

1 **Extraction and trapping of truffle flavoring compounds into food**
2 **matrices using supercritical CO₂**

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13 **Keywords:** *Tuber melanosporum*, supercritical fluids, response surface methodology,
14 trapping material, fatty acids, sterols.

15 **Abbreviations:** SFE, supercritical fluid extraction; RSM, response surface methodology;
16 UHPSFC-QTOF-MS, ultra-high-performance supercritical fluid chromatography
17 coupled to quadrupole- time-of-flight mass spectrometry.

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22 **Abstract**

23 A supercritical fluid extraction methodology was used to extract flavoring and bioactive
24 compounds from truffles. Some parameters such as CO₂ flow rate (1-3 mg/mL),
25 extraction time (15-90 min) and different trapping food matrices (grape seed oil, gelatin,
26 agar agar and water) were optimized using response surface methodology to enhance
27 extraction and trapping yields. The optimal conditions (2.27 mg/mL CO₂ flow rate, 82.5
28 min when using 40 °C and 30 MPa, with 1mL grape seed oil as trapping matrix) obtained
29 with *Tuber melanosporum* were applied to three different truffle species: *Terfezia*
30 *claveryi*, *Tuber aestivum* and *Tuber indicum*. A total of 32 metabolites were profiled in
31 the extracts using ultra-high-performance supercritical fluid chromatography coupled to
32 quadrupole time-of-flight mass spectrometry. Compounds such as brassicasterol ergosta-
33 7,22-dienol, oleic and linoleic acid were found at similar amounts in all the extracts but
34 other molecules (*e.g.* fungal sterols) showed a particular distribution depending on the
35 specie studied and whether a trapping matrix was used at the SFE outlet.

36

37 **1. Introduction**

38 Truffles are considered one of the world's most highly prized and valued foods mainly
39 due to their organoleptic properties. Recently, they are also being investigated because of
40 their bioactive compounds with potential to induce beneficial effects on human health
41 (Khalifa et al., 2019; Patel, Rauf, Khan, Khalid, & Mubarak, 2017). Many of the studies
42 were focused on large molecules such as β -glucans (Tejedor-Calvo, Morales, Marco, et
43 al., 2020), however, truffles also include approximately 5–8 % lipids (Harki, Bouya, &
44 Dargent, 2006). Although lipids are present in lower concentrations than polysaccharides,
45 many of them showed interesting biological activities besides their main role in the
46 characteristic truffle flavor (N. Shah, Usvalampi, Chaudhary, Seppänen, & Sandesh,
47 2020)).

48 In *Tuber* species, unsaponifiable lipids such as ergosterol (ergosta-5,7,22-trienol) and
49 brassicasterol (ergosta-5,22- dienol) are the main sterols in the hyphal membranes
50 representing approximately respect. 60 and 30 % of total sterols (Harki, Klæbe, Talou,
51 & Dargent, 1996; Tejedor-Calvo, Morales, García-Barreda, et al., 2020; Tejedor-Calvo
52 et al., 2019), whereas in *Terfezia* genus, brassicasterol content reaches 98 % (Weete,
53 Kulifaj, Montant, Nes, & Sancholle, 1985). Other minor sterols were also detected in
54 *Tuber* sp., such as cholesterol, 5-dihydroergosterol, campesterol, 24(28)-
55 dehydroergosterol, stigmasterol, stigmasta-7,24(28)-dienol, fungisterol, lanosterol, β -
56 sitosterol, 24-methylene-24,25-dihydro- drolanosterol and 4 α -methylergosta-8(9),24(28)-
57 dienol) (Tang, Li, & Tang, 2012). Some of these compounds were associated with a wide
58 range of health-promoting properties, such as antioxidant, anti-inflammatory,
59 hypolipidemic, hypocholesterolemic and hypoglycemic activities (Gil-Ramírez et al.,
60 2013; Roncero-Ramos & Delgado-Andrade, 2017). Additionally, they also contribute to
61 the truffle flavor as they are precursors of the C19-sterols that generate the characteristic
62 boar meat smell (e.g. 5 α -androst-16-en-3 α -ol) (Wang, Li, Li, & Tang, 2008).

63 Apart from sterols, other biologically active compounds such as tocopherols and
64 unsaturated fatty acids were studied in detail for edible mushrooms, but only scarcely in
65 the truffle species. Shah et al (2020) found palmitic (C16:0), stearic (C18:0), oleic (C18:1)
66 and linoleic (C18:2) acids as main fatty acids in truffles and only traces of polyunsaturated
67 fatty acids (PUFAs) (N. N. Shah, Hokkanen, Pastinen, Eljamil, & Shamekh, 2020). These
68 fatty acids were detected in several *Tuber* (*T. aestivum*, *T. brumale*, *T. indicum*, *T.*
69 *gennadii*, *T. magnatum*, *T. melanosporum*, and *T. oligospermum*) and *Terfezia* species (*T.*
70 *arenaria*, *T. leptoderma*, *T. magnusii*) together with lower levels of shorter fatty acids,
71 such as C6:0, C8:0, C10:0, C12:0, C14:0, C14:1, C15:0, C15:1 (Tejedor-Calvo, Amara,
72 et al., 2020). Moreover, the fatty acid profile can be used to distinguish between truffles
73 from different geographical regions (Hilszczańska et al., 2016; N. N. Shah et al., 2020;
74 N. Shah, Usvalampi, Chaudhary, Seppänen-Laakso, et al., 2020) showing different and
75 characteristic flavors. In fact, a new taste known as ologustus or pinguis was associated
76 with activation of gustatory neurons by some fatty acids (Reed & Xia, 2015; Running,
77 Craig, & Mattes, 2015).

78 Supercritical fluid extraction (SFE) is frequently used to obtain lipid enriched fractions.
79 SFE is an environmentally friendly technology that leaves no traces of organic solvents
80 in the extracted material. This technique was recently utilized to obtain aromatic extracts
81 from truffles (Tejedor-Calvo et al., 2021), however, the obtained lipid fraction was not
82 examined in detail and was not associated to truffle flavors. Therefore, in this work, SFE
83 methods were optimized and applied to three different truffles species and the lipids
84 obtained in their extracts were identified and relatively quantified. Depressurization was
85 carried out through different food grade matrices as an attempt to enhance extraction yield
86 trapping aromatic volatiles.

87

88 **2. Materials and methods**

89 *2.1. Biological material*

90 *Tuber melanosporum* (Vittad.) and *Tuber aestivum* ascocarps were collected at Gúdar-
91 Javalambre woods (Teruel, Spain) and *Tuber indicum* was supplied by Espora Gourmet
92 company (Soria, Spain). *Terfezia claveryi* Chatin was collected from an experimental
93 field in Caravaca de la Cruz (Murcia, Spain). Truffles (20 units/species) were
94 taxonomically authenticated by morphological features (Montecchi & Sarasini, 2000;
95 Rioussset, 2001), selected and processed under refrigeration (Rivera, Venturini, Marco,
96 Oria, & Blanco, 2011). Truffles were lyophilized, ground and sieved to obtain a particle
97 size lower than 0.5 mm and stored at -20 °C until further use.

98 Grapeseed oil was purchased from Dietisa company (Barcelona, Spain). Food grade agar-
99 agar and powder gelatine were purchased from a local supermarket. The agar-agar
100 proximal composition was 85% fiber, <0.5% carbohydrates, <0.5% lipids, 1.3% proteins,
101 0.7% salt. Gelatin package indicated 89% protein and 1.2% salt in its nutritional table.

102

103 *2.2. Reagents*

104 Methyl *tert*-butyl ether (MTBE; 99.8%), MS-grade ammonium formate and
105 ammoniumacetate were provided by Sigma-Aldrich (Steinheim, Germany). Hexane
106 (95%) was obtained from LAB-SCAN (Gliwice, Poland). LC-HPLC grade methanol and
107 water LC-MS grade was provided by VWR (VWR Chemicals, France). MilliQ water was
108 produced using a Milli-Q purification system (Millipore, Billerica, MA). Ultrapure CO₂
109 was provided by AGA industrial gases (Lidingö, Sweden). All other solvents were of
110 analytical grade.

111

112 *2.3. Supercritical fluid extraction (SFE)*

113 An analytical SFE system (Waters MV-10, Milford, MA) including a fluid delivery
114 module for pumping CO₂, an oven for placing the extraction vessels, an automated back
115 pressure regulator, a pump, and a fraction collector module was used to conduct the
116 experiments. The heads of the CO₂ pump were cooled using a chiller operated at 4 °C
117 (Figure 1). Samples (1 g) were loaded together with glass beads into a 10 mL stainless
118 steel extraction vessel. After each extraction the system was flushed with a CO₂ flow for
119 5 min. The system was controlled by ChromScope™ software (Waters, Milford, MA).
120 The collected extracts were stored at -20 °C before further analysis.
121 Two face centered central composite designs (FC-CCDs) with three center points were
122 created in MODDE 10.1 (Sartorius Stedim Biotech, Malmö, Sweden). Extraction time
123 (15-90 min) and flow rate (1-3 mL/min) were selected as independent factors, with
124 extraction yield as response variable. Pressure and temperature were kept at 300 bar and
125 40 °C, respectively (Tejedor-Calvo et al., 2021). The design was adjusted to maximize
126 the extraction yield, considering flow rate and time being of equal importance. In total,
127 12 experiments were performed in a random order (Table 1). All extracts were recovered
128 in a glass bottle (10 mL) by inserting the SFE outlet into a trapping solution of grapeseed
129 oil (OIL), agar-agar (AA), gelatin (GE) or MilliQ water (W). The AA (2.5 mg/mL) and
130 GE (20 mg/mL) solutions were prepared by heating them at 100 °C for 10 min. Different
131 matrix volumes were tested (1, 3 and 5 mL). The optimal conditions providing the highest
132 extraction yields were defined by using graphical and numerical analyses based on
133 response surface plots and a desirability function.

134

135 *2.4. Lipid extraction*

136 Lipid extraction was carried out as described by Matysah et al. (2008) (Matyash, Liebisch,
137 Kurzchalia, Shevchenko, & Schwudke, 2008). Briefly, each SFE extract or residue

138 (material remaining in the extraction cell after SFE; approximately 40 mg) was mixed
139 with 1300 μ l of MTBE:MeOH (10:3, v/v) and stirred for 1 h at room temperature. Then,
140 MS-grade water (250 μ l) was added to induce phase separation. After 10 min incubation
141 at room temperature, the sample was centrifuged at 1,200 g for 10 min. The upper
142 (organic) phase was collected, and the lower phase was mixed with 1300 μ l of a solvent
143 mixture equivalent to the expected composition of the initial upper phase
144 (MTBE/methanol/water (10:3:2.5, v/v/v)). Afterward, the solution was again stirred, the
145 upper phase collected, pooled together with the first extraction and dried in a vacuum
146 centrifuge. To facilitate evaporation, MS-grade methanol (200 μ l) was added to the
147 organic extract after 25 min of vacuum centrifugation. Extracted lipids were dissolved in
148 200 μ l CHCl₃/methanol/water (60:30:4.5, v/v/v) and stored at -80 °C until analysed.

149

150 2.5. UHPSFC/ESI-QTOF-MS

151 Lipid extracts were analysed using an ultrahigh performance supercritical fluid
152 chromatography system (UHPSFC; ACQUITY UPC2, Waters, Milford, MA) using an
153 ACQUITY UPC2 Torus DIOL column (130 Å, 1.7 μ m, 3 mm x 100 mm, Waters)
154 equipped with a Torus DIOL (130 Å, 1.7 μ m, 2.1 mm x 5 mm, Waters) guard column.
155 The column temperature and the active back pressure regulator (ABPR) were set at 50 °C
156 and 110 bar, respectively. The injection volume was 1 μ L and the flow rate was set at 1.6
157 mL/min. Methanol containing 20 mM ammonium formate was used as mobile phase
158 modifier using the following gradient: 0 min, 1% modifier; 18 min, 50%; 19 min, 50%;
159 19.5 min 1%; 21 min 1%.

160 The UHPSFC was connected to a Xevo 2G QTOF-MS (Waters). Two T-pieces (Waters)
161 were used to control the back pressure at the column outlet and to infuse methanol (0.25
162 mL/min) as a makeup liquid. The MS was operated in positive and negative electrospray

163 ionization (ESI) mode with a scan range of m/z 80–1200. In positive ESI mode, the
164 capillary voltage was 3.2 kV, the sampling cone voltage 34 V, the source temperature 120
165 °C, the drying gas temperature 420 °C, the cone gas flow 40 L/h and the drying gas flow
166 690 L/h. In negative ESI mode, the capillary voltage was 2.6 kV, the sampling cone
167 voltage 42 V, the source temperature 120 °C, the drying gas temperature 390 °C, the cone
168 gas flow 40 L/h and the drying gas flow 740 L/h. A collision energy ramp between 15
169 and 55 eV was used to acquire data independent, targeted and data dependent MS/MS
170 data in both ESI modes. Data was acquired using MassLynx v4.1 (Waters). Raw data
171 were processed using the open-source software package MZmine 2.16
172 (<http://mzmine.sourceforge.net/>) [31]. Data were generated by targeted peak detection
173 with a m/z tolerance of 0.5 mDa or 5 ppm and a retention time tolerance of 0.2 min.
174 Tentative compound identification was carried out by comparing experimental with
175 theoretical MS data in negative mode using the exact mass of the [M-H]⁻ ion
176 (<http://www.lipidmaps.org>). Moreover, MS/MS data were compared with spectra in
177 Massbank (<https://massbank.eu/MassBank/>), CRC, Wiley and reference literature. The
178 data presented for tentatively identified compounds include the name of the compounds,
179 CAS number (sterols) or common abbreviation (fatty acids), molecular formula, retention
180 time, and theoretical and experimental m/z data (Table 2). Blank samples, produced from
181 extraction of pure trapping matrices, were analysed using the same method.

182

183 2.6. *Statistical analysis*

184 Differences between groups were evaluated using one-way analysis of variance
185 (ANOVA) followed by Tukey's multiple comparison test *post hoc* in GraphPadPrism
186 version 5.01 (GraphPad Software, San Diego, CA). Experimental designs were evaluated

187 using multiple linear regression in Statgraphics Centurion XVI (Statpoint Technologies,
188 Warrenton, VA).

189

190 **3. Results and discussion**

191 *3.1. SFE method and relative lipid profile*

192 Firstly, *Tuber melanosporum* was submitted to SFE extractions and according to the
193 obtained results (Table 1), the optimal extraction conditions generating the highest
194 extraction yield (1.44%), was 2.3 mL/min CO₂ flow rate and 82.5 min (Figure 2a). A low
195 flow rate (1 ml/min) yielded the lowest desirability (below 0.7) whereas the highest was
196 found with long extraction times (50 min) and an intermediate flow rate (2-3 mL/min).
197 Time, more than flow rate, was positively influencing the extraction yields (Figure 2b).
198 Same effect was shown in a study using similar SFE conditions in sunflower oil. They
199 also described that the extraction rate was linearly associated with time in the early
200 extraction stages (Salgin, Döker, & Çalimli, 2006). CO₂ extraction abilities depends
201 mainly on parameters such as pressure and temperature therefore by modulating SFE
202 conditions, CO₂ can selectively extract several compounds depending on their specific
203 solubility in this solvent (Salgin et al., 2006). Thus, depending on the targeted
204 compounds, the extraction processes show different extraction yields, extracts
205 compositions etc.

206 Compare to previous studies, the extraction yield was slightly lower (2.1 %) than those
207 obtained by Tejedor-Calvo et al. (2021) for *T. melanosporum* (Tejedor-Calvo et al.,
208 2021). Similar yields were reported for edible mushrooms, such as *Agaricus bisporus*
209 (1.4-2.1%) (Gil-Ramírez et al., 2013) and *Lentinula edodes* (1.1-1.7%) (Morales, Piris,
210 Ruiz-Rodriguez, Prodanov, & Soler-Rivas, 2018). Differences between truffle batch, SFE
211 parameters (CO₂ flow rate), scale and device designs might be the reason for these

212 differences.

213 A total of 32 compounds were detected in the extracts from *T. melanosporum*. Most of
214 the compounds were fatty acids, including DHA and EPA, and fungal sterols (Figure 3).
215 Surprisingly, compounds not previously described in truffles, such as brassicasterone
216 (ergosta-4,22-dien-3-one), ergone (ergosta-4,6,8(14),22-tetraen-3-one, (22E)-),
217 fecosterol (24-methylene-cholest-8-en-3 β -ol) and demosterol (3 β -cholesta-5,24-dien-3-
218 ol) were also detected.

219 Some fatty acids might be linked to truffle flavor and smell since many aromatic
220 compounds are produced downstream within the fatty acid metabolic pathways. For
221 instance, the odor generated when a food containing linolenic acid is decomposing is not
222 only ascribed to the fatty acid oxidation, but also to the smell reduction noticed when the
223 fatty acid is transformed into carbonyl compounds (with low odor threshold values), such
224 as (Z)-3-hexenal, (E,Z)-2,6-nonadienal and (Z)-1,5-octadien-3-one (Anklam, 2005).
225 Moreover, food products containing linoleic acid usually include other strong flavoring
226 compounds such as (E)-2-nonenal, trans-4,5-epoxy-(E)-2-decenal and 1-octen-3-one
227 (Anklam, 2005). Aldehydes with exceptionally strong aromas can be produced in foods
228 by the autoxidation of some fatty acids and yield an exceptionally strong smell even when
229 present in very low amounts. An example is octadeca-(Z,Z)-11,15-dienoic acid (the
230 precursor for (Z)-4-heptenal) that can be found in beef and mutton and often in butter
231 (Anklam, 2005). Some of these aromatic compounds were described in truffles species
232 and in truffle extracts obtained using SFE (Campo, Marco, Oria, Blanco, & Venturini,
233 2017; Tejedor-Calvo et al., 2021). But no reports were found correlating flavor with lipid
234 compounds of fresh truffle.

235 A heat map was constructed to illustrate the lipid patterns and their variation within the
236 extractions carried out using *T. melanosporum* (Figure 3). Among the identified sterols,

237 brassicasterol, ergone, ergosta-7,22-dienol and fecosterol were present in the obtained
238 extracts in high levels, whereas desmosterol was more abundant in the SFE residues.
239 Although ergosterol is the main constituent of fungal hyphae (Harki et al., 1996), it was
240 detected in low percentage as compared to the other sterols. However, in other SFE
241 studies (Tejedor-Calvo et al., 2021) this compound was the main contributor to the
242 obtained extracts. The different extraction conditions utilized might modulate the CO₂
243 viscosity enhancing the preferential extraction of other compounds before ergosterol.
244 Moreover, only traces of desmosterol were extracted with SFE, with the larger amounts
245 of this compound remaining in the truffle residue. The solubility of desmosterol decreased
246 with increasing polarity of the solvents. This sterol was reported to be more soluble in
247 ethanol and acetone than in strongly polar solvents (Chen, Su, Xing, Yang, & Ren, 2008)
248 and supercritical CO₂ resembled polarities close to polar solvents (dielectric constant 1-
249 .0-1.4) (Leeke et al., 2005; Raveendran, Ikushima, & Wallen, 2005) being then not the
250 most convenient solvent to extract it.

251 Extracts and residues showed a similar fatty acid profile. Oleic (C18:1) and linoleic
252 (C18:2) acid were found as the main compounds in all the obtained fractions
253 independently of the extraction conditions tested, suggesting that no preferential
254 extraction of certain fatty acids was promoted with any condition, they can be extracted
255 but not all the content (Table 3). The fatty acid profile was similar to that observed in
256 previous studies (Shah et al., 2020; Tejedor-Calvo, Amara, et al., 2020) for a raw truffle
257 suggesting that the SFE extraction did not significantly modify the lipid composition.
258 Lipids containing oleate and linoleate groups were also pointed by other authors (Shah et
259 al., 2020; Tang, Li, Li, Wan, & Tang, 2011; Tejedor-Calvo, Amara, et al., 2020) as main
260 truffle constituents. Hence, SFE seems to be a promising technique for production of
261 extracts with high content of bioactive omega-9 and -3 fatty acids. These unsaturated fatty

262 acids are involved in many health promoting and disease preventing processes (Calder,
263 2013; Guo, 2009).

264 The CO₂ flow rate was the only parameter that induced significant alterations in extraction
265 yield. An increase in the flow rate (E10, E11 and E12 figure 3) was beneficial for the
266 extraction of some sterols (*i.e.* brassicasterol, ergone, fecosterol, ergosterol,
267 brassicasterone) and fatty acids (*i.e.* hexadecenoic, stearic and oleic acids, DHA and
268 11,14-eicosadienoic acid). In SFE, the extraction yield and mass transfer coefficient
269 increase with higher CO₂ flow rate (Salgin et al., 2006). Therefore, a high CO₂ flow rate
270 generally extract a larger number of compounds. Nevertheless, a lower CO₂ flow rate
271 could be particularly used to obtain extracts enriched in *e.g.* erogsta-7,22-dienol.

272

273 3.2. Influence of trapping matrices

274 After the SFE method was optimized, different matrices were tested to trap aromas and
275 flavoring compounds to prevent their release into the atmosphere during CO₂
276 depressurization. Four matrices with different chemical composition were selected
277 covering the most frequent molecules found in food matrices to test different binding or
278 scavenging mechanisms such as a high fat, high protein, high polysaccharide and aqueous
279 based matrices. Results indicated that a higher trapping capacity was observed when a
280 low quantity of trapping matrix was used (Figure 3). The matrices tested showed
281 significantly different trapping capacities. The highest yield was obtained using GE (1.42
282 %) followed by OIL (1.16 %); W and AA showed the lowest trapping ability (Figure 4a).
283 Polysaccharides present in the AA were not able to trap lipophilic compounds, being even
284 less effective than a water matrix. Guichard (2002) studied the interactions between
285 flavoring compounds and food ingredients and demonstrated that the physical state of
286 carbohydrates is an important parameter influencing flavor retention (E. Guichard, 2002).

287 In this work, the AA viscosity was low (liquid) due to the continuous supply of CO₂.
288 Whether a higher viscosity would benefit trapping of lipids remain unknown. Matrices
289 including triglycerides, proteins, polysaccharides and their combinations forming
290 emulsions, can be designed to fine-tune trapping of a wide range of molecules (E.
291 Guichard, 2002; Elisabeth Guichard, 2006; Mao, Roos, Biliaderis, & Miao, 2017).
292 Designing of a specific matrix combination is challenging, being the lipid composition
293 the factor highly influencing the affinity of the other food molecules, such as proteins, to
294 trap volatiles (Roberts, Pollien, & Watzke, 2003; Seuvre, Espinosa Díaz, & Voilley,
295 2000). Moreover, synergistic and antagonistic effects might also play a major role.
296 Under the conditions tested, the number of total sterols collected was independent of the
297 trapping matrix, although the contribution of each individual sterol to the total content
298 slightly differed (Figure 5a). For instance, the SFE extract from *T. melanosporum*
299 collected using water as trapping matrix contained almost 50% ergosta-7,22-dienol,
300 values that were significantly higher than the amounts collected without any matrix.
301 However, ergone levels were higher when matrices were used, except water. When
302 grapeseed oil was used, a larger variety of compounds were detected, with a lipid profile
303 differing from that observed in absence of matrix. The observed differences may be due
304 to transformations of some sterols into new derivatives via interactions with the trapping
305 matrix. The fatty acid content was also highly influenced by the trapping matrix. Oleic
306 and linoleic acid, followed by hexadecenoic acid, remained the most abundant fatty acids
307 in the SFE extracts of *T. melanosporum* without trapping material (Table 3). When GE
308 and AA were used as trapping matrix a completely different lipid profile was observed
309 since behenic and linoleic acid were present in high concentrations and oleic acid was
310 lacking. Presumably, GE proteins or AA polysaccharides might easily bind or trap long
311 chain fatty acids but with particular stearic conformation. OIL was the trapping material

312 showing the highest extraction yield with linoleic acid contributing to almost 60% of the
313 total extract. Water showed the widest selectivity and was able to concentrate
314 hexadecenoic acid, stearic acid, vaccenic acid, 11,14-eicosadienoic acid, cis-erucid acid
315 and lignoceric acid, as compared with extractions conducted in absence of trapping
316 matrix. These results were surprising, as water provides the most hydrophilic trapping
317 solvent. However, it is possible that the hydrophobic extracts were suspended in water
318 forming micelles or other colloidal structure.

319

320 3.3. SFE extraction and trapping of flavoring compounds from other truffle species

321 The optimal extraction conditions defined using *T. melanosporum* were applied to
322 additional truffles species as it was previously indicated (Tejedor-Calvo et al., 2021) that
323 SFE yields and general compound contents of the obtained extracts were similar
324 (including *Terfezia* species)(Tejedor-calvo, Garc, & Sergio, 2022). Extractions were
325 conducted using grapeseed oil as trapping matrix, as this matrix retained most of the
326 flavoring compounds from *T. melanosporum* and only marginally altered its lipid
327 composition. Hence, SFE extractions of *Tuber aestivum*, *Tuber indicum* and *Terfezia*
328 *claveryi* were carried out with and without the oil at the depressurization outlet.

329 The extraction yields ranged from 1.0 to 1.3% and were independent of placing or not
330 trapping oil (Figure 4b). The extraction yield from *T. claveryi* was higher compared to
331 the other *Tuber* truffles, but lower than those observed for *T. melanosporum* (1.4%). The
332 use of oil at the outlet modified the observed sterol profile for all 3 species (Figure 5b).
333 In the selected species but particularly in *T. claveryi*, a notable increase in the content of
334 ergone concomitant with a reduction of ergosterol and cholesterol was noticed in presence
335 of oil compared to when it was absent. *T. aestivum* extracts obtained without oil in the trap
336 showed a wide variety of sterols, with some of them such as ergoste or stigmasterol being

337 lower in concentration as compared to when oil was used. The sterol profile of the extract
338 from *T. indicum* was less affected by the presence of oil, with only levels of cholesterol
339 and fecosterol being reduced when trapping matrix was used.

340 When changes in the fatty acid profile of the studied truffles were evaluated, one of the
341 most dramatic changes was noticed in the *T. indicum* extracts. In this truffle presence of
342 oil in the trap increased the yield of behenic acid almost 8-fold compared to when the trap
343 was left empty. In addition, several minor changes in the fatty acid profile were also
344 observed. For instance, levels of lauric acid, myristic acid, 14Z-eicosenoic acid, and cis-
345 erucid acid slightly increased in the presence of oil in the trap. Conversely, levels of
346 hexadecanoic acid, palmitoleic acid, margaric acid, stearic acid, linoleic acid, and DHA
347 decreased in presence of oil. The latter compound is worldwide known as responsible for
348 several important biological functions, including cell membranes structure, protein
349 function, production of lipid mediators and gene expression, thereby potentially
350 contributing to improve human health (Calder, 2018; Guo, 2009; Sande et al., 2019).

351 Moreover, behenic acid is able to bind glycerol-based lipid backbones impairing
352 pancreatic lipase activity and therefore preventing obesity (Moreira, Santos, Gambero, &
353 Macedo, 2017; Silva et al., 2020). Hence, the use of grapeseed oil as trapping material
354 enables the production of SFE extracts with different fatty acid profiles than without it
355 modulating their different bioactivities and therefore their potential applications.

356 Moreover, the fatty acid profile can also be related to the aroma compounds and their
357 perception (Roberts et al., 2003). Some of these odorous compounds such as linoleic and
358 oleic were detected in SFE extracts from black truffle (*T. melanosporum*) (Tejedor-Calvo
359 et al., 2021) as well as in the other selected species and they were/were not modified when
360 extracted with oil. Apparently, C8 volatiles, such as 3-octanone, 1-octen-3-one, 3-octanol,
361 Z-5-octen-1-ol, 1-octen-3-ol were assigned as potential signaling molecules produced by

362 truffle mycelium and fruiting bodies (Splivallo, Bossi, Maffei, & Bonfante, 2007). Some
363 of these compounds were pointed as mushroom-like aroma and considered as major
364 contributors to the characteristic flavor of several mushrooms and truffles, including *T.*
365 *indicum*. These compounds are mainly formed by the oxidation of linoleic and linolenic
366 acid in the presence of enzymes, such as lipoxygenase and hydroperoxidelyase (Assaf,
367 Hadar, & Dosoretz, 1997). Those enzymes play a key role in fruits, vegetables and
368 mushrooms, releasing some odorous compounds like the green-grassy or cucumber-like
369 smelling aldehydes hexanal, (Z)-3-hexenal (leafy aldehyde), (Z,Z)-3,6-nonadienal or
370 mushroom like of 1-octen-3-ol (Anklam, 2005). In the obtained extracts, enzymatic
371 reactions might not be involved because the high pressure and temperature utilized for
372 extraction will denature them however, during depressurization, the extract is released
373 sprayed into the air and it might enhance its contact with oxygen promoting autoxidation
374 or transformation of small volatile molecules such as aromas, that might also be the reason
375 that when depressurization is carried out inside a lipid matrix (with less access to oxygen)
376 the extract composition was modified. Therefore, the trapping matrices might also be used
377 to modulate the aromatic profile of the obtained extracts however, more studies are
378 needed to explore all the potential combinations.

379 **4. Conclusions**

380 Supercritical fluid technology allowed an efficient generation of extracts enriched in lipid
381 molecules involved in truffle aroma and flavor. The optimal conditions (2.27 mg/mL CO₂
382 flow rate and 82.5 min extraction) were able to extract fatty acids (mainly oleic and
383 linoleic acid) and sterols (brassicasterol, ergosta-7,22-dienol and fecosterol) from *Tuber*
384 *melanosporum* truffles. The use of trapping matrices such as grapeseed oil placed at the
385 SFE outlet enhanced capture of some key truffle aromatic molecules but also modify their
386 content of sterols and fatty acids. These results were also obtained when other truffles

387 species were submitted to similar extractions generating their own particular profile of
388 flavoring and beneficial compounds depending on the specie considered and the use or
389 not of a trapping matrix. These results indicated that environmentally friendly extraction
390 and trapping methodologies can be used to produce tailor-made natural truffle aromatic
391 and functional extracts. At the present, further studies are being carried out to test their
392 food stability before commercialization.

393

394 **Declaration of Competing Interest**

395 The authors declare that they have no known competing financial interests or personal
396 relationships that could have appeared to influence the work reported in this paper.

397 **CRediT authorship contribution statement**

398 **Eva Tejedor-Calvo:** Conceptualization, Investigation, Formal analysis, Data curation,
399 Methodology, Writing - original draft. **Pedro Marco:** Supervision, Validation, Writing -
400 review & editing. **Peter Spégel:** Software, Supervision, Visualization, Writing - review
401 & editing. **Cristina Soler-Rivas:** Visualization, Writing - review & editing.

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573

574 **TABLES**

575 **Table 1.** Experimental design for SFE extractions from *T. melanosporum* and obtained
576 extraction yield

Independent factors				Response
N	Run order	Time (min)	Flow (ml/min)	Yield (%)
1	4	15	1	0.85
2	7	52.5	1	1.17
3	8	90	1	1.17
4	9	15	2	0.94
5	3	52.5	2	1.35
6	5	52.5	2	1.35
7	11	52.5	2	1.44
8	12	52.5	2	1.35
9	10	90	2	1.44
10	1	15	3	1.13
11	6	52.5	3	1.35
12	2	90	3	1.44

577

578 **Table 2.** Lipid compounds identified in the SFE extract obtained from *T. melanosporum*
 579 using the optimized conditions (40 °C, 2.27 ml/min CO₂ and 300 bar). Experimental and
 580 theoretical data for sterols in negative [M-H]⁻ are shown. *CAS Numbers are provided for
 581 sterols and abbreviations for fatty acids. RT correspond to retention time.

Compound identity			Negative ESI [M-H] ⁻		
Common name	Other information*	Molecular formula (neutral)	RT (min)	m/z experimental	m/z theoretical
<i>Sterols</i>					
Brassicasterol	474-67-9	C ₂₈ H ₄₆ O	2.08	395.3353	395.3319
Ergone	19254-69-4	C ₂₈ H ₄₀ O	2.32	397.2233	397.2257
Ergosta7,22-dienol	96391-64-9	C ₂₈ H ₄₄ O	2.32	395.2979	395.3392
Fecosterol	516-86-9	C ₃₀ H ₅₀ O	2.7	411.3095	411.3632
Cholesterol	57-88-5	C ₂₇ H ₄₆ O	3.47	395.3886	395.3079
Stigmasterol	83-48-7	C ₂₉ H ₄₈ O	3.54	445.3292	445.3967
Ergosterol	57-87-4	C ₂₈ H ₄₄ O	3.85	385.3476	385.2955
Desmosterol	313-04-2	C ₂₇ H ₄₄ O	4.4	383.3319	383.3523
Brassicasterone	4030-92-6	C ₂₈ H ₄₆ O	4.99	425.3789	425.3633
9,19-cyclolanostane-3,7-diol	91851-80-8	C ₃₀ H ₅₂ O	5.92	395.3319	395.3884
<i>Fatty acids (FA)</i>					
Caprylic acid	C8:0	C ₈ H ₁₆ O ₂	2.22	143.1078	143.0692
Capric acid	C10:0	C ₁₀ H ₂₀ O ₂	2.22	171.1391	171.6391
Lauric acid	C12:0	C ₁₂ H ₂₄ O ₂	2.22	199.1704	199.1311
Myristic acid	C14:0	C ₁₄ H ₂₈ O ₂	2.22	227.2017	227.1297
Hexadecanoic acid	C16:0	C ₁₆ H ₃₂ O ₂	2.08	255.2311	255.2304
cis-9-palmitoleic acid	C16:1	C ₁₆ H ₃₀ O ₂	2.11	253.2173	253.2159
Margaric acid	C17:0	C ₁₇ H ₃₄ O ₂	2.11	269.2486	269.2483
Stearic acid	C18:0	C ₁₈ H ₃₆ O ₂	2.11	283.2643	283.2607
Oleic acid	C18:1n9	C ₁₈ H ₃₄ O ₂	2.15	281.2486	281.2494
cis-vaccenic acid	C18:1n11	C ₁₈ H ₃₄ O ₂	2.60	281.2486	281.2476
Linoleic acid	C18:2n6	C ₁₈ H ₃₂ O ₂	2.22	279.2313	279.2321
alpha-Linolenic acid	C18:3n3	C ₁₈ H ₃₀ O ₂	2.57	277.2173	277.2169
14Z-eicosenoic acid	C20:1	C ₂₀ H ₃₈ O ₂	2.22	309.2799	309.2797
11,14-eicosadienoic acid	C20:2	C ₂₀ H ₃₆ O ₂	2.29	307.2643	307.2645
5,8,11-eicosatrienoic acid	C20:3n3	C ₂₀ H ₃₄ O ₂	2.32	305.2486	305.2498
Heneicosylic acid	C21:0	C ₂₁ H ₄₂ O ₂	2.74	325.3112	325.2372
EPA	C20:5n3	C ₂₀ H ₃₀ O ₂	2.39	301.2173	301.2170
Behenic acid	C22:0	C ₂₂ H ₄₄ O ₂	2.26	339.3269	339.3265
cis-erucic acid	C22:1n9	C ₂₂ H ₄₂ O ₂	2.29	337.3112	337.3117
DHA	C22:6n3	C ₂₂ H ₃₂ O ₂	2.22	327.2330	327.2158
Lignoceric acid	C24:0	C ₂₄ H ₄₈ O ₂	2.29	367.3582	367.3579
Nervonic acid	C24:1	C ₂₄ H ₄₆ O ₂	2.32	365.3425	365.8224

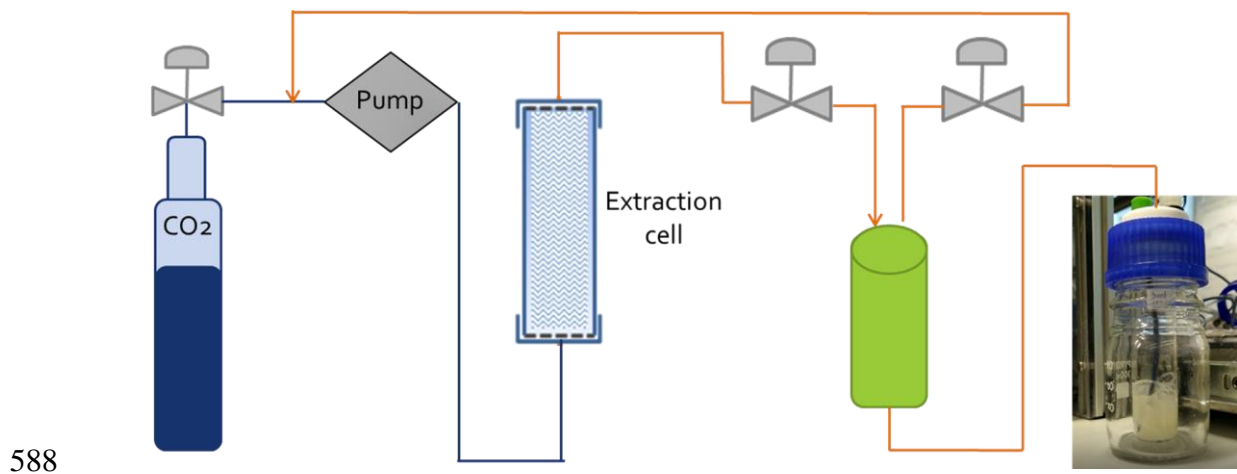
583 **Table 3.** Fatty acid composition of the SFE extracts from selected truffles: no trapping matrix (none), water (W), gelatin (GE), agar-agar (AA),
 584 grapeseed oil (OIL). Indicated results are expressed as percentage of each fatty acid from the total fatty acid content.

Truffle specie	<i>T. melanosporum</i>					<i>T. clavaryi</i>		<i>T. aestivum</i>		<i>T. indicum</i>	
	None	W	GE	AA	OIL	None	OIL	None	OIL	None	OIL
caprylic acid	0.07	0.06	1.58	0.94	0.00	0.46	0.94	0.37	1	0.8	0.8
capric acid	0.00	0.00	0.18	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00
lauric acid	0.12	0.11	0.84	1.25	0.11	0.53	1.25	0.56	1.24	0.71	0.61
myristic acid	0.03	0.04	0.54	0.74	0.08	0.21	0.74	0.17	0.43	0.4	0.54
hexadecanoic acid	4.27	5.66	0.66	0.64	1.55	2.1	0.64	1.37	0.55	3.69	0.78
cis-9-palmitoleic acid	0.48	0.90	0.63	0.12	0.19	0.64	0.12	0.62	0.21	0.52	0.43
margaric acid	0.18	0.26	0.38	0.16	0.09	0.22	0.16	0.19	0.17	0.42	0.5
stearic acid	3.04	8.19	0.38	5.43	1.55	0.99	0.43	0.92	0.33	0.96	0.88
oleic acid	29.23	18.85	10.20	11.53	29.37	35.12	38.64	29.34	28.23	21.34	30.3
vaccenic acid	1.72	8.73	6.02	3.00	0.68	0.31	0.00	3.29	0.00	4.89	0.00
linoleic acid	49.64	34.42	51.19	47.01	59.38	38.56	32.01	43.14	30.2	31.86	31.9
α -linoleic acid	1.74	1.87	1.37	0.26	0.43	0.16	0.26	1.15	0.25	2.66	0.76
14Z-eicosenoic acid	0.70	0.88	0.75	5.17	0.57	3.58	5.17	3.07	4.72	2.93	3.93
11,14-eicosadienoic acid	1.84	6.27	0.75	0.32	0.41	0.8	0.32	1.58	1.05	1.14	1.01
5,8,11-eicosatrienoic acid	0.08	0.46	0.30	0.13	0.04	0.03	0.13	0.1	0.33	0.6	0.5
heneicosylic acid	0.03	0.24	3.49	0.34	0.03	0.09	0.34	0.15	0.69	0.00	0.00
EPA	0.05	0.81	0.20	0.13	0.04	0.03	0.13	0.22	0.15	0.44	0.34
behenic acid	0.30	0.92	17.03	13.33	2.46	1.72	13.33	1.74	26.52	11.29	22.29
cis-erucid acid	0.31	1.37	1.35	0.41	0.11	0.35	0.41	0.35	0.41	0.62	0.72
DHA	5.22	4.96	0.91	8.64	2.70	13	4.53	10.92	2.12	8.24	3.34
lignoceric acid	0.93	4.99	0.89	0.41	0.20	1.07	0.41	0.73	1.27	6.32	0.21
nervonic acid	0.00	0.00	0.33	0.06	0.02	0.03	0.06	0.02	0.14	0.19	0.19

585

586 **FIGURES**

587 **Figure 1.** Supercritical fluid extraction and trapping system with oil as trapping matrix.



590 **Figure 2.** A) 2D response surface plot of SFE extractions from *T. melanosporum*. The
591 responses variable was the extraction yield. B) Effect of time (min) and flow rate (ml/min)
592 on the SFE extraction yield.

593 A)

594

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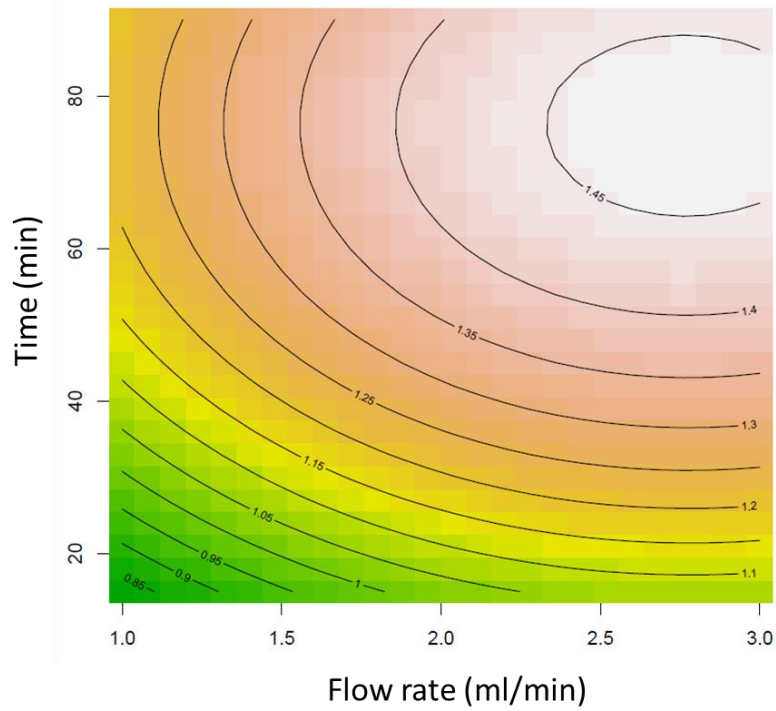
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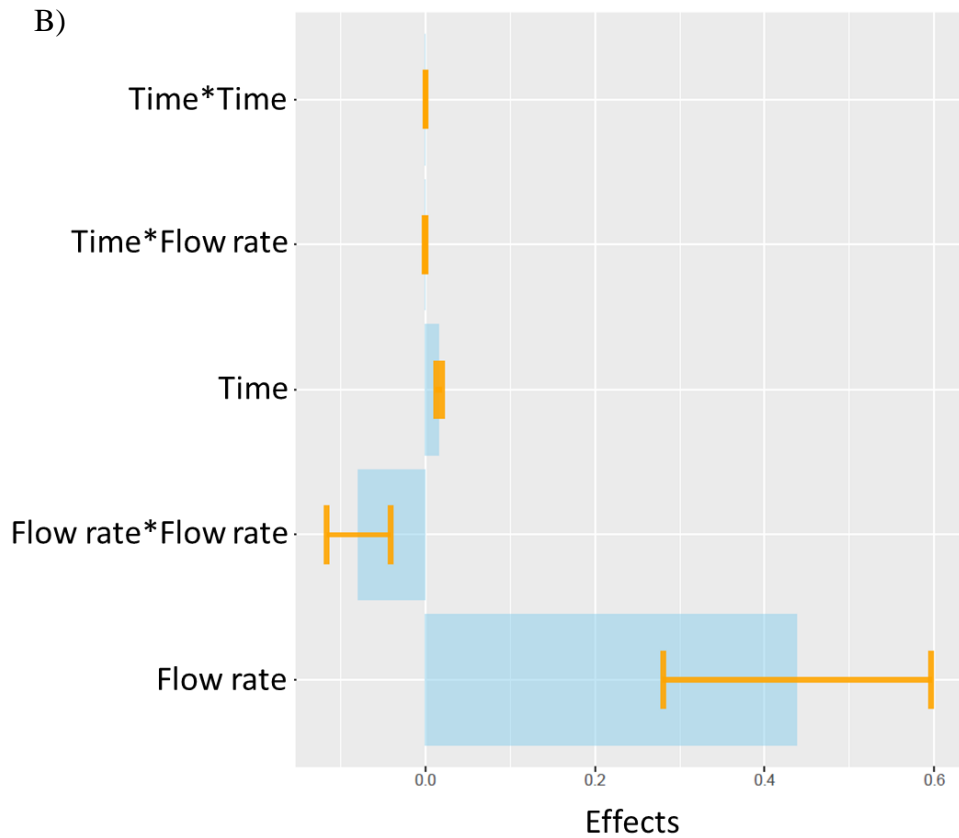
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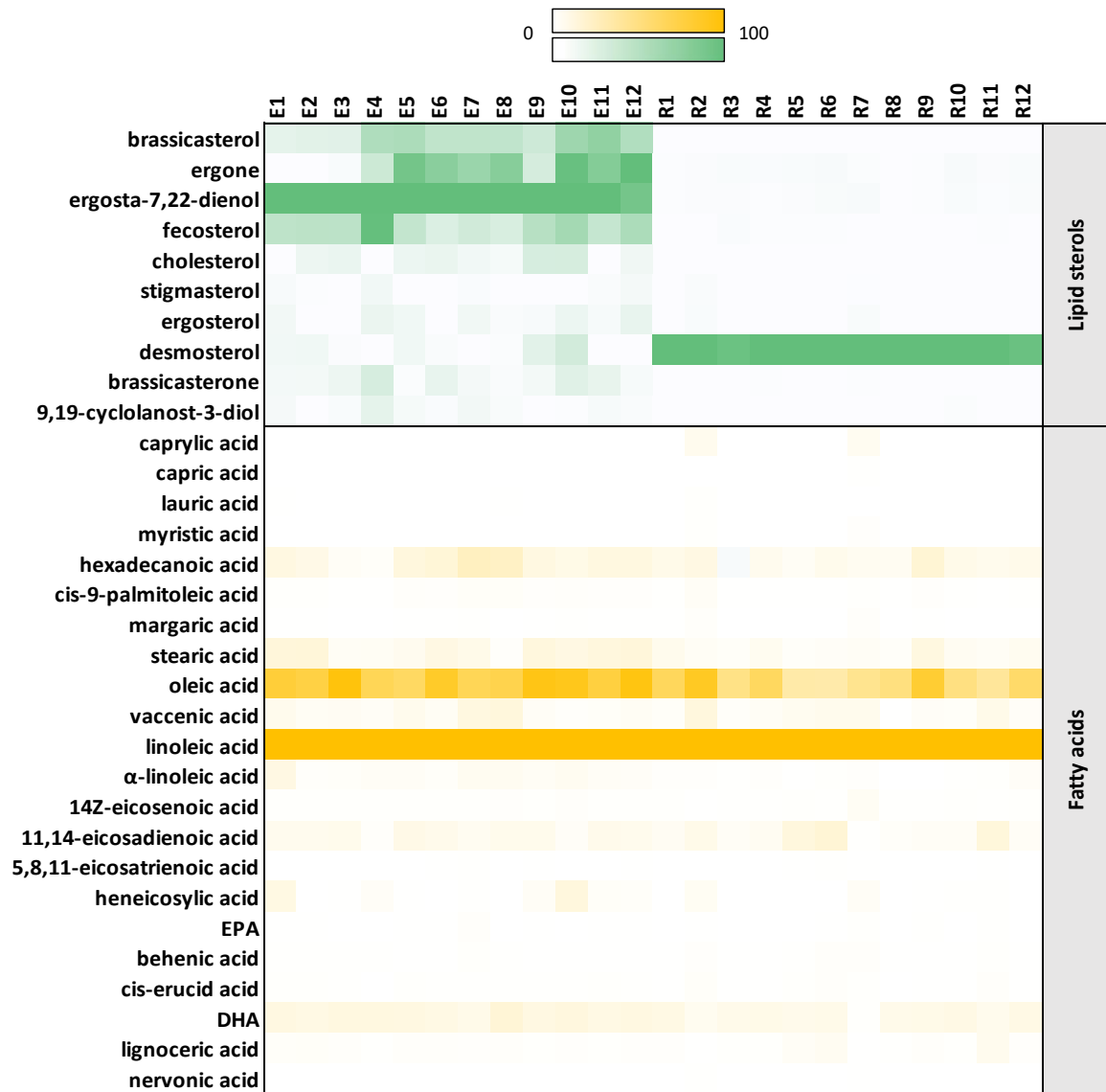
601 B)

602



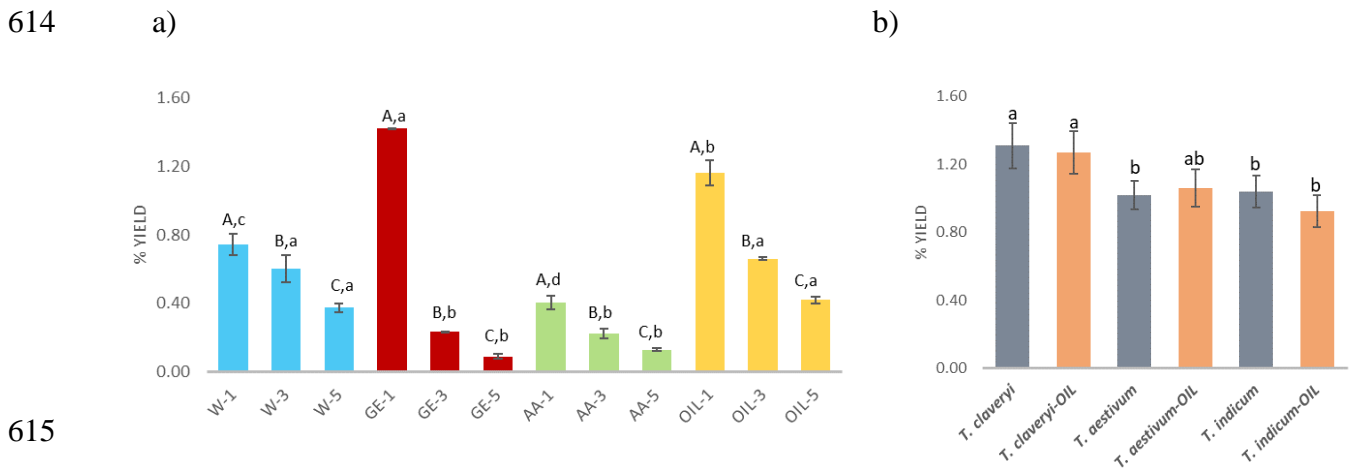
603 **Figure 3.** Fatty acids (yellow) and sterols (green) from the SFE extracts (E) and residues
 604 remaining in the extraction cell (R) obtained from *T. melanosporum* under the
 605 experimental conditions indicated in table 1. Color intensity refers to the percentage (0-
 606 100 %) of total sterol and total fatty acids respectively.

607



608

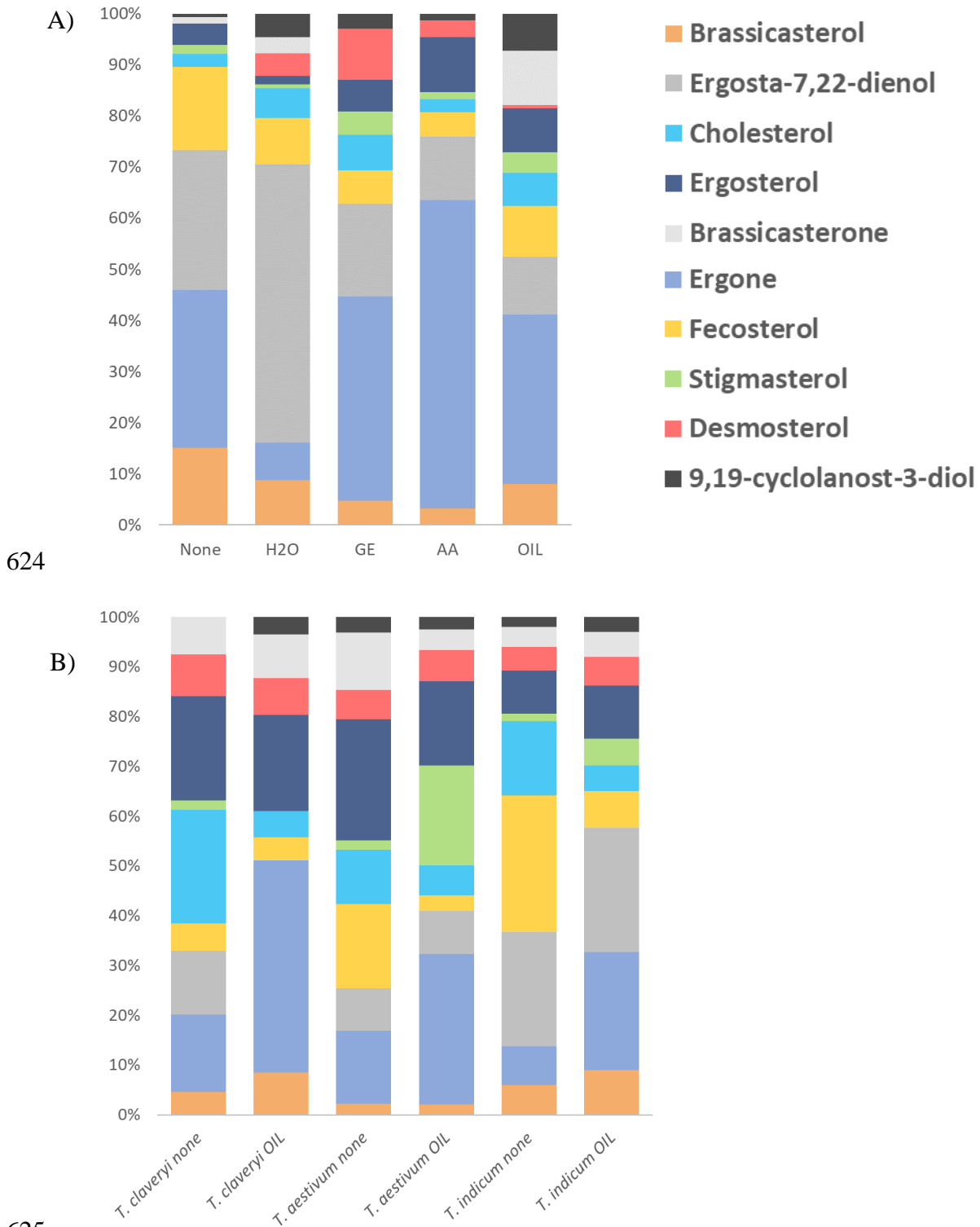
609 **Figure 4.** Extraction yield (%) of a) *T. melanosporum* extracts obtained at optimal
 610 conditions using as trapping matrix water (W, blue), gelatin (GE, red), agar agar (AA,
 611 green), and grapeseed oil (OIL, yellow). The number next to the trapping material indicate
 612 the volume utilized (mL). b) Extracts from 3 truffle species with (orange) and without
 613 (grey) grapeseed oil as trapping material.



616 A, B Different letters denote significant differences ($p \leq 0.05$) between mL added among
 617 the same trapping matrices.

618 a, b Different letters denote significant differences ($p \leq 0.05$) between same volume within
 619 different matrices (Figure-a), between the use of oil or not (Figure-b)

620 **Figure 5.** Sterol composition (%) of A) *T. melanosporum* SFE extracts obtained in
 621 absence of trapping matrix or in presence of 1mL trapping matrix: without matrix (none),
 622 water (W), gelatin (GE), agar-agar (AA), grapeseed oil (OIL). B) SFE extracts from three
 623 different truffle species with and without oil as a trapping matrix.



625