

27110 - Physical Chemistry

Información del Plan Docente

Academic Year	2016/17
Academic center	100 - Facultad de Ciencias
Degree	446 - Degree in Biotechnology
ECTS	6.0
Course	2
Period	First semester
Subject Type	Compulsory
Module	---

1.Basic info

1.1.Recommendations to take this course

1.2.Activities and key dates for the course

For students enrolled in the subject, places, times and dates of lectures and practical sessions will be public via Bulletin Board advertisements of the grade on the platform Moodle at the University of Zaragoza , <https://moodle2.unizar.es/add/>, and in the moodle page for the course. These routes will be also used to communicate enrolled students their distribution by groups of practical sessions, which will be organized by the coordination of degree. Provisional dates will be available on the website of the Faculty of Sciences in the corresponding section of the Degree in Biotechnology : <https://ciencias.unizar.es/grado-en-biotecnologia> .

In this web there will be also available the dates of exams.

2.Initiation

2.1.Learning outcomes that define the subject

2.2.Introduction

3.Context and competences

3.1.Goals

3.2.Context and meaning of the subject in the degree

3.3.Competences

3.4.Importance of learning outcomes

4.Evaluation

27110 - Physical Chemistry

5. Activities and resources

5.1. General methodological presentation

The learning process for this subject is based on the following:

Activity 1: Acquisition of theoretical knowledge of Physical Chemistry (3 ECTS) in participatory lectures.

Activity 2: Classes solving exercises and seminars (1.5 ECTS) in small groups.

Activity 3: Laboratory practice (1.5 ECTS).

5.2. Learning activities

Training activity 1. Lectures

Training activity 2. Exercises

Training activity 3. Laboratory

5.3. Program

THEORY PROGRAM

The kinetics of reactions

The definition of reaction rate and empirical chemical kinetics. Measurement of reactions rate. Integrated rate laws: simple and complex reactions. The determination of the rate law.

Reaction mechanisms. The temperature dependence of reaction rates.

Unimolecular and trimolecular reactions. Reactions in solution.

Homogeneous catalysis. Enzymatic catalysis.

Transport phenomena.

Transport properties. Diffusion, dialysis and ultracentrifugation. Transport across biological membranes. Sedimentation. Viscosity; rheology. The conductivities of electrolyte solutions.

Thermodynamic.

Fundamentals. The first law; Thermochemistry. The second law; Entropy. The

Helmholtz and Gibbs energies. Variations of thermodynamic functions. The chemical potential of a pure substance.

27110 - Physical Chemistry

Phase equilibria for pure and mixtures fluids. Phase diagrams. Solutions. Partial molar quantities. Ideal and no-ideal solutions. Vapour-liquid equilibrium. Distillation. Liquid-liquid equilibrium. Colligative properties.

Chemical equilibrium. Equilibrium constant. The response of equilibria to the conditions.

Equilibrium electrochemistry.

Electrochemical cells. The electromotive force. Electrodes; standard potentials. Reversible electrodes. Liquid-liquid junction potential. Selective membrane electrodes. Potentiometric measurements: Applications. Electrokinetic phenomena; electrophoresis.

Surfaces.

Interfaces and surface tension. Pressure inside bubble, cavities and drops. Capillarity. Adsorption of gases on solid surfaces. Physical adsorption and chemisorption. Adsorption isotherms. adhesion and cohesion.

LABORATORY PROGRAM

1. Acid-catalized inversion of sucrose. Study of the reaction by polarimetric measurements.
2. Enzyme inhibition and poisoning
3. Absorption spectrum of p-methoxyphenol. Calculation of its acid dissociation constant.
4. Measurement of the electromotive force cell.

5.4.Planning and scheduling

Schedules of lectures and problems will coincide with the officially established and will be available at:
<https://ciencias.unizar.es/grado-en-biotecnologia> .

The places, calendar and groups for training and practical sessions will be established in coordination with the rest of maters at beginning of course. The Coordinator will produce the groups of students for these activities at beginning of course to avoid overlaps with other subjects. Exercises groups will be distributed at the beginning of the course.

5.5.Bibliography and recomended resources