

Información del Plan Docente

Academic Year	2016/17
Academic center	100 - Facultad de Ciencias
Degree	452 - Degree in Chemistry
ECTS	9.0
Course	2
Period	Annual
Subject Type	Compulsory
Module	---

1.Basic info**1.1.Recommendations to take this course****1.2.Activities and key dates for the course****2.Initiation****2.1.Learning outcomes that define the subject****2.2.Introduction****3.Context and competences****3.1.Goals****3.2.Context and meaning of the subject in the degree****3.3.Competences****3.4.Importance of learning outcomes****4.Evaluation****5.Activities and resources****5.1.General methodological presentation****5.2.Learning activities**

The learning process for this subject is based on the following:

60 hours of Lectures.

30 hours of Problem-based learning. The group is divided into two subgroups.

20 hours of mentoring work.

5.3.Program

Part I: Aim, methodology and purpose of Analytical Chemistry.

Chapter 1: Introduction to Analytical Chemistry.

Chapter 2- The analytical process.

Chapter 3- Quality assurance in the analytical process.

Chapter 4- Equilibrium chemistry in Analytical Chemistry.

Part II: General operations in the analytical process.

Chapter 5- Sampling.

Chapter 6- Sample preparation.

Chapter 7- Sample dissolution.

Chapter 8- Calibration.

Chapter 9- Evaluation of analytical data.

Part III. Chemical methods of analysis.

Chapter 10- Gravimetric methods.

Chapter 11- Overview of titrimetric methods.

Chapter 12- Acid-base titrations.

Chapter 13- Precipitation titrations.

Chapter 14- Redox titrations.

Chapter 15- Complexation titrations.

Part IV- Electroanalytical methods.

Chapter 16- Overview of electroanalysis.

Chapter 17- Potentiometric methods.

Chapter 18- Voltammetric methods.

Chapter 19- Other electroanalytical methods

5.4. Planning and scheduling

Lectures, practical sessions and examination dates will follow the scheduling fixed by the Science Faculty, which is published in its website (<https://ciencias.unizar.es/calendario-y-horarios>) and in the learning platform Moodle within the *Química Analítica I* course.

5.5. Bibliography and recommended resources

BB	Consultar información/recursos incorporados en el ADD de la asignatura.
BB	Harris, Daniel C.. Análisis químico cuantitativo / Daniel C. Harris . - 3 ^a ed. Barcelona [etc.] : Reverté, cop. 2007
BC	Belarra Piedrafita, Miguel Ángel. Cálculos rápidos para los equilibrios químicos en disolución / Miguel Angel Belarra Piedrafita Zaragoza : Prensas Universitarias de Zaragoza, 2002
BC	Skoog, Douglas A.. Principios de análisis instrumental / Douglas A. Skoog, F. James Holler, Stanley R. Crouch ; traductor, María Bruna Josefina Anzures ; revisión técnica Francisco Rojo Callejas, Juan Alejo Pérez Legorreta . - 6 ^a ed. México, D. F. : Cengage Learning, cop. 2008
BC	Valcárcel Cases, Miguel. Principios de química analítica / Miguel Valcárcel Barcelona : Springer-Verlag Ibérica, D.L. 1999

Online resources:

Analytical Chemistry 2.0 [Bibliografía

Básica de la asignatura] -

[<http://www.asdlib.org/onlineArticles/ecourseware/Analytical%20Chemistry%20200>]

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