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> Debenzylative cycloetherification as synthetic tool in the diastereoselective synthesis of 3,6-disubstituted hexahydro-2H-furo[3,2-b]pyrroles, PDE1 enzyme inhibitors with antiproliferative effect on melanoma cells

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Debenzylative cycloetherification as synthetic tool in the diastereoselective synthesis of 3,6-disubstituted hexahydro-2H-furo[3,2-b]pyrroles, PDE1 enzyme inhibitors with antiproliferative effect on melanoma cells.

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#### Abstract

  - Active as inhibitor of PDE1 isoforms

PDE1A $\mathrm{IC}_{50}=4664 \mathrm{nM}$ PDE1B IC $_{50}=2820 \mathrm{nM}$ PDE1C $\mathrm{IC}_{50}=573 \mathrm{nM}$ - With antiproliferative activity on A375 melanoma cells $\mathrm{IC}_{50}=4+/-1 \mu \mathrm{M}$

Two series of novel chiral hexahydro-2H-furo[3,2-b]pyrroles, 4-(7,8-dimethoxyquinazolin-4-yl) series $A$ and 4-(6,7-dimethoxyquinazolin-4-yl) series $B$, were synthesized in enantiomerically pure form and evaluated for their inhibitory effects upon phosphodiesterase 1 (PDE1) and phosphodiesterase 4 (PDE4) as well as for their inhibitory activity on cell proliferation in A375 melanoma and 3T3fibroblast cells in vitro. Key steps of synthesis were i) diastereoselective nucleophilic addition of vinylmagnesium bromide to N -allylimine derived from conveniently protected D-glyceraldehyde, ii) ring closing metathesis, iii) debenzylative cycloetherification and iv) aromatic nucleophilic substitution. Some of the obtained compounds were proven to be active as inhibitors of PDE1 isoforms, with IC50 values in the high nanomolar/low micromolar concentration range, and showed antiproliferative activity on A375 melanoma cells.


## INTRODUCTION

3',5'-Cyclic adenosine monophosphate (cAMP) and 3',5'-cyclic guanosine monophosphate (cGMP) act as second messengers in hormone-regulated
processes and play an important role in signal transduction, synaptic transmission and other biological processes. ${ }^{1}$ Phosphodiesterases (PDEs) are the only enzymes that catalyse the hydrolysis of cyclic nucleotides. They are grouped into 11 broad families, PDE1-PDE11, which in turn are further divided into isoforms on the basis of encoding gene (e.g., PDE4A-D) and splicing isoforms (e.g., PDE4D1 - PDE4D9); so that more than 100 PDE isoforms can be distinguished. Some are cAMP specific (PDE4, 7, and 8), other specifically metabolize cGMP (PDE5, 6, and 9), and some possess dual specificity (PDE1, 2, 3, 10, and 11). ${ }^{2}$ Cyclic nucleotides play an important role in a variety of cellular mechanisms and, therefore, regulation of PDEs hydrolytic activity is a very important matter as its alteration can affect multiple cellular processes. That is why there has been an increasing interest in the development of phosphodiesterase inhibitors as tools for specific manipulation of cyclic nucleotide signalling for therapeutic use. ${ }^{3}$ In this context, the development of PDE inhibitors, which are selective for PDE families, isoforms and splicing isoforms, could be lead to novel specific therapeutic strategies for various pathologies. Several studies have revealed that the levels of PDE activity in a variety of tumours is altered, which affects the ratio of cGMP to cAMP and, thereby, their regulatory roles. The observed effects of PDE inhibition in in vitro and in vivo studies in tumour models suggest a potential role for PDE inhibitors as anticancer drugs. ${ }^{4}$ The calcium-dependent PDE1 family was first isolated in 1970 from rat brain ${ }^{5}$ and bovine brain ${ }^{6}$ and consists of several isoforms that are mainly expressed in brain, cardiac tissues, and smooth muscles. ${ }^{3 e}$ Their inhibition may provide new therapeutic strategies for various diseases. ${ }^{4 c, 7}$ Moreover PDE1 has been characterized in
melanoma cell lines and it has been shown that PDE1A inhibition exerts antiproliferative effects. ${ }^{8}$

It has been reported that compounds with an hexahydro-2H-furo[3,2-b]pyrrole core (Figure 1) are PDE1 inhibitors that are of potential utility for the treatment of neurodegenerative and psychiatric disorders. ${ }^{9}$ When $R^{1}=R^{2}=R^{3}=R^{4}=H$ compound with $3 \mathrm{aS}, 6 \mathrm{aS}$ configuration showed a higher ability to inhibit PDE1 isoforms than compound with $3 \mathrm{aR}, 6 \mathrm{aR}$ configuration $\left(\mathrm{IC}_{50}(\mathrm{nM})\right.$ ranging from 120 to 160 versus $\mathrm{IC}_{50}(\mathrm{nM})$ ranging from 2800 to 3700 , depending on the subtype). All these findings prompted us to develop a stereoselective synthesis of novel compounds with a (3aS,6aS)-hexahydro-2H-furo[3,2-b]pyrrole core and to evaluate their activity as PDE1 inhibitors and as anticancer agents.

Figure 1. General structure of PDE1 enzyme inhibitors with the hexahydro-2H-furo[3,2-b]pyrrole core. ${ }^{9}$


3aS,6aS


$3 \mathrm{a} R, 6 \mathrm{a} R$

## RESULTS AND DISCUSSION

Among the different synthetic strategies to build the tetrahydrofurane ring ${ }^{10}$ debenzylative cycloetherification ${ }^{10 \mathrm{c}}$ is an underexploited approach for the synthesis of tetrahydrofuran-containing molecules. In this context, the activation of alkenes in
$\delta$ position with respect to the benzyl ether by halogenation (usually iodination and occasionally bromination) or epoxidation leads to tetrahydrofurans through debenzylative cycloetherification (Scheme 1).


Scheme 1. Debenzylative cycloetherification of $\delta$ benzyloxyalkenes

During our the development of versatile synthetic methodologies for the synthesis of chiral nitrogen-containing heterocycles using chiral imines that are readily available from renewable sources as starting materials, we have found ${ }^{11}$ that ring closing metathesis of diallylic amines obtained from 2,3-di-O-benzylglyceraldehyde N benzylimines provides 2,5-dihydropyrroles, in which the alkene moiety is in the $\delta$ position with respect to the primary benzyloxy group. This has provided the basis for the development of a new synthetic procedure to prepare key intermediates in the synthesis of new potential PDE1 inhibitors with the hexahydro-2H-furo[3,2-b]pyrrole core.

Synthesis of 2,5-dihydropyrrole 5 was performed as outlined in scheme 2. To gain access to diallylamine 3, we added vinylmagnesium bromide to $N$-allylimine 2 obtained in situ by oxidative cleavage of 1,2,5,6-tetra-O-benzyl-D-mannitol $\mathbf{1}$ with sodium periodate and performed a subsequent reaction with allylamine. When the imine 2 reacted with vinylmagnesium bromide in ether at $-20^{\circ} \mathrm{C}$, the desired
diallylamine $\mathbf{3}$ with syn configuration was obtained with total diastereoselectivity after 12 h reaction. ${ }^{12}$ This stereochemical outcome is in accordance with previous results for the addition of organometallic reagents to $N$-benzyl imines derived from glyceraldehyde with benzyl ether as hydroxyl protecting group. ${ }^{11-13}$ As performance of ring closing methathesis (RCM) using secondary amines as substrates is generally improved by $N$-protection, ${ }^{14}$ diallylamine 3 was converted into its corresponding $N$-tert-butylcarbamate. Addition of excess of di-tert-butyl dicarbonate to a methanol/triethylamine (9:1) solution of 3 at $50^{\circ} \mathrm{C}$ led to N -Boc diallylamine 4 in $88 \%$ isolated yield. Then 2,5-dihydropyrrole 5 was cleanly obtained in $94 \%$ yield from a solution of compound 4 in dichloromethane at room temperature using first generation Grubbs Catalyst ( $10 \% \mathrm{~mol}$ ) to promote ring closing methathesis.


Scheme 2. Synthesis of 2,5-dihydropyrrole 5.

With 2,5-dihydropyrrole 5 in hand and taking into account previous results on biological activity ${ }^{9}$, we continued with the synthesis of two series of novel hexahydro$2 H$-furo[3,2-b]pyrrole derivatives, $N$-7,8-dimethoxyquinazolin-4-yl (series A) and $N$ -6,7-dimethoxyquinazolin-4-yl (series B), of $3 \mathrm{aS}, 6 \mathrm{a}$ R configuration with different substituents at C3 and C6 position (Figure 2). The construction of the required
hexahydro-2H-furo[3,2-b]pyrrole core was performed by debenzylative cycloetherification procedures using different $\mathrm{C}=\mathrm{C}$ activation modes depending on the nature of the heteroatom to be introduced at C 6 position.

Figure 2. General structure of hexahydro-2H-furo[3,2-b]pyrrole derivatives of series $A$ and $B$


Series A


Series B

First we performed dioxirane epoxidation of 5 using 3-methyl-3(trifluoromethyl)dimethyldioxirane, ${ }^{15}$ generated in situ by oxidation of 1,1,1trifluoroacetone with oxone ${ }^{\circledR}$. The reaction in acetonitrile/water (2:1) at $0^{\circ} \mathrm{C}$ provided the desired epoxide 6 derived from the addition of oxygen to the double bond on the side opposite to the substituent at C2 with total diastereoselectivity in $87 \%$ isolated yield. We also observed the formation of a small amount of compound $\mathbf{7}$ derived from debenzylative cycloetherification. Upon treatment with a catalytic amount of trifluoroacetic acid in methanol/water (3:1) at $40^{\circ} \mathrm{C}$, compound 6 evolved to the formation of 7, which was isolated in $75 \%$ yield. Hydrolysis of 7 with hydrogen chloride in ethyl acetate at room temperature yielded 3-benzyloxy-6-hydroxy hexahydro-2H-furo[3,2-b]pyrrole 8 with a $93 \%$ isolated yield. (Scheme 3)



Scheme 3. Synthesis of 3-benzyloxy-6-hydroxy hexahydro-2H-furo[3,2-b]pyrroles.

From compound 7 we obtained other hexahydro- $2 H$-furo[3,2-b]pyrroles with oxygenated substituents on C-3 and C-6 according to Scheme 4 .


Scheme 4. Synthesis of 3,6-dihydroxy hexahydro-2H-furo[3,2-b]pyrrole derivatives.

Chlorine is present in a significant number of bioactive compounds and nowadays plays a prominent role in drug design. ${ }^{16}$ This prompted us to explore activation of the $\mathrm{C}=\mathrm{C}$ bond by chlorination. In this way, a chlorine atom is installed at C 6 . This position
in the bicyclic system is essentially unreactive towards nucleophilic substitution, which prevents any behaviour of the compound as alkylating agent.

Reaction of compound 5 with $N$-chlorosuccinimide in acetonitrile at room temperature led to the desired 6-chloro hexahydro-2H-furo[3,2-b]pyrrole 13 but conversion was incomplete. When the reaction was performed in dimethylformamide at $35^{\circ} \mathrm{C}$ and in the presence of trifluoroacetic acid as an additive, we observed total conversion of the starting material and compound 13 was obtained in $80 \%$ yield. As far as we know, this is the first report in the literature on the synthesis of a 2chloromethyl tetrahydrofuran by debenzylative cycloetherification. Next hexahydro$2 H$-furo[3,2-b]pyrrole 14 was obtained from compound 13 according to Scheme 5.


Scheme 5. Synthesis of 3-hydroxy-6-choro hexahydro-2H-furo[3,2-b]pyrrole derivatives.

N-Boc hexahydro-2H-furo[3,2-b]pyrroles 7, 9, 12, 13 and 14 were hydrolysed and coupled with 4-chloro-7,8-dimethoxyquinazoline (A) and 4-chloro-6,7dimethoxyquinazoline $(B)$ to prepare the corresponding compounds of series $A$ and B, respectively. The nucleophilic aromatic substitution reaction was performed by heating the amine and the chloroquinazoline at the appropriate temperature in the presence of diisopropylethylamine and using dimethylformamide as solvent (Table 1). As a general trend, we observed that 4-chloro-7,8-dimethoxyquinazoline was
more reactive than 4-chloro-6,7-dimethoxyquinazoline providing higher yields working at lower temperatures.

Table 1. Synthesis of compounds of the series A and B

|  |  | $\mathrm{H}^{+}$ |  <br> $A=7,8$-dimethroxy derivative $B=6$, ,-dimethxoxy derivative | $\xrightarrow[\substack{\text { DMF, } \Delta, 12 h}]{\text { DIPEA }} \text { MeC }$ |  | $H_{A}^{n}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Substrate | X | Y | Hydrolysis conditions ${ }^{\text {a }}$ | quinazoline | T [ ${ }^{\circ} \mathrm{C}$ ] | Product | Overall Yield [\%] |
| 7 | BnO | OH | I | A | 60 | 15A | 93 |
| 7 | BnO | OH | 1 | B | 80 | 15B | 70 |
| 9 | BnO | OAc | 11 | A | 60 | 16A | 96 |
| 9 | BnO | OAc | 11 | B | 80 | 16B | 71 |
| 12 | AcO | OMe | III | A | 45 | 17A | 86 |
| 12 | AcO | OMe | 11 | B | 80 | 17B | 49 |
| 13 | BnO | Cl | III | A | 45 | 18A | 91 |
| 13 | BnO | Cl | III | B | 70 | 18B | 55 |


| 14 | MeO | Cl | III | A | 45 | 19A | 77 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 14 | MeO | Cl | III | B | 70 | 19B | 42 |

${ }^{\text {a }}$ Hydrolysis conditions: I, HCl in EtOAc, rt; II, TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt, III, HCl in EtOAc, $0^{\circ} \mathrm{C}$

The structures and stereochemistries of epoxide 6 and compounds obtained by debenzylative cycloetherification process were unambiguously established on the basis of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NMR coupling interactions and X -ray diffraction analysis. In epoxide 6, the $\mathrm{H}_{\mathrm{A}}$ resonance of both rotamers ${ }^{17}$ appears as a doublet at 4.19 and 4.36 ppm with a coupling constant of 3.6 and 4.0 Hz respectively corresponding the ${ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}$ coupling with benzylic proton $\mathrm{H}_{\mathrm{E}}$ as determined in the COSY spectrum. The absence of coupling between vicinal nuclei $\left({ }^{3} J_{\mathrm{H}-\mathrm{H}} \approx 0 \mathrm{~Hz}\right)$ across a single bond (Figure 3 ) is due to a spatial disposition with protons with a $90^{\circ}$ dihedral angle which in compound 6 is a clear indicative of the trans disposition between $\mathrm{H}_{\mathrm{A}}$ and $\mathrm{H}_{\mathrm{B}}$.

Figure 3. DQF-COSY of compound 6


For compound 14 (obtained from 13), $H_{A}$ and $H_{B}$ resonances of both rotamers appear as doublets due to its mutual ${ }^{3} J_{\mathrm{H}-\mathrm{H}}$ coupling as determined in the COSY spectra. The absence of vicinal coupling in signals corresponding to $H_{B}$ and $H_{C}$ nuclei in compounds 14 (Figure 4) is again a clear indicative of the trans disposition between both protons with a $90^{\circ}$ dihedral angle.

Figure 4. DQF-COSY of compound 14


The structure of compound 7 was unambiguously determined by X-ray crystallography.

We evaluated compounds 15A-19A and 15B-19B for their ability to inhibit PDE1 and PDE4. Furthermore, we tested most compounds of series $A$ and series $B$ for their ability to inhibit cell proliferation in A375 melanoma cell lines and 3 T3 fibroblast cells. Inhibition of PDE1 and PDE 4 enzymatic activities was initially evaluated as a single point at a concentration of $10 \mu \mathrm{M}$ using standard in vitro enzymatic assays. ${ }^{18}$ Compounds with high inhibitory activity were selected to determine their $\mathrm{IC}_{50}$ values. As can be seen from the data in Table 2, most of the assayed hexahydro-2H-furo[3,2-b]pirroles were more potent against PDE1 isoenzyme than against PDE4 isoenzyme with 18A being the most active compound. The potency of 18 A in inhibiting PDE1C isoform is higher than in inhibiting PDE1A and PDE1B isoforms. In addition compounds 17B and 19B preferentially inhibit the PDE1C isoform of PDE1.

Table 2. Inhibitory activity of compounds of the series $A$ and $B$

| Compound | PDE1A $\text { (\% inh @ } 10$ <br> $\mu \mathrm{M}$ ) | PDE1A <br> ( $\mathrm{IC}_{50}$, <br> nM) | $\begin{gathered} \text { PDE1B } \\ (\% \text { inh @ } 10 \\ \mu \mathrm{M}) \end{gathered}$ | $\begin{gathered} \text { PDE1B } \\ \left(\mathrm{IC}_{50},\right. \\ \mathrm{nM}) \end{gathered}$ | $\begin{gathered} \text { PDE1C } \\ (\% \text { inh @ } 10 \\ \mu \mathrm{M}) \end{gathered}$ | $\begin{gathered} \text { PDE1C } \\ \left(\mathrm{IC}_{50},\right. \\ \mathrm{nM}) \end{gathered}$ | $\begin{gathered} \text { PDE4 } \\ (\% \text { inh @ } 10 \\ \mu \mathrm{M}) \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 15A | 32 |  | 22 |  | 68 |  | 0 |
| 15B | 22 |  | 11 |  | 19 |  | 14 |
| 16A | 22 |  | 19 |  | 36 |  | 11 |
| 16B | 8 |  | -2 |  | 12 |  | 9 |
| 17A | 14 |  | 16 |  | 12 |  | 0 |
| 17B | 48 |  | 39 |  | 87 | 2167 | 0 |
| 18A | 70 | 4664 | 72 | 2820 | 97 | 573 | 27 |
| 18B | 28 |  | 22 |  | 56 |  | 24 |
| 19A | 41 |  | 31 |  | 56 |  | 13 |
| 19B | 30 |  | 40 |  | 83 | 2367 | 3 |


#### Abstract

With the exception of compound $17 \mathrm{~A}^{19}$, we tested the compounds for antiproliferative activity in the 3 T 3 fibroblast cell line and the A375 melanoma cell line, using the Janus-Green in vitro assay. ${ }^{20}$ Cell proliferation and survival depicted in Figure 5 A and 5 B show that none of the compounds tested at $10 \mu \mathrm{M}$ were cytotoxic per se in both lines. However, in 3T3 fibroblasts, 18A virtually abolished cell proliferation, while 18B had a lower anti-proliferative activity (Figure 5A, left panel). 15A, 15B, 16A, 16B, 19A and 19B had no statistically significant activity (Figure 5A, both panels). In A375 melanoma cells, we saw the same activity profile, but potencies were generally higher and 18A virtually abolished cell proliferation, while 18B had a lower anti-proliferative activity. Like in 3T3 fibroblasts, 19A and 19B had no statistically significant activity. The most active compound 18A was assayed over a concentration range of $100 \mathrm{nM}-50 \mu \mathrm{M}$ and we found significant anti-proliferative activity in A375 melanoma cells at $1 \mu \mathrm{M}$, again, a strong anti-proliferative activity at $10 \mu \mathrm{M}$, and a profound anti-proliferative activity and overall lower survival at $50 \mu \mathrm{M}$, the highest concentration tested. We calculated an $\mathrm{IC}_{50}$ of $4+/-1 \mu \mathrm{M}$ (Figure 5C).






Figure 5. In vitro anti-proliferative activity of hexahydrofuropyrroles: A) Time course of inhibition of proliferation in 3T3 fibroblasts. B) Time course of inhibition of cell proliferation in A375 melamoma cells. C) Concentration-dependence of 18A and $\mathrm{IC}_{50}$. Note that at 50 uM absorbance value below initial values (day 0 ) indicate a reduced survival. Data are mean +/- SEM; $n=4-30$ measurements from 1-5 independent experiments. *P < 0.05 vs. control ( 0.1 \% DMSO as vehicle), Student's T test.


#### Abstract

In summary, we have shown debenzylative cycloetherification is a powerful synthetic tool in the highly diastereoselective synthesis of 3,6-disubstituted hexahydro-2H-furo[3,2-b]pyrroles with four stereocenters. This structural motif is present in compounds with inhibitory activity against PDE1 enzyme with potential use as medicaments for the treatment of neurodegenerative and psychiatric disorders. ${ }^{9}$ Optimization of the reactions conditions allowed the development of a new and efficient activation of the alkene in $\delta$ position with respect to the benzyl ether that allows the introduction of a chlorine atom in 3 position of the bicyclic core. Some of the obtained compounds were proven to act as inhibitors of PDE1 isoforms being more active against PDE1C and revealed antiproliferative activity on A375 melanoma cells.


## EXPERIMENTAL SECTION

General Details. Unless otherwise specified, all reagents were obtained from commercial suppliers and were used without purification. For anhydrous conditions, reactions were carried out under Ar in solvents dried using a Solvent Purification System (SPS). Whenever possible, the reactions were monitored by TLC. TLC analysis was performed on precoated silica gel polyester plates with an $\mathrm{F}_{254}$ indicator and products were visualized using UV light (254 nm) and ninhydrin, anisaldehyde, potassium permanganate or ethanolic phosphomolybdic acid solutions followed by heating. Column chromatography was performed on silica gel ( $60,40-63 \mu \mathrm{~m}$ ) with air pressure.

Melting points were determined in open capillary tubes and are not corrected. FT-IR spectra of oils were recorded as thin films on NaCl plates and FT-IR spectra of solids were recorded on pressed KBr pellets, $v_{\max }$ values expressed in $\mathrm{cm}^{-1}$ are given for the main absorption bands. Optical rotations were measured on a digital polarimeter at $\lambda 589 \mathrm{~nm}$ and $25^{\circ} \mathrm{C}$ in cells with 1 or 10 cm path length, $[\alpha]_{D}$ values are given in $10^{-1}$ deg $\cdot \mathrm{cm}^{2} \cdot \mathrm{~g}^{-1}$ and concentrations are given in $\mathrm{g} / 100 \mathrm{~mL} .{ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were acquired in deuterated solvent at room temperature unless otherwise stated at 400 and 100 MHz , respectively using a $5-\mathrm{mm}$ probe. All chemical shifts ( $\delta$ ) are reported in parts per million (ppm) with the solvent resonance as the internal standard, ${ }^{20}$ and coupling constants $(\mathrm{J})$ in hertz $(\mathrm{Hz})$. High-resolution mass spectra were recorded from methanolic solutions on a MICROTOF-Q (quadrupole time-offlight) micro instrument using the positive electrospray ionization mode (ESI+). The

X-ray diffraction data were collected at room temperature on a four-circle diffractometer, using graphite-monochromated Mo-K $\alpha$ radiation $(\lambda=0.71073 \AA)$.

## (3R,4S)-N-Allyl-4,5-bis(benzyloxy)pent-1-en-3-amine (3)

Solid $\mathrm{NaIO}_{4}(3.15 \mathrm{~g}, 14,74 \mathrm{mmol})$ and water ( 3.7 mL ) were added successively to a stirred solution of 1,2,5,6-tetra-O-benzyl-D-mannitol ( $4.0 \mathrm{~g}, 7.37 \mathrm{mmol}$ ) in THF ( 35 ml ) and the resulting mixture was vigorously stirred at room temperature for $1-2 \mathrm{~h}$. The reaction mixture was filtered and the filtrate was concentrated in vacuo. The obtained crude was immediately dissolved in anhydrous diethyl ether ( 25 mL ) and anhydrous $\mathrm{MgSO}_{4}(3.55 \mathrm{~g}, 29.5 \mathrm{mmol}$ ) and allylamine ( $842 \mathrm{mg}, 14.74 \mathrm{mmol}$ ) were successively added to the resulting solution. The reaction mixture was stirred at room temperature for 2 h , filtered and concentrated in vacuo. The resulting crude imine was dissolved in anhydrous diethyl ether ( 65 mL ), cooled to $-20^{\circ} \mathrm{C}$, and then slowly added under an argon atmosphere to a stirred 1.0 M solution of vinylmagnesium bromide in THF ( $29.5 \mathrm{~mL}, 29.5 \mathrm{mmol}$ ) at $-20^{\circ} \mathrm{C}$. Stirring was continued for 12 h at the same temperature. The reaction mixture was quenched at $0^{\circ} \mathrm{C}$ by slow addition of water ( 40 mL ) and filtered through Celite ${ }^{\circledR}$ pad. The organic layer was separated and the aqueous layer was extracted with ethyl ether ( $2 \times 80$ mL ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: $2 / 3$ containing $\mathrm{Et}_{3} \mathrm{~N}(1 \% \mathrm{v} / \mathrm{v})$ yielded diastereomerically pure $\mathbf{3}$ $\left(2.30 \mathrm{~g}, 68 \%\right.$ yield) as a yellowish oil. $[\alpha]_{\mathrm{D}}{ }^{25}=-11.0\left(c=1.02\right.$ in $\left.\mathrm{CHCl}_{3}\right) ;$ IR (neat) 3332, $1642 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.95$ (bs, 1H), 3.08 (dddd, $\mathrm{J}=14.2$,
$6,6,1.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.28-3.35(\mathrm{~m}, 2 \mathrm{H}), 3.60-3.65(\mathrm{~m}, 2 \mathrm{H}), 3.74-3.80(\mathrm{~m}, 1 \mathrm{H}), 4.55$ $(\mathrm{d}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.5(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~d}, \mathrm{~J}=11.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.84(\mathrm{~d}, \mathrm{~J}=$ $11.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10$ (dddd, $J=10.2,1.4,1.4,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.17$ (ddd, $J=17.2,1.7$, $1.7,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.20-5.26(\mathrm{~m}, 1 \mathrm{H}), 5.62-5.72(\mathrm{~m}, 1 \mathrm{H}), 5.90(\mathrm{dddd}, \mathrm{J}=17.2,10.3$, $6.6,5,2 \mathrm{~Hz}, 1 \mathrm{H}), 7.29-7.42(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 49.6$, $62.5,70.6,73.1,73.4,81.0,115.7,118.6,127.7,127.7,127.7,128.0,128.4,128.5$, 137.0, 137.7, 138.4, 138.6; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}_{2} 338.2115$, found 338.2120 .
tert-Butyl allyl[(3R,4S)-4,5-bis(benzyloxy)pent-1-en-3-yl]carbamate (4)
Di-tert-butyl dicarbonate $(4.98 \mathrm{~g}, 22.82 \mathrm{mmol})$ was added to a solution of compound $3(3.35 \mathrm{~g}, 9.93 \mathrm{mmol})$ in methanol/triethylamine $9 / 1(30 \mathrm{~mL})$ at room temperature and the reaction mixture was stirred at $50^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: diethyl ether/hexane: $1 / 7,2^{\text {nd }}$ eluent: diethyl ether/hexane: $1 / 5$ ) yielded diastereomerically pure $4(3.82 \mathrm{~g}$, $88 \%$ yield) as a colourless oil. $[\alpha]_{\mathrm{D}}{ }^{25}=4.3\left(c=1.02\right.$ in $\mathrm{CHCl}_{3}$ ); IR (neat) 1691,1642 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, 333 \mathrm{~K}\right) \delta 1.49(\mathrm{~s}, 9 \mathrm{H}), 3.59(\mathrm{dd}, \mathrm{J}=10.6,5.3 \mathrm{~Hz}$, $1 \mathrm{H}), 3.71$ (dd, $J=10.6,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.84$ (dddd, $J=16.0,6.2,1.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.95$4.14(\mathrm{~m}, 2 \mathrm{H}), 4.39(\mathrm{dd}, \mathrm{J}=7.6,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.53(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.58(\mathrm{~d}, \mathrm{~J}=$ $12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.61(\mathrm{~d}, \mathrm{~J}=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~d}, \mathrm{~J}=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.06$ (dddd, $\mathrm{J}=$ $10.2,1.5,1.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}) 5.12$ (dddd, $J=17.2,1.6,1.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.15-5.28(\mathrm{~m}$, 2 H ), 5.86 (dddd, $J=17.1,10.2,5.8,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.08$ (ddd, $J=17.7,10.4,7.7 \mathrm{~Hz}$, 1H), 7.24-7.40 (m, 10H); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}, 333 \mathrm{~K}\right) \delta 28.7,50.4$, $62.0,71.4,73.2,73.7,79.6,79.8,115.8,118.2,127.5,127.6,127.8,127.8,128.3$,
128.4, 135.2, 136.1, 138.7, 139.2, 155.6; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{NNaO}_{4} 460.2458$, found 460.2454 .
tert-Butyl (R)-2-[(S)-1,2-bis(benzyloxy)ethyl]-2,5-dihydro-1H-pyrrole-1carboxylate (5)

A solution of compound $4(2.19 \mathrm{~g}, 5.0 \mathrm{mmol})$ in dichloromethane $(25 \mathrm{~mL})$ was added to a solution of Grubbs first generation catalyst ( $412 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) in dichloromethane $(25 \mathrm{~mL})$ at room temperature and the resulting solution was stirred for 2 h at the same temperature and then evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: 1/4) yielded diastereomerically pure $5(1.99 \mathrm{~g}, 97 \%$ yield $)$ as a colourless oil. $[\alpha]_{D^{25}}=121.6$ ( $c=$ 1.04 in $\mathrm{CHCl}_{3}$ ); IR (neat) $1698,1624 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 333 \mathrm{~K}\right) \delta 1.51$ (s, 9H), 3.50-3.58 (m, 2H), 3.98 (dddd, J = 15.6, 5.6, 2.1, 2.1 Hz, 1H), 4.16-4.38 (m, $2 \mathrm{H}), 4.52(\mathrm{~d}, \mathrm{~J}=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{~d}, \mathrm{~J}=12.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.74-4.88(\mathrm{~m}, 3 \mathrm{H}), 5.79-$ $5.85(\mathrm{~m}, 1 \mathrm{H}), 5.86-5.92(\mathrm{~m}, 1 \mathrm{H}), 7.25-7.44(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}-\operatorname{APT}\{1 \mathrm{H}\} \mathrm{NMR}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}, 333 \mathrm{~K}\right) \delta 28.7,54.3,65.3,71.4,73.1,73.5,79.2,79.9,126.7,127.4,127.5$, 127.6, 127.7, 128.4, 128.4, 138.8, 139.3, 154.4; HRMS (ESI+) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{31} \mathrm{NNaO}_{4} 432.2145$, found 432.2163.
tert-Butyl (1S,2S,5R)-2-[(S)-1,2-bis(benzyloxy)ethyl]-6-oxa-3-azabicyclo[3.1.0]hexane-3-carboxylate (6)
0.1 M Aqueous solution of $\mathrm{Na}_{2}$ EDTA $(504 \mu \mathrm{~L})$, 1,1,1-trifluoroacetone ( $3.05 \mathrm{~g}, 27.22$ mmol ) and water ( 16 mL ) were added successively to a stirred solution of compound $5(1.03 \mathrm{~g}, 2.52 \mathrm{mmol})$ in acetonitrile $(30 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. Then a mixture of oxone ${ }^{\circledR}(7.90$ $\mathrm{g}, 12.85 \mathrm{mmol})$ and $\mathrm{NaHCO}_{3}(1.48 \mathrm{~g}, 17.62 \mathrm{mmol})$ as a solid was added slowly in
small portions over a period of 1 h at $0^{\circ} \mathrm{C}$. After being stirred for 2 h at $0^{\circ} \mathrm{C}$, the solid material was removed by filtration and the filtrate was diluted with water ( 10 mL ) and extracted with dichloromethane ( $3 \times 40 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: $1 / 2$ yielded diastereomerically pure $6\left(933 \mathrm{mg}, 87 \%\right.$ yield) as a white solid. M.p. $=66.4-67.6^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{25}=63.5\left(c=1.0\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \operatorname{IR}(\mathrm{KBr}) 1679,1109 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 1.40$ and $1.46(\mathrm{~s}, 9 \mathrm{H}), 3.21$ and $3.24(\mathrm{dd}, \mathrm{J}=12.8,1.2 \mathrm{~Hz}$ and $J=12.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.48-3.53(\mathrm{~m}, 1 \mathrm{H}), 3.55-3.63(\mathrm{~m}, 1 \mathrm{H}), 3.63-3.67(\mathrm{~m}, 1 \mathrm{H})$, 3.69-3.73 (m, 1H), 3.78 and $3.88(\mathrm{~d}, J=12.4 \mathrm{~Hz}$ and $J=12.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.94$ and $4.07(\mathrm{ddd}, \mathrm{J}=6.4,3.6,3.6$ and $J=7.2,3.6,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.19$ and $4.36(\mathrm{~d}, \mathrm{~J}=4.0$ Hz and $J=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.51-4.61(\mathrm{~m}, 2 \mathrm{H}), 4.68$ and $4.71(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}$ and $J=$ $12.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.80 and $4.81(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}$ and $J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.28-7.40(\mathrm{~m}$, 10 H ); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 28.4$ and 28.5 , 47.9 and $48.5,55.0$ and $55.6,57.1$ and $57.4,59.0$ and $59.5,70.3$ and $70.7,72.8$ and 73.1, 73.7 and $73.7,77.6$ and $77.7,80.1$ and $80.4,127.7$ and 127.7, 127.8 and 127.9, 128.4 and 128.5, 128.5 and 128.5, 138.0 and 138.2, 138.4 and138.6, 154.9 and 155.2; HRMS (ESI+) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{31} \mathrm{NNaO}_{5} 448.2094$, found 448.2081.
tert-Butyl (3S,3aS,6R,6aR)-3-(benzyloxy)-6-hydroxyhexahydro-4H-furo[3,2-b]pyrrole-4-carboxylate (7)

Trifluoroacetic acid ( $50 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) was added to a stirred dispersion of compound $6(910 \mathrm{mg}, 2.14 \mathrm{mmol})$ in methanol$/$ water $3 / 1(18 \mathrm{~mL})$ and the mixture
was stirred at $40^{\circ} \mathrm{C}$ for 24 h and then evaporated in vacuo. The residue was diluted with water ( 15 mL ) and extracted with dichloromethane ( $3 \times 30 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: diethyl ether/hexane: $1 / 1,2^{\text {nd }}$ eluent: diethyl ether) yielded diastereomerically pure 7 ( $538 \mathrm{mg}, 75 \%$ yield) as a white solid. M.p. $=124.3-128.5^{\circ} \mathrm{C} ;[\alpha]^{25}=44.1\left(c=0.63\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}(\mathrm{KBr})$ 3340, $1680 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 1.49(\mathrm{~s}, 9 \mathrm{H})$, 3.01 (bs, 1H), 3.27-3.33(m, 1H), 3.61 and $3.72(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}$ and $\mathrm{J}=12.6 \mathrm{~Hz}$, $1 \mathrm{H}), 3.76-3.83(\mathrm{~m}, 1 \mathrm{H}), 3.92$ and $4.00(\mathrm{~d}, \mathrm{~J}=10.0 \mathrm{~Hz}$ and $J=10.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.16-$ $4.22(\mathrm{~m}, 1 \mathrm{H}), 4.16-4.22$ and $4.28-4.31(\mathrm{~m}, 1 \mathrm{H}), 4.47-4.56(\mathrm{~m}, 2 \mathrm{H}), 4.59$ and 4.67 $(d, J=11.8 \mathrm{~Hz}$ and $J=10.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.70$ and $4.82(\mathrm{~d}, J=11.5 \mathrm{~Hz}$ and $J=11.8 \mathrm{~Hz}$, 1H), 7.26-7.40 (m,5H); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 28.5$ and $28.6,53.2$ and $53.5,65.1,71.4$ and $71.5,72.5$ and $73.0,72.7$ and 73.4 , 80.2 and $80.8,82.4$ and $83.2,86.4$ and $87.0,127.5$ and 127.7, 127.9, 128.4 and 128.6, 137.8 and 138.2, 154.4 and 154.6; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NNaO}_{5} 358.1625$, found 358.1636 .
(3S,3aS,6R,6aR)-3-(Benzyloxy)hexahydro-2H-furo[3,2-b]pyrrol-6-ol (8) A solution of compound $7(167 \mathrm{mg}, 0.50 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(1 \mathrm{~mL})$ was stirred at room temperature for 2 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(8 \mathrm{~mL})$ and extracted with dichloromethane ( $3 \times 15 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo to give compound 8 ( 109 mg , $93 \%$ yield $)$ as a yellowish solid. M.p. $=112.1-115.3^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=-41.1(c=1.0$ in
$\mathrm{CHCl}_{3}$ ); IR (KBr) 3298, 3113, 1118, 1097, 1080, $1043 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 2.76(\mathrm{bs}, 2 \mathrm{H}), 2.85(\mathrm{dd}, \mathrm{J}=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{~d}, \mathrm{~J}=12.4 \mathrm{~Hz}, 1 \mathrm{H})$, 3.80 (dd, $J=10.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{dd}, \mathrm{J}=10.0,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.92-3.94(\mathrm{~m}, 1 \mathrm{H})$, 4.01 (d, J = 5.2 Hz, 1H), 4.16-4.18 (m, 1H), $4.43(\mathrm{~d}, \mathrm{~J}=5.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.55(\mathrm{~d}, \mathrm{~J}=$ $11.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.60(\mathrm{~d}, \mathrm{~J}=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.27-7.38(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}-\operatorname{APT}\{1 \mathrm{H}\} \operatorname{NMR}(100$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 53.2,66.7,71.5,72.0,76.3,85.3,88.4,127.8,127.9,128.6,137.9 ;$ HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{NO}_{3}$ 236.1281, found 236.1288.
tert-Butyl (3S,3aS,6R,6aR)-6-acetoxy-3-(benzyloxy)hexahydro-4H-furo[3,2-b]pyrrole-4-carboxylate (9)

Acetic anhydride ( $384 \mathrm{mg}, 3.76 \mathrm{mmol}$ ) was added to a stirred solution of compound 7 ( $315 \mathrm{mg}, 0.94 \mathrm{mmol}$ ), DMAP ( $57.4 \mathrm{mg}, 0.47 \mathrm{mmol}$ ) and triethylamine ( 380 mg , $3.76 \mathrm{mmol})$ in acetonitrile $(3 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 5 min first and then at room temperature for 2 h . The reaction mixture was diluted with dichloromethane $(30 \mathrm{~mL})$ and water $(15 \mathrm{~mL})$. The aqueous layer was then extracted with dichloromethane $(2 \times 20 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 15 mL ), dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: $1 / 2$ ) yielded diastereomerically pure 9 ( $350 \mathrm{mg}, 99 \%$ yield) as a yellowish solid. M.p. $=78.1-80.5^{\circ} \mathrm{C} ;[\alpha]_{D^{25}}=16.7\left(c=1.0\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \operatorname{IR}(\mathrm{KBr}) 1736,1699,1239,1114$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 1.47$ and $1.50(\mathrm{~s}, 9 \mathrm{H}), 2.03$ and $2.04(\mathrm{~s}, 3 \mathrm{H}), 3.38$ and $3.41(\mathrm{dd}, \mathrm{J}=13.0,3.9 \mathrm{~Hz}$ and $J=12.8,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.62$ and $3.78(\mathrm{~d}, \mathrm{~J}=13.0 \mathrm{~Hz}$ and $J=12.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.76-3.85(\mathrm{~m}, 1 \mathrm{H}), 3.92$ and $4.08(\mathrm{~d}$, $J=10.0 \mathrm{~Hz}$ and $J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.20$ and $4.30(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}$ and $J=4.0 \mathrm{~Hz}, 1 \mathrm{H})$,
4.40 and $4.48(\mathrm{~d}, \mathrm{~J}=4.5 \mathrm{~Hz}$ and $J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.56$ and $4.60(\mathrm{~d}, \mathrm{~J}=4.5 \mathrm{~Hz}$ and $J$ $=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.59$ and $4.65(\mathrm{~d}, \mathrm{~J}=11.8 \mathrm{~Hz}$ and $\mathrm{J}=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.67$ and $4.80(\mathrm{~d}$, $J=11.8 \mathrm{~Hz}$ and $J=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.08$ and $5.11(\mathrm{~d}, J=4.0 \mathrm{~Hz}$ and $J=3.9 \mathrm{~Hz}, 1 \mathrm{H})$, 7.23-7.37 (m, 5H); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 21.0,28.5$ and $28.6,50.9$ and $51.3,65.5$ and $65.5,71.4$ and $71.4,72.4$ and 73.0 , 74.5 and $75.2,80.4$ and $80.7,82.4$ and $83.2,84.0$ and $84.8,127.4$ and 127.8, 127.6 and 127.9, 128.4 and 128.5, 137.7 and 138.1, 153.8 and 154.2, 169.7 and 169.9; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{20} \mathrm{H}_{27} \mathrm{NNaO}_{6} 400.1731$, found 400.1745 .
tert-Butyl (3S,3aS,6R,6aR)-3-(benzyloxy)-6-methoxyhexahydro-4H-furo[3,2-b]pyrrole-4-carboxylate (10)

A $60 \%$ dispersion of sodium hydride in mineral oil ( $90 \mathrm{mg}, 2.24 \mathrm{mmol}$ ) was added in a single portion under an Ar atmosphere at $0^{\circ} \mathrm{C}$ to a solution of compound 7 (300 $\mathrm{mg}, 0.89 \mathrm{mmol}$ ) in anhydrous THF ( 12 mL ). After being stirred for 1 h at $0^{\circ} \mathrm{C}$, methyl iodide ( $765 \mathrm{mg}, 5.39 \mathrm{mmol}$ ) was added and stirring was continued under an Ar atmosphere at room temperature for additional 2 h . The reaction was quenched at 0 ${ }^{\circ} \mathrm{C}$ with water $(7.5 \mathrm{~mL})$ and saturated $\mathrm{NaHCO}_{3}(7.5 \mathrm{~mL})$ aqueous solution and then extracted with dichloromethane $(3 \times 30 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 10 mL ), dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: 3/2) yielded diastereomerically pure 10 ( $301 \mathrm{mg}, 96 \%$ yield) as a colourless oil. $[\alpha]_{D^{25}}=35.9$ ( $c=1.0$ in $\mathrm{CHCl}_{3}$ ); IR (neat) 1702, 1699, $1100 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 1.50(\mathrm{~s}, 9 \mathrm{H}), 3.22$ and $3.27(\mathrm{dd}, \mathrm{J}=$ $12.6,3.8 \mathrm{~Hz}$ and $J=12.7,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}), 3.69$ and $3.86(\mathrm{~d}, \mathrm{~J}=12.6 \mathrm{~Hz}$
and $J=12.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.75-3.85(\mathrm{~m}, 2 \mathrm{H}), 3.93$ and $4.00(\mathrm{~d}, \mathrm{~J}=10.0 \mathrm{~Hz}$ and $\mathrm{J}=10.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.20$ and $4.31(\mathrm{~d}, \mathrm{~J}=3.5 \mathrm{~Hz}$ and $\mathrm{J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.41$ and $4.48(\mathrm{~d}, \mathrm{~J}=4.5$ Hz and $J=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.60$ and $4.84(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}$ and $J=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.63-$ $4.65(\mathrm{~m}, 1 \mathrm{H}), 4.69$ and $4.71(\mathrm{~d}, \mathrm{~J}=11.8 \mathrm{~Hz}$ and $J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.26-7.40(\mathrm{~m}, 5 \mathrm{H})$; ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 28.5$ and 28.6, 49.8 and 50.7, 57.0, 65.1 and $65.3,71.4$ and $71.5,72.4$ and $72.9,80.0$ and $80.5,81.9$ and 82.5, 82.5 and $83.4,83.5$ and $84.7,127.5$ and $127.6,127.9$ and 127.9, 128.4 and 128.6, 137.8 and 138.3, 154.0 and 154.3; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{NNaO}_{5} 372.1871$, found 372.1789 .
tert-Butyl (3S,3aS,6R,6aR)-3-hydroxy-6-methoxyhexahydro-4H-furo[3,2-b]pyrrole-4-carboxylate (11)

A solution of compound 10 ( $298 \mathrm{mg}, 0.85 \mathrm{mmol}$ ) in ethanol ( 10 mL ) was hydrogenated with molecular hydrogen for 2 h at atmospheric pressure and room temperature and in the presence of $20 \% \mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}(74,5 \mathrm{mg})$ as a catalyst. The catalyst was removed by filtration through a Celite ${ }^{\circledR}$ path and the solvent evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: diethyl ether/hexane: $2 / 1,2^{\text {nd }}$ eluent: diethyl ether) yielded diastereomerically pure 11 (220 $\mathrm{mg}, 99 \%$ yield) as a colourless oil. $[\alpha]_{\mathrm{D}}{ }^{25}=51.9$ ( $c=1.0$ in $\mathrm{CHCl}_{3}$ ); IR (neat) 3437, $1700,1168,1102 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 1.44$ and $1.47(\mathrm{~s}, 9 \mathrm{H}), 3.20$ and $3.26(\mathrm{dd}, \mathrm{J}=12.5,3.9 \mathrm{~Hz}$ and $J=12.4,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.34(\mathrm{~s}$, $3 \mathrm{H}), 3.59$ and $3.75-3.81(\mathrm{~d}, \mathrm{~J}=12.4 \mathrm{~Hz}$ and $\mathrm{m}, 1 \mathrm{H}), 3.73(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.75-$ $3.81(\mathrm{~m}, 1 \mathrm{H}), 3.75-3.81$ and $3.85(\mathrm{~m}$ and $\mathrm{dd}, \mathrm{J}=9.8,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.17-4.21(\mathrm{~m}$, 1H), 4.35-4.38 and 4.41-4.43(m, 1H), 4.58-4.61 (m, 1H); ${ }^{13} \mathrm{C}-A P T\{1 \mathrm{H}\}$ NMR ( 100
$\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 28.5$ and $28.6,49.7$ and $50.8,57.1,68.1$ and $68.8,74.2$ and $74.3,75.6$ and $76.5,80.4$ and $80.6,81.9$ and $82.5,83.4$ and 84.6 , 154.9 and 155.1; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{21} \mathrm{NNaO}_{5}$ 282.1312, found 282.1318.
tert-Butyl (3S,3aS,6R,6aR)-3-acetoxy-6-methoxyhexahydro-4H-furo[3,2-b]pyrrole-4-carboxylate (12)

Acetic anhydride ( $150 \mathrm{mg}, 1.47 \mathrm{mmol}$ ) was added to a stirred solution of compound 11 ( $95,5 \mathrm{mg}, 0.37 \mathrm{mmol})$, DMAP ( $22.5 \mathrm{mg}, 0.18 \mathrm{mmol}$ ) and triethylamine ( 149 mg , $1.47 \mathrm{mmol})$ in acetonitrile $(1 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 5 min first and then at room temperature for 1 h . The reaction mixture was diluted with dichloromethane ( 12 mL ) and water $(5 \mathrm{~mL})$. The aqueous layer was then extracted with dichloromethane ( $2 \times 6 \mathrm{~mL}$ ). The combined organic extracts were washed with brine ( 6 mL ), dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: $2 / 1)$ yielded diastereomerically pure 12 ( $110 \mathrm{mg}, 99 \%$ yield) as a colourless oil. $[\alpha]_{D^{25}}$ $=16.7$ ( $c=1.0$ in $\mathrm{CHCl}_{3}$ ); IR (neat) 1747, 1699, 1233, $1102 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}(400 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 1.41(\mathrm{~s}, 9 \mathrm{H}), 2.02(\mathrm{~s}, 3 \mathrm{H}), 3.15-3.21(\mathrm{~m}, 1 \mathrm{H}), 3.30(\mathrm{~s}$, 3 H ), 3.61 and $3.70-3.81(\mathrm{~d}, \mathrm{~J}=12.4 \mathrm{~Hz}$ and $\mathrm{m}, 1 \mathrm{H}$ ), 3.70-3.81(m,2H), 3.85 (dd, J $=10.7,3.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.19$ and $4.31(\mathrm{~d}, \mathrm{~J}=3.7 \mathrm{~Hz}$ and $J=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.49-4.54(\mathrm{~m}$, $1 \mathrm{H}), 5.25-5.30(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 20.9,28.4,49.5$ and $50.5,57.0,65.5$ and $65.5,72.5$ and $72.9,77.3$ and $77.6,80.4$ and 80.6, 81.5 and 82.4, 83.7 and 84.9, 153.9 and 154.1, 169.7 and 170.0; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{NNaO}_{6} 324.1418$, found 324.1423 .

## tert-Butyl <br> (3S,3aS,6R,6aR)-3-(benzyloxy)-6-chlorohexahydro-4H-furo[3,2-

 b]pyrrole-4-carboxylate (13)N -Chorosuccinimide ( $80 \mathrm{mg}, 0.60 \mathrm{mmol}$ ), trifluoroacetic acid ( $40 \mathrm{mg}, 0.35 \mathrm{mmol}$ ) and water ( $120 \mu \mathrm{~L}$ ) were added successively to a stirred solution of compound 5 (123 $\mathrm{mg}, 0.30 \mathrm{mmol})$ in dimethylformamide $(1.5 \mathrm{~mL})$ at room temperature. The resulting mixture was stirred at $35^{\circ} \mathrm{C}$ until the complete disappearance of the starting material. The progress of the reaction was monitored by TLC and additional portions of N chorosuccinimide were added if reaction stopped before completion. Reaction was complete in about 5 h , which was followed by removal of the solvent in vacuo. Purification of the residue by flash chromatography (eluent: ethyl acetate/hexane: $1 / 9$ yielded diastereomerically pure $13(85 \mathrm{mg}, 80 \%$ yield) as a white solid. M.p. $=$ $57.8-59.5^{\circ} \mathrm{C} ;[\alpha] \mathrm{D}^{25}=43.2\left(c=1\right.$ in $\left.\mathrm{CHCl}_{3}\right)$; $\mathrm{IR}(\mathrm{KBr}) 1702,1164,1115 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ $\operatorname{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 1.52$ and $1.53(\mathrm{~s}, 9 \mathrm{H}), 3.54$ and 3.56 (dd, $J=13.0,3.8 \mathrm{~Hz}$ and $J=13.0,3.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.83 and $3.85-3.89(\mathrm{dd}, \mathrm{J}=10.4$, 3.6 Hz and $\mathrm{m}, 1 \mathrm{H}$ ), $3.85-3.89$ and $4.02(\mathrm{~m}$ and $\mathrm{d}, \mathrm{J}=13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.99$ and 4.07 (d, $J=10.0 \mathrm{~Hz}$ and $J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.21$ and $4.31(\mathrm{~d}, \mathrm{~J}=3.4 \mathrm{~Hz}$ and $J=3.4 \mathrm{~Hz}$, $1 \mathrm{H}), 4.25$ and $4.26(\mathrm{~d}, \mathrm{~J}=3.8 \mathrm{~Hz}$ and $\mathrm{J}=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.56$ and $4.66(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}$ and $J=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.62$ and $4.68(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}$ and $J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.71$ and $4.84(\mathrm{~d}, \mathrm{~J}=11.6 \mathrm{~Hz}$ and $J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.73$ and $4.76(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}$ and $J=4.0$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 4.27-4.41 (m, 5H); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 28.4$ and $28.5,53.2$ and $53.8,58.3$ and $58.8,64.9,71.4$ and $71.5,73.4$ and 73.9, 80.6 and $80.9,82.1$ and $82.8,86.9$ and $87.8,127.5$ and 127.8, 127.7 and
127.9, 128.4 and 128.6, 137.6 and 138.0, 153.9 and 154.2; $\mathrm{HRMS}(\mathrm{ESI}+) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$ calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{CINNaO}_{4}$ 376.1286, found 376.1292.

## b]pyrrole-4-carboxylate (14)

tert-Butyl (3S,3aS,6R,6aR)-6-chloro-3-methoxyhexahydro-4H-furo[3,2-

A solution of compound $13(78 \mathrm{mg}, 0.22 \mathrm{mmol})$ in ethanol ( 2.6 mL ) was hydrogenated with molecular hydrogen for 2.5 h at atmospheric pressure and room temperature and in the presence of $20 \% \mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}(19,4 \mathrm{mg})$ as a catalyst. The catalyst was removed by filtration through a Celite ${ }^{\circledR}$ path and the solvent evaporated in vacuo. The obtained residue was dissolved in anhydrous THF ( 1.4 mL ) and a $60 \%$ dispersion of sodium hydride in mineral oil ( $13.4 \mathrm{mg}, 0.33 \mathrm{mmol}$ ) was added in a single portion under an Ar atmosphere at $0^{\circ} \mathrm{C}$. After being stirred for 1 h at $0^{\circ} \mathrm{C}$, methyl iodide ( $185 \mathrm{mg}, 1.30 \mathrm{mmol}$ ) was added and stirring was continued under an Ar atmosphere at room temperature for additional 2 h . The reaction was quenched at $0{ }^{\circ} \mathrm{C}$ with water $(3 \mathrm{~mL})$ and saturated $\mathrm{NaHCO}_{3}(2 \mathrm{~mL})$ aqueous solution and then extracted with dichloromethane ( $3 \times 10 \mathrm{~mL}$ ). The combined organic extracts were washed with brine ( 10 mL ), dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. Purification of the residue by flash chromatography (eluent: diethyl ether/hexane: 2/3) yielded diastereomerically pure 14 ( $49.5 \mathrm{mg}, 81 \%$ yield) as a yellowish solid. M.p. $=73.1-74.5^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=89.7\left(c=1.0\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \operatorname{IR}(\mathrm{KBr}) 1693$, 1116, $1076 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (mixture of rotamers) $\delta 1.47$ and 1.52 (s, 9H), 3.44 and $3.48(\mathrm{~s}, 3 \mathrm{H}), 3.50$ and $3.53(\mathrm{dd}, \mathrm{J}=12.8,4.0 \mathrm{~Hz}$ and $\mathrm{J}=12.8,, 4.0$ $\mathrm{Hz}, 1 \mathrm{H}), 3.78$ and $3.91-3.95(\mathrm{dd}, \mathrm{J}=10.4,3.6 \mathrm{~Hz}$ and m, , 1 H ), 3.79 and $3.91-3.95$ $(\mathrm{d}, \mathrm{J}=10.4 \mathrm{~Hz}$ and $\mathrm{m}, 1 \mathrm{H}), 3.84$ and $3.98(\mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}$ and $\mathrm{J}=13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.91-$
3.95 and $4.06(\mathrm{~m}$ and $\mathrm{d}, \mathrm{J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.21$ and $4.23(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}$ and $\mathrm{J}=4.0$ $\mathrm{Hz}, 1 \mathrm{H}), 4.45$ and $4.53(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}$ and $\mathrm{J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.63$ and $4.66(\mathrm{~d}, \mathrm{~J}=4.4$ Hz and $J=4.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ (mixture of rotamers) $\delta 28.5$ and 28.6, 53.2 and $53.9,57.3$ and $57.4,58.4$ and $58.9,64.1$ and $64.4,73.5$, 80.6 and $80.9,83.8$ and $84.7,86.9$ and $87.8,153.9$ and 154.2; HRMS (ESI+) $m / z$ $[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{CINNaO}_{4} 300.0973$, found 300.0980 .
(3S,3aS,6R,6aR)-3-(Benzyloxy)-4-(7,8-dimethoxyquinazolin-4-yl)hexahydro-2H-furo[3,2-b]pyrrol-6-ol (15A)

A solution of compound $7(40 \mathrm{mg}, 0.12 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at room temperature for 2 h and then evaporated in vacuo. The residue was diluted with 1 m aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane ( $3 \times 8 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 26 \mathrm{mg}$ ) was dissolved in anhydrous DMF ( 1 mL ) and DIPEA ( $75 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) and 4-chloro7,8 -dimethoxyquinazoline ( $34 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $60^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: ethyl acetate/dichloromethane: $1 / 1,2^{\text {nd }}$ eluent: ethyl acetate/dichloromethane/ ethanol: $5 / 5 / 1$ ) yielded diastereomerically pure 15A ( $47 \mathrm{mg}, 93 \%$ yield) as a white solid. M.p. $=178.8-180.2^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}} 25=100.0\left(c=0.25\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \operatorname{IR}(\mathrm{KBr}) 3245,1494,1102 \mathrm{~cm}^{-}$ ${ }^{1} ;{ }^{1} \mathrm{H} \operatorname{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.81(\mathrm{~s}, 3 \mathrm{H}), 3.90-3.94(\mathrm{~m}, 1 \mathrm{H}), 3.93(\mathrm{~s}, 1 \mathrm{H}), 4.02$ (d, $J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{dd}, \mathrm{J}=12.0,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.57$ $(d, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.80(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.86(\mathrm{~d}, \mathrm{~J}=$
$12.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~d}, \mathrm{~J}=12.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.39(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{~d}, \mathrm{~J}=9.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.28(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{dd}, \mathrm{J}=7.4,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, 7.67 (d, J = $9.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.27(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 56.2$, 58.3, 61.6, 68.5, 71.4, 73.7, 74.2, 82.2, 84.9, 110.2, 110.7, 122.5, 127.8, 127.9, 128.6, 138.3, 139.3, 153.0, 154.0, 160.0; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{5}$ 424.1867, found 424.1871.
(3S,3aS,6R,6aR)-3-(Benzyloxy)-4-(7,8-dimethoxyquinazolin-4-yl)hexahydro-2H-furo[3,2-b]pyrrol-6-yl acetate (16A)

A solution of compound $9(64 \mathrm{mg}, 0.17 \mathrm{mmol})$ in dichloromethane/ trifluoroacetic acid $5 / 1(1.2 \mathrm{~mL})$ was stirred at room temperature for 3 h and then evaporated in vacuo. The residue was diluted with 0.6 M potassium carbonate/bicarbonate buffer $(4 \mathrm{~mL})$ and extracted with dichloromethane ( $3 \times 8 \mathrm{~mL}$ ). The combined organic extracts were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated under reduced pressure. The residue ( $\approx 45 \mathrm{mg}$ ) was dissolved in anhydrous DMF ( 1.3 mL ) and DIPEA ( $112 \mathrm{mg}, 0.87 \mathrm{mmol}$ ) and 4-chloro-7,8-dimethoxyquinazoline ( $49 \mathrm{mg}, 0.22$ mmol ) were added successively. The reaction mixture was stirred at $60^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ethyl acetate/ethanol: $9 / 1$ containing $\mathrm{Et}_{3} \mathrm{~N} \quad(1 \% \mathrm{v} / \mathrm{v})$ ) yielded diastereomerically pure 16A (76 mg, 96\% yield) as a white solid. M.p. $=81.3-84.7$ ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=78.8\left(c=1.0\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}(\mathrm{KBr}) 1750,1492,1098,1046 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.97(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{dd}, \mathrm{J}=10.2,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.98-4.01(\mathrm{~m}, 1 \mathrm{H})$, $3.99(\mathrm{~s}, 3 \mathrm{H}), 4.03(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.05(\mathrm{~s}, 3 \mathrm{H}), 4.23-4.27(\mathrm{~m}, 2 \mathrm{H}), 4.67(\mathrm{~d}, \mathrm{~J}=$ $4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{~d}, \mathrm{~J}=12.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.93(\mathrm{~d}, \mathrm{~J}=12.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}$,
$1 \mathrm{H}), 4.34(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{~d}, \mathrm{~J}=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.23-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.30-7.35$ (m, 2H), 7.39-7.41 (m, 2H), $7.74(\mathrm{~d}, \mathrm{~J}=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.71(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 21.0,56.4,56.5,61.7,68.2,71.3,73.7,75.5,81.9,82.1,111.7$, $112.0,121.3,127.8,127.9,128.5,138.1,142.2,146.9,154.1,154.2,159.9,170.0 ;$ HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{NaO}_{6} 488.1792$, found 488.1770. (3S,3aS,6R,6aR)-4-(7,8-Dimethoxyquinazolin-4-yl)-6-methoxyhexahydro-2H-furo[3,2-b]pyrrol-3-yl acetate (17A)

A solution of compound 12 ( $51 \mathrm{mg}, 0.17 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(1 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 3 h and then evaporated in vacuo. The residue was dissolved in anhydrous DMF ( $1,4 \mathrm{~mL}$ ) and DIPEA ( $145 \mathrm{mg}, 1.12 \mathrm{mmol}$ ) and 4-chloro-7,8dimethoxyquinazoline ( $50 \mathrm{mg}, 0.22 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $45^{\circ} \mathrm{C}$ for 2.5 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: ethyl acetate/dichloromethane: 1/1, $2^{\text {nd }}$ eluent: ethyl acetate/dichloromethane/ ethanol: 9/9/2) yielded diastereomerically pure 17A ( $57 \mathrm{mg}, 86 \%$ yield) as a white solid. M.p. $=180.1-181.6^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=88.4$ ( $c=1.0$ in $\mathrm{CHCl}_{3}$ ); $\mathrm{IR}(\mathrm{KBr}) 1731,1485,1244,1111 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 2.11$ (s, 3H), 3.29 (s, 3H), 3.89 (d, J = $10.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.94-3.97 (m, 1H), 3.96 (s, $3 \mathrm{H}), 4.00(\mathrm{dd}, \mathrm{J}=10.8,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H}), 4.02-4.07(\mathrm{~m}, 2 \mathrm{H}), 4.66(\mathrm{~d}, \mathrm{~J}=4.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.21(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.26(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=9.4 \mathrm{~Hz}, 1 \mathrm{H})$, 7.77 (d, J = $9.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), $8.63(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 21.1, $55.9,56.5,57.2,61.7,67.4,73.5,77.4,82.0,82.5,111.8,111.9,121.3,142.2,146.9$, 154.1, 154.1, 160.2, 170.1; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{6}$ 390.1660 , found 390.1663 .
(3S,3aS,6R,6aR)-3-(Benzyloxy)-6-chloro-4-(7,8-dimethoxyquinazolin-4-yl)hexahydro-2H-furo[3,2-b]pyrrole (18A)

A solution of compound $13(50 \mathrm{mg}, 0.14 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 3 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane ( 3 x 8 mL ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 36 \mathrm{mg}$ ) was dissolved in anhydrous DMF (1 mL ) and DIPEA ( $98 \mathrm{mg}, 0.76 \mathrm{mmol}$ ) and 4-chloro-7,8-dimethoxyquinazoline ( 43 mg , $0.19 \mathrm{mmol})$ were added successively. The reaction mixture was stirred at $45^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: diethyl ether, $2^{\text {nd }}$ eluent: ethyl acetate containing $E t_{3} \mathrm{~N}$ $(0.5 \% \mathrm{v} / \mathrm{v})$ ) yielded diastereomerically pure 18A ( $57 \mathrm{mg}, 91 \%$ yield) as a yellowish solid. M.p. $=49.6-52.3^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=125.7\left(c=0.5\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}(\mathrm{KBr}) 1492,1104$, $1041 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.97(\mathrm{dd}, \mathrm{J}=10.4,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H})$, $4.05(\mathrm{~d}, \mathrm{~J}=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.07(\mathrm{~s}, 3 \mathrm{H}), 4.21-4.23(\mathrm{~m}, 1 \mathrm{H}), 4.23(\mathrm{~d}, \mathrm{~J}=11.6 \mathrm{~Hz}, 1 \mathrm{H})$, 4.41 (dd, $J=11.6,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.47(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}, 1 \mathrm{H})$, $4.83(\mathrm{~d}, \mathrm{~J}=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.92(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{~d}$, $J=9.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.29(\mathrm{~m}, 1 \mathrm{H}), 7.31-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.38-7.42(\mathrm{~m}, 2 \mathrm{H}), 7.70(\mathrm{~d}$, $J=9.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.74(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 56.6,58.9$, $59.0,61.8,67.5,71.4,74.6,81.6,85.4,111.6,112.2,121.1,127.9,128.0,128.6$, 138.0, 142.3, 146.8, 154.1, 154.3, 160.1; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{CIN}_{3} \mathrm{O}_{4} 442.1528$, found 442.1549 .
(3S,3aS,6R,6aR)-6-Chloro-4-(7,8-dimethoxyquinazolin-4-yl)-3-methoxyhexahydro-2H-furo[3,2-b]pyrrole (19A)

A solution of compound $14(48 \mathrm{mg}, 0.17 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 5 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane ( 3 x 8 mL ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 29 \mathrm{mg}$ ) was dissolved in anhydrous DMF (1 mL ) and DIPEA ( $112 \mathrm{mg}, 0.87 \mathrm{mmol}$ ) and 4-chloro-7,8-dimethoxyquinazoline (49 $\mathrm{mg}, 0.22 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $45^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: diethyl ether, $2^{\text {nd }}$ eluent: ethyl acetate/ethanol: 9/1 containing $\mathrm{Et}_{3} \mathrm{~N}(0.5 \% \mathrm{v} / \mathrm{v})$ ) yielded diastereomerically pure 19A ( $49 \mathrm{mg}, 77 \%$ yield) as a yellowish solid. M.p. $=224.7-226.9^{\circ} \mathrm{C} ;[\alpha]{ }^{25}=156.5\left(c=0.5\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}$ (KBr) 1492, $1105 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.56(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{dd}, \mathrm{J}=10.0$, $4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.99-4.02(\mathrm{~m}, 1 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.05(\mathrm{~s}, 3 \mathrm{H})$, $4.20(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.38(\mathrm{dd}, \mathrm{J}=12.0,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{~d}, \mathrm{~J}=3.8 \mathrm{~Hz}, 1 \mathrm{H})$, $4.74(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.44(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.16(\mathrm{~d}, \mathrm{~J}=9.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.71(\mathrm{~d}, \mathrm{~J}$ $=9.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.69(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 56.6,57.6,58.9$, $59.0,61.8,66.9,74.1,84.1,85.4,111.7,112.2,121.1,142.4,147.1,154.2,154.2$, 160.2; HRMS (ESI+) m/z $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{17} \mathrm{H}_{21} \mathrm{CIN}_{3} \mathrm{O}_{4} 366.1215$, found 366.1206.
(3S,3aS,6R,6aR)-3-(Benzyloxy)-4-(6,7-dimethoxyquinazolin-4-yl)hexahydro-2H-furo[3,2-b]pyrrol-6-ol (15B)

A solution of compound $7(64 \mathrm{mg}, 0.19 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at room temperature for 2 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane ( $3 \times 8 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 42 \mathrm{mg}$ ) was dissolved in anhydrous DMF ( 1.3 mL ) and DIPEA ( $125 \mathrm{mg}, 0.97 \mathrm{mmol}$ ) and 4-chloro-6,7-dimethoxyquinazoline ( $54 \mathrm{mg}, 0.24 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: ethyl acetate, $2^{\text {nd }}$ eluent: ethyl acetate/ethanol: 9/1) yielded diastereomerically pure 15B ( $57 \mathrm{mg}, 70 \%$ yield) as a white solid. M.p. $=163.7-164.9^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=92.2\left(c=0.5\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}$ (KBr) 3361, 1510, 1077, $1003 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.92$ (s, 3H), 3.95 (s, 3H), 3.95-3.99 (m, 1H), 4.01 (dd, J = 10.2, 1.0 Hz, 1H), 4.14-4.18 (m, 2H), 4.24 $(\mathrm{d}, \mathrm{J}=11.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{~d}, \mathrm{~J}=3.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~d}, \mathrm{~J}=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~d}, \mathrm{~J}=$ $12.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.89(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{~d}, \mathrm{~J}=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.03(\mathrm{~s}, 1 \mathrm{H}), 7.21$ $(\mathrm{s}, 1 \mathrm{H}), 7.25-7.40(\mathrm{~m}, 5 \mathrm{H}), 8.31(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 56.3$, $56.3,58.3,68.1,71.5,73.5,74.1,82.3,85.1,104.2,105.9,109.7,127.9,127.9$, 128.6, 138.2, 147.0, 147.8, 151.9, 154.3, 158.8; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{5} 424.1867$, found 424.1872.
(3S,3aS,6R,6aR)-3-(Benzyloxy)-4-(6,7-dimethoxyquinazolin-4-yl)hexahydro-

## 2H-furo[3,2-b]pyrrol-6-yl acetate (16B)

A solution of compound $9(46.5 \mathrm{mg}, 0.12 \mathrm{mmol})$ in dichloromethane/ trifluoroacetic acid $5 / 1(1.2 \mathrm{~mL})$ was stirred at room temperature for 3 h and then evaporated in vacuo. The residue was diluted with 0.6 M potassium carbonate/bicarbonate buffer $(4 \mathrm{~mL})$ and extracted with dichloromethane $(3 \times 8 \mathrm{~mL})$. The combined organic extracts were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated under reduced pressure. The residue ( $\approx 33 \mathrm{mg}$ ) was dissolved in anhydrous DMF ( 1 mL ) and DIPEA $(80 \mathrm{mg}, 0.62 \mathrm{mmol})$ and 4-chloro-6,7-dimethoxyquinazoline ( $36 \mathrm{mg}, 0.16 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ethyl acetate) yielded diastereomerically pure 16B ( $41 \mathrm{mg}, 71 \%$ yield) as a white solid. M.p. $=170.1-172.0^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=104.0\left(c=0.05\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \operatorname{IR}(\mathrm{KBr}) 1760$, 1504, 1229, 1110, $1061 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ HMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 2.00(\mathrm{~s}, 3 \mathrm{H}), 3.95$ (s, $3 \mathrm{H}), 3.97-4.03(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~s}, 3 \mathrm{H}), 4.04(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.21-4.23(\mathrm{~m}, 1 \mathrm{H})$, 4.28 (dd, $J=12.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.81(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}, 1 \mathrm{H})$, $4.90(\mathrm{~d}, \mathrm{~J}=12.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.35(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.36(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~s}$, 1H), 7.27 (s, 1H), 7.26-7.29 (m, 1H), 7.31-7.36 (m, 2H), 7.37-7.40 (m, 2H), 8.62 (s, 1H); ${ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 21.1,56.1,56.3,56.4,68.2,71.5,73.7$, 75.4, 82.2, 82.6, 103.7, 107.4, 110.3, 127.9, 127.9, 128.6, 138.1, 148.2, 148.8, 152.8, 154.5, 159.0, 170.0; HRMS (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{6}$ 466.1973, found 244.1970.
(3S,3aS,6R,6aR)-4-(6,7-Dimethoxyquinazolin-4-yl)-6-methoxyhexahydro-2H-furo[3,2-b]pyrrol-3-yl acetate (17B)

A solution of compound $12(85 \mathrm{mg}, 0.28 \mathrm{mmol})$ in dichloromethane/ trifluoroacetic acid $2.5 / 1(2.5 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 10 min first and then at room temperature for 2 h . The reaction mixture was evaporated in vacuo and the residue was diluted with 0.6 M potassium carbonate/bicarbonate buffer ( 9 mL ) and extracted with dichloromethane ( $3 \times 15 \mathrm{~mL}$ ). The combined organic extracts were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated under reduced pressure. The residue ( $\approx$ 28 mg ) was dissolved in anhydrous DMF ( 1 mL ) and DIPEA ( $94 \mathrm{mg}, 0.73 \mathrm{mmol}$ ) and 4-chloro-6,7-dimethoxyquinazoline ( $43 \mathrm{mg}, 0.19 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $80^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: ethyl acetate/dichloromethane: $1 / 1,2^{\text {nd }}$ eluent: ethyl acetate/dichloromethane/ethanol: 10/10/1) yielded diastereomerically pure 17B ( $54 \mathrm{mg}, 49 \%$ yield) as a white solid. M.p. $=65.2-66.5^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=58.7\left(c=0.25\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{IR}(\mathrm{KBr})$ 1736, 1509, 1240, $1094 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ HMR (400 MHz, CDCl ${ }_{3}$ ) $\delta 2.13(\mathrm{~s}, 3 \mathrm{H}), 3.35(\mathrm{~s}, 3 \mathrm{H}), 3.91(\mathrm{~d}, \mathrm{~J}=10.8$ $\mathrm{Hz}, 1 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.99-4.01(\mathrm{~m}, 1 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H}), 4.04-4.12(\mathrm{~m}, 3 \mathrm{H}), 4.71(\mathrm{~d}, \mathrm{~J}$ $=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.24(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.25(\mathrm{~d}, \mathrm{~J}=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.27(\mathrm{~s}, 1 \mathrm{H}), 7.28$ (s, 1H), $8.54(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 21.2, 55.7, 56.2, 56.4, $57.3,67.5,73.6,77.6,82.4,82.5,103.9,107.0,110.2,148.1,152.5,154.5,159.3$, 170.2; HRMS (ESI+) m/z [M+Na] ${ }^{+}$calcd for $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{NaO}_{6}$ 412.1479, found 412.1476.
(3S,3aS,6R,6aR)-3-(Benzyloxy)-6-chloro-4-(6,7-dimethoxyquinazolin-4-yl)hexahydro-2H-furo[3,2-b]pyrrole (18B)

A solution of compound $13(67 \mathrm{mg}, 0.19 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 3 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane ( 3 x 8 mL ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 48 \mathrm{mg}$ ) was dissolved in anhydrous DMF (1.1 mL ) and DIPEA ( $129 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and 4-chloro-6,7-dimethoxyquinazoline ( 56 mg , 0.25 mmol ) were added successively. The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ethyl acetate/hexanes: 3/1) yielded diastereomerically pure 18B (46 mg, $55 \%$ yield) as a yellowish solid. M.p. $=62.0-63.0^{\circ} \mathrm{C} ;[\alpha]^{25}=70.8(c=$ 0.15 in $\mathrm{CHCl}_{3}$ ); $\mathrm{IR}(\mathrm{KBr}) 1509,1219,1104 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.95$ (s, 3H), $3.99(\mathrm{dd}, \mathrm{J}=10.4,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.03(\mathrm{~s}, 3 \mathrm{H}), 4.06(\mathrm{~d}, \mathrm{~J}=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.18$ $(d, J=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{~d}, \mathrm{~J}=11.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{dd}, \mathrm{J}=11.6,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.48$ $(\mathrm{d}, \mathrm{J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.78(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.84-4.86(\mathrm{~m}, 1 \mathrm{H}), 4.87(\mathrm{~d}, \mathrm{~J}=12.0$ $\mathrm{Hz}, 1 \mathrm{H}), 5.53(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 7.25-7.39(\mathrm{~m}, 6 \mathrm{H}), 8.63(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-$ APT\{1H\} NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 56.3,56.4,58.5,58.9,67.5,71.5,74.5,81.8$, 85.8, 103.6, 107.3, 110.2, 128.0, 128.0, 128.6, 137.9, 148.3, 148.6, 152.7, 154.6, 159.1; $\mathrm{HRMS}(\mathrm{ESI}+) \mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{CIN}_{3} \mathrm{O}_{4} 442.1528$, found 442.1539 .
(3S,3aS,6R,6aR)-6-Chloro-4-(6,7-dimethoxyquinazolin-4-yl)-3-methoxyhexahydro-2H-furo[3,2-b]pyrrole (19B)

A solution of compound $14(48 \mathrm{mg}, 0.17 \mathrm{mmol})$ in 4 M HCl in ethyl acetate $(0.75 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 5 h and then evaporated in vacuo. The residue was diluted with 1 M aqueous solution of $\mathrm{NaOH}(4 \mathrm{~mL})$ and extracted with dichloromethane (3 x 8 mL ). The combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and evaporated in vacuo. The residue ( $\approx 29 \mathrm{mg}$ ) was dissolved in anhydrous DMF (1 mL ) and DIPEA ( $112 \mathrm{mg}, 0.87 \mathrm{mmol}$ ) and 4-chloro-6,7-dimethoxyquinazoline (49 $\mathrm{mg}, 0.22 \mathrm{mmol}$ ) were added successively. The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 12 h and then evaporated in vacuo. Purification of the residue by flash chromatography ( $1^{\text {st }}$ eluent: ethyl acetate/dichloromethane: $1 / 1,2^{\text {nd }}$ eluent: ethyl acetate/dichloromethane/ethanol: 10/10/1) yielded diastereomerically pure 19B (27 $\mathrm{mg}, 43 \%$ yield) as a yellowish solid. M.p. $=106.8-108.4^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=161.3(c=0.5$ in $\left.\mathrm{CHCl}_{3}\right)$; $\mathrm{IR}(\mathrm{KBr}) 1511,1249,1109,1002 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{HMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.51$ $(\mathrm{s}, 3 \mathrm{H}), 3.93(\mathrm{dd}, \mathrm{J}=10.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.97-3.99(\mathrm{~m}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~s}, 3 \mathrm{H})$, $4.02(\mathrm{~d}, \mathrm{~J}=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{~d}, \mathrm{~J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{dd}, \mathrm{J}=12.0,3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.44(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.43(\mathrm{~d}, \mathrm{~J}=4.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~s}$, $1 \mathrm{H}), 7.27(\mathrm{~s}, 1 \mathrm{H}), 8.58(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{APT}\{1 \mathrm{H}\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 56.2,56.4$, $57.5,58.4,58.7,67.1,73.7,84.2,86.0,103.5,107.4,110.2,148.2,148.7,152.8$, 154.5, 159.0; HRMS (ESI+) $m / z[M+H]^{+}$calcd for $\mathrm{C}_{17} \mathrm{H}_{21} \mathrm{ClN}_{3} \mathrm{O}_{4} 366.1215$, found 366.1227.

## Inhibition assays

PDE1A, PDE1B and PDE1C inhibition assays were performed in $60 \mu \mathrm{~L}$ samples containing a fixed amount of the PDE1 enzyme (sufficient to convert 20-25\% of the tritiated substrate), a buffer ( 50 mM HEPES, $\mathrm{pH} 7.6 ; 10 \mathrm{mM} \mathrm{MgCl} 2 ; 0.02 \%$ Tween20), $0.1 \mathrm{mg} / \mathrm{mL}$ BSA, $\left[{ }^{3} \mathrm{H}\right]$-cAMP to a final concentration of 15 nM and varying amounts of inhibitors. Reactions were initiated by addition of $\left[{ }^{3} \mathrm{H}\right]-\mathrm{cAMP}$, and reactions were allowed to proceed for 1 h at room temperature before being terminated through mixing with $20 \mu \mathrm{~L}(0.2 \mathrm{mg})$ yttrium silicate SPA beads (PerkinElmer, Waltham, MA, USA). The beads were allowed to settle for 1 h in the dark before the plates were counted in a Wallac 1450 Microbeta counter (PerkinElmer). The measured signals were converted to activity relative to an uninhibited control (100\%), and IC50 values were calculated using XIFit (IDBS) extension to excel. PDE4 inhibition assay was performed in a similar fashion using [ $\left.{ }^{3} \mathrm{H}\right]$-labelled cAMP.

## Cell proliferation assays

Cell proliferation/survival was spectrophotometrically assessed using the Janus Green B green assay as described previously with some modifications. ${ }^{20}$ Briefly, cells (1500 cells/well) were seeded in 96-well plates and the compound(s) or the vehicle, DMSO, were added. Vehicle concentrations were kept the same for all concentrations of compounds. Cells were formalin-fixed at day 0 (immediately after addition of compounds), 1,2, 3, and 4 . Fixed cells were stained for 5 min with 50 $\mu / /$ well of $0.3 \%$ Janus B Green dye (Acros Organics, Belgium) at room temperature with continuous stirring followed by a washing step with water. The dye was eluted
with $200 \mu \mathrm{l} /$ well of 0.5 M HCl of hydrochloric acid and top-read measurements of absorbance were performed in a microplate reader (Sinergy HT, Biotek, USA) at 595 nm . Data in figures are presented as \% of control (DMSO). We used absorption values for statistical comparisons and, for calculation of IC50 values, we fitted data points using the concentration-response equation: $y=A 2+(A 1-A 2) /\left(1+(x / x 0)^{p}\right.$.

## ASSOCIATED CONTENT

## Supporting Information

The supporting information is available free of charge via the internet at http://pubs.acs.org.

Copies ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of products and X-ray crystallographic data (ORTEP) for compound 7 (PDF).

X-ray crystallographic data for compound 7 (CIF).

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## REFERENCES

(1) Pittenger, C.; Nestler, E. J.; Duman, R. S. "Cyclic Nucleotides in the Nervous System", in Basic Neurochemistry: Principles of Molecular, Cellular, and Medical Neurobiology, 8th Edition, eds. Brady, S. T.; Siegel, G. J.; Albers, R. W.; Price, D. L. Academic Press: New York, 2012, 423-441.
(2) (a) Bender, A. T.; Beavo, J. A. Cyclic Nucleotide Phosphodiesterases: Molecular Regulation to Clinical Use. Pharmacol. Rev. 2006, 58, 488-520. (b) Francis, S. H.; Blount, M. A.; Corbin, J. D. Mammalian Cyclic Nucleotide Phosphodiesterases: Molecular Mechanisms and Physiological Functions. Physiol. Rev. 2011, 91, 651-690.
(3) (a) Francis, S. H.; Conti, M.; Houslay, M. D. in Phosphodiesterases as Drug Targets, Springer-Verlag: Berlin, 2011. (b) S. Liras, A. S. Bell, in Phosphodiesterases and Their Inhibitors, Wiley-VCH: Weinheim, 2014. (c) Zhang, H. T.; Xu, Y.; O'Donnell, J. M. in Phosphodiesterases: CNS Functions and Diseases, Springer, 2017. (d) DeNinno, M. P. Future Directions in Phosphodiesterase Drug Discovery. Bioorg. Med. Chem. Lett., 2012, 19, 6794-6800. (e) Keravis, T.; Lugnier, C. Cyclic Nucleotide Phosphodiesterase (PDE) Isozymes as Targets of the Intracellular Signalling Network: Benefits of PDE Inhibitors in Various Diseases and Perspectives for Future Therapeutic Developments. British J. Pharmacol. 2012, 165, 1288-1305. (f) Blokland, A.; Menniti, F. S.; Prickaerts, J. PDE Inhibition and Cognition Enhancement. Expert Opin. Therapeutic Patents 2012, 22, 349-354. (g) Sharma, S.; Kumar, K.; Deshmukh R.; Sharma, P. L. Phosphodiesterases: Regulators of Cyclic Nucleotide Signals and Novel Molecular Target for Movement Disorders. Eur.
J. Pharmacol. 2013, 714, 486-497. (h) Maurice, D. H.; Ke, H.; Ahmad, F.; Wang, Y.; Chung, J. C. Manganiello, V. C. Advances in Targeting Cyclic Nucleotide Phosphodiesterases. Nat. Rev. Drug. Discov. 2014,13, 290-314.
(4) (a) Savai, R.; Pullamsetti, S. S.; Banat, G. A.; Weissmann, N.; Ghofrani, H.; Grimminger F. A.; Schermuly; R. T. Targeting Cancer with Phosphodiesterase Inhibitors. Expert Opin. Investig. Drugs 2010, 19, 117-131. (b) Fajardo, A. M.; Piazza G. A.; Tinsley, H. N. The Role of Cyclic Nucleotide Signaling Pathways in Cancer: Targets for Prevention and Treatment. Cancers 2014, 6, 436-458. (c) Peng, T.; Gong, J.; Jin, Y.; Zhou, Y.; Tong, R.; Wei, X.; Bai, L.; Shi, J. Inhibitors of phosphodiesterase as cancer therapeutics. Eur. J. Med. Chem. 2018, 150, 742-756.
(5) Cheung, W. Y. Cyclic 37:57-Nucleotide Phosphodiesterase. Determination of an Activator. Biochem. Biophys. Res. Commun. 1970, 38, 533-538.
(6) Kakiuchi, S.; Yamazaki, R. Calcium-dependent Phosphodiesterase Activity and its Activating Factor (PAF) from Brain Studies on Cyclic 37:57-Nucleotide Phosphodiesterase. Biochem. Biophys. Res. Commun. 1970, 41, 1104-1110.
(7) (a) Medina, A. E. Therapeutic Utility of phosphodiesterase Type I Inhibitors in Neurological Conditions. Frontiers in Neuroscience 2011, 5, article 21. (b) Chan, S.; Yan, C. PDE1 Isozymes, Key Regulators of Pathological Vascular Remodeling. Curr. Opin. Pharmacol. 2011, 11, 720-724. (c) Kokkonen, K.; Kass, D. A. Nanodomain Regulation of Cardiac Cyclic Nucleotide Signaling by Phosphodiesterases. Annu. Rev. Pharmacol. Toxicol. 2017, 57, 455-479. (d) Kim G. E.; Kass D. A. "Cardiac Phosphodiesterases and Their Modulation for Treating Heart Disease" in Heart Failure. Handbook of Experimental

Pharmacology, eds. Bauersachs J.; Butler, J.; Sandner, P. vol 243. Springer International Publishing, AG 2016, 249-269. (e) Nthenge-Ngumbau, D. N.; Mohanakumar, K. P. Can Cyclic Nucleotide Phosphodiesterase Inhibitors Be Drugs for Parkinson's Disease?. Mol. Neurobiol. 2018, 55, 822-834.
(8) (a) Shimizu, K.; Murata, T.; Watanabe, Y.; Sato, C.; Morita, H.; Tagawa, T. Characterization of phosphodiesterase 1 in human malignant melanoma cell lines.Anticancer Res. 2009, 29, 1119-1122. (b) Abusnina, A.; Keravis, T.; Yougbaré, I.; Bronner, C.; Lugnier, C. Anti-proliferative Effect of Curcumin on Melanoma Cells is Mediated by PDE1A Inhibition that Regulates the Epigenetic Integrator UHRF1. Mol. Nutr. Food Res. 2011, 55, 1677-1689.
(9) Kehler, J.; Rasmussen, L. K.; Langgard, M. WO2015118097A1. (Hexahyrofuropyrroles as PDE1 Inhibitors)
(10) Wolfe, J. P.; Hay, M. B. Recent Advances in the Stereoselective Synthesis of Tetrahydrofuranes. Tetrahedron 2007, 63, 261-290. (b) Jalce, G.; Franck, X.; Figadère, B. Diastereoselective Synthesis of 2,5-Disubstituted Tetrahydrofurans. Tetrahedron: Asymmetry 2009, 20, 2537-2581. (c) Tikad, A.; Delbrouck, J. A.; Vincent, S. P. Debenzylative Cycloetherification: An Overlooked Key Strategy for Complex Tetrahydrofuran Synthesis. Chem. Eur. J. 2016, 22, 9456-9476.
(11) Badorrey, R.; Cativiela, C.; Díaz-de-Villegas, M. D.; Díez, R.; Gálvez, J. A. The First Asymmetric Synthesis of 1,4-Dideoxy-1,4-imino-D-talitol. Synlett, 2005, 1734-1736.
(12) In this reaction is important to add the imine to the nucleophile, in this way the formation of triallylimine $\mathbf{X}$ resulting from an elimination/nucleophilic addition
process is avoided. (a) Badorrey, R.; Cativiela, C.; Díaz-de-Villegas, M. D.; Díez, R.; Gálvez, J. A. Stereodivergent Addition of Allylmetal Reagents to Imines Derived from ( R )-2,3-Di-O-benzylglyceraldehyde by Appropriate Selection of Metal and Double Stereodifferentiation. Eur. J. Org. Chem. 2002, 3763-3767. (b) Badorrey, R.; Cativiela, C.; Díaz-de-Villegas, M. D.; Gálvez, J. A. Study of the Lewis acid-promoted addition of silylenol ethers to imines derived from glyceraldehyde. Tetrahedron Lett. 2003, 44, 9189-9192.

x
(13) (a) Cativiela, C.; Díaz-de-Villegas, M. D.; Gálvez, J. A. Stereoselective Synthesis of $\alpha$-Hydroxy- $\beta$-amino Acids Using D-Glyceraldehyde as the Homochiral Source. Tetrahedron: Asymmetry 1996, 7, 529-536. (b) Badorrey, R.; Cativiela, C.; Díaz-de-Villegas, M. D.; Gálvez, J. A. Reversal of the Stereochemical Course of the Addition of Phenylmagnesium Bromide to N benzylimines Derived from R-Glyceraldehyde Depending on the O-Protecting Group and its Application to the Synthesis of Both Enantiomers of Phenylglycine. (c) Badorrey, R.; Cativiela, C.; Díaz-de-Villegas, M. D.; Gálvez, J. A. Highly Convergent Stereoselective Synthesis of Chiral Key Intermediates in the Synthesis of Palinavir from Imines Derived from L-Glyceraldehyde. Tetrahedron 2002, 58, 341-354. (d) Badorrey, R.; Cativiela, C.; Díaz de Villegas, M. D.; Díez, R.; Gálvez, J. A. Study of the Reactions between Vinylmagnesium Bromide and Imines Derived from (R)-Glyceraldehyde - The

Key Step in the Stereodivergent Synthesis of Conveniently Protected, Enantiopure syn- and anti-2-Amino-1,3,4-butanetriol Derivatives. Eur. J. Org. Chem. 2003, 2268-2275.
(14) (a) Felpin, F. X.; Lebreton, J. Recent Advances in the Total Synthesis of Piperidine and Pyrrolidine Natural Alkaloids with Ring-Closing Metathesis as a Key Step. Eur. J. Org. Chem. 2003, 3693-3712. (b) Compain, P. Olefin Metathesis of Amine-Containing Systems: Beyond the Current Consensus. Adv. Synth. Catal. 2007, 349, 1829-1846. (c) Compain, P.; Hazelard, D. "Synthesis of Amine-Containing Heterocycles by Metathesis Reactions: Recent Advances and Opportunities" in Synthesis of Heterocycles by Metathesis Reactions, ed Prunet J. Top. Heterocycl. Chem. 2014, 47, 111153.
(15) Adam, W.; Saha-Moller, C.; Zhao, C. G. Dioxirane Epoxidation of Alkenes. Org. React. 2003, 61, 219-516.
(16) (a) Naumann, K. Influence of Chlorine Substituents on Biological Activity of Chemicals. J. Prakt. Chem. 1999, 341, 417-435. (b) Hernandes, M. Z.; Cavalcanti, S. M. T.; Moreira, D. R. M.; de Azevedo Jr, W. F.; Leite, A. C. L. Halogen Atoms in the Modern Medicinal Chemistry: Hints for the Drug Design. Current Drug Targets, 2010, 11, 303-314. (c) Kosjek, T.; Heath, E. Halogenated Heterocycles as Pharmaceuticals. Top. Heterocycl. Chem. 2012, 27, 219-246.
(17) NMR spectra were recorded in $\mathrm{CDCl}_{3}$ at room temperature as in these conditions spectra showed perfectly resolved the signals corresponding to both $N$-Boc rotamers, which allowed an easier interpretation.
(18) Laursen, M.; Beck, L.; Kehler, J.; Christoffersen, C. T.; Bundgaard, C.; Mogensen, S.; Mow, T. J.; Pinilla, E.; Knudsen, J. S.; Hedegaard, E. R.; Grunnet, M.; Simonsen, U. Novel Selective PDE Type 1 Inhibitors Cause Vasodilatation and Lower Blood Pressure in Rats. Br. J. Pharmacol. 2017, 174, 2563-2575.
(19) Compound 17A was insoluble in the assay conditions and its antiproliferative activity could not be determined.
(20) Oliván-Viguera, A., Valero, M. S., Murillo, M. D., Wulff, H., García-Otín, A. L., Arbonés-Mainar, J. M., Köhler, R. Novel phenolic inhibitors of small/intermediate-conductance $\mathrm{Ca}^{2+}$-activated $\mathrm{K}^{+}$channels, $\mathrm{KCa3} .1$ and KCa2.3. PLoS One 2013, 8, e58614.
(21) Residual solvent signals set according to Fulmer, G. R.; Miller, A. J. M.; Sherden, N. H.; Gottlieb, H. E.; Nudelman, A.; Stolz, B. M.; Bercaw, J. E.; Goldberg, K. I. NMR Chemical Shifts of Trace Impurities: Common Laboratory Solvents, Organics, and Gases in Deuterated Solvents Relevant to the Organometallic Chemist. Organometallics 2010, 29, 2176-2179.

