

1 **Changes in pools of organic matter and major elements in the soil following**
2 **prescribed pastoral burning in the central Pyrenees**

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15 **ABSTRACT**

16 High–mountain soils are rich in partially decomposed organic matter, which is highly
17 sensitive to mineralization and fire. Prescribed burning is performed in the Pyrenees to keep
18 subalpine grasslands open for grazing. The compositions of the ash, litter and duff layers,
19 and the particulate organic matter (POM) of the topsoil in the 0–1, 1–2, 2–3, and 3–5 cm
20 depths were analyzed in relation to the nutrient availability after the prescribed burning of
21 a stand encroached by erizón (*Echinospartum horridum*). The concentrations of C, N, P,
22 and S and organic components (nonstructural, hemicellulose, cellulose, and lignin–type)
23 were determined before the prescribed burn and 0, 6, 12, 18, and 24 months after the
24 prescribed burn. The fire consumed the aboveground biomass, the litter and part of the duff
25 layer, and the most thermostable (i.e., lignin–type) components and the least volatile

26 elements (P, S) were selectively preserved in the resulting ash. Prescribed burning caused
27 significant losses of organic-C (-72%) and of N (-68%) only in the 0–1 cm depth. The
28 organic-C loss was mostly (82%) from the POM, whereas the N loss was from more similar
29 proportions of the POM (57%) and the nonparticulate organic matter (NPOM) (43%).
30 However, few changes were observed in the composition of the organic matter, which
31 pointed to a largely uneven combustion that resulted in a substantial part of the organic
32 matter remaining largely untouched by the fire. After 6 months, the duff layer was depleted
33 in hemicellulose by 32% compared to immediately after the burn, and fragmentation of the
34 POM into the NPOM was observed. During the second spring, N- and P-rich charred POM
35 were incorporated into the top 1 cm, while C-rich charcoal particles underwent
36 fragmentation and vertical transport into the deeper soil. The preburn ecosystem was limited
37 by P, and likely also by S. The plant available N showed transient increases of 1.5–2.1
38 times the immediate postburn levels for nitrate-N at 12 months after burning, and of up to
39 10–20 times for ammonium-N at 18 months. In contrast, the concentrations of plant-
40 available P and S gradually declined to 1.8–3.3 and 1.8–4.0 times, respectively, lower at 24
41 months after the burn. Results indicated that fire-induced increases in the nutrient
42 availability can be short-lived in high-mountain habitats, but steadier and likely more
43 persistent nutrient inputs can derive from the gradual breakdown of charred organic matter.

44

45 **KEYWORDS** litter, duff, particulate organic matter, mountain soils, ecological
46 stoichiometry

47

48 1. INTRODUCTION

49 Fire is a major disturbance around the globe (Thonicke et al., 2001) and is one of the more
50 accessible and powerful tools available for manipulating habitats through the removal of
51 unwanted plant biomass (Kull, 2008). Fire has been used for millennia in the mountain
52 areas of southern Europe to clear land for grazing, being largely responsible for the
53 expansion of alpine and subalpine grasslands at the expense of woodlands (Favilli et al.,
54 2010; González–Sampériz et al., 2017). In the Pyrenees, burning practices have remained
55 until modern times as an indispensable tool for keeping the grazing lands open, and have
56 resurged in recent years in the form of prescribed burns (Faerber, 2009). Fire considerably
57 affects soils, either directly by heating the soil components during burning or indirectly due
58 to the removal of plant cover and the subsequent erosion after the fire (Santin and Doerr
59 2016). Prescribed fires have been shown to have variable effects depending on the soil and
60 vegetation conditions and on the fire type, but are generally considered less harmful to the
61 soils than wildfires because of the limited intensity and severity and low soil heating
62 (Alcañiz et al., 2018; Fernandes et al., 2013).

63

64 High–mountain soils are generally poor and are often shallow or skeletal, and their fertility
65 depends to a large extent on the soil organic matter (SOM) and its role as a water reservoir,
66 source of nutrients, support for root growth, and soil–structuring agent (Bojko and Kabala,
67 2017; Prietzel and Christophel, 2014). Due to the low temperatures in the environment,
68 organic matter cycling in the high–mountain areas is slow, resulting in the accumulation
69 and temporal immobilization of large reservoirs of carbon (C) (Bojko and Kabala, 2017;
70 Ward et al., 2014) and nutrients (Gavazov, 2010; Weintraub, 2011), mainly in the form of
71 particulate organic matter (POM), i.e., SOM larger than 50 μm (Puissant et al., 2017;
72 Saenger et al., 2015). However, POM is generally labile, i.e., easily decomposable by

73 microorganisms, which makes these C and nutrient reservoirs sensitive to changes in the
74 climate or land management (García–Pausas et al., 2017; Hagedorn et al., 2010). Due to
75 their richness in POM, mountain soils can be particularly vulnerable to management
76 practices that have potential impacts on the C cycle and ecosystem performance (Bing et
77 al., 2016; Hagedorn et al., 2010).

78

79 Several recent studies have investigated the impact of prescribed pastoral fires on high–
80 mountain soils in the southern Pyrenees and have focused on microbiological and
81 biochemical properties (Armas–Herrera et al., 2018; Girona–García et al., 2018a, 2019;
82 Múgica et al., 2018), bioavailable nutrients (Girona–García et al., 2018c; Múgica et al.,
83 2018), and total C and nitrogen (N) reservoirs in the whole topsoil layer (Nuche et al., 2018;
84 Girona–García et al., 2019) and in different aggregate–size classes (Girona–García et al.,
85 2018b). However, little is known about the effects of high–mountain pastoral burns on the
86 composition and nutrient concentration of organic matter in the high–mountain soils.

87

88 Moreover, studies examining the effects of burning on nutrient reservoirs in POM are rare.
89 Soil POM mainly consists of hemicellulose, cellulose, and lignin–type structural
90 components, which are listed in increasing order of their stabilities against microbial
91 (Jensen et al., 2005; Kwiatkowska–Malina 2018) and thermal degradation (Yang et al.,
92 2007). When the fire intensity is not too large as is the case with prescribed fires, the
93 preferential preservation of the more resistant (i.e., lignin–type) fractions over the more
94 unstable fractions (hemicellulose, cellulose) can be expected (Fernández et al., 2001). Fire
95 will also likely cause more intense effects on free POM than on nonparticulate organic
96 matter (NPOM), which is chemically stable and associated with soil minerals (Almendros
97 and González–Vila, 2012).

98

99 The aim of this study was to analyze the short-term (two-year) effects of prescribed
100 burning on the composition and the nutrients contained in the organic matter near the soil
101 surface and in the topsoil layer and on the nutrient availability in the topsoil in a subalpine
102 pastureland area in the Pyrenees. It was hypothesized that (i) burning has greater effects on
103 the most labile components of decomposing organic matter than on those more stable and
104 on the POM than on the NPOM and that (ii) fire-induced mineralization of large amounts
105 of organic matter would lead to increased nutrient availability immediately and shortly after
106 burning.

107

108 2. MATERIAL AND METHODS

109 2.1. Experimental setting

110 Research was conducted in a subalpine summer pasture area located in Buisán-Fanlo
111 (province of Huesca, Spain) in the south-central Pyrenees. The elevation at the study site
112 is approximately 1760 m above sea level and the climate in the area is continental with
113 some oromediterranean influence with a mean atmospheric temperature of 6 °C and mean
114 rainfall of 1500 mm·year⁻¹. The lithology is composed of calcareous Eocene rocks,
115 including detrital sediments of limestone and marlstone. The slope is 10–25% and is
116 exposed to the south. The soils are Calcaric Cambisols of varying thickness with a silt–
117 loam texture in the top 5 cm and a clay–loam texture below 5 cm. During recent decades,
118 the livestock density has decreased by approximately 50%, and undergrazing has resulted
119 in a mosaic-patterned landscape, with nearly monospecific thickets of erizón
120 [*Echinopartum horridum* (Vahl) Rothm.], a thorny shrub encroaching into grasslands
121 dominated by mesophytic grasses such as common bent (*Agrostis capillaris* L.), quaking

122 grass (*Briza media* L.), erect brome (*Bromus erectus* Huds.), Chewing's fescue (*Festuca*
123 *nigrescens* Lam.), and red clover (*Trifolium pratense* L.).

124

125 On 12 November 2015, the wildfire prevention team (EPRIF, the Spanish acronym) of
126 Huesca and the firefighting reinforcement brigades (BRIF) of Daroca (Spain) conducted
127 the prescribed burn of a nearly monospecific stand of *E. horridum* with an area of 3.8 ha
128 (Fig. 1a). The shrub cover of the stand was 75%, and the fuel load was estimated at 9.2 kg
129 m⁻² from aboveground biomass and 1.6 kg m⁻² from the litter (undecomposed or only
130 slightly decomposed plant debris) and duff (partly decomposed organic matter) layers. The
131 topsoil was moist at the surface (1.3–1.7 kg water·kg⁻¹ dry soil in the 0–1 cm depth) and
132 less moist below (0.5–0.8 kg water·kg⁻¹ dry soil in the 1–5 cm depth). Burning was
133 conducted under weak winds (< 8 km·h⁻¹) using a point–source technique; the shrubs were
134 ignited one by one when the shrub cover was not continuous. Temperatures were recorded
135 during burning at a single point on the soil surface and at depths of 1, 2, and 3 cm into the
136 soil using k–type thermocouples. At the soil surface, a maximum temperature of 438 °C
137 and a residence time with temperatures above 100 °C of 12.5 min were recorded, but the
138 heating was only slight at the 1–cm depth (peak of 31 °C) and negligible at the 2– and 3–
139 cm depths.

140

141 On the day of the burn, a smaller area was selected within the stand before burning, near,
142 but not too close to, the starting ignition point, with sufficient soil depth to avoid rocks at
143 or under the soil surface when sampling soils. Within the area, three microsites were
144 selected that were separated by 5 m, as wide as possible and distant enough from rocks to
145 avoid border effects, forming an upside–down equilateral triangle. The spacing was judged
146 in the field to be enough to ensure sample independence, since the triangular design

147 prevented flows of runoff or sediments between the microsites. At each microsite, a 0.25–
148 m² surface was delimited, and samples were collected immediately before burning
149 (preburn) from the litter, duff, and topsoil. Litter was collected by hand–picking from the
150 ground surface. Then duff was obtained by gently scrapping the surface of the underlying
151 topsoil using a trowel. Finally, topsoil samples were obtained by digging with a trowel from
152 the 0–1, 1–2, 2–3, and 3–5 cm depths. Immediately after burning (T0), samples were
153 collected in adjacent 0.25–m² areas at the same microsites from the ash deposited on the
154 soil surface, the remaining duff layer, and the topsoil from the 0–1, 1–2, 2–3, and 3–5 cm
155 depths. Sampling of the duff and topsoil was repeated every six months over the two
156 following years in May 2016 (T6) (Fig. 1e), November 2016 (T12) (Fig. 1f), May 2017
157 (T18) (Fig. 1g), and November 2017 (T24) (Fig. 1h). The study used the preburn state as
158 the reference point of comparison to assess temporal changes, due to either the fire effects
159 or environmental conditions that vary with time, particularly from seasonal and weather
160 effects.

161

162 2.2. Laboratory procedures

163 Research aimed to assess changes in the composition of the organic matter and the labile
164 reservoirs of C, N, phosphorus (P), and sulfur (S) in the litter and duff layers and in the
165 POM within the topsoil layer. The nutrient status of the soil was also investigated through
166 analysis of the plant–available N, P, and S concentrations and related soil properties, such
167 as pH and electric conductivity. A schematic overview of the main analyses is provided in
168 Figure 2.

169

170 The litter, duff, and ash samples were dried at 60 °C in an oven to constant weight and were
171 ground using a cutting mill to pass a 1–mm sieve. Soil samples were divided into two parts:

172 one part was air-dried for 3–5 days until visibly dry, sieved to < 2 mm and stored at room
173 temperature (18–25 °C), and the second part was sieved while still fresh and stored at 4 °C
174 until analysis. For all samples, the water content was determined gravimetrically to constant
175 weight in an oven at 103 °C. The total concentration of the organic matter was assessed
176 from weight-loss-on-ignition at 550 °C for 4 h in a muffle furnace, and the total C and N
177 concentrations were determined using a CN Vario Max elemental analyzer (Elementar,
178 Hanau, Germany). For soil and ash samples, the calcium carbonate equivalent was
179 determined by the volumetric method using a Bernard calcimeter (Proeti, Madrid, Spain),
180 and the concentration of carbonate-C was subtracted from that of the total C to obtain the
181 organic-C concentration.

182

183 The litter, duff, and ash were analyzed for the total concentrations of P and S by ashing in
184 the presence of magnesium nitrate and redissolution in 2*N* hydrochloric acid (Kalra and
185 Maynard, 1991). The total P concentration was spectrometrically determined by the blue-
186 molybdate method (Murphy and Riley, 1962) and the total S concentration was determined
187 by turbidimetry with barium chloride (Rhoades, 1982). The organic composition of the
188 litter, duff, and ash was assessed using an Ankom 200 Fiber Analyzer (Ankom Technol.,
189 Fairport, NY, USA) according to the Van Soest method, which uses neutral-detergent,
190 acid-detergent, and 72% sulfuric acid solutions to sequentially determine the nonstructural
191 (cell content), hemicellulose, cellulose, and lignin-type components of the organic matter,
192 where lignin-type components include lignin as well as other substances with comparable
193 recalcitrance, such as cutin, suberin or waxes (Berg and McClaugherty, 2014). The
194 insoluble mineral residue was separated by ashing in a muffle furnace at 550 °C, and its
195 concentration was subtracted from that of the lignin-type components.

196

197 The soils and ash were analyzed for chemical properties and nutrient concentrations. The
198 pH was measured in a 1:2.5 air-dried soil:water or ash:water mass suspension, and the
199 electric conductivity was measured in a 1:5 air-dried soil:water or ash:water mass extract.
200 The bioavailable N in the forms of ammonium (ammonium-N) and nitrate (nitrate-N) was
201 extracted from the fresh soil or from ash with a 2 M potassium chloride solution at a 1:10
202 mass:volume ratio and was determined with steam distillation through the magnesium
203 oxide-Devarda alloy method (Bremner, 1965). Plant-available P was extracted from the
204 air-dried soil or ash following the Olsen procedure using a 0.5 M sodium bicarbonate
205 solution with pH 8.5 at a 1:20 mass:volume ratio (Watanabe and Olsen, 1965) and was
206 spectrometrically determined by the molybdate-blue method (Murphy and Riley, 1962).
207 The soluble (plant-available) S was determined in the 1:5 mass:volume aqueous extract by
208 turbidimetry with barium chloride (Rhoades, 1982).

209

210 For the POM analysis, soil subsamples were dispersed using a sodium hexametaphosphate
211 solution, and the sand-sized fraction was separated using a 50- μ m sieve. From the sand-
212 sized fraction, a portion was ground using a mortar and was analyzed for POM, C
213 [particulate organic-C (POC)] and N [particulate organic N (PON)] concentrations
214 following the methods described above. Another part was placed in water and stirred to
215 separate the floating POM from the mineral sand (Saenger et al., 2015). In the isolated
216 POM, the total P [particulate organic P (POP)] and S [particulate organic S (POS)]
217 concentrations and the nonstructural, hemicellulose, cellulose, and lignin-type organic
218 fractions were determined using the methods described above.

219

220 2.3. Data processing and statistical analyses

221 Results of the analyses were expressed in concentration units on a mass per oven-dry mass
 222 basis. The concentrations of nonstructural, hemicellulose, cellulose, and lignin-type
 223 components were expressed as percentages of the total (ash-free) organic matter. The
 224 concentrations of the POC, PON, POP, and POS were expressed on a mass per mass of
 225 total soil basis. The values of nonparticulate organic-C (NPOC) and nonparticulate organic
 226 N (NPON) were obtained by subtracting the POC and PON concentrations from the total
 227 concentrations of soil organic-C and N. The C:N, C:P, C:S, N:P, N:S, and P:S
 228 stoichiometric mass ratios were calculated for the litter, duff, ash, and POM and the C:N
 229 ratio was calculated for the NPOM.

230

231 The initial weight loss resulting from the combustion and volatilization of the soil
 232 components during burning was calculated for each of the soil depths from the total SOM
 233 values, according to the following formula [Eq. (1)]:

$$234 \text{ Mass loss} = 1000 \times \frac{SOM_{preburn} - SOM_{T0}}{1000 - SOM_{T0}} \quad (1)$$

235 where the weight loss is expressed as $\text{g} \cdot \text{kg}^{-1}$ initial concentration, and $SOM_{preburn}$ and
 236 SOM_{T0} are the SOM concentrations just before and after burning, expressed as $\text{g} \cdot \text{kg}^{-1}$ soil.

237 The weight loss resulted in the apparent enrichment of any remaining components in the
 238 soil after burning, which was estimated with the following calculation [Eq. (2)]:

$$239 \text{ Concentration effect} = \frac{1000}{1000 - \text{Mass loss}} \quad (2)$$

240 Taking into account the concentration effect, the actual losses of total N, P, and S from the
 241 POM were estimated using the following calculation [Eq. (3)]:

$$242 \Delta(N, P, S) = \frac{POM_{preburn} \times PO(N, P, S)_{preburn}}{1000} - \frac{POM_{T0} \times PO(N, P, S)_{T0}}{1000 \times \text{Concentration effect}} \quad (3)$$

243 and of total N from the NPOM according to the following calculation [Eq. (4)]:

$$244 \Delta N = \frac{NPOM_{preburn} \times NPON_{preburn}}{1000} - \frac{NPOM_{T0} \times NPON_{T0}}{1000 \times \text{Concentration effect}} \quad (4)$$

245 where $POM_{pre-burn}$, POM_{T0} , $NPOM_{pre-burn}$, and $NPOM_{T0}$ are the concentrations of POM and
246 NPOM before and after burning, expressed as $g \cdot kg^{-1}$ soil; $PO(N, P, S)$ and $NPON$ are the
247 total concentrations of N, P, and S in the POM and of N in the NPOM, expressed as $mg \cdot kg^{-1}$
248 ¹ organic matter; and $\Delta(N, P, S)$ and ΔN are the losses estimated in $g \cdot kg^{-1}$ initial soil.

249

250 The changes in the composition of the duff and the topsoil over the study period (preburn,
251 T0, T6, T12, T18, and T24) across the distinct layers (duff and topsoil at 0–1, 1–2, 2–3,
252 and 3–5 cm depths) were analyzed using a linear mixed model of repeated measures in IBM
253 SPSS Statistics v. 22 (IBM Corporation, Armonk NY, USA).

254

255 An initial model was built including time and layer as fixed effects, and the sampling
256 microsite within the stand as a random effect to deal with the risk of spatial dependence.
257 For each variable, the corrected Akaike information criterion (AICc) and the Bayesian
258 information criterion (BIC) were used to select the best model among various alternatives.
259 Thus, the AIC and BIC were used to test for the need of the random “microsite”, for which
260 models either including or not the random “microsite” term were compared (Supplementary
261 Table 1). Lower AIC and BIC values were obtained when the random factor was included
262 than when not included (with the only exceptions of the models for some stoichiometric
263 ratios), which indicates that the soil properties varied across the microsites, hence the
264 inclusion of the random term was fully justified. The AICc and BIC were also used to
265 determine the most suitable covariance structure (Onyiah, 2008), which was a first-order
266 self-regressive structure in all cases. The assumptions of normality and homoscedasticity
267 were checked through Shapiro–Wilk tests and visually with residual plots, and, when
268 necessary, the variables were square root- or log-transformed for statistical analyses.

269

270 In almost all cases, the interaction between the factors time and layer was statistically
271 significant or near significant at $P < 0.05$ (Supplementary Table 2). Given interactions,
272 examining the simple effects of one of the factors in the interaction at the individual levels
273 of the other factor is recommended (Onyiah, 2008). Because the focus of this study was
274 on temporal changes, all analyses were repeated separately for each layer through
275 consideration of only the microsite and time factors. The results of the analyses for the
276 effects of time on the individual layers are shown in Supplementary Table 3. When
277 significant differences were identified for time, pairwise comparisons were performed with
278 Bonferroni tests at $P < 0.05$. For ease of interpretation, means given in the results were
279 based on the nontransformed data.

280

281 3. RESULTS

282 3.1. Characteristics of the ground surface and vegetation

283 Fire almost completely consumed the aboveground biomass and the litter layer (Fig. 1a, b).
284 After burning, the soil surface was covered with a mix of unconsumed material from the
285 duff layer (Fig. 1c) and ashes of various colors (white, gray, black) indicative of an irregular
286 fire intensity (Fig. 1d), which were intercalated with some patches of bare ground.

287 The ash produced by the fire was alkaline in nature and was rich in carbonates and soluble
288 substances (Table 1). The concentration of the organic matter was less than 20% and was
289 proportionally rich in recalcitrant (lignin-type) components and much poorer in the most
290 labile (hemicellulose, nonstructural) components. On average, 22% of the total S, 4% of
291 total P, and 0.7% of total N in the ash were in forms available to plants.

292

293 Six months after the burn (Fig. 1d), i.e. after the winter and snow-covered periods, a large
294 amount of burnt residue remained at the soil surface, covering 90% of the area. At 12

295 months after burning (Fig. 1e), mixed ash and duff material and limited incorporation of
296 charred organic matter into the topsoil were noticeable. Plant recolonization (as shown in
297 Fig. 1e, 1f, 1g, 1h) started slowly (5% cover after 6 months), but spread to most of the land
298 surface after 24 months (Fig. 1h). The recolonizing vegetation was dominated by plants
299 that resprout from rhizomes or bulbs, such as glaucous sedge (*Carex flaca* Schreb.), dwarf
300 sedge (*C. humilis* Leysser), cypress spurge (*Euphorbia cyparissias* L.), English iris [*Iris*
301 *latifolia* (Mill.) Voss.], wall germander (*Teucrium chamaedrys* L.), and Teesdale violet
302 (*Viola rupestris* F.W. Schmidt), along with *E. horridum*, which sprouts from seeds.

303

304 3.2. Organic compositions of the litter, duff, and topsoil

305 The litter and duff layers had similar organic compositions before the burn and differed
306 mainly in that the lignin-type components were at a greater concentration in the duff than
307 in the litter matter (Fig. 3). Burning removed the litter layer, but most of the duff layer
308 remained at the soil surface (Fig. 1c, 1d). During the study period, the composition of the
309 duff layer showed little change, except for a tendency toward selective preservation of the
310 most-recalcitrant polymers over time (Fig. 3). Indeed, the hemicellulose concentration
311 decreased slightly, but significantly at T6 and T24 compared to preburn and T0, whereas
312 the cellulose concentration increased by T24 compared to preburn. The lignin-type
313 predominant fraction did not change.

314

315 Burning did not immediately affect the organic composition of the topsoil. During the study
316 period, early (T6) and late (T24) enrichments in the lignin-type components were recorded
317 in the 0–1 cm depth. Similarly, the proportions of nonstructural components in organic
318 matter significantly increased first (T12) in the 3–5 cm and afterwards (T18) in the 1–2 and

319 2–3 cm depths. The NPOM changed little during the study period, showing only a decrease
320 in its relative proportion in the 3–5 cm depth at T12.

321

322 3.3. Elemental composition of the organic matter

323 The topsoil underwent an average loss of 71% of its total SOM (POM + NPOM) in the 0–
324 1 cm depth (Fig. 4a). The loss equaled 39% of the soil mass in the top 1 cm before burning
325 and resulted in an apparent enrichment of 65% of the soil components remaining after
326 burning, solely as a result of their concentration in the burned soil. The total organic-C and
327 N underwent mean losses of 72% (Fig. 4b) and 68% (Fig. 4c), respectively, of their preburn
328 concentrations in the same depth. Burning most severely impacted the POM, which
329 accounted for 73% of the total loss of organic matter, 82% of the total C loss, and 60% of
330 the total N loss, on average. Conversely, the NPOM only lost 27% of its initial
331 concentration, which, added to the mass-concentration effect, caused its concentration in
332 the soil to increase from $321 \pm 31 \text{ g}\cdot\text{kg}^{-1}$ to $463 \pm 109 \text{ g}\cdot\text{kg}^{-1}$ (mean \pm standard deviation)
333 after burning. In the underlying soil, the 1–2, 2–3, and 3–5 cm depths, the changes in SOM
334 as result of burning were much lower, resulting in little or no concentration effect (+4%,
335 +0%, and –0%, on averages).

336

337 The combustion of the POM and NPOM in the 0–1 cm depth released, on average, 12 g of
338 N per kg of initial soil, of which 57% was released from the POM and 43% from the NPOM.
339 From the POM alone, 0.17 g P kg^{-1} and 0.44 g of S $\cdot\text{kg}^{-1}$ initial soil were released, on
340 average. The N, P, and S releases were approximately 120, 40, and 7.5 times larger,
341 respectively, than their bioavailable concentrations before burning in the 0–1 cm depth.
342 Despite this, no noticeable increases were observed in the concentrations of bioavailable N
343 or P immediately after burning. An increase in the soluble concentration of S was observed,

344 although the increase was 12 times smaller than the amount of S released from the POM
345 combustion.

346

347 The immediate impact of burning on the POC concentration was followed by further losses
348 during the first months after burning and the POC concentration reached the lowest in the
349 0–1 and 1–2 cm depths at T6 (Fig. 5a). The NPOC concentrations decreased significantly
350 in the 0–1 and 1–2 cm depths just after burning, but increased at T6 up to levels that were
351 no longer lower than the preburn levels. The NPOC reached its greatest concentration at
352 T18 after burning, which was statistically significant in the 2–3 and 3–5 cm depths. Like
353 the POC, the PON decreased during the first months following burning in the 0–1 and 1–2
354 cm depths, reaching its significantly lowest concentrations at T6 and greatest
355 concentrations at T18 (Fig. 5b). In contrast, the NPON in the 0–1 cm depth, unlike the
356 NPOC, did not show a fast recovery, but continued to decrease after burning, reaching a
357 minimum at T12, and did not revert to preburn levels during the study period. No changes
358 over time occurred for PON or NPON at greater depths. The soil POP concentrations (Fig.
359 5c) did not decrease as a result of burning and showed transient variations during the study
360 period, reaching its lowest concentrations at T6 and its maximum concentrations at T18.
361 The POS concentration (Fig. 5d) decreased at T0 to its lowest concentration in the 0–1 cm
362 depth, which were significantly lower than the preburn, T18, and T24 concentrations.
363 Conversely, POS concentration in the 1–2, 2–3, and 3–5 cm depths was lowest at preburn,
364 then increased thereafter until reaching larger concentrations at T18.

365

366 The C:N ratio of the duff layer (Fig. 6a) did not show significant variations during the study
367 period. The C:N ratio of the POM decreased significantly in all depths at T6 and at later
368 sampling times, whereas the C:N ratio of the NPOM increased. The C:P ratio (Fig. 6b) was
369 lowest for the duff layer at T0 and increased significantly to the largest values at T18,

370 whereas the C:P ratio of the POM was largest before burning and decreased significantly
371 at T6 in the 0–1 cm depth and at T18 in the 1–2, 2–3, and 3–5 cm depths. The C:S ratio
372 (Fig. 6c) increased in the duff at T6 and decreased in the POM at T6. The N:P ratio (Fig.
373 6d) was largest in the duff layer at T18, when it was significantly greater than that preburn.
374 In contrast, the N:P of the POM was lowest at T18, when its values were significantly lower
375 than that preburn in the 0–1 cm depth and generally lower than those at T12 and T24. The
376 N:S ratio (Fig. 6e) of the duff layer tended to increase after T6; the values at T18 were
377 significantly lower than the preburn values. The N:S ratio of the POM did not show
378 significant variations during the study period. Finally, the P:S ratio (Fig. 6f) increased after
379 the burn until reaching values significantly greater than the initial values in the duff at T12
380 and in the POM at T18.

381

382 3.4. Soluble and plant–available elements

383 Electrical conductivity was observed to increase significantly in the top 3 cm of the soil
384 (Fig. 7a) immediately after burning (T0). The largest conductivity was observed at T0 and
385 T12 and was larger than that at T6 in the 0–1 cm depth and that at T24 in the 1–2 and 2–3
386 cm depths. The soil pH (Fig. 7b) remained alkaline (7.4–7.7) during the entire study period
387 and was only affected by burning at T0 in the 0–1 cm depth when pH was slightly, but
388 significantly greater than preburn, T6, T18, and T24 values. The calcium carbonate
389 equivalent did not differ during the study period and was significantly lower in the 0–1 cm
390 depth ($82 \pm 65 \text{ g}\cdot\text{kg}^{-1}$) than in greater (1–5 cm) depths ($115 \pm 73 \text{ g}\cdot\text{kg}^{-1}$).

391

392 The concentrations of ammonium– and nitrate–N showed large variability before burning,
393 which made their changes immediately after burning undetectable. Ammonium–N (Fig. 7c)
394 increased postburning until peaking at T18 before decreasing at T24. Apart from this trend,

395 the spring ammonium–N concentrations were generally, though not significantly, greater
396 than those of the preceding autumn. For nitrate–N (Fig. 7d), the largest values in the 2–3
397 and 3–5 cm depths were reached at T6, and the largest concentrations in the 0–1 and 1–2
398 cm depths were reached at T12. There was a sharp decrease at T18, although the
399 concentrations seemed to increase again, though not significantly, at T24. In this case, the
400 autumn concentrations were generally, though not significantly, greater than those in the
401 preceding spring. The plant–available P (Fig. 7e) did not vary during burning, but decreased
402 gradually thereafter in all depths, reaching its minimum concentrations at the end of the
403 study period (T24). Within the general decline, the concentrations were numerically,
404 though not significantly, greater in spring (T6, T18) than in autumn (T12, T24). At T0, the
405 soluble S (Fig. 7f) reached its maximum concentration, which was significantly greater
406 than the preburn concentrations in the 0–1 cm depth. Soluble S concentrations decreased
407 with time since burning in the 1–2, 2–3 and 3–5 cm depths, reaching their minimums at
408 T24. In addition, concentrations were generally, though not significantly, greater in autumn
409 (T12, T24) than in spring (T6, T18).

410

411 4. DISCUSSION

412 4.1. Direct effects of fire on the organic layers

413 The maximum temperature recorded at the soil surface (438 °C) was intermediate between
414 the characteristic temperatures for the formation of black carbon (approximately 350 °C)
415 and for complete ashing (550 °C) (González–Pérez et al., 2004) and were sufficient to
416 produce the formation of calcite from organic–C and Ca and the partial volatilization of N
417 and S, but not of P (Bodí et al., 2014). The combustion produced an ash material where the
418 organic matter was a minor component that was dominated (69%) by lignin–type
419 components. The ash composition contrasted with that of the litter before burning, where

420 the lignin–type components accounted for 30% of the organic matter and that reported for
421 the fine twigs of *E. horridum* (Marinas et al., 2003; Mora et al., 2018), where these
422 components ranged from 15 to 18% on average. The larger concentration of lignin–type
423 components in the ash indicated a selective preservation of the lignin–type components,
424 likely as result of their lower flammability, greater resistance to heat and ease of
425 carbonization in comparison with the other structural polymers in plants (Yang et al., 2007).
426 These results were consistent with the decreasing concentration of O–alkyl C, which has
427 been reported in ¹³C nuclear magnetic resonance analyses of the burning of litter and is
428 attributed to the intense thermal degradation of hemicellulose and cellulose (Alexis et al.,
429 2010; Merino et al., 2015). However, compared with litter, the ash had only slightly lower
430 concentrations of cellulose but much lower concentrations of hemicellulose, which likely
431 reflected a lower contribution to the ash from the litter and duff layers than from the
432 biomass of *E. horridum*, where the cellulose concentration was double that of the
433 hemicellulose (Marinas et al., 2003; Mora et al., 2018).

434

435 With respect to the elemental composition, the ash had N, S, and P total concentrations that
436 were 2.1, 6.7, and 36 times lower, respectively, than that in the litter before burning and
437 2.3, 20, and 26 times lower than the fine twigs of *E. horridum* (Marinas et al., 2003; Mora
438 et al., 2018). These results highlighted a lower loss of total P relative to S, of S relative to
439 N, and of N relative to C, which was consistent with their different volatilization
440 temperatures. As a consequence, the ash was depleted in mineral N, but rich in soluble,
441 plant–available S, even though some of the S that may have been insolubilized by contact
442 with the carbonates under the alkaline conditions of the ash (Pereira et al., 2019). Due to
443 its high volatilization temperature, P was abundant in ash, but only a small part was in a
444 form available for plants. The ratio of the bioavailable P to the total P has been reported to

445 decrease in ash with increasing temperatures as inorganic phosphate ions become bound to
446 cations in the ash, forming insoluble metal phosphates (Caon et al., 2014; Gray and
447 Dighton, 2006).

448

449 However, the burning of the duff layer did not modify the proportions of the distinct
450 fractions in the organic matter or the elemental stoichiometry. These results could be related
451 to the uneven incidence of fire in the duff layer, which may, in some cases, have been
452 completely removed, while in others, was preserved, possibly with few changes to its
453 composition. Results from this study were in agreement with those of other studies (e.g.,
454 Alexis et al., 2010; Butler et al., 2017), which reported few differences between the
455 composition of the litter before burning and that remaining at the soil surface and
456 emphasized the influence of unburnt organic matter on the composition of the organic
457 layers after the application of prescribed burning.

458

459 4.2. Direct effects of fire on the topsoil

460 Burning considerably affected the upper 1 cm of the topsoil, whereas, at greater depth, the
461 immediate effects on the soil properties were minimal or insignificant. This result was in
462 agreement with several studies on the effects of burning shrublands (e.g., Neary et al., 1999;
463 Caon et al., 2014) or controlled burning under laboratory conditions (Badía –Villas et al.,
464 2014), which reported no element volatilization from below 2 cm.

465

466 Results showed that C losses came mostly (82%) from the combustion of POM, but the
467 release of N came from both the POM and NPOM in comparable proportions (57% and
468 43%, respectively) because of the greater concentration of total N in NPOM in comparison
469 to that in the POM (Fig. 6a). Most of the released N was lost through volatilization from

470 the soil, as judged by the lack of significant increases in the levels of the mineral N in the
471 soil after burning.

472

473 The release of S was estimated from only the POM, but a considerable release of S may
474 have also occurred from the NPOM, since, as with N, S is usually more abundant in NPOM
475 than in POM, especially in soils from cool climates (Amelung et al., 1998). The S release
476 was associated with an increase in S concentration in soluble, plant-available forms, but
477 the increase was approximately 40 times lower than the estimation of the S released from
478 the POM in the 0–1 cm depth in this study. Significant S immobilization was unlikely to
479 have occurred since the immobilization of this element normally takes places at pH values
480 close to 12 (Pereira et al., 2019), which was much greater than the pH of the topsoil in the
481 0–1 depth, which ranged from 7.3–7.8. Thus, it can be deduced that most of the mineralized
482 S was lost through volatilization from the soil. The loss of S from the upper soil layer is
483 generally not expected in low- or moderate-intensity fires (Neary et al., 1999) because soil
484 temperatures rarely rise enough in those fires to generate S losses from the topsoil
485 (Tiedemann, 1987).

486

487 The P released from the POM, to which an unknown amount of P released from NPOM
488 should be added, did not immediately increase the concentration of the plant-bioavailable
489 P in the soil. This may be due to increase in the pH in the 0–1 cm depth, which at the
490 calcareous study site might contribute to making P unavailable via calcium fixation. The P
491 enrichment may have also been partially overridden by the loss of some amount of P in the
492 form of fine-ash particles, which are highly susceptible to removal from a site by
493 convection in the smoke columns during fire (Raison et al., 1985). Ash is highly enriched
494 in low-volatile elements, hence ash transport in the smoke can result in substantial P export

495 (Raison et al., 1985). Dense smoke was observed during burning (Fig. 2b) and is typical
496 when densely branched, shrub thickets are burned (Bell and Adams, 2008).

497

498 Few variations were identified in the proportions of the distinct fractions of the SOM in the
499 0–1 cm depth as a result of burning, which contrasted with the results of other studies that
500 reported a preferential removal of carbohydrates and selective preservation of lignin and
501 humified SOM (Badía–Villas et al., 2014; Jiménez–Morillo et al., 2016). In this study, it
502 was likely that, as in the duff layer, the most surficial layer of the topsoil experienced
503 uneven exposure to the fire, leading to an almost complete removal of the SOM in some
504 areas, but resulted in few significant effects in the SOM in other areas. Furthermore, some
505 of the organic matter may have been protected within the soil aggregates, which was
506 consistent with the results presented for this same burn by Girona–García et al. (2018b)
507 who reported fewer fire effects on the SOM in those soil fractions with greater degrees of
508 aggregation.

509

510 4.3. Postburn evolution of the organic composition

511 During the first two years following the burn, the composition of the duff layer remained
512 fairly stable. The main observable trend was the decrease in the hemicellulose by the spring
513 after the first winter after burning (T6). Hemicellulose is most labile among the major
514 components of litter and has been reported to be actively degraded during the first year after
515 exposure to fire (Pyke, 2003). At the same time (T6), the C in the topsoil was affected by a
516 disintegration process, as can be inferred from the decrease in POC and increase in NPOC.
517 A similar process likely affected the POP, which also reached its minimum concentration
518 by T6. Because the microbial biomass of the topsoil was severely affected by the burn
519 (Girona–García et al., 2018a), such disintegration can be related to physical mechanisms,

520 such as gelifluxion and cryoturbation, which play important roles in the breakdown of
521 charcoal in soils in cold regions (Preston and Schmidt, 2006). Physical alteration, however,
522 did not affect the N within the POM or the NPOM, as was also observed by Naisse et al.
523 (2015), who attributed the result to the greatest N stability within the heterocyclic aromatic
524 structures in the charcoal.

525

526 In the second spring (T18), a large incorporation of sand-sized, N-rich ash into the
527 shallowest part of the topsoil occurred, which manifested as increases in the POC and PON
528 in the 0–1 and 1–2 cm depths. Simultaneously, in 1–2 and 2–3 cm depths, enrichment in
529 fine C-rich organic particles occurred, as evidenced by the increases in the NPOC, but not
530 in NPON, and large C:N ratios. The fine particles may have resulted from the breakdown
531 of the particulate charcoal and the subsequent transport to deeper soil layers by drainage
532 and/or cryoturbation (Hobley, 2019).

533

534 In contrast, the hemicellulose and the nonstructural components of the POM increased in
535 the 3–5 cm depth at T12 and extended to shallower depths (1–2 and 2–3 cm depths) at T18.
536 The low thermal stability of these compounds makes it unlikely that they were of pyrogenic
537 origin, and their spreading from the subsoil suggests that they probably originated from the
538 belowground biomass of the plants that resprouted via rhizomes and bulbs, which were
539 observed to be predominant in the recolonizing vegetation. Hemicelluloses and
540 nonstructural carbohydrates act as mobile reservoirs of energy and nutrients for many
541 plants and play an important role in the capacity for resprouting after burning (Martínez–
542 Vilalta et al., 2016).

543

544 4.4. Stoichiometric effects

545 Mora et al. (2018) analyzed the major nutrients in the fine twigs of *E. horridum* in the study
546 region, obtaining an average N:P ratio of 46, which was well above the threshold of 16–20
547 for P limitation (Güsewell, 2004; Koerselman and Meuleman, 2006). In this study, the litter
548 layer, which mainly consisted of fine twigs, showed a similar N:P ratio of approximately
549 53, which related to the symbiotic N inputs of leguminous plants and to the presence of
550 calcite in the soil, which favors the retrogradation of P into forms that are not plant–
551 available. Mora et al. (2018) also reported an average N:S ratio of 6.5, which was much
552 lower than the recommended critical ratio of 17 (Kozłowska–Strawska and Kaczor, 2009),
553 indicating a greater limitation of S relative to N. In the present study, greater N:S ratios
554 (16.5 on average) were measured in the litter, suggesting strong resorption of S during
555 senescence, which was consistent with a strong plant S demand. Total S concentrations
556 were also observed to decrease in the organic matter from the litter to the duff layer, which
557 pointed to a large microbial S demand. Both findings were indicative of a certain degree of
558 S limitation in the preburn ecosystem.

559

560 The stoichiometric ratios of the decomposing organic matter provide insight into the ease
561 with which the contained nutrients can be mobilized (Prescott, 2005; Berg and
562 McClaugherty, 2014). In this study, the C:N ratio averaged approximately 30 in the litter,
563 approximately 20 in the duff during the entire study period and approximately 20 in the
564 POM immediately before and after burning. According to the criteria compiled by Prescott
565 (2005), these values allow for a moderate release of N. Furthermore, since T6, the C:N of
566 the POM decreased to mean values ranging from 5 to 13, allowing even faster N
567 mobilization, which coincided with the largest levels of nitrate–N that were observed at T6
568 and T12.

569

570 In the case of the C:P ratio, the litter showed large initial values, with an average close to
571 1600. The duff also showed large values before burning (700 on average), immediately
572 after the fire (400) and, especially after the fire (700 to 1400). The POM had C:P ratios
573 between 400 and 1200 before burning. All of these values indicate a low capacity for P
574 release. After burning, the C:P of the POM sharply decreased to between 40 and 200
575 between T6 and T18, indicating large potential for P mobilization. This phenomenon
576 showed signs of weakening at T24, when large C:P ratios, 300 to 400 in the 0–1 and 1–2
577 cm depths, were measured again.

578

579 The N:P ratio of the POM was observed to decrease to below 15 at T18, which indicated a
580 switch in the nutrient limitation from P to N at this time (Prescott, 2005). It is well known
581 that the incorporation of P-rich, charred materials into the soil can induce a shift from P to
582 N limitation after fire (Toberman et al., 2014), which has led to recognize fire as an essential
583 element for maintaining productivity in P-limited ecosystems (Dijkstra and Adams, 2015;
584 Butler et al., 2017). In this study, the shift in limitation was short-lasting, as the N:P ratio
585 increased in the next sampling period (T24), although the increase occurred in autumn,
586 when large quantities of nutrients were immobilized by microorganisms. Microorganisms
587 are known to immobilize large quantities of key elements, such as P (Weintraub, 2011),
588 which has been suggested to affect low-nutrient habitats beyond the simple effect of the
589 ash layer (Huang et al., 2013). This result was in agreement with the observations by
590 Múgica et al. (2018) in the western Pyrenees, who reported temporary increases in plant-
591 bioavailable N only during the first year after fire, but identified more persistent (> 18
592 months) effects in increasing the N concentration in the microbial biomass and decreasing
593 the microbial N demand.

594

595 4.5. Postburn evolution of soil nutrient availability

596 Results of this study showed tightly inter-related seasonal variations of low statistical
597 significance in the plant-available concentrations of N, P, and S in the topsoil, which
598 pointed to a greater concentration of ammonium-N and available P in spring and nitrate-
599 N and soluble S in autumn. The variations mirrored the typical seasonal oscillations of plant
600 and microbial nutrient resources in high-mountain habitats. For example, during spring
601 snowmelt, high-mountain soils receive considerable nutrient inputs from the release of the
602 nutrients immobilized in the microbial biomass since autumn and from the discharge of
603 nutrients mineralized by microorganisms during winter (Bardgett et al., 2005). However,
604 melting is associated with intense leaching, so more soluble nutrients, such as nitrate-N
605 (Hagedorn et al., 2001; Clement et al., 2012) or soluble S, are removed, whereas other less
606 mobile nutrients, such as ammonium-N (Makarov et al., 2010; Clement et al., 2012) and P
607 (Costa et al., 2019), are retained. During summer, nutrients are taken up in large amounts
608 by plants (Bardgett et al., 2005), but the nitrate concentrations increase in summer and until
609 autumn due to nitrification (Makarov et al., 2010), while sulfate concentrations can increase
610 due to the deposition of S-rich aerosols in the Pyrenean ranges during summer (Ripoll et
611 al., 2015).

612

613 Seasonal soil nutrient dynamics superpose and interact with the fire effects. Thus, during
614 the second spring (T18), the maximum ammonium-N concentration coincided with the
615 incorporation of large amounts of N-rich ash into the soil. Likewise, the minimum nitrate-
616 N concentrations occurred at the same time as the translocation of the fine particles of
617 pyrogenic organic matter, which were both caused by intense leaching that accompanied
618 snowmelt. In general, the increases in the N availability that could be attributed to burning
619 were short-lived and rapidly reverted back to preburn levels during the following few

620 months. For the soluble S and plant-available P concentrations, in contrast, the apparent
621 trend was a gradual decline after burning, where both reached their minimum
622 concentrations by the end of the study period (T24).

623

624 5. CONCLUSIONS

625 The analysis of the ash that resulted from the burning revealed a strong enrichment in the
626 most recalcitrant (lignin-type) fractions and the least-volatile elements (S and especially
627 P) relative to those of the plant biomass and the litter and duff layers. There was also a
628 much greater effect of burning on the POM than on the NPOM in the topsoil. These results
629 supported the hypothesis that the different thermal stabilities of the organic components
630 should lead to a heterogeneous effect of the controlled burn on the organic matter. The
631 effects of combustion on the NPOM, despite being minor, were responsible for releasing
632 large amounts of elements, such as N and S, that are more abundant in NPOM than in POM.
633 In contrast, the organic matter that remained at the soil surface and in the topsoil after
634 burning showed only slight changes relative to that before burning, which could be related
635 to an uneven influence of burning, as a result of which some of the organic matter was
636 maintained with few or no modifications.

637

638 The investigated habitat was shown to be P-limited, following the study area's calcareous
639 nature, and also likely S-limited. The immediate effects of burning included strong N and,
640 to a lesser degree S, volatilization from the organic matter and the immobilization of P into
641 forms unavailable to plants. During the two years following burning, the bioavailable
642 concentrations of the studied nutrients showed short-lived increases in N or gradual
643 decreases in S and P, which were likely related to the strong demand for these elements
644 from microorganisms and plants and, probably, to intense leaching. More consistently, the

645 soil nutrient status was also modified by the incorporation and breakdown of nutrient-rich
646 ash particles during the first year and especially during the second year after burning. The
647 gradual disintegration of the particulate charcoal and its transport to depth by cryoturbation
648 and/or leaching may have extended the effect of burning on the nutrients to beyond the
649 period for which the effects on bioavailable concentrations were observed.

650

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659

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896 **FIGURE CAPTIONS**

897 **Figure 1** Images of the study site before (a) and during (b) burning; of the soil surface after
898 burning (c, d); and of the study site 6 (e), 12 (f), 18 (g), and 24 months (h) after burning.

899 **Figure 2** Schematic diagram of the main analyses carried out on the litter, duff, ash, and
900 soil samples collected immediately before (preburn) and after burning (T0) and 6, 12, 18,
901 and 24 months after burning (T6, T12, T18, and T24). OM = organic matter, POM =
902 particulate organic matter.

903 **Figure 3** Postburn evolution of the proportions (mean \pm standard error) of nonstructural,
904 hemicellulose, cellulose, lignin-type, and nonparticulate organic matter (NPOM) of the
905 litter and duff layers and of the topsoil in the 0–1, 1–2, 2–3, and 3–5 cm depths. The
906 headings of the columns denote the sampling time: immediately before (preburn) and after
907 burning (T0) and 6, 12, 18, and 24 months after fire (T6, T12, T18, and T24). Within each
908 layer or depth, values followed by the same letter do not differ among sampling periods.

909 **Figure 4** Mean changes (\pm standard deviation) resulting from combustion and apparent
910 enrichment in the concentrations of the particulate (POM) and nonparticulate (NPOM)
911 organic matter (a), particulate (POC) and nonparticulate (NPOC) organic carbon (b), and
912 particulate (PON) and nonparticulate (NPON) nitrogen (c) in the topsoil at the 0–1, 1–2, 2–
913 3, and 3–5 cm depths. The headings of the columns denote the sampling time: immediately
914 before (preburn) and after burning (T0) and 6, 12, 18, and 24 months after burning (T6,
915 T12, T18, and T24). Values (mean \pm SD) are expressed relative to the initial (preburn)
916 concentrations.

917 **Figure 5** Postburn evolution of the concentrations (mean \pm standard error) of the particulate
918 (POC) and nonparticulate (NPOC) organic carbon (a), particulate (PON) and nonparticulate
919 (NPON) nitrogen (b), particulate organic phosphorus (POP) (c), and particulate organic
920 sulfur (POS) (d) in the topsoil at the 0–1, 1–2, 2–3, and 3–5 cm depths. The headings of the

921 columns denote the sampling time: immediately before (preburn) and after burning (T0)
922 and 6, 12, 18, and 24 months after burning (T6, T12, T18, and T24). Within each depth, the
923 values followed by the same letter do not differ among sampling periods.

924 **Figure 6** Postburn evolution of the values (mean \pm standard error) of the C:N ratio of the
925 litter and duff layers and of the particulate (POM) and nonparticulate (NPOM) organic
926 matter of the topsoil in the 0–1, 1–2, 2–3 and 3–5 cm depths; and of the C:P (b), C:S (c),
927 N:P (d), N:S (e), and P:S (f) ratios of the litter and duff layers and of the POM in the topsoil
928 in the 0–1, 1–2, 2–3 and 3–5 cm depths. The headings of the columns denote the sampling
929 time: immediately before (preburn) and after burning (T0) and 6, 12, 18, and 24 months
930 after burning (T6, T12, T18, and T24). Within each layer or depth, the values followed by
931 the same letter do not differ among sampling periods.

932 **Figure 7** Postburn evolution of the soil properties (mean \pm standard error) in the topsoil at
933 the 0–1, 1–2, 2–3, and 3–5 cm depths for (a) electric conductivity, (b) pH and
934 concentrations of ammonium–N (c), nitrate–N (d), Olsen–P (e), and water–soluble S (f).
935 The headings of the columns denote the sampling time: immediately before (preburn) and
936 after burning (T0) and 6, 12, 18, and 24 months after burning (T6, T12, T18, and T24).
937 Within each depth, the values followed by the same letter do not differ significantly
938 between sampling periods.

939 **Table 1** Chemical composition (mean \pm standard deviation) of the ash in comparison with
 940 that of the litter layer before the fire along with the average values of the fine twigs of
 941 *Echinopartum horridum* reported by Marinas et al. (2003) (1) and Mora et al. (2018) (2).

	Ash	OL layer	Twigs of <i>E. horridum</i>	
			(1)	(2)
pH (1: 2.5 solid: water)	11.6 \pm 0.8			
Electric cond. (mS \cdot cm ⁻¹ , 1: 5)	8.8 \pm 2.0			
Carbonates (g \cdot kg ⁻¹)	325 \pm 45			
Ammonium-N (mg \cdot kg ⁻¹)	38.4 \pm 1.7			
Nitric-N (mg \cdot kg ⁻¹)	6.6 \pm 0.6			
Olsen-P (mg \cdot kg ⁻¹)	419 \pm 32			
Soluble S (mg \cdot kg ⁻¹)	1254 \pm 710			
Total P (g \cdot kg ⁻¹)	9.89 \pm 7.05	0.27 \pm 0.03		0.38 \pm 0.10
Total S (g \cdot kg ⁻¹)	5.79 \pm 1.85	0.86 \pm 0.07		0.28 \pm 0.08
Total organic matter (g \cdot kg ⁻¹)	223 \pm 90	948 \pm 7		967 \pm 6
Total organic-C (g \cdot kg ⁻¹)	131 \pm 46	431 \pm 28		
Total N (g \cdot kg ⁻¹)	6.87 \pm 3.14	14.2 \pm 1.08	15.7 \pm 6.1	17.0 \pm 1.5
C:N ratio	20.5 \pm 3.1	31.2 \pm 1.4		
C:P ratio	20 \pm 13	1600 \pm 100		
C:S ratio	25 \pm 13	501 \pm 35		
N:P ratio	1.1 \pm 0.8	52.7 \pm 5.6		46.4 \pm 10.3
N:S ratio	1.4 \pm 0.8	16.5 \pm 1.9		6.5 \pm 1.7
P:S ratio	1.58 \pm 0.61	0.31 \pm 0.01		1.43 \pm 0.45
Lignin (% of organic matter)	69.2 \pm 9.3	29.3 \pm 1.1	18.6 \pm 3.4	15.2 \pm 0.5
Cellulose (% of organic matter)	21.3 \pm 6.3	24.0 \pm 5.7	26.3 \pm 5.0	32.6 \pm 0.6



Fig. 1

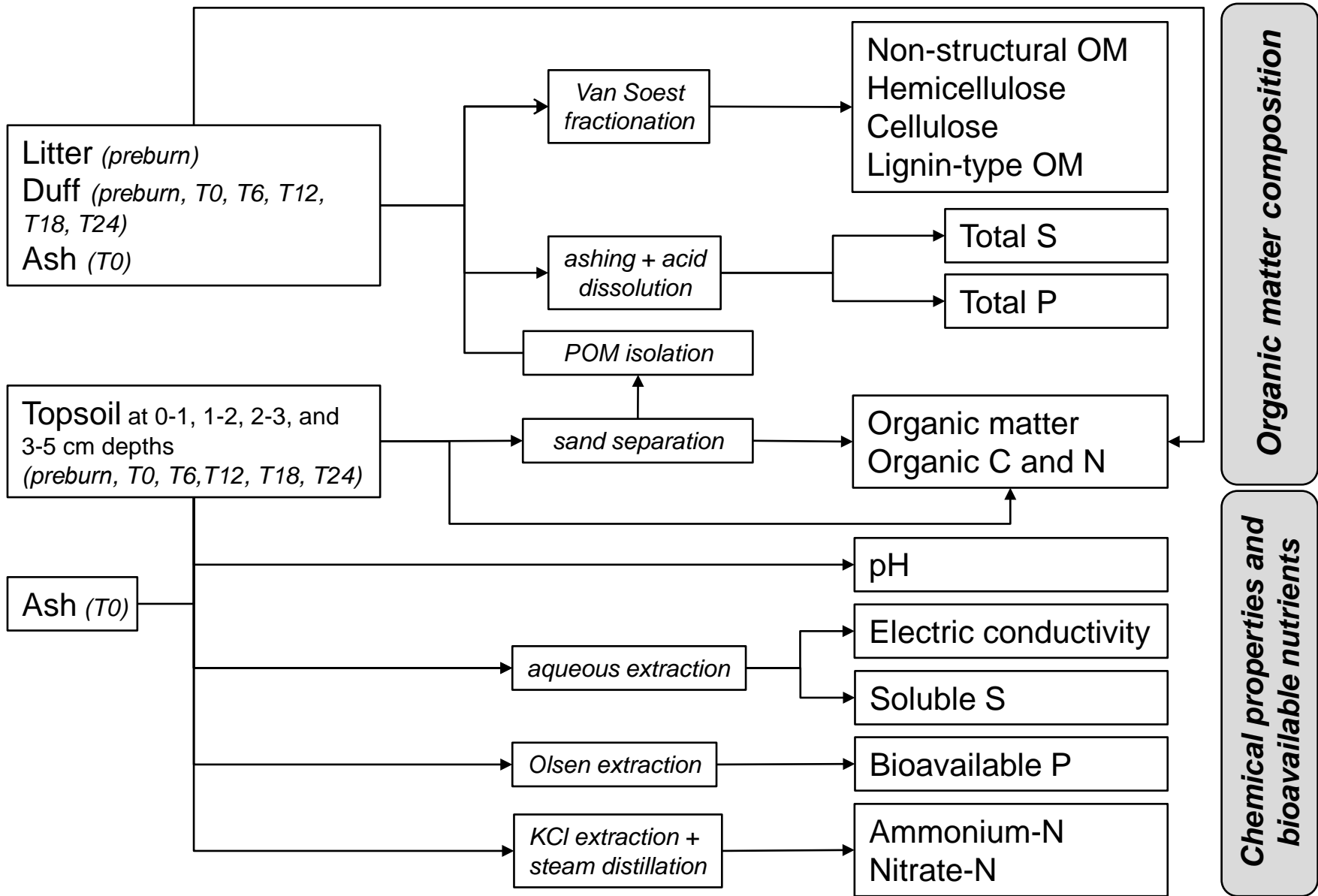


Fig. 2

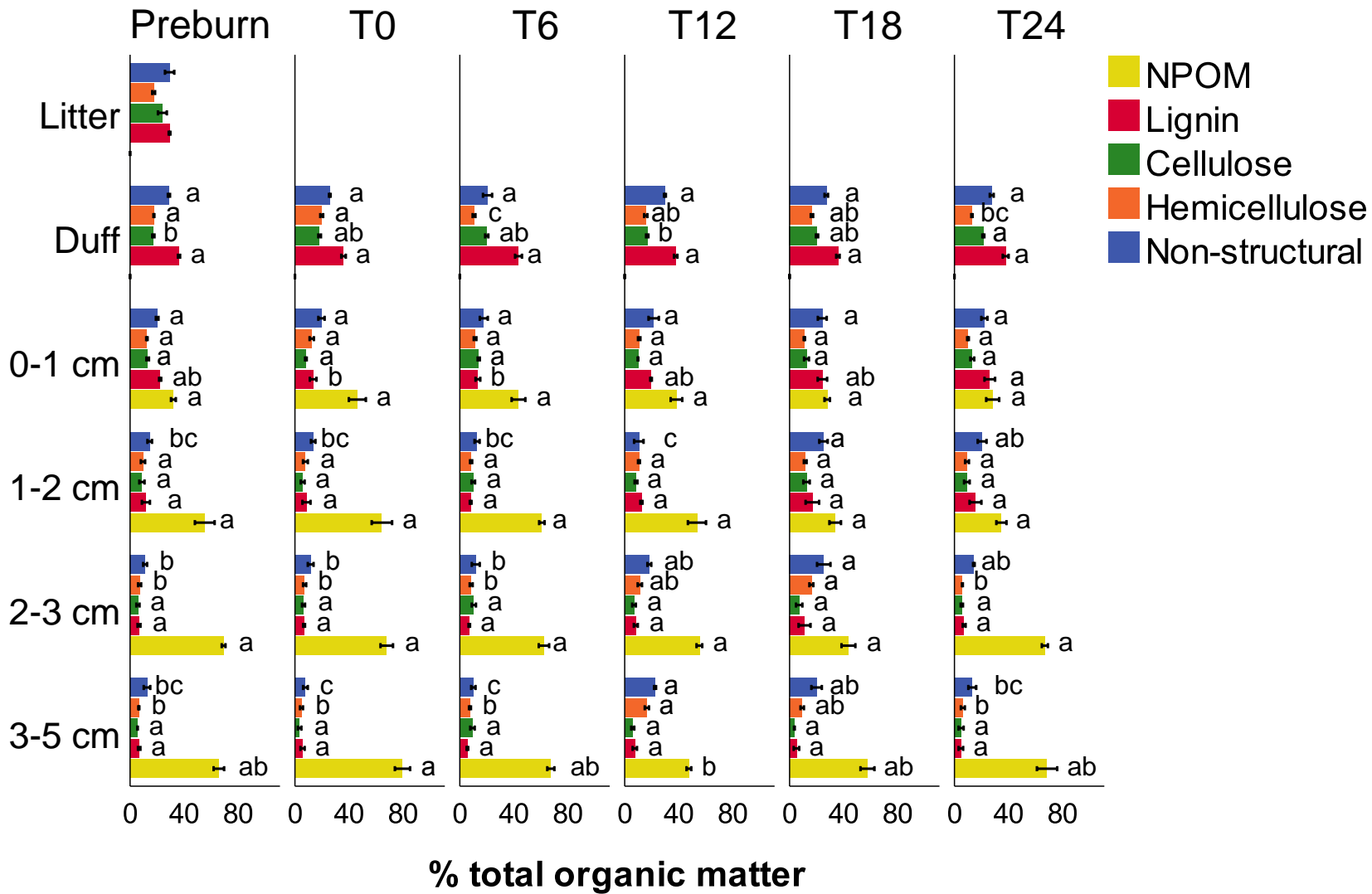


Fig. 3

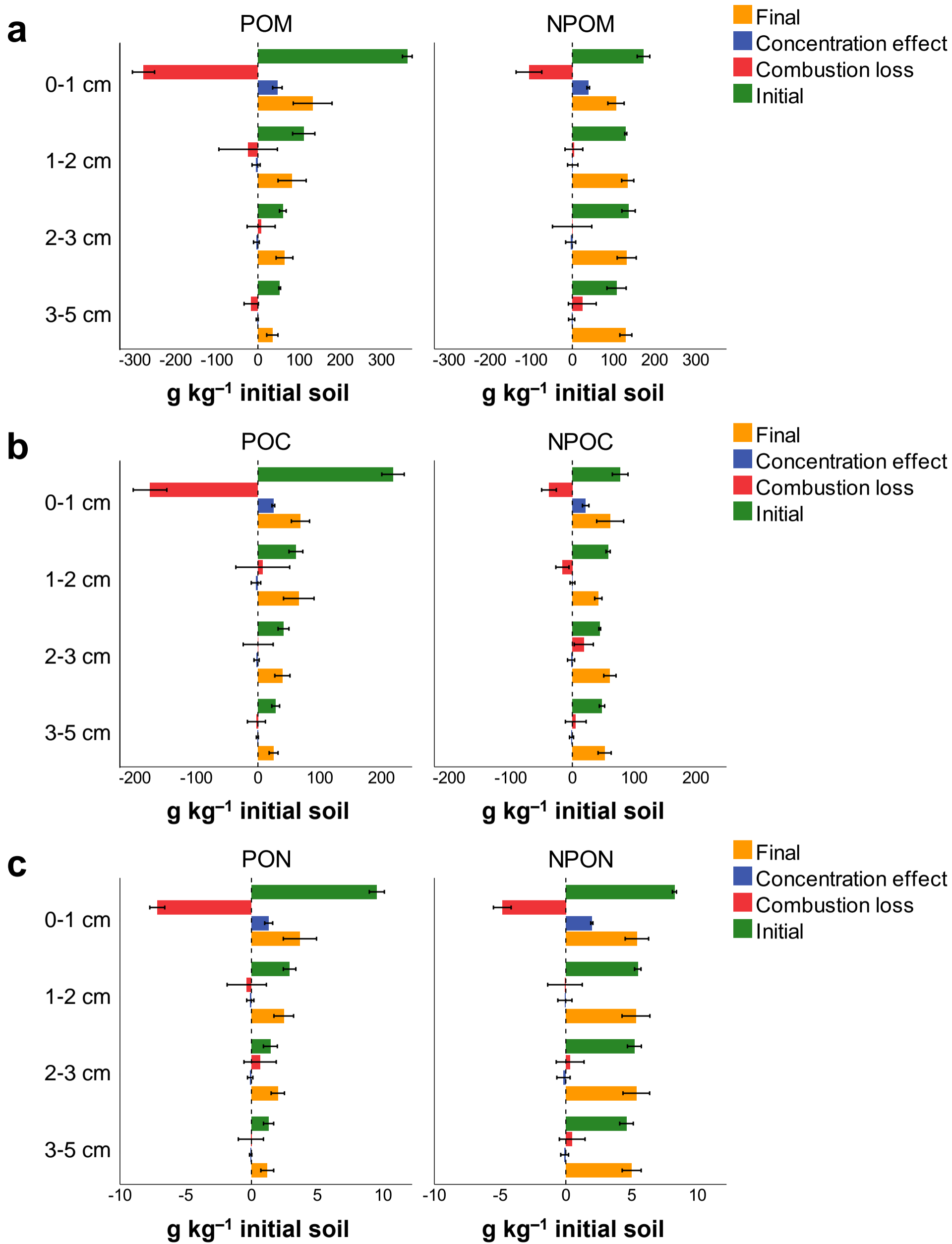


Fig. 4

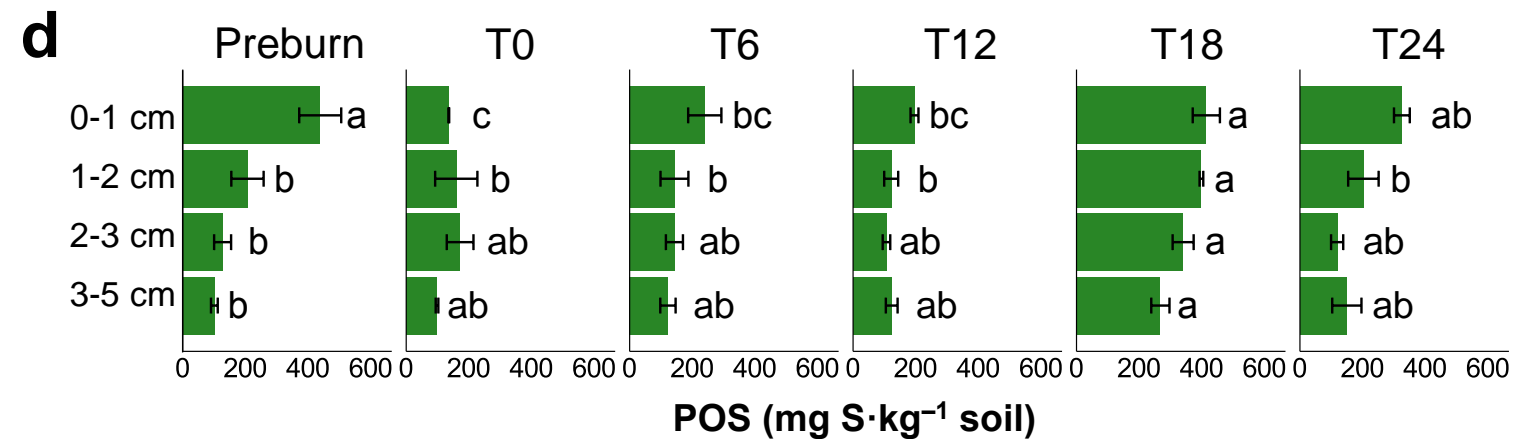
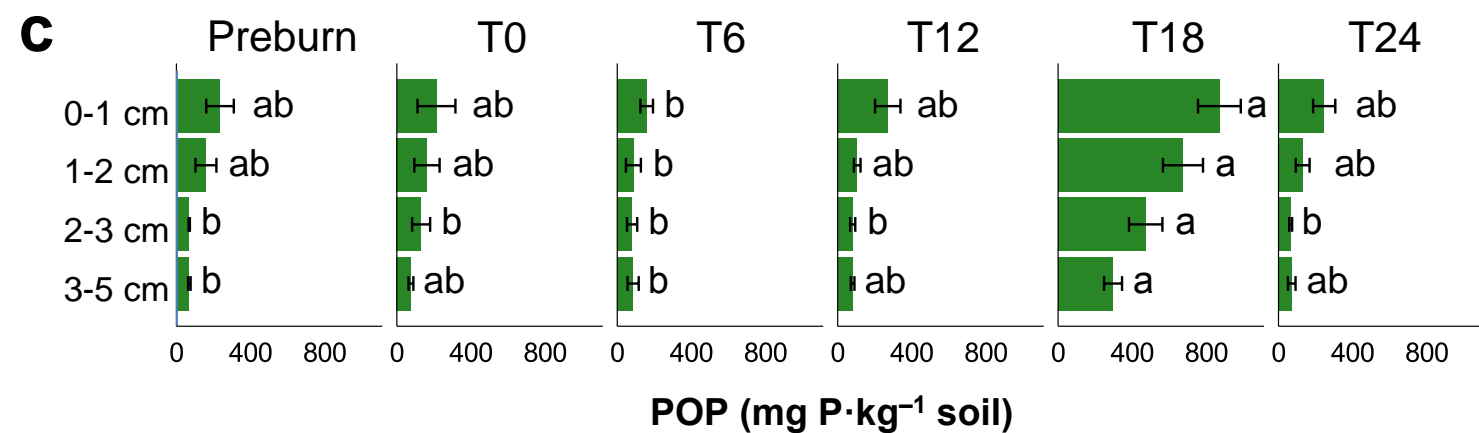
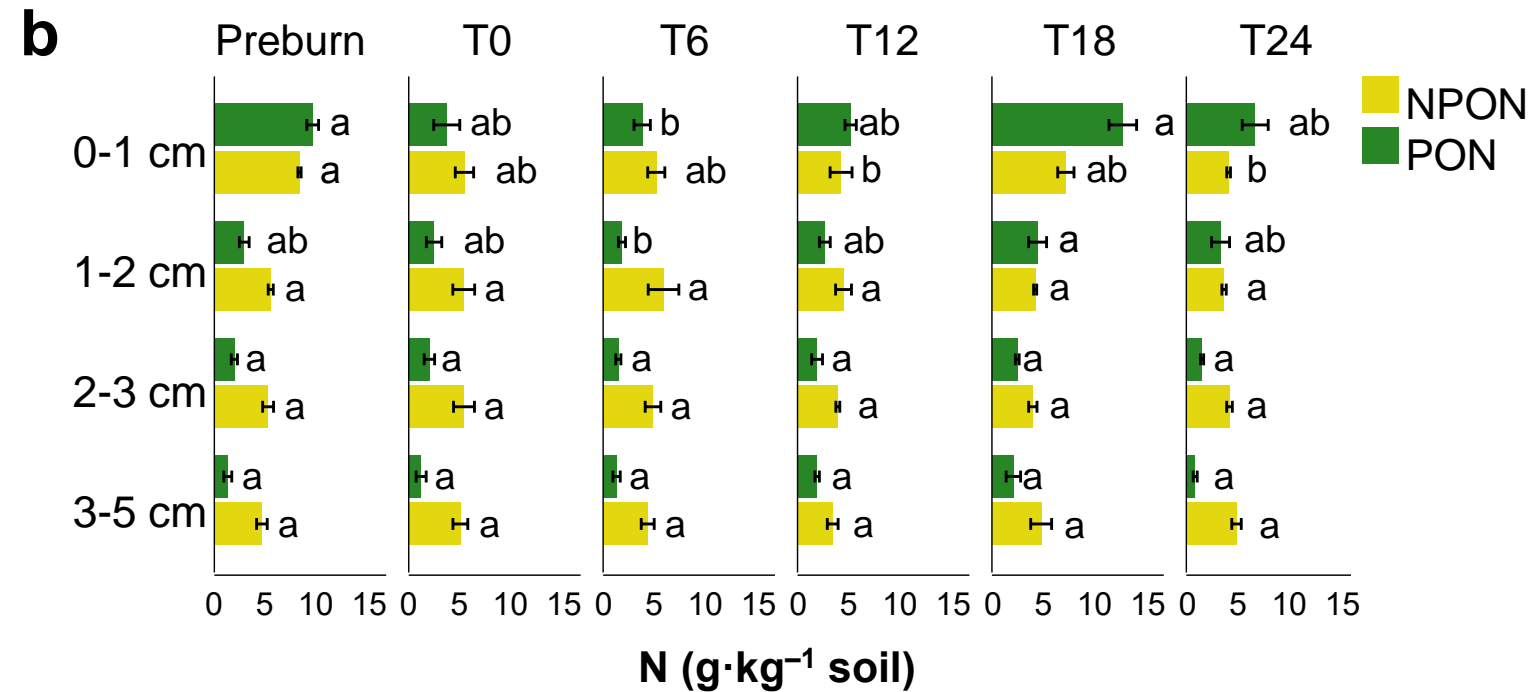
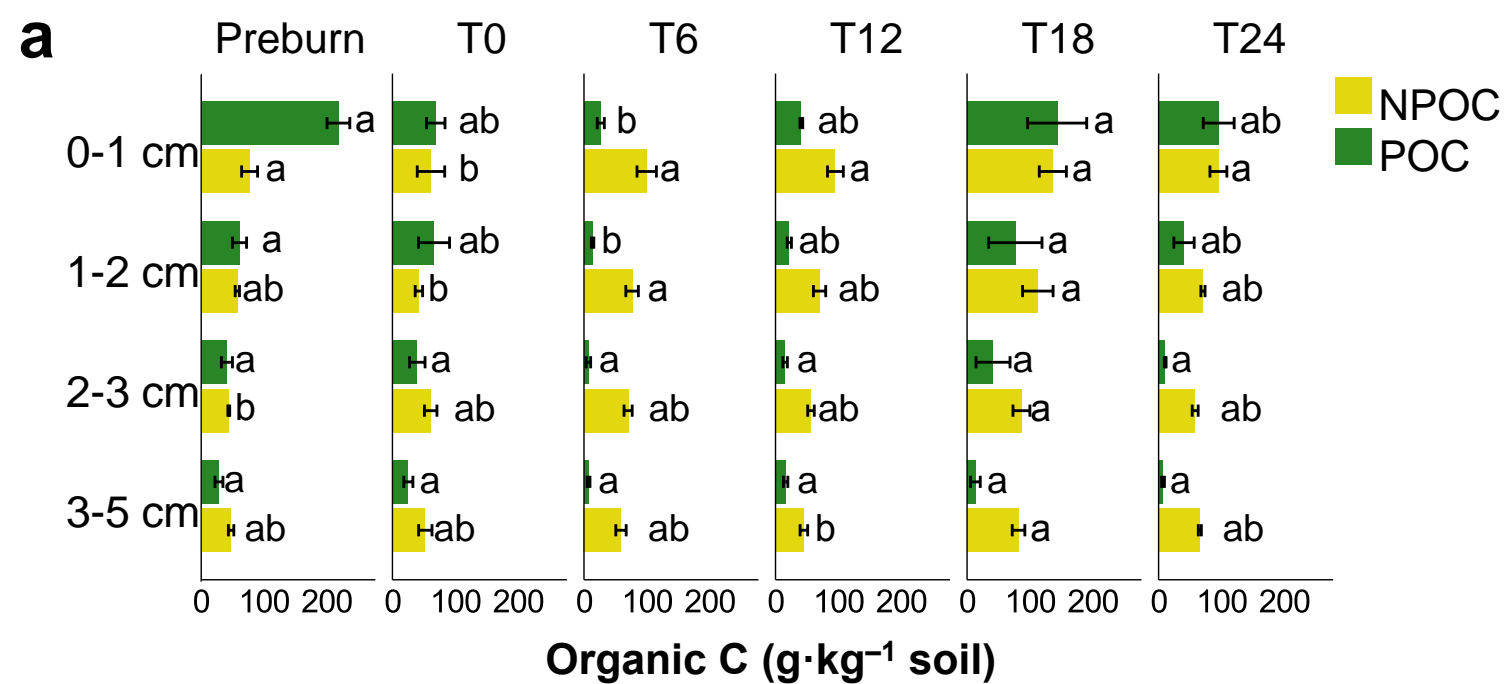


Fig. 5

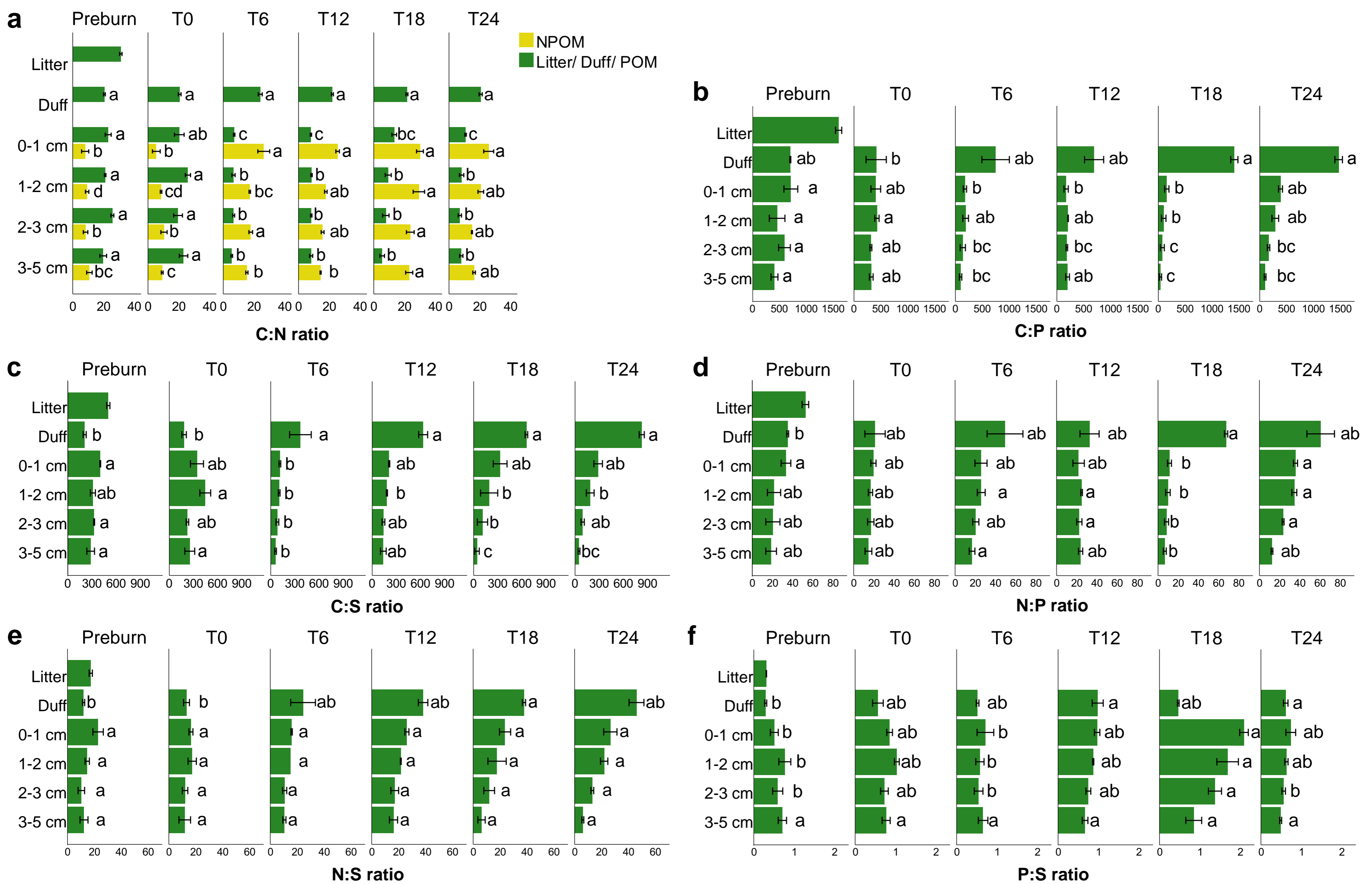


Fig. 6

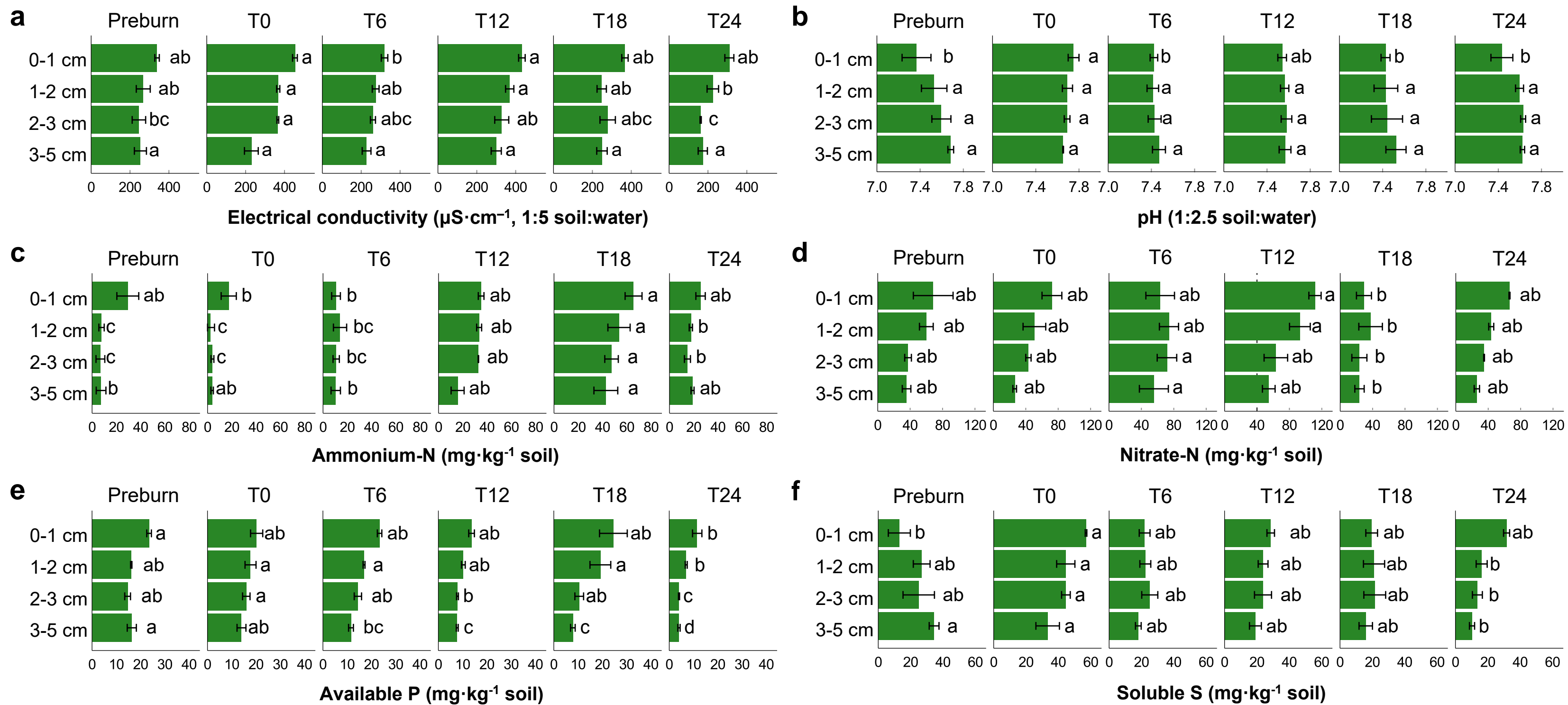


Fig. 7

HIGHLIGHTS

- Prescribed fire caused considerable C losses at the 0–1 cm depth, mostly from POM
- Uneven combustion resulted in few qualitative changes in the remaining organic matter
- Bioavailable N showed transient increases, while S and P gradually declined after fire
- Steadier nutrient inputs were derived from the breakdown of particulate charcoal