements [6]. However, their TE performances are lower than those

determined in intermetallic materials. Consequently, most of the works published in this field are dealing with the improvement of their TE performances through many different ways [7–13].

Nevertheless, TE properties of materials are not the only parameters determining the efficiency of a TE module. One of them is the so-called manufacturing factor (MF) which is a function of the ideal resistance in the module ($R_{\rm id}$, taking into account the resistance of legs), and the real internal resistance ($R_{\rm int}$, where the contact resistance is also present), as MF = $R_{\rm id}/R_{\rm int}$ [14]. This MF is usually determined in the as-fabricated modules, but it can be decreased during the working life of these modules due to the differential expansion of the different components (p, and n thermoelectric legs, metallic parts, and insulating ceramic) which can damage the contacts among them, increasing the internal resistance. Consequently, before building a TE module, it is very convenient choosing materials with very close thermal expansion coefficients to avoid internal stresses when working at high temperature. Besides, high mechanical properties are also desired in these TE legs in order to maintain the module integrity and increasing its useful life.

In this work, $Ca_3Co_4O_9$ material will be prepared, with the addition of small amounts of B_4C (0, 0.10, 0.25, 0.50, and 0.75 wt.%), via the classical solid-state route. The effect of these additions on the structure and microstructure of $Ca_3Co_4O_9$ sintered pellets will be studied and related to the modifications produced in parameters such as linear expansion coefficient, mechanical, and thermoelectric properties.

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2. Experimental

 $Ca_3Co_4O_9 + x \text{ wt.}\% B_4C$, with x = 0.0, 0.10, 0.25, 0.50, and 0.75,have been prepared using the classical solid-state route from CaCO₃ (Panreac, 98 + %), and Co₂O₃ (Aldrich, 98 + %) commercial powders. They were weighed in the adequate amounts, mixed and ball milled under water media for 30 min at 300 rpm to produce a suspension, which was dried under infrared radiation. The resulting dry mixture was manually milled to produce a very fine powder, subsequently thermally treated at 750 and 800 °C for 12 h under air, with an intermediate manual milling. This thermal process is adequate to decompose CaCO₃ and to produce intermediate by-products, in agreement with previously reported data [15]. After milling these calcined powders. B₄C (Alfa Aesar, 99 + %) has been added in the stoichiometric proportions and the mixture has been ball milled 30 min at 300 rpm to homogenize the mixture. After this process, the different materials were cold-pressed in form of pellets with around $3 \times 3 \times 12$ mm³, under 400 MPa applied pressure perpendicular to the 3×12 mm² surface. Finally, the compacts were sintered at 900 °C for 24 h under air atmosphere, followed by a final furnace cooling.

Powder X-ray diffraction (XRD) patterns were determined in a theta-theta PANalyticalX'Pert Pro diffractometer (CuK_{α} radiation, λ = 1.54059 Å) between 10 and 40 degrees, where the main peaks of $Ca_3Co_4O_9$ phase, and the most intense of B_4C , appear. Density measurements were performed using Archimedes' method on several samples for each B_4C addition. The relative density values were calculated with respect to the theoretical one, assuming a dense and perfect mixture of $Ca_3Co_4O_9$ (4.677 g/cm³ [16]) and B_4C (2.50 g/cm³ [17]).

Microstructural observations have been performed on the samples surfaces and fractured sections in a Field Emission Scanning Electron Microscope (FESEM, Carl Zeiss Merlin) combined with an energy-dispersive spectrometry (EDS) system. Mechanical properties have been determined in several samples for each composition through flexural strength, Young's modulus, compression tests and Vickers microhardness measurements. Flexural strength and Young's modulus have been obtained from the three point bending test in an Instron 5565, with 30 µm/s punch displacement and 10 mm span; compressive strength was performed parallel to the long pellet axis at 30 μm/s; and Vickers microindentation tests were accomplished under 0.05 Kg applied load. Seebeck coefficient and electrical resistivity were simultaneously determined through the four-point contact, along the large pellet axis, using a LSR-3 (Linseis GmbH) under He atmosphere in the 50-800 °C temperature range. Thermal diffusivity (α) has been measured in the same direction used for the electrical characterization in a laser-flash system (Linseis LFA 1000). Thermal conductivity (κ) has been calculated as $\kappa = \alpha \cdot C_p \cdot d$, being C_p specific heat and d, sample density. C_p has been determined through Dulong-Petit law. From d, C_p , E, and κ_l , phonon mean free path has been calculated for each composition. In order to stablish the samples TE performances, ZT was determined from Seebeck coefficient, electrical resistivity, and thermal conductivity data. These data were used to evaluate the properties evolution as a function of the dopant content, and were also compared with previously reported values in the literature for similar compounds. Finally, dilatometric behaviour of samples has been studied in the 25-800 °C temperature range in a L79 HCS dilatometer (Linseis GmbH) to determine their linear expansion coefficient evolution as a function of B₄C content.

3. Results and discussion

3.1. Structural and microstructural characterization

Fig. 1 illustrates the XRD patterns obtained in $Ca_3Co_4O_9 + x$ wt.% B_4C powdered samples. As it can be easily observed in the figure, all samples present very similar patterns, independently of the B_4C addition. Moreover, all peaks can be associated to the diffraction planes

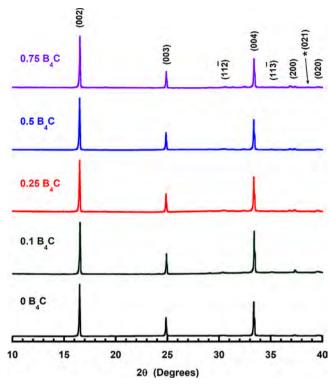


Fig. 1. Powder X-ray diffraction patterns of $Ca_3Co_4O_9 + x$ wt.% B_4C samples. The diffraction planes indicate the reflections of $Ca_3Co_4O_9$ phase. * identifies the B4C diffraction peak.

(indicated in the graph) of $Ca_3Co_4O_9$ phase with monoclinic symmetry [18,19]. Furthermore, the most intense diffractions correspond to its ab-planes, which could be associated to preferential grain orientation. However, it should be highlighted that XRD patterns were obtained on powdered samples and, consequently, this grain orientation is produced by the samples preparation, as observed in previous works [20]. On the other hand, B_4C has been identified with this technique through its (021) reflection peak at 37.818 degrees, appearing with very low intensity for B_4C content ≥ 0.25 wt.% [21].

Microstructural studies performed on the samples surfaces have shown that all samples are composed of flake-like particles with no preferential orientation, which is the characteristic situation in this system, as reported in previous works [22,23]. Moreover, EDS analysis has confirmed the XRD results, showing that these grains correspond to the $Ca_3Co_4O_9$ phase. On the other hand, the $Ca_3Co_4O_9$ grain sizes are decreased when the amount of B_4C is increased, while porosity seems to be increased. In order to confirm the raise in porosity when the B_4C content is larger, the samples density was evaluated and the mean results, together with their standard error, are displayed in Table 1. As it is clear from these data, samples density is slightly decreased when the amount of B_4C is increased, corroborating the SEM observations. In any case, the relative density values are in the order of the typically reported in materials prepared through the classical solid-state method [23–25]. This low densification can be explained when considering the

Table 1Apparent density, standard error, and relative density determined through Archimedes' method.

Density (g/cm ³)	Std. error	Relative density
3.4532	0.0317	73.8
3.4358	0.0830	73.6
3.3742	0.0627	72.3
3.3205	0.0289	71.3
3.3009	0.0502	71.0
	.4532 .4358 .3742 .3205	.4532 0.0317 .4358 0.0830 .3742 0.0627 .3205 0.0289

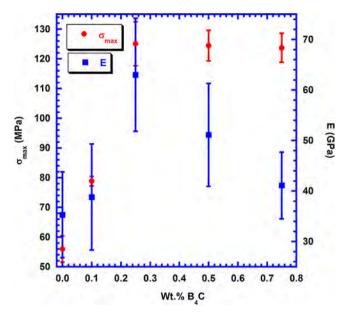


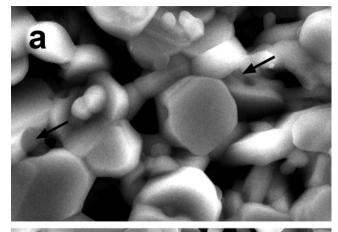
Fig. 2. σ_{max} , and E values obtained through three point bending tests, with their standard error, as a function of B_4C content.

phase equilibria diagram [26], which shows that $\text{Ca}_3\text{Co}_4\text{O}_9$ phase is stable up to 926 °C. Consequently, sintering procedure has to be performed below this temperature, which is, in turn, much lower than the eutectic temperature (1350 °C), drastically limiting the samples densification. Even if porosity is undesirable to get very high mechanical properties, it can help reducing thermal conductivity of these samples and, consequently, increasing thermoelectric performances.

3.2. Mechanical characterization

Fig. 2 presents the three point bending tests results for the $\rm Ca_3Co_4O_9$ and their standard errors, together with the Young's modulus values extracted for these tests, as a function of $\rm B_4C$ content. In spite of the slight decrease of density induced by $\rm B_4C$ addition, mechanical properties are drastically increased, when compared to the pure samples. Flexural strength is increased in around 50 % for 0.1 wt.% $\rm B_4C$ addition, and more than 100 % for higher contents, while Young's modulus is only raised in around 10 % for 0.1 wt.% $\rm B_4C$ addition, and about 75 % for 0.25 wt.% $\rm B_4C$ samples. These results seem to be contradictory with the porosity evolution and, consequently, the fractured sections of samples were microstructurally studied.

Fig. 3 shows representative micrographs at high magnification of 0.25, and 0.75 B₄C samples. As it can be observed in the pictures, besides the grey contrast (Ca₃Co₄O₉ phase), a new contrast (dark grey) is appearing in higher proportions when the amount of B₄C is increased, which is connecting thermoelectric grains. EDS analysis has shown that this contrast contains B, C, O, Ca, and Co. These cations are present in a relationship (using B as reference) 1.0:0.485 ± 0.005:2.85 \pm 0.09:0.77 \pm 0.02:0.66 \pm 0.04, for B, C, O, Ca, and Co, respectively, independently of B₄C content. The formation of this new phase can be explained when considering that B₄C starts to be oxidized under air at around 450 °C, producing CO2 and B2O3 [27]. This effect could lead to the total B₄C oxidation, however, the formed B₂O₃ has melting point of 450 °C [28] and, between 600 and 1000 °C, forms a protective layer on the particles surface avoiding further oxidation [29]. On the other hand, the presence of this liquid phase, making close contact with the thermoelectric grains, lead to Ca-B-O [30], and Co-B-O [31] compounds formation on the B₄C particles surface. Consequently, these compounds form bridges between the thermoelectric grains, enhancing the samples mechanical properties. Bending stresses obtained in pure samples (56 MPa) are more than three times higher than the reported



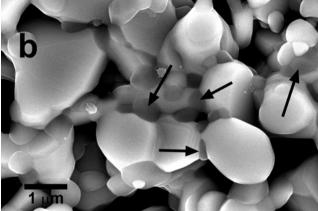


Fig. 3. Representative SEM micrographs obtained in fractured surfaces of $Ca_3Co_4O_9 + x$ wt.% B_4C , with x = a) 0.25; and b) 0.75. The arrows show the dark grey contrast which corresponds to a B-rich phase.

for sintered samples (18.4 MPa) [32]. However, despite the adverse porosity effect, the best bending stress values were obtained in samples with B_4C content ≥ 0.25 wt.% (125 MPa). These values are about a half of the reported for high density materials with well oriented grains (254 MPa) [32]. Moreover, Young's modulus determined in 0.25 wt.% B_4C samples is only 35 % lower than the best reported values in this kind of materials [32].

Fig. 4 displays compressive strength of the different samples, as a function of B_4C content. As it is expected, the stresses determined through this test are higher than the obtained through three point bending tests. The evolution illustrated in the graph clearly confirms the effect of B_4C addition observed in the three point bending tests. In spite of the slight increase of samples porosity when B_4C is added, compressive strength is drastically increased by a factor 2 for B_4C content ≥ 0.25 wt.%, when compared to the pristine samples prepared in this work. On the other hand, slight differences with the three point bending test can be found in the graph, as stresses obtained by compression are asymptotically increased when B_4C content is higher than 0.25 wt %

Other important mechanical characteristic is the material hardness, which is responsible for the machinability and wear resistance of individual thermoelectric elements [32–34]. This property has been determined through Vickers microhardness, and the obtained results are presented, together with their standard error, in Fig. 5, as a function of B_4C content. As it can be seen in the graph, in spite of the porosity influence on the measurements, a clear increase of microhardness with the B_4C amount is produced. This evolution is in agreement with the microstructural features already discussed in Fig. 3, which enhance all mechanical properties determined in this work. Moreover, the values for the pristine samples (0.11 GPa) are the same reported in previous

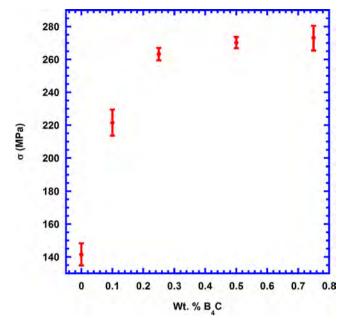


Fig. 4. σ values obtained through compression tests, with their standard error, as a function of B_4C content.

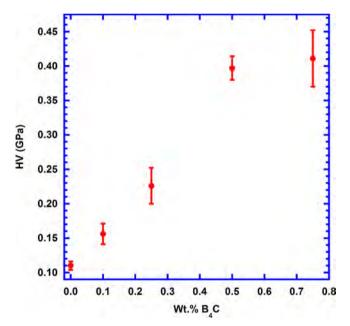


Fig. 5. Vickers microhardness values, with their standard error, as a function of $B_4 C$ addition.

works (0.12 GPa) [32], while the best results in samples with B_4C addition (0.41 GPa) can reach one third of the reported for nearly fully dense materials (1.3 GPa after hot pressing at 30 MPa) [32].

It is important to highlight that these porous samples with B_4C addition reach much higher mechanical properties than the expected in this kind of materials, not very far from the obtained in nearly fully dense materials, but using a more simple, economic, and rapid process. This is a very advantageous situation when considering these materials for practical applications.

3.3. Thermoelectric characterization

Electrical resistivity variation with temperature, as a function of B₄C content is presented in Fig. 6. As shown in the plot, all samples possess

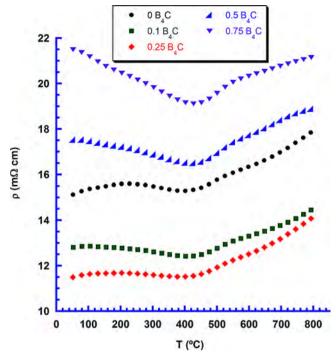


Fig. 6. Electrical resistivity variation with temperature, as a function of B_4C content, in $Ca_3Co_4O_9 + x$ wt.% B_4C samples.

very similar behaviour in the whole measured temperature range. The samples display a minimum at around 450 °C, which corresponds to the change between semiconducting $(d\rho/dT < 0)$ and metallic $(d\rho/dT > 0)$ behaviour. This temperature indicates the change from a hole hopping from Co⁴⁺ to Co³⁺ [35] to a charge carriers transport in the valence or conduction band [36]. This is a very common behaviour observed in Ca₃Co₄O₉ sintered materials [23,37]. On the other hand, B₄C addition up to 0.25 wt.% decreases electrical resistivity, while further addition drastically increases it, when compared to samples without additions. Different assumptions can be made to account for this. It can be due to a chemical effect such as the one already observed in CCO thin films, where a slightly lower Co content, with respect to the stoichiometric one, leads to lower electrical resistivity [38]. The latter mechanism may explain the decrease of the electrical resistivity of CCO, for the lowest B₄C addition, as this addition induces a slight departure of cobalt from the structure. However, other phenomena may be considered such as those related to the microstructure changes (grain size modification with B₄C addition, inter-granular insulating phase formation, etc.). It is clear that further investigations would be required to address this issue in detail. The lowest electrical resistivity values have been obtained in samples with 0.25 wt.% B₄C. The minimum resistivity values measured in these samples at 800 °C (14 m Ω cm) are slightly lower than the best values reported for Ca₃Co₄O₉ samples sintered or textured through spark plasma sintering (15–18 m Ω cm) [39] and around 20 % lower than in pure samples. However, they are higher than the measured in highly dense materials prepared using alternative methods (10 m Ω cm) [40].

Fig. 7 shows the S variation with temperature and B_4C content. S displays positive values in all samples, indicating that conduction is mainly produced by holes. Moreover, the behaviour is the same for all samples, the values are increased when the temperature rises, and these values are the same for all samples within the measurement errors. These very close S values, independently of B_4C content can be explained by the very small compositional modification produced in the samples discussed previously. Furthermore, the cations drainage induced by the liquid B_2O_3 is probably limited to the zones close to the grain boundaries, maintaining the core of the grains practically

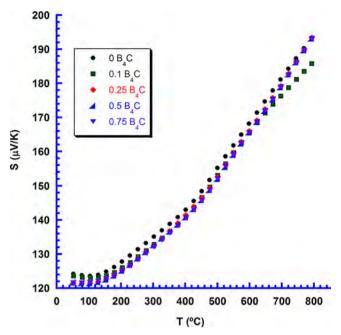


Fig. 7. Seebeck coefficient variation with temperature, as a function of B_4C content, in $Ca_3Co_4O_9\,+\,x$ wt.% B_4C samples.

unchanged. Consequently, no significant modifications in S values should be expected. In addition, the fact that the B_4C content does not impact the Seebeck coefficient suggests that the carrier mobility may be modified by the B_4C addition. Therefore, the modification of the electrical resistivity with B_4C addition may be linked to the carrier mobility changes. The highest values at 800 °C (195 $\mu V/K$) are higher than the reported in pure $Ca_3Co_4O_9$ sintered or textured through spark plasma sintering (170–175 $\mu V/K$) [39], but lower than the reported in very dense materials obtained by alternative methods (205 $\mu V/K$) [40].

Total κ consists in the addition of two components, $\kappa = \kappa_1 + \kappa_e$, where κ_l is the lattice thermal conductivity contribution and κ_e is the electronic counterpart. κ_e can be estimated from the Wiedemann-Franz's law [41], which is expressed as κ_e = LoT, where L, and σ are Lorenz number $(2.45 \times 10^{-8} \text{ V}^2/\text{K}^2)$, and electrical conductivity, respectively. The variation with temperature and B₄C content of the electronic and total thermal conductivity are presented in Fig. 8. When comparing both graphs, it is clear that electronic thermal conductivity is increased when the temperature rises. In contrast, the total thermal conductivity decreases from room temperature to around 600 – 650 °C induced by the increase of lattice vibrations which enhance phonon scattering. At higher temperatures, the total thermal conductivity increases due to the larger influence of electronic thermal conductivity. Moreover, B₄C addition leads to a clear decrease of total thermal conductivity, when compared to the pure samples, especially at high temperatures. These results are in agreement with previous works in Ca_{3-x}B_xCo₄O₉ which showed that thermal conductivity is decreased with B-content up to 0.5 [42]. At 800 °C, the minimum total thermal conductivity value determined in 0.25 wt.% B₄C samples (1.2 W/K m) is about 20 % higher than the lowest measured in these samples (1.0 W/K m at 650 °C). These results are in accordance with the phonon mean free path calculated from $\kappa_{l}, \, sound \, \, velocity, \, and \, heat \, capacity, \, as \, it$ decreases from the undoped samples (2.5 $\mu m)$ to the 0.25 wt.% $B_4 C$ ones (1.8 μm), increasing up to 2.2 μm for 0.75 wt.% B₄C samples. These mean free paths are within the range of the reported in the literature [43]. Moreover, these values are comparable to the reported in classically sintered materials (0.9 W/K m with \sim 60 % theoretical density) which show higher amount of porosity [44] than the samples in this work. On the other hand, it is lower than the best values obtained by spark plasma sintering (2.1 W/K m) [45], and much lower

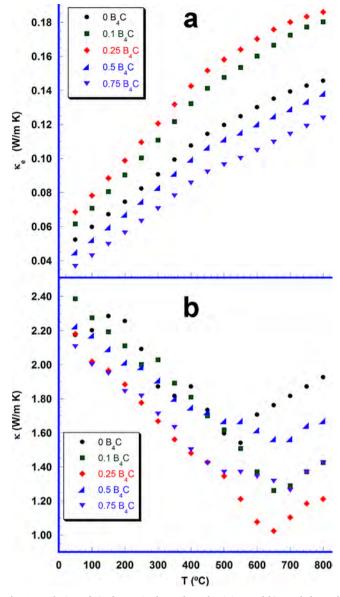


Fig. 8. Evolution of a) Electronic thermal conductivity; and b) Total thermal conductivity, with temperature, as a function of B_4C content, in $Ca_3Co_4O_9 + x$ wt.% B_4C samples.

than the measured in textured materials by hot-uniaxial pressing (2.2, 3, and 4.7 W/K m) [13,44,46]. Other interesting observation is that total thermal conductivity is decreasing with temperature in all samples, reaching a minimum, and then increasing again at higher temperatures. It can be seen that these minima appear at lower temperatures when the B_4C content is increased. This behaviour can be related to the fact that probably κ_e is typically underestimated in this kind of compounds, as Lorentz number (2.45 $10^{-8}\ V^2/K^2$) used for its calculation is only valid for metallic materials. Consequently, as shown in previous works, it should be calculated as $L=1.5\ +\ exp(-S/116)$, where S is the absolute Seebeck coefficient in the maximum [47]. In this work, this expression has not been used, as no S maximum has been found for any of the prepared samples.

With all the previously discussed data, ZT has been calculated and plotted in Fig. 9. As displayed in the graph, ZT values are increased with temperature in the whole measured temperature range, and also with B_4C content up to 0.25 wt.%, decreasing for higher additions. The highest value has been measured at 800 °C in 0.25 wt.% B_4C samples (0.24), which is more than two times higher than the calculated for

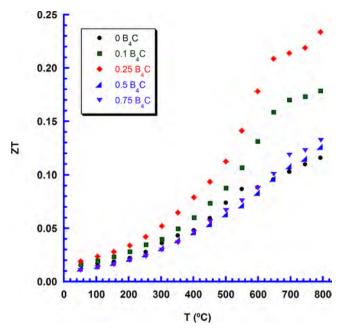


Fig. 9. ZT evolution with temperature, as a function of B_4C addition in $Ca_3Co_4O_9 + x$ wt.% B_4C samples.

Table 2 Thermal expansion coefficient (α) determined in Ca₃Co₄O₉ + x wt.% B₄C samples.

α (ppm/K)
10.32
9.65
9.36
9.56
9.68

pure ones. Moreover, it is higher than the measured in classically (0.08 [44]), or spark plasma sintered samples (0.15 [45]), and even than the determined in textured samples (0.18, and 0.16 [13,44]). However, this value is lower than the reported one for materials prepared through non-conventional methods (0.35 [15]) or in $\text{Ca}_{3\text{-x}}\text{Sr}_x\text{Co}_4\text{O}_9$ textured materials (0.29 [46]).

Thermal expansion coefficient has been determined in all samples and presented in Table 2. As it can be observed in this table, thermal expansion of pure samples (10.32 ppm/K) is in the range of the reported value for $\text{Ca}_3\text{Co}_4\text{O}_9$ materials (10.6 ppm/K [25]) and lower than the measured in Bi-substituted materials (12.8 ppm/K [25]). On the other hand, B₄C addition, up to 0.25 wt.%, decreases thermal expansion coefficient, increasing with further additions. The minimum value (9.3 ppm/K) measured in this work corresponds to a decrease of only around 10 % but closer to the one reported for Al_2O_3 (7.5 ppm/K [48]). This reduction, even if it does not seem to be significant, may reduce the differential thermal expansion between two of the main components of thermoelectric modules, decreasing the internal stresses at working temperatures, and increasing the life span of these devices.

4. Conclusions

 $\text{Ca}_3\text{Co}_4\text{O}_9 + x$ wt.% B_4C (x = 0, 0.1, 0.25, 0.5, and 0.75) polycrystalline materials have been prepared through the classical ceramic method. XRD analysis has only identified $\text{Ca}_3\text{Co}_4\text{O}_9$ phase in all samples, independently of the B_4C content. SEM observations have shown that B_4C has reacted with oxygen producing liquid B_2O_3 on its surface, slightly reacting with $\text{Ca}_3\text{Co}_4\text{O}_9$ grains and forming bridges between the grains. These microstructural modifications have been reflected in the

mechanical properties, which were drastically enhanced when compared to the pure samples, despite the slight increase of porosity. Seebeck coefficient has not been affected by B_4C addition, while electrical resistivity, thermal conductivity and thermal expansion have been decreased, leading to maximum ZT values close to the best reported in the literature, and about two times higher than the measured in pure samples in this work. It is worth to mention that these results have been obtained through a simple and scalable process which can be easily transferred to industry.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:https://doi.org/10.1016/j.jeurceramsoc.2020.08.024.

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