

1 **Reductive off-odors in wines: Formation and release of H₂S and**
2 **Methanethiol during the accelerated anoxic storage of wines**

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Abstract

In order to better understand the processes involved in the development of H₂S and methanethiol (MeSH) along anoxic storage of wines, 24 wines were stored in strict anoxia at 50°C for 3 weeks. Free and total forms of H₂S and MeSH were measured at different times. Results showed that: 1) All wines contain relevant proportions of bonded forms of H₂S and MeSH (93% and 47% on average); 2) such % decreases with age; 3) levels of total forms are related to wine metal composition; 4) anoxic storage brings about an increase of free forms, a strong decrease in the percentage of bonded forms, and except for H₂S in red wines, an increase in total forms. Both de novo formation and release contribute to reductive off-odors. Release is predominant for reds and H₂S, while at 50°C, de novo formation dominates for whites and rosés and MeSH

Keywords: Reduction, Copper, Sulfur, Bonded H₂S, Wine aging, de novo formation

1. Introduction

Reductive off-odors are a not infrequent outcome of wine production, particularly of bottle aging (Mestres, Busto, & Guasch, 2000; Park, Boulton, Bartra, & Noble, 1994; Siebert, Solomon, Pollnitz, & Jeffery, 2010; Ugliano, Dieval, Siebert, Kwiatkowski, Aagaard, Vidal, et al., 2012; Ugliano, Kwiatkowski, Vidal, Capone, Siebert, Dieval, et al., 2011) and are responsible for an important proportion of faulty wines. Such a problem is mostly caused by the development of H_2S and MeSH (Ugliano, et al., 2011), although a number of different other volatile sulfur compounds (VSCs) have been also identified (Mestres, Busto, & Guasch, 2000; Park, Boulton, Bartra, & Noble, 1994). A third relevant molecule, dimethyl sulfide (DMS), is also frequently included within the group of reductive problems. However, DMS strongly differs from H_2S and MeSH both in sensory effects (Escudero, Campo, Farina, Cacho, & Ferreira, 2007; Lytra, Tempere, Zhang, Marchand, de Revel, & Barbe, 2014; Segurel, Razungles, Riou, Salles, & Baumes, 2004) and in chemical origin and properties (Segurel, Razungles, Riou, Trigueiro, & Baumes, 2005), and should be considered apart.

It is usually thought that the most relevant source of reductive off-odors is the alcoholic fermentation and in fact, the development of the characteristic H_2S and MeSH odors during this key wine making step is sometimes clearly observed. H_2S can be directly formed by *Saccharomyces* from elemental sulfur (Schutz & Kunkee, 1977), sulfates or more easily from the sulfite (Jiranek, Langridge, & Henschke, 1995) usually added as antioxidant and antimicrobial agent. The formation is typically stronger in musts with low

levels of assimilable nitrogen (Jiranek, Langridge, & Henschke, 1995), although the factors determining its synthesis are far from being clearly understood (Ugliano, Fedrizzi, Siebert, Travis, Magno, Versini, et al., 2009). The demonstrated influence of methionine (Barbosa, Mendes-Faia, & Mendes-Ferreira, 2012; Spiropoulos, Tanaka, Flerianos, & Bisson, 2000) or cysteine levels (Jiranek, Langridge, & Henschke, 1995; Moreira, Mendes, Pereira, de Pinho, Hogg, & Vasconcelos, 2002) on the formation of H₂S and MeSH undeniably suggests that the formation of these components in fermentation depends on many factors. In the event of an excessive formation of these compounds, winemakers try to control their levels by copper fining, aeration or addition of lees (Clark, Grant-Preece, Cleghorn, & Scollary, 2015; Ugliano, Kwiatkowski, Travis, Francis, Waters, Herderich, et al., 2009; Viviers, Smith, Wilkes, & Smith, 2013).

The reasons why these molecules accumulate during bottle aging, more often in those wines in which these compounds were previously formed in fermentation, are not clearly known (Ugliano, Kwiatkowski, et al., 2009). Several hypotheses have been formulated along the years, such as the hydrolysis of thioacetates (Leppanen, Denslow, & Ronkainen, 1980) or thioethers (Waterhouse & Laurie, 2006), the reduction of disulfides (Bobet, Noble, & Boulton, 1990), the reaction between cysteine and wine α -dicarbonyls (PripisNicolau, deRevel, Bertrand, & Maujean, 2000), the transition-metal catalyzed reduction of sulfate or sulfite (Swiegers, Bartowsky, Henschke, & Pretorius, 2005) or the metal catalyzed desulfhydration of sulfur containing amino acids (Gruenwedel & Patnaik, 1971; Viviers, Smith, Wilkes, & Smith, 2013). All these possibilities involve the “de novo”

formation of H₂S or MeSH from a specific precursor through a specific chemical route. Because the problem is most often observed in the absence of oxygen it is generally assumed that those processes involving a chemical reduction are the most likely candidates, and notably the reduction of disulfides is most often mentioned in the technical literature.

However, an important factor that should be additionally taken into account in order to understand the observed increases of H₂S and MeSH along wine bottle aging is the presence of relevant amounts of these two compounds under the form of non-volatile complexes with Cu(II) and other metal cations, as recently demonstrated (Franco-Luesma & Ferreira, 2014). The finding is further supported by the recent observation that Cu(II) is not easily removed from the wine by precipitation as CuS at the normal levels found in wineries (Clark, Grant-Preece, Cleghorn, & Scollary, 2015). These complexes are reversible, so that the simple dilution of the wine with brine releases back into the headspace H₂S and MeSH. This dilution coupled to a SPME preconcentration constitutes the base for the analytical determination of total (=free+bonded) forms (Lopez, Lapena, Cacho, & Ferreira, 2007). Such determination has to be complemented with a gentle direct headspace injection for the exclusive analysis of free forms to have a clear picture of the total balance of free and bonded forms (Franco-Luesma & Ferreira, 2014). The existence of those interconvertible species means that studies using a method sensitive exclusively to free forms, or exclusively to total forms, will see just half of the picture and most surely will not be able to make a clear diagnosis. For instance, in a previous aging

experiment involving 16 different red wines, and in which VSCs were monitored using the strategy measuring total forms, no H₂S increase was observed even in the samples stored under complete anoxia (Ferreira, Bueno, Franco-Luesma, Cullere, & Fernandez-Zurbano, 2014). Such an unexpected result raised the question of whether the increases of H₂S observed in other reports using methods sensitive to free forms were in fact the unnoticed result of the release of already present bonded forms or whether they were the result of a genuine “de novo” formation which simply did not happen in that sample set. The present paper aims at determining whether the often observed increases of H₂S and MeSH along the anoxic storage of wines are the result of the de novo formation from precursors or are rather the result of the release of complexed forms.

2. Material and methods

2.1. Solvents and Chemical Standards

Ethanol and methanol were purchased from Merk (Darmstadt, Germany). Water with resistance of 18.2 MΩ·cm at 25 °C was purified in a Milli-Q system from Millipore (Bedford, Germany).

Chemicals used for the analytical characterization were of analytical reagent grade and were supplied by Merck (Darmstadt, Germany), Panreac (Barcelona, Spain), Sigma-Aldrich (Madrid, Spain), Lancaster (Eastgate, UK), Scharlau (Barcelona, Spain), Oxford Chemicals (Hartlepool, UK), Fluka (Madrid, Spain), ChemService (West Chester, PA, USA), Extrasynthèse (Genay, France) and SAFC (Steinheim, Germany). Purity of chemical

standards is over 95% in all cases and most of them are over 99%. TSK Toyopearl gel HW-50F was purchased from Tosohaas (Montgomery-ville, PA, USA).

2.2. Wines

Twenty-four Spanish wines were purchased from a local retailer in Zaragoza. Denomination of origin, vintage, alcohol % (v/v), free and total SO₂, pH and metal content are detailed in Table 1 of the Supplementary material.

2.3. Accelerated reduction

Wines were opened inside an anoxic glove box (Jacomex, France) containing Argon atmosphere. Average levels of O₂ inside the anoxic chamber were below 2 ppm (v/v). The levels of dissolved O₂ in the recently opened wines were measured with a calibrated OptiOx SG 9 oxygen sensor (Mettler Toledo-España, Barcelona). The sensor was calibrated daily following the manufacturer directions with water-saturated air by introducing it in a test tube containing a wetted sponge for 5 minutes and activating the calibration function. Wines were stirred for some minutes within the Ar chamber just letting the very small amounts of dissolved oxygen (<0.2 mg/L) dissipate to ensure that at the beginning of the experiment the levels of this molecule were below the limit of detection of the sensor (1 µg/L). The wine was then distributed in 60 mL tubes with highly hermetic closures (WIT, Blanquefort, France). The tubes were completely filled avoiding any headspace. The tubes were further enclosed in a vacuum thermos-sealed plastic bag with known O₂ permeability (< 9 cm³ / m² 24 h) supplied by Amcor (Barcelona, Spain). After this, the bagged tubes were taken out of the Ar chamber and were then incubated in a water bath

at 50°C (Grant Instruments, Cambridge, UK) for 1.5, 5.5, 12.5 or 21 days, prior to further analysis. Owing to the complicated logistics of the experiment, only one tube per wine was used for analysis at each sampling point and was further discarded. Analyses of VSCs were carried out in each tube in triplicate as described below.

The reproducibility of the accelerated reduction procedure was checked in an independent experiment in which 20 tubes, 10 filled with the same red wine and 10 others with the same white wine were incubated at 50°C for 12.5 (5 tubes/wine) and 21 days (5 tubes/wine) and analyzed as described below. Some tubes filled with a wine model (12% ethanol (v/v); pH 3.5; 5 g/L tartaric acid) and containing also a PreSens Pst6 oxygen sensor (Presens GmbH, Regensburg, Germany) were similarly prepared within the anoxic chamber and were further incubated at 50°C for 21 days. The O₂ permeated was measured with the Nomasense O₂ analyzer (NomaCorc, Aubel, Belgium). The readings were in all cases below the detection limits of the system, implying that the O₂ permeated into the tubes was below 60 ng, which for our purposes can be considered negligible.

2.4. Chemical analysis

Total and free VSCs, color, Total Polyphenol Index, pH, total and free SO₂, Trolox equivalent antioxidant capacity (TEAC) and Folin-Ciocalteu were measured in all sampling points. Metals, polyphenols and amino acids were measured only in the initial wines.

The methods and instruments for the analysis of free and total VSCs are described in references (Franco-Luesma & Ferreira, 2014) and (Lopez, Lapena, Cacho, & Ferreira, 2007). Briefly, for the analysis of total forms, the wine is introduced into the anoxic

chamber and 0.1 mL are pipetted into a 20 mL standard headspace vial, already containing 4.9 mL of a NaCl brine. The internal standard solution (EMS, PrSH and thiophene at 20 $\mu\text{g/L}$) is also added, the vial is sealed, taken out of the chamber and immediately (without idle time in the sampler tray) preconcentrated by headspace SPME and further analyzed by GC with pulsed flame photometric detection (pFPD).

For the analysis of free forms, the wine is also introduced into the anoxic chamber, and 12 mL with 40 μL of the internal standard solution (EMS, PrSH and thiophene at 2 mgL^{-1}) are directly transferred into a 20 mL standard headspace vial. The vial is then sealed, taken out of the chamber and immediately analyzed. The analysis consists of a direct headspace injection in the GC-pFPD system. In both determinations, the areas are normalized to those of the internal standard (PrSH for H_2S , MeSH and EtSH and EMS for DMS, DES, DMDS and DEDS) and interpolated in calibration graphs built by the analysis of a synthetic calibrated wine.

Color was quantified as the absorbances at 420, 520 and 620 nm. Red wines were diluted 10 times with purified water, and rosés and whites were measured directly. Total Polyphenol Index was estimated as absorbance at 280 nm, red wines were diluted 100 times, rosés 50 and whites 20 with purified water. All absorbance measurements were made in triplicate using 1 cm cuvettes and a UV-VIS spectrophotometer UV-17000 Pharma Spec from Shimadzu (Kyoto, Japan).

TEAC and Folin-Ciocalteu assays were adapted from procedures described by Rivero-Perez et al. (Rivero-Perez, Muniz, & Gonzalez-Sanjose, 2007) and Singleton et al.

(Singleton, Orthofer, & Lamuela-Raventos, 1999) respectively. Absorbance measurements were taken using 1 cm cuvettes at a UV-VIS spectrophotometer UV-17000 Pharma Spec from Shimadzu (Kyoto, Japan).

Cu, Fe, Mn, Ni, Sn and Zn in wines were determined by ICP-OES with a Thermo Elemental IRIS Intrepid, as indicated in the method proposed by Gonzalez et al. (Gonzalez, Armenta, Pastor, & de la Guardia, 2008), microwave-assisted digestion was used as sample treatment.

Analyses of the polyphenolic matter was performed following the method described by Gonzalez-Hernandez et al. (Gonzalez-Hernandez, Avizcuri-Inac, Dizy, & Fernandez-Zurbano, 2014). Two mL of wine were filtered by 0.45 μ m and fractionated by Gel Permeation Chromatography (GPC) in an automated fraction collector from Gilson (Middleton, WI, USA) with a Vantage L column (120 mm x 12 mm) from Millipore (Bedford, Ma, USA) packed with TSK Toyopearl gel HW-50F (Tosohaas, Montgomery Ville, PA, USA). Two fractions were collected and brought to dryness under vacuum. Fraction 1 was dissolved in 2 mL of formic acid/water (5:95, v/v) and it was further analyzed by UPLC–DAD-MS for quantifying anthocyanins and by UPLC–MS for quantifying flavonols, flavanols, hydroxycinnamic acids, phenolic acids, aconitic acid and resveratrol. Fraction 2 was dissolved in 2 mL of methanol.

The analyses of amino acids were made following a precolumn derivatization procedure with aminoquinolyl-N-hydrosysuccinimidyl carbamate (AQC) using a quaternary high-performance liquid chromatography (HPLC) eluent system as it is described in (Hernandez-

Orte, Ibarz, Cacho, & Ferreira, 2003). A summary of all this data can be found in Table 2 of the supplementary material.

2.5. Statistical analysis and data treatment

Simple statistical calculations were carried out with Excel 2013 (Microsoft, WA, USA). For the comparison of means of different groups of samples, normal t-test assuming equal variances were carried out. For the comparison of means of the same samples at different times, paired t-test statistics were used. Regression analysis was used in order to assess the significance of temporal evolutions. Models were built using Partial Least-Squares (PLS) regression with The Unscrambler 9.7 (CAMO Software AS, Oslo, Norway). Models have been produced combining correlation analysis and further PLS modeling. The simplest models explaining as much as possible the variability of the parameter and producing minimum predictive errors were searched by progressively reducing the number of variables in the model. In all the cases models were developed and validated by cross-validation, a strategy in which the model is built iteratively leaving out one sample per iteration. The predictive power of these models is assessed via the root mean square error (RMSE), which is estimated as well by cross validation, computing the predictive error for each sample in the iteration in which it was left out.

3. Results and discussion

Free and complexed forms of H₂S and mercaptans in wines

Twenty four different wines were analyzed for their levels in free and total VSCs following a previously developed procedure (Franco-Luesma & Ferreira, 2014). With the analytical methods used, only H₂S, MeSH and DMS were found above the method detection limits. Results of the analysis are summarized in Table 1. As seen in the table, most wines contain small amounts of free H₂S and MeSH and only in four cases the levels of these molecules were not detectable. At these levels, these molecules are most likely not causing a reductive off-odor problem. In contrast, levels of DMS are high enough so that this molecule will exert a notable sensory effect, and depending on the aromatic environment will enhance fruity notes or will even produce an unpleasant effect (Lytra, Tempere, Zhang, Marchand, de Revel, & Barbe, 2014; Segurel, Razungles, Riou, Salles, & Baumes, 2004).

Confirming previous results (Franco-Luesma & Ferreira, 2014), both H₂S and MeSH are present under non-volatile forms, most likely as complexes with Cu²⁺, Fe²⁺ and even Zn²⁺. Results in the table indicate that, on average, free H₂S constitutes just a 6% of the total amount of H₂S in a red wine and 8% in a white or rosé. Similarly, free MeSH only accounts on average for 38% of the total levels of a red wine while a 69% of the MeSH contained in whites and rosés is under free forms. In clear contrast, DMS was not found present in complexed forms in agreement with previous work (Franco-Luesma & Ferreira, 2014). Levels of total H₂S are relatively high, likely more than enough to cause an aromatic

problem if those complexed forms would be released. Levels of free and total H₂S are not correlated, while in the case of MeSH, there is a positive correlation (significant at $P < 0.01$) between free and complexed forms, suggesting that in this case the differential metal content of the wine is not that critical in determining the proportion in bonded forms.

Another interesting question is the relationship between VSCs content and wine age, which applies only to the 16 red wines in the study, since whites and rosés were all from the same vintage. Regression analysis revealed that the average level of free H₂S increased by 0.38 ± 0.11 µg/L per year of aging (significant at $P < 0.01$), while the average level of total H₂S remained approximately constant (see supplementary material Figure 1a). Consequently, the bonded fraction decreases from $96.9 \pm 0.89\%$ in wines from 2012 to $87.8 \pm 4.0\%$ in wines from 2008 or older, which suggests that there is an average release of $1.9 \pm 0.7\%$ of bonded forms per year of aging (significant at $P < 0.05$).

Similar results were observed in the case of MeSH and regression analysis revealed the existence of an average increase of 0.23 ± 0.06 µg/L per year of aging (significant at $P < 0.01$), while levels of total forms were not found to significantly increase with age (see supplementary material Figure 1b). The proportion of bonded forms changed from $74.1 \pm 7.2\%$ in 2012 wines to $40 \pm 6.2\%$ in wines from 2008 or older (significant at $P < 0.01$), suggesting an average release of $8.1 \pm 2.7\%$ of bonded forms per year of aging. Finally, in the case of DMS, regression analysis revealed that its levels increase on average 6.9 ± 1.7 µg/L per year of aging (significant at $P < 0.01$). No further comment on this aroma molecule will be presented in this paper.

Relationship between actual contents and wine composition

Total H₂S levels of whites and rosés are strongly related to the wine content in metals. Although univariately there is only a significant correlation with wine Zn levels ($r = -0.67$, significant at $P < 0.05$), PLS modeling shows that in fact the wine content in total H₂S is nearly completely determined by its contents in four metals. As the model reveals (model 2 in Table 2), the total H₂S levels of wine is directly related to its levels of copper and manganese and negatively related to its levels of iron and zinc. The variance explained by the model (by cross validation) is 89.6% and the predictive error is just 1.37 µg/L. In case of red wines, wines from 2008 or older had to be excluded from the model, but for the other 11 younger wines PLS modeling confirmed that the levels of total H₂S were positively related to wine copper levels and negatively related to the levels of Zn and Fe, as shown in Table 2 (model 1).

While there are two possible known causes explaining the positive correlation between total H₂S and copper, the negative role played by Zn and Fe is less obvious. In terms of catalytic action, addition of Zn to commercial wines did not produce but an increase in H₂S levels, while the effects of iron were not clear (Viviers, Smith, Wilkes, & Smith, 2013); hence, catalytic action is not consistent with the negative coefficients in the models. The same happens to the demonstrated potential ability of Zn²⁺ and Fe²⁺ to reversibly trap H₂S in synthetic hydroalcoholic solutions (Franco-Luesma & Ferreira, 2014). A potential explanation would be that must deficiencies in Zn and Fe would be responsible for yeast overproduction of H₂S. In fact, it is known that Zn is an essential element for

281 *Saccharomyces cerevisiae* and that its deprivation can cause sluggish fermentations
282 (Bromberg, Bower, Duncombe, Fehring, Gerber, Lau, et al., 1997; Gauci, Beckhouse,
283 Lyons, Beh, Rogers, Dawes, et al., 2009), and there is also a report demonstrating that the
284 activity of Zn-Cu superoxide dismutases is enhanced by H₂S (Searcy, Whitehead, &
285 Maroney, 1995). However, and to the best of our knowledge there are no literature
286 reports demonstrating a direct relationship between low levels of this element or of iron
287 and high production of H₂S in fermentation, as the models may suggest.

288 In the case of copper, it is known that winemakers use it to remove excesses of H₂S, so
289 that wines coming from fermentations with high production of this molecule may have
290 more residual copper. Besides, wines naturally containing more copper may have
291 accumulated more H₂S produced in fermentation.

292 In the case of reds, the model suggests that total H₂S levels are related to the wine
293 chromatic parameters and to the wine levels in methionine and cysteine (Table 2, model
294 1). Chromatic parameters may be just indicating the negative influence of wine oxidation
295 (A420 and A520) on total H₂S levels. The negative relationship between wine total levels
296 of H₂S and methionine may be attributed to the demonstrated suppressing effect of this
297 amino acid on H₂S production by yeast (Barbosa, Mendes-Faia, & Mendes-Ferreira, 2012;
298 Spiropoulos, Tanaka, Flerianos, & Bisson, 2000). The similar relationship of cysteine is
299 more difficult to explain, because in fact there is a known relationship between the must
300 levels of this amino acid and the formation of H₂S (Jiranek, Langridge, & Henschke, 1995;
301 Moreira, Mendes, Pereira, de Pinho, Hogg, & Vasconcelos, 2002). On the other hand, it

has been reported that high cellular levels of cysteine were related to H₂S suppression (Spiropoulos & Bisson, 2000). Our poor understanding about the factors that determine the residual amounts of these amino acids in wine after fermentation does not make it possible to propose a clear hypothesis.

In the case of methanethiol, two independent models with a quite similar structure have been also derived for reds and for whites and rosés, as seen in Table 2 (models 3 and 4). One of the most remarkable observations is that, in clear contrast with the models for H₂S, the levels of total MeSH contained in the wine are inversely related to copper levels, confirming that the role of copper as trapping agent of MeSH is not really important in this case. Secondly, as was previously observed, in both cases total MeSH levels are related to the actual content in free MeSH. Thirdly, methionine plays a negative role in both models, as was already observed for H₂S in red wines, but in apparent contrast with the known relationship between must methionine levels and the production of methanethiol by *Saccharomyces* (Perpete, Duthoit, De Maeyer, Imray, Lawton, Stavropoulos, et al., 2006) or by lactic bacteria (Pripis-Nicolau, de Revel, Bertrand, & Lonvaud-Funel, 2004) and with the relationship between wine methionine and de novo formation of MeSH during anoxic storage (Ferreira, Bueno, Franco-Luesma, Cullere, & Fernandez-Zurbano, 2014). This suggests that the residual amount of methionine in wine is not a major source of the total MeSH found in wine before anoxic aging, which would support a major fermentation origin for this compound.

Finally, the models also assign a consistent negative influence of the levels of some important antioxidants on the wine total MeSH content. The negative correlation between MeSH formation and resveratrol has been previously observed in an independent experiment (Ferreira, Bueno, Franco-Luesma, Cullere, & Fernandez-Zurbano, 2014).

Evolution of H₂S and MeSH during anoxic aging

The wines were stored in strict anoxia at 50°C for 3 weeks, and samples were analyzed at days 1.5, 5.5, 12.5 and 21. This type of accelerated aging was selected because in preliminary trials we could observe that the rates of increase of H₂S and MeSH were fast enough for the purposes of the experiment and that there was no evident formation of artefacts. At higher temperatures such as 70°C, strong increases of free and total forms of H₂S and MeSH together with decreases in DMS were observed, suggesting strong decomposition of S-amino acids (data not shown). In addition, an aging experiment at 25°C with a set of similar wines has been also conducted. Results will be presented in a forthcoming paper, but confirm that the accumulation of free H₂S and free MeSH at 25°C and 50°C are strongly correlated. The reproducibility of the accelerated anoxic aging turned out to be very high, as revealed by the low RSDs for the overall process obtained in an independent experiment (see methods). The overall variability (given as RSD(%)) was 3.9%, 2.2% and 1.6% for the levels of free H₂S, MeSH and DMS, respectively. For total forms RSDs were 5.3% and 3.3 % for total H₂S, and MeSH, respectively. In all cases the major contributor to uncertainty was the analytical determination.

The anoxic aging brought about important increases in the levels of free H₂S, as summarized in Figure 1 and Table 3. The figure gives the evolution of the average contents of free and total H₂S for the 16 red wines (Figure 1a) and the 8 whites and roses (Figure 1b) along the anoxic storage. Figure 1c shows the evolution of the proportion of complexed forms of H₂S during the storage. In the case of red wines the total content of H₂S remained relatively stable along the whole process. The average increment, as detailed in Table 3, was 1.58 µg/L which did not reach the level of statistical significance (paired t test, P=0.07). It can be concluded that for red wines and on average, there is no significant de novo formation of this molecule during the storage. On the contrary, the average levels of free H₂S increased continuously becoming at the end of the storage just slightly smaller than total levels. The average total increase of free H₂S was above 16 µg/L, as detailed in Table 3 (paired t test, significant at P<10⁻⁸). As seen in Figure 1c, the average proportion of complexed forms decreases from the initial 94% (Table 1) to a meager 23% after the 3 weeks (paired t test, significant at P<10⁻⁸). The consequence of the process is therefore a neat transformation of complexed non-volatile forms into free H₂S.

In the case of white and rose wines, shown in Figure 1b, levels of total H₂S increased slightly but significantly, becoming on average 9.4 µg/L higher than the initial contents after the storage (paired t test, significant at P<0.0001). For these types of wines the levels of free H₂S increased very fast in the first week of storage and then the rate of increase slowed down. This is also seen in Figure 1c, which confirms that for whites and rosés, the release of complexed forms is very fast and seems to be nearly completed in less than one

week. Remarkably, in the last sampling point there is a slight increase in the proportion of bonded forms. Such increase is statistically significant (paired t test, $P=0.026$).

Going into more detail, in a reduced number of 4 red wines there was a slight but significant (regression analysis, in all cases $P<0.05$) increment of total H_2S , as can be seen in Table 3. Increments were in three of the cases smaller than $4\text{ }\mu\text{g/L}$ and in the other case it was of $6.7 \pm 0.8\text{ }\mu\text{g/L}$ (Significant at $P<0.01$). The increments of free H_2S of red wines were in all cases but one, between 10 and $26\text{ }\mu\text{g/L}$. The odd sample released only $2.9\text{ }\mu\text{g/L}$ in the three weeks and after the process it still contained 89% of H_2S in bonded forms. The proportion of complexed forms at the end of the process was smaller than 12% for half of the samples, and smaller than 35% for all but two. For whites and roses, increments in total H_2S level were significant ($P<0.05$) in all the wines and ranged from 4.6 to $12.6\text{ }\mu\text{g/L}$, as detailed in Table 3, while increments in free forms were relatively homogeneous ranging from 14 to $33\text{ }\mu\text{g/L}$. For these wines, at the end of the process the % in bonded forms ranged from 0 to 54%.

The mass balance of the process is summarized in Table 4. As detailed in the table, it can be estimated that out of the $16.2\text{ }\mu\text{g/L}$ of free H_2S that on average will accumulate in red wines during the anoxic storage, $14.7\text{ }\mu\text{g/L}$ can be attributed to the release. This value has been estimated by applying the decrease of the fraction bonded (-71%) to the total initial content ($20.8\text{ }\mu\text{g/L}$ in Table 1). Therefore, it can be said that attending to our data, the release of H_2S from complexes is the dominant process in reds, explaining on average 90.3% of the observed increase in free H_2S potentially responsible for an off-odor. In the

case of whites and rosés, however, the de novo formation of H₂S becomes more relevant, so that the release from complexed forms explains just 58% of the free H₂S accumulated in the wine during the aging.

Results for the evolution of MeSH along the anoxic aging are shown in Figure 2 and in Table 3. In this case there are clear average increments of total forms both in red and in white and rosé wines. In red wines, the average increment is 1.4 µg/L (paired t test, significant at $P < 10^{-10}$), while in whites and rosés it reaches nearly 3 µg/L (paired t test, significant at $P < 0.0001$). In both types of wines the proportion of complexed forms significantly decreases (paired t test, significant at $P < 10^{-10}$ for reds and < 0.01 for whites and rosés), faster in the case of whites and rosés as shown in Figure 2c, so that in the last sampling points average levels of free and total forms are coincident. The release and formation of this molecule in red wines is relatively homogeneous, and increments of free values range from 1.8 to 4.3 µg/L, while increments of total forms range between 0.88 and 1.91. In white and roses there is more variability, and increments in free forms range a factor 4 (from 1.9 to 7.7), while total forms range from 1.58 to 4.38, as shown in Table 3. The overall balance of the process for MeSH is summarized in Table 4. In this case, it is de novo formation the dominant process causing the accumulation of the free form of this molecule in the headspaces of wines. It can be seen that in the case of reds, the release from complexed forms just accounts for a 47.5% of the free MeSH accumulated, while in whites and rosés, release from complexes become nearly irrelevant, since only 24 % of the amounts of free MeSH accumulated in these types of wines are attributed to the release.

It should be taken into account that the mass balance summarized in table 4 refers to accelerated anoxic storage at 50°C. It can be expected that the pattern will not change dramatically at lower temperatures, at least in relative terms, but different proportions of de novo formation and release will surely be observed. It is remarkable that two of the main observations derived from a previous work carried out at 25°C with red wines (Ferreira, Bueno, Franco-Luesma, Cullere, & Fernandez-Zurbano, 2014), namely the inexistence of relevant de novo formation of H₂S and the significant de novo formation of MeSH, are coincident with the results presented here. All those data make it possible to state that the release of free forms from already present complex forms and the de novo production of small quantities of H₂S (except in reds) and MeSH during storage is a fairly common outcome of wines. A corollary to this is that if the wine is bottled using a perfectly hermetic closure, both processes may lead to the accumulation in the media of significant amounts of free H₂S and MeSH, which undoubtedly will have consequences on the wine sensory profiles. Questions which will have to be further addressed are the time frame at which these phenomena take place at normal wine storage temperatures, the chemical origin of bonded forms and the chemical nature of the processes causing de novo formation.

In summary, this paper has revealed that the observed accumulation of free H₂S and MeSH along the accelerated anoxic storage of wines is caused both by de novo formation from precursors and by release from already existent Cu(II)-H₂S and Cu(II)-MeSH complexes. While release from complexes seems to be the major cause explaining the

accumulation of H₂S, particularly in red wines, de novo formation becomes the dominant source of MeSH, particularly in white and rosé wines.

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FIGURE CAPTIONS

Figure 1. Evolution of the mean total (1a) and free (1b) contents and of the mean proportions in complexed forms (1c) of H_2S along the accelerated reductive aging period. Experimental points represent the average of the 16 (reds) and 8 (whites and rosés) samples. Error bars are the standard error of the corresponding means ($S/\sqrt{16}$ for reds and $S/\sqrt{8}$ for whites and rosés).

Figure 2. Evolution of the mean total (2a) and free (2b) contents and of the mean proportions in complexed forms (2c) of MeSH along the accelerated reductive aging period. Experimental points represent the average of the 16 (reds) and 8 (whites and rosés) samples. Error bars are the standard error of the mean ($S/\sqrt{16}$ for reds and $S/\sqrt{8}$ for whites and rosés).

Table 1. Free and total contents of H₂S, MeSH and DMS (for this compound free=total) of the wines in the study. Data expressed in µg/L

	H ₂ S			MeSH			DMS
	free	total	% bonded	free	total	% bonded	
Red Wines							
AA	3.4 ± 0.1	44.9 ± 2.9	92%	1.6 ± 0.0	2.7 ± 0.1	42%	40.6 ± 0.0
BL	2.2 ± 0.3	20.4 ± 0.0	89%	1.8 ± 0.0	4.2 ± 0.0	57%	35.8 ± 0.2
CH	0.9 ± 0.0	13.3 ± 0.0	94%	1.5 ± 0.0	2.2 ± 0.1	33%	40.9 ± 0.6
FG	0.7 ± 0.1	15.1 ± 0.3	95%	0.9 ± 0.0	1.7 ± 0.0	45%	23.1 ± 0.1
LH	1.1 ± 0.1	19.4 ± 2.5	94%	1.1 ± 0.0	2.7 ± 0.2	60%	19.6 ± 0.3
MF	nd	30.2 ± 0.3	>99%	nd	1.9 ± 0.0	>79%	11.7 ± 0.1
MZ	nd	17.1 ± 1.7	>99%	nd	1.1 ± 0.1	>63%	13.8 ± 0.1
PC	0.6 ± 0.0	22.0 ± 1.8	97%	0.7 ± 0.1	2.4 ± 0.0	72%	12.4 ± 3.3
RD	0.7 ± 0.1	14.9 ± 2.0	95%	0.8 ± 0.0	2.7 ± 0.3	72%	9.2 ± 0.1
SL	0.3 ± 0.3	28.4 ± 1.4	99%	nd	2.3 ± 0.1	>83%	23.8 ± 1.1
TP	0.7 ± 0.0	21.9 ± 3.4	97%	0.8 ± 0.0	2.0 ± 0.1	60%	47.3 ± 0.5
TS	0.6 ± 0.1	21.3 ± 4.3	97%	0.7 ± 0.0	2.3 ± 0.2	68%	26.2 ± 0.2
TZ	2.9 ± 0.0	12.1 ± 0.1	76%	1.3 ± 0.0	1.9 ± 0.0	29%	57.3 ± 0.4
UB	0.7 ± 0.0	19.9 ± 1.9	97%	1.3 ± 0.0	2.2 ± 0.2	42%	62.0 ± 0.2
VN	1.8 ± 0.2	16.9 ± 0.7	90%	0.9 ± 0.0	1.9 ± 0.1	51%	36.4 ± 0.2
VV	0.8 ± 0.0	15.0 ± 0.3	95%	0.9 ± 0.0	2.0 ± 0.1	54%	18.8 ± 0.1
Average	1.1 ± 0.1	20.8 ± 0.7	94%	0.9 ± 0.0	2.3 ± 0.1	62%	29.9 ± 0.3
SD	0.99	8.1	6%	0.55	0.67	23%	16.4
Maximum	3.44	44.9	>99%	1.82	4.20	>83%	62.0
Minimum	<0.2	12.1	76%	<0.4	1.08	29%	9.2
Whites & Roses							
AB	0.4 ± 0.4	28.0 ± 0.2	99%	1.5 ± 0.03	2.1 ± 0.1	27%	10.9 ± 0.1
CC	1.1 ± 0.3	21.4 ± 0.5	95%	2.4 ± 0.01	3.0 ± 0.0	22%	14.1 ± 0.0
EN	3.8 ± 0.4	23.9 ± 0.1	84%	1.8 ± 0.07	2.7 ± 0.2	33%	44.8 ± 1.2
FR	3.9 ± 0.9	29.1 ± 0.6	86%	2.3 ± 0.06	2.5 ± 0.1	10%	40.1 ± 1.2
HJ	3.4 ± 0.0	19.9 ± 1.4	83%	1.1 ± 0.07	2.3 ± 0.1	54%	24.8 ± 0.3
IL	nd	24.0 ± 0.4	>99%	0.8 ± 0.04	2.0 ± 0.0	60%	22.6 ± 0.1
MR	3.6 ± 0.9	28.5 ± 2.7	87%	2.0 ± 0.05	3.3 ± 0.4	39%	28.5 ± 1.1
VT	nd	31.0 ± 3.5	>99%	2.7 ± 0.02	2.8 ± 0.2	4%	7.0 ± 0.0
Average	2.0 ± 0.3	25.7 ± 0.8	92%	1.8 ± 0.02	2.6 ± 0.1	31%	24.1 ± 0.4
SD	1.81	4.0	7%	0.66	0.44	20%	13.5
Maximum	3.94	31.0	>99%	2.73	3.26	60%	44.8
Minimum	<0.2	19.91	83%	0.80	1.99	4%	7.04

nd, not detected. The estimated limit of detection are 0.2 µg/L (H₂S) and 0.4 µg/L (MeSH)

Table 2. Summary of the PLS models relating the actual wine content in total H₂S and total MeSH with the wine chemical composition

Nº	Parameter	EVar	RMSE	Model (regression coefficients)
1	Reds H ₂ Stot	78%	3.90	5.00 + 0.227 Cu + 0.237 A620 -0.205 A420 – 0.195 A520 – 0.075 Fe – 0.145 Zn – 0.43 Cysteine – 0.146 Methionine
2	Wh&Ros H ₂ Stot	90%	1.37	6.41 + 0.303 Cu + 0.359 Mn – 0.487 Fe – 0.583 Zn
3	Reds MeSHtot	82% ¹	0.19	7.11+ 0.117 FreeMeSH + 0.169 H₂Stot/Cu – 0.140 Cu – 0.14 t-caftaric acid -0.153 caffeic acid – 0.177 c-resveratrol – 0.176 t-resveratrol -0.175 proanthocyanidin A2 -0.167 TEAC– 0.160 Methionine
4	Wh&Ros MeSHtot	81%	0.20	6.18 + 0.179 FreeMeSH + 0.217 H₂Stot/Cu – 0.164 Cu -0.20 t-resveratrol -0.18 vitisin A – 0.188 pyranoanthocyanins - 0.058 Methionine

¹One sample with maximum contents excluded

Table 3. Increments of free and total contents of H₂S and MeSH measured in the wines in the study after 21 days of anoxic aging at 50°C. Data expressed in µg/L

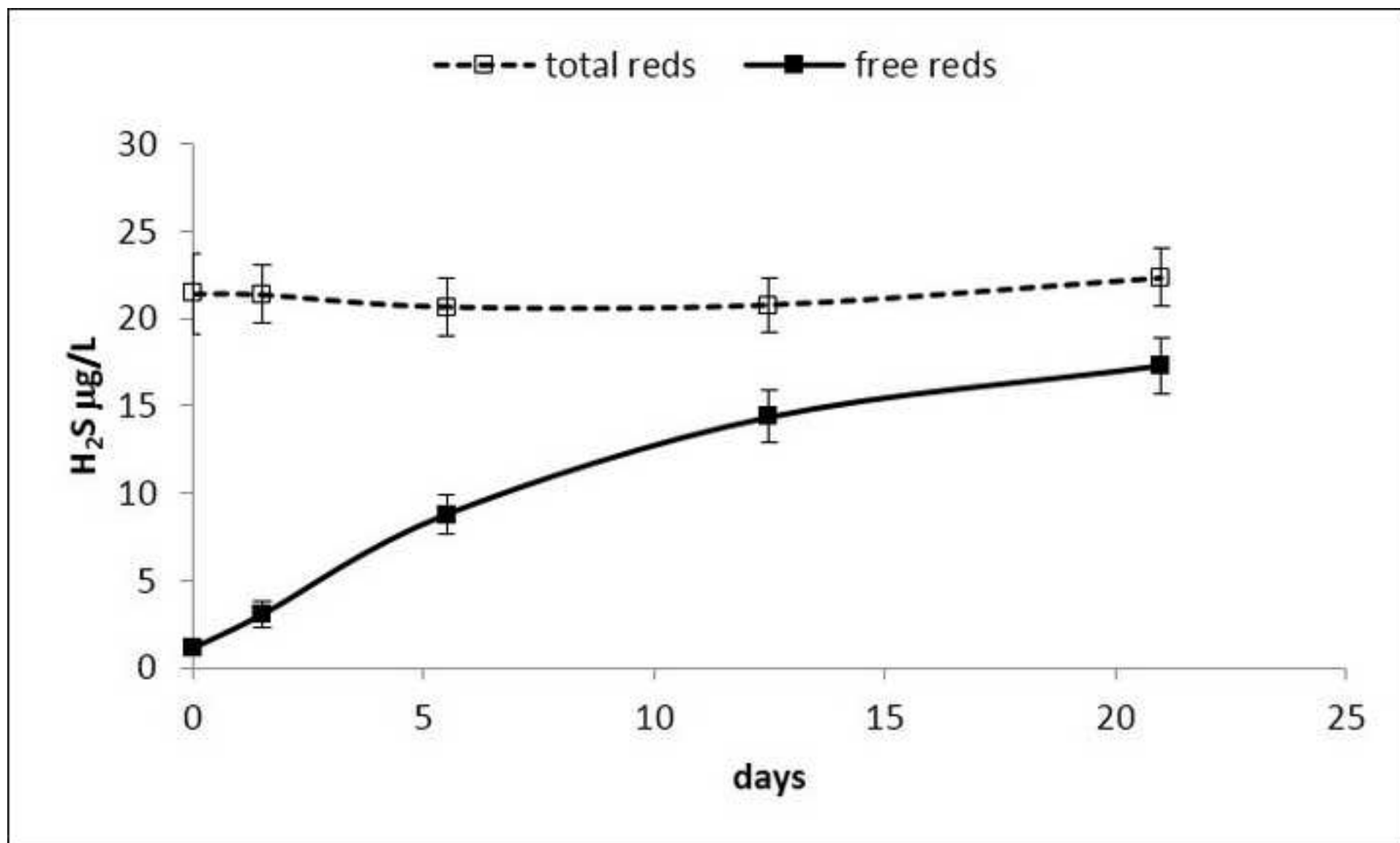
	H₂S			MeSH		
	Free	Total	%bonded	Free	Total	%bonded
Red wines						
AA	25.7 ± 0.2	ns	-59%	2.8 ± 0.09	1.3 ± 0.3	<-42%
BL	22.6 ± 0.8	ns	-89%	4.3 ± 0.12	1.4 ± 0.6	<-57%
CH	17.6 ± 0.1	3.9 ± 1.0	-94%	3.5 ± 0.06	1.9 ± 0.1	<-33%
FG	14.0 ± 0.2	ns	-67%	2.3 ± 0.04	1.7 ± 0.3	-41%
LH	23.0 ± 0.2	ns	-93%	2.6 ± 0.03	1.1 ± 0.4	-58%
MF	2.9 ± 0.2	ns	-11%	2.0 ± 0.003	1.2 ± 0.6	-66%
MZ	10.5 ± 0.9	ns	-67%	2.2 ± 0.02	1.4 ± 0.2	-89%
PC	19.3 ± 0.3	ns	-86%	1.8 ± 0.08	1.0 ± 0.3	-45%
RD	13.7 ± 0.2	4.0 ± 2.2	-72%	2.3 ± 0.05	1.3 ± 0.4	-47%
SL	13.6 ± 0.7	ns	-52%	3.4 ± 0.06	1.8 ± 0.3	-82%
TP	15.9 ± 0.3	ns	-85%	2.1 ± 0.04	0.9 ± 0.2	-60%
TS	13.9 ± 0.2	ns	-67%	2.5 ± 0.06	1.1 ± 0.2	-65%
TZ	17.5 ± 0.01	3.4 ± 7.8	-76%	2.4 ± 0.01	1.3 ± 1.5	<-29%
UB	10.0 ± 0.3	ns	-41%	4.2 ± 0.09	1.6 ± 0.8	<-42%
VN	19.3 ± 0.6	6.7 ± 2.5	-79%	2.3 ± 0.10	1.8 ± 0.2	-38%
VV	20.6 ± 0.4	3.9 ± 0.3	-95%	2.6 ± 0.03	1.3 ± 0.2	<-54%
Average	16.2 ± 0.4	1.6±3.8	-71%	2.7 ± 0.06	1.4 ± 0.5	-60%
SD	5.73	3.06	22%	0.75	0.31	16%
Maximum	25.70	6.65	-95%	4.33	1.91	-89%
Minimum	2.87	-1.61	-11%	1.77	0.88	->29%
Whites & Roses						
AB	20.3 ± 1.7	12.6 ± 8.7	-50%	4.4 ± 0.09	3.4 ± 0.5	<-27%
CC	32.9 ± 0.2	9.3 ± 0.8	-95%	7.7 ± 0.14	4.4 ± 0.4	<-22%
EN	16.8 ± 1.6	9.4 ± 4.1	-46%	3.1 ± 0.10	2.7 ± 0.2	-25%
FR	14.0 ± 0.7	8.6 ± 6.1	-34%	3.9 ± 0.07	3.7 ± 0.3	-9%
HJ	24.9 ± 0.03	12.2 ± 5.6	-71%	2.8 ± 0.09	2.7 ± 0.5	-32%
IL	27.5 ± 2.0	12.3 ± 7.8	-76%	3.8 ± 0.06	3.0 ± 0.3	-52%
MR	26.9 ± 0.5	6.5 ± 3.7	-75%	1.9 ± 0.09	1.6 ± 0.4	-19%
VT	16.2 ± 0.5	4.6 ± 4.9	-46%	2.7 ± 0.05	1.6 ± 0.2	<-4%
Average	22.5 ± 1.4	9.4±	-61%	3.8 ± 0.09	2.9 ± 0.4	-31%
SD	6.64	2.89	21%	1.76	0.97	15%
Maximum	32.89	12.58	-95%	7.66	4.38	-52%
Minimum	13.98	4.64	-34%	1.89	1.58	<-4%

ns: not significant

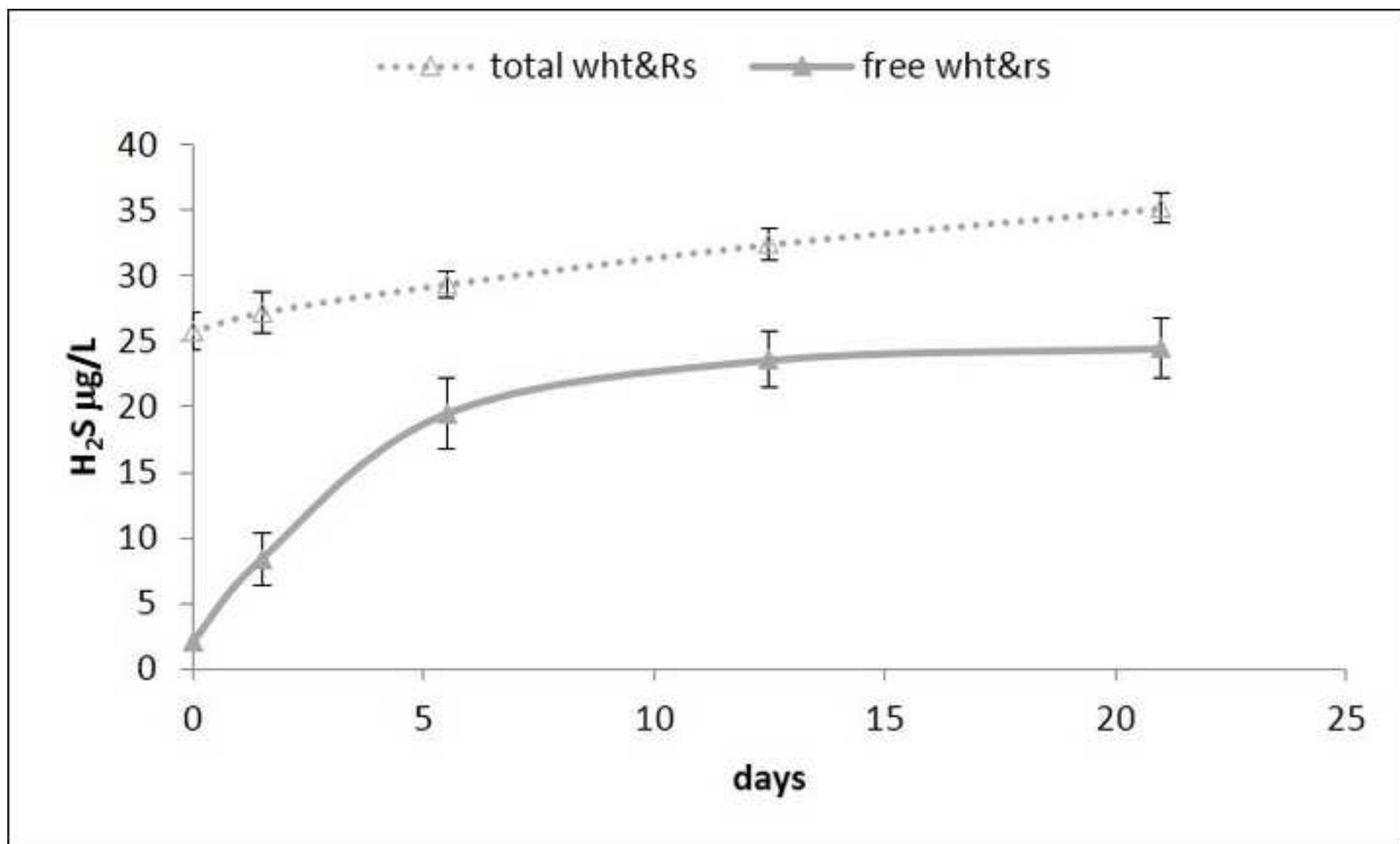
Table 4. Mean amounts of H₂S and MeSH formed along the anoxic storage and estimated contributions of the processes of de novo formation and release from complexes

	H₂S		MeSH	
	Reds	Whites & rosés	Reds	Whites & rosés
Total increase of free forms (release + de novo) (µg/L)	16.2	22.5	2.7	3.8
Increase attributed to release (µg/L)	14.7	13.0	1.3	0.9
% increase attributed to release	90.3%	58.0%	47.5%	24.1%
% increase attributed to de novo formation	9.7%	42.0%	52.5%	75.9%
Bonded fraction at the beginning (%)	94%	92%	62%	31%
Bonded fraction after the storage (%)	23%	31%	2%	0%
Decrease of the fraction bonded (%)	71%	61%	60%	31%

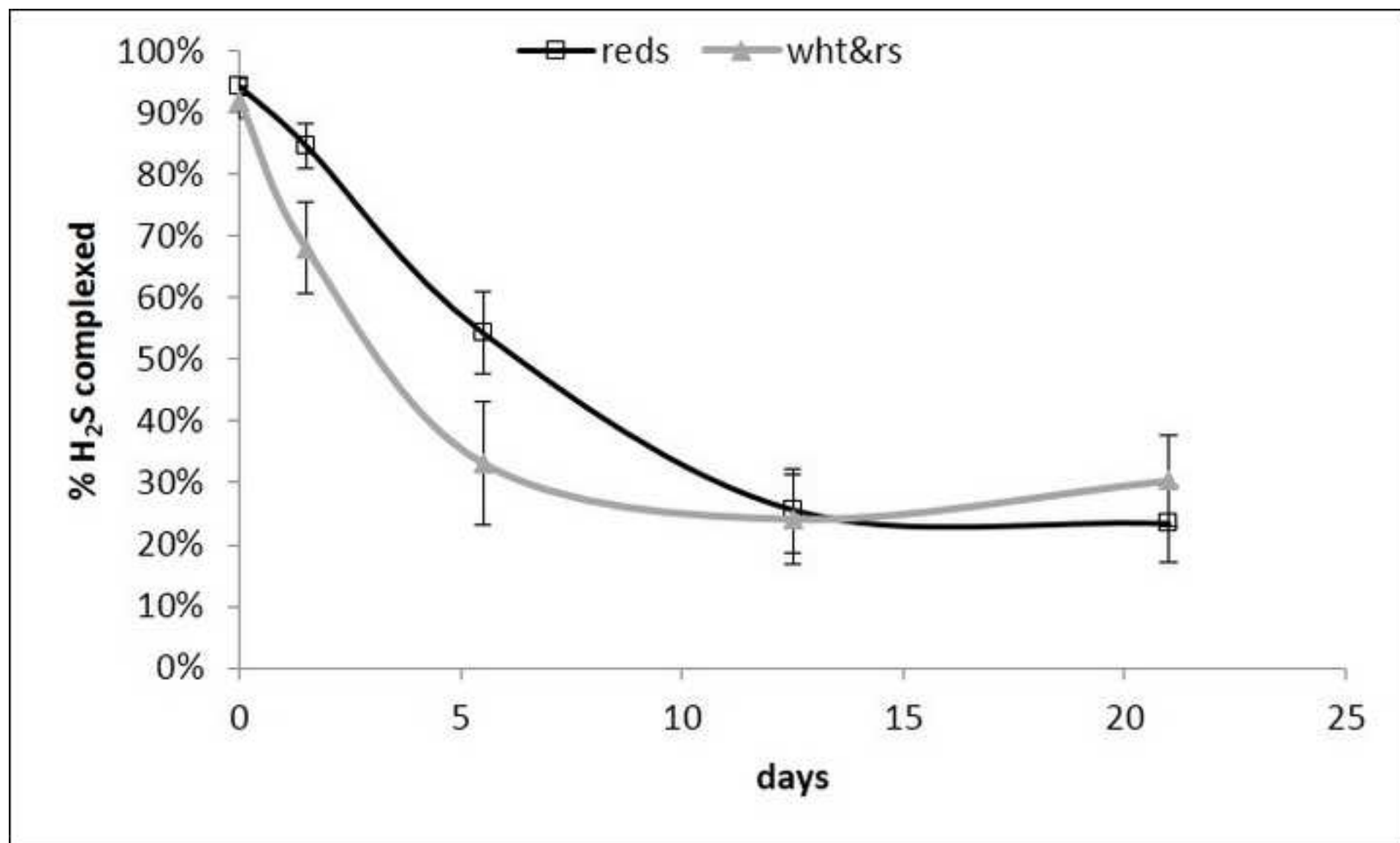
Figure(s) 1a



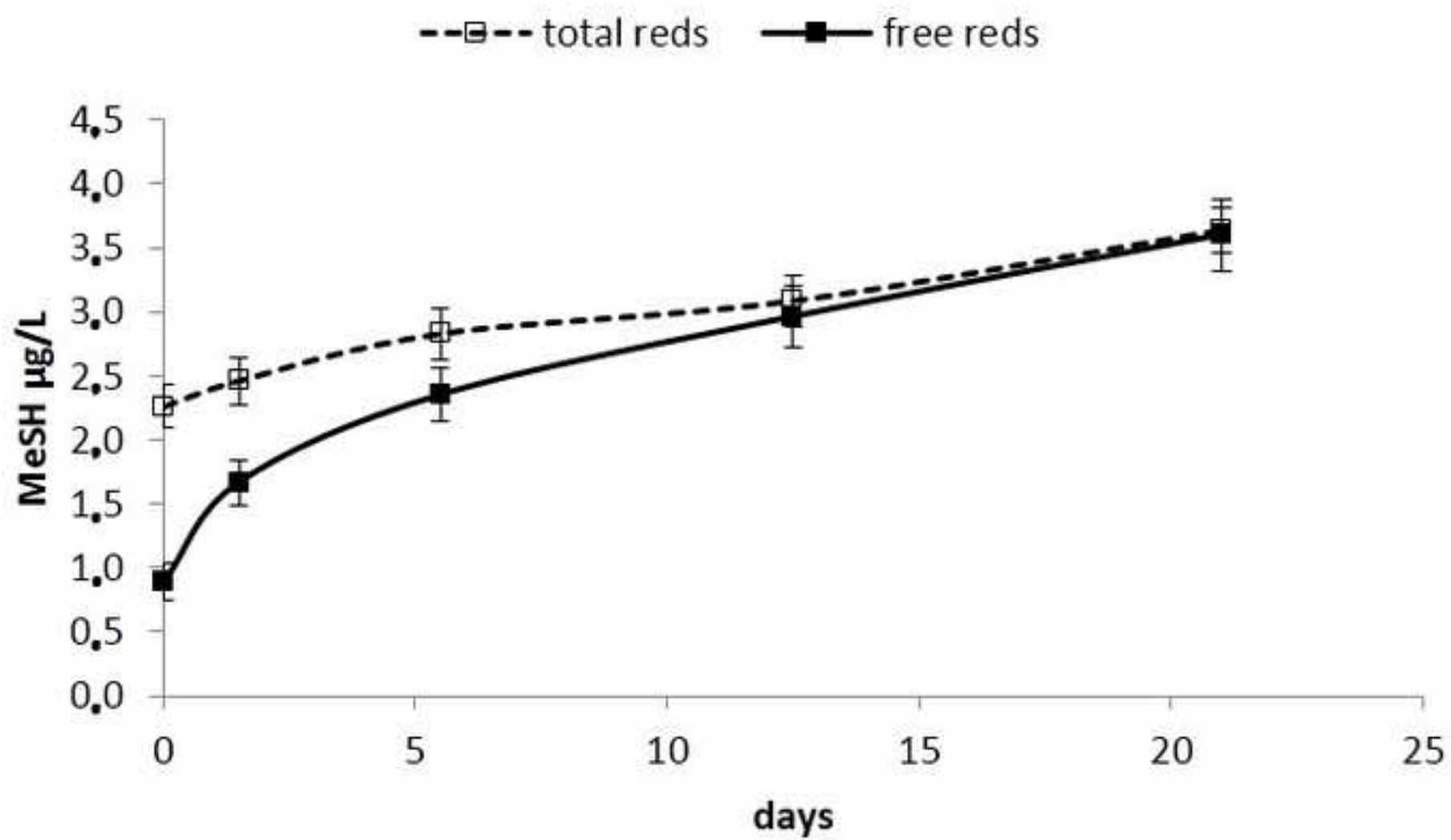
Figure(s) 1b



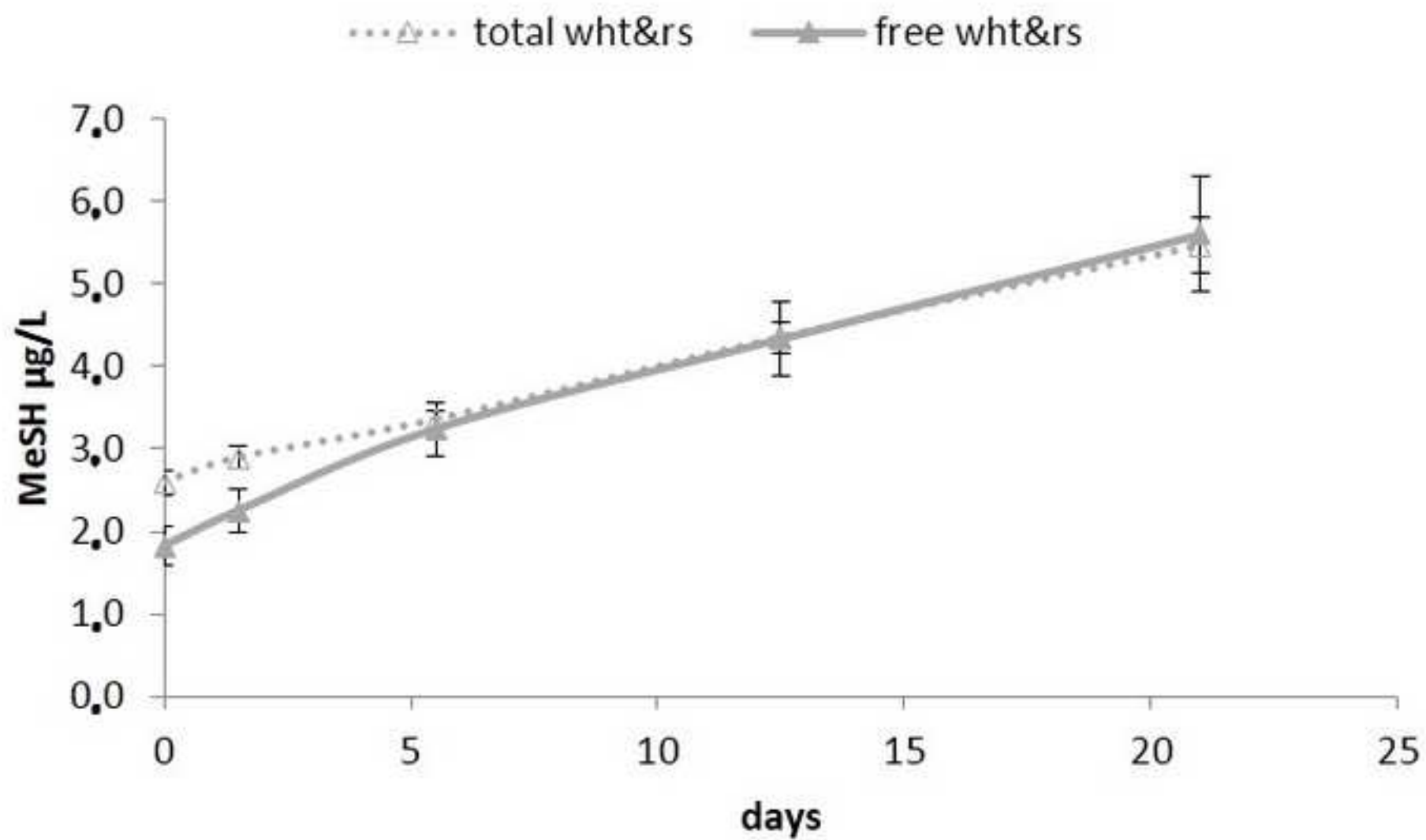
Figure(s) 1c



Figure(s) 2a



Figure(s) 2b



Figure(s) 2c

