

### Información del Plan Docente

Academic Year 2016/17

Academic center 100 - Facultad de Ciencias

**Degree** 543 - Master's in Molecular Chemistry and Homogeneous Catalysis

**ECTS** 6.0

Course

Period First semester

Subject Type Compulsory

Module ---

### 1.Basic info

### 1.1.Recommendations to take this course

The basic knowledge about chemical structure, bonding and reactivity of coordination and organometallic compounds is highly recommended.

Class attendance along with continued work facilitates to pass the course.

## 1.2. Activities and key dates for the course

The subject *Molecular Design in Inorganic and Organometallic Chemistry* will be taught in the first half term, in addition to the other 3 compulsory subjects of the Master and the optional subjects *Basic methodologies in synthesis* and *Bibliographic resources and databases.* Throughout the course some individual or team-based controls, as well as presentations of some literature works, will be carried out in order to delve into some issues. The presentation dates will be communicated in advance.

The laboratory section of the subject together with those corresponding to other subjects of the module *Molecular Chemistry and Catalysis* constitute an integrated block. Laboratory sessions will be held in the second half of the term. The dates and place will be announced well in advance.

### 2.Initiation

### 2.1.Learning outcomes that define the subject

The student must apply basic concepts of coordination chemistry to the synthesis of complexes that contain small molecules.

The student should evaluate the possible modes of coordination, bonding, properties, reactivity and potential applications.

The student must know and apply concepts of metal-metal bonding to di- or polynuclear complexes and clusters.



The student must identify the methods of synthesis, the utility and the applications of nanoparticles.

The student must know the principles and characteristics of the different M-C bonds and recognize the different families of organometallic compounds.

The student must know the usual methods in the synthesis of organometallic complexes and their general properties.

The student must predict the stability and reactivity of the different types of transition metal organometallic complexes and be able to propose methods of synthesis.

The student must be able to use the information provided from spectroscopic and analytical techniques to the characterization of the organometallic complexes.

The student must solve problems and discuss critically questions about structure and reactivity of organometallic complexes.

The student must recognize the utility of some organometallic compounds for the synthesis of organic molecules and as catalysts in chemical processes.

### 2.2.Introduction

Molecular Design in Inorganic and Organometallic Chemistry is a subject dedicated to the study of synthesis, bond systems and models, properties and reactivity of coordination and organometallic complexes, being them mono- or polynuclear, and containing or not metal-metal bonds. This subject also includes the study of the cluster and nanoparticles.

A more detailed study will be paid to the organometallic complexes because of their greater involvement in the homogeneous catalysis. process The complexes will be classified for the study in two main groups: complexes with sigma C-M bonds, and complexes with C-M sigma-pi and pi-cyclic bonds. The study pays attention to the structural aspects, the unusual reactivity and also to the current applications: therapeutic properties, luminescent character and possible future developments.

## 3.Context and competences

#### 3.1.Goals

This subject is one of the central topics in the framework of the *Master in Molecular Chemistry and Homogeneous Catalysis*, because provides knowledge of coordination compounds and organometallic complexes, including clusters and nanoparticles, about synthesis, bond models, properties, reactivity and current applications like the catalytic processes that are studied in another subjects as *Catalysis*, *Asymmetric Catalysis*, *Sustainable Chemistry and Catalysis* and *Supramolecular Chemistry*.

One of the main goals of the subject is that students increase their knowledge in Coordination Chemistry to be able to evaluate the effect of different complexes over some small molecules, some of which are widely used in catalytic reactions and processes. In addition, the student will realize how organic compounds change its reactivity when act as ligands bonded to the metal centers. The knowledge of the different types of M-C bond, depending on the organic groups



involved, its structure and related properties will provide to the students a global perspective of the Organometallic Chemistry of the transition elements.

# 3.2. Context and meaning of the subject in the degree

The subject *Molecular Design in Inorganic and Organometallic Chemistry* is included in the compulsory module of the *Master in Molecular Chemistry and Homogeneous Catalysis*. This subject is taught in the first half term of the course and it has assigned 6 ECTS credits: 4 ECTS for lectures, 1.5 ECTS for solving problems sessions with student participation, and 0.5 ECTS credits to be developed in the laboratory in coordination with other subjets of the module *Molecular Chemistry and Catalysis*. The program of the subject extend the knowledge acquired in the subjects of Inorganic Chemistry along the Degree in Chemistry and is essential for understanding the modifications in the properties and reactivity of molecules or organic fragments caused by the presence of metal, thus influencing processes of Organic Chemistry, Biochemistry and Catalysis.

## 3.3.Competences

To know and predict the synthesis and stability of coordination compounds.

Capacity to recognize and make sense to the modifications suffered by the molecules coordinated to a metal center and expect its new reactivity.

Capacity to recognize the presence of metal-metal bonds in polynuclear coordination compounds or clusters.

To be able to select the method for the preparation of metal nanoparticles and for the prediction of potential applications.

To use the nomenclature and specific terminology within the framework of Inorganic Chemistry.

Capacity to recognize the different types of organometallic complexes and predict their stability.

To redict the different reactivity of organometallic complexes.

To be able to select the synthetic methods to prepare organometallic complexes of transition metals.

To be able to design and synthesize new organic, inorganic or organometallic molecules that could be of industrial and technological interest.

To be able to select and use the analytical and spectroscopic data got from standard techniques to elucidate the composition and structure of organometallic complexes.

To understand, explain and study the mechanisms of stoichiometric and catalytic reactions.

To apply methods, procedures and experimental techniques in advanced synthesis.

To know the applications of organometallic complexes in synthesis and catalysis.



To know the applications of coordination compounds and organometallic complexes in medicine.

To be able to assimilate and critically evaluate research results in Molecular Chemistry relating them with the theoretical knowledge.

## 3.4.Importance of learning outcomes

The learning results of this subject are very important because they will allow the graduate to predict the stability and reactivity of different types of coordination and and organometallic compounds. Also, the learning results will allow the graduate to propose methods of synthesis and get criteria to select the appropriate information from the analytical and spectroscopic techniques for the characterization of complexes. In summary, they will provide tools to the students to develop their creative ability to perform his job as a researcher and the knowledge to predict the usefulness of inorganic and organometallic complexes in different actual aspects of Science and Technology.

### 4.Evaluation

The evaluation of this course is based on the following activities, weighted as indicated:

- 1.- A mid-term examination containing theoretical, theoretic-practical questions and problems from the themes 1 to 5 of the program (**P1**). Students who get a mark higher than 5 points will avoid these themes in the final exam.
- 2.- Elaboration and presentation of a supervised individual or team-based practical work on a scientific paper (T1).

There will be a final exam at the end of the term containing theoretical and theoretic-practical questions or problems from the themes 6 to 10 of the program. (**P2**). Students who had not been done the control **P1** or who had not get a 5 points mark, must have an additional exam with questions from themes 1 to 5 (**P1'**), which will represent the 35 % of the final grade.

The final grade will be the best of the following notes:

NOTE 1= 0.35\*( **P1** or **P1'** )+ 0.25\* **T1** + 0.40\* **P2** 

NOTE 2= 0,45\* P1' + 0,55\* P2

The rating of the students in the second annual examination session will consist in a single written exam that cover all themes of theory, problems or laboratory sessions defined as learning activities.

The number of official examination calls per registration and their use will be subjected to the statements of the Regulation of Permanence in Master Studies and Regulation of the Learning Assessment . The latest document will also regulate the general design and scoring criteria of the assessment activities, as well as the exam schedules and timetable for the post-examination review.

### 5. Activities and resources

## 5.1.General methodological presentation



- 1. Theoretical classes (4 ECTS)
- 2. Problem-solving sessions and seminars (1.5 ECTS)
- 3. Laboratory sessions (0.5 ECTS)

# 5.2.Learning activities

Training activity of acquiring advanced knowledge in Inorganic and Organometallic Chemistry. This activity includes 40 hours of interactive teaching in the classroom in a large group.

Seminars and problem-solving sessions. The activity includes 15 classroom hours in which the students will work individually or in small groups several case studies and scientific papers related with the subject's content.

Laboratory sessions. This activity includes 5 hours in the chemistry laboratory. The laboratory sessions of this subject are of compulsory attendance and, along with other subjects of the *Chemistry and Molecular Catalysis* module, form an integrated block.

Guided Works.

Tutorials: the students will have 3 hours per week for individualized tutoring.

## 5.3.Program

The lecture topics to be covered are as follows:

**Chapter 1** .- **Fundamental concepts in Coordination Chemistry.** General considerations of transition metals. Structure and properties of coordination compounds. Electronic structure and bonding. Synthesis and reactivity of coordination compounds.

Chapter 2 .- Activation of small molecules by coordination to metal centers. Coordination of CO 2 to metal centers, reactivity. Carboxylation reactions. Complexes with NO and N 2 O, coordination modes of nitrogen and reduction processes. Coordination and activation of diatomic molecules. Inter - and intramolecular hydrogen bonds.

**Chapter 3** .- **Metal-metal bonds in coordination compounds.** Introduction and bonding types. Multiple metal-metal bond: bond order, strength and M-M distance. Preparation and reactivity of complexes with quadruple bond. Preparation and reactivity of complexes with triple bond. Complexes with quintuple bond. Other types of metal-metal bonds.

**Chapter 4** .- **Clusters Compounds.** Carbonyl Clusters of high and low nuclearity. The CO bond in metal carbonyls. Application of NAE and Wade rules to the structure of the clusters. Isolobular concept. Halide clusters. Synthesis of clusters and reactivity.

**Chapter 5** .- **Huge clusters and nanoparticles.** Classification of the nanoparticles according to the size and shape. Preparation, properties and applications of nanoparticles.



**Chapter 6** .- **Types of ligands.** Transition metal organometallic complexes: σ-donor ligands. Preparation of transition-metal-alkyl and -aryl compounds. Thermodinamic versus kinetic lability. Reactivity: Insertion reactions. Alkenyl and alkynyl complexes: synthesis, properties and applications.

**Chapter 7** .- **Transition metal complexes with M-C multiple bonds** . Transition metal carbene complexes. Types of carbenes: preparation, structure, binding and reactivity. Alenylidene and vinylidene complexes. Transition metal carbyne complexes: preparation, structure, bonding and reactivity. Applications.

Chapter 8 .- Transition metal complexes with sigma-pi M-C bonds (linear or cyclic but non aromatic ligands). Synthesis, structure, bonding, reactivity and applications of complexes with olefins and conjugated di-olefins. Alkyne complexes. Allyl and enyl derivatives. Synthesis, structure, bonding and reactivity. Applications.

Chapter 9 .- Transition metal complexes with sigma-pi M-C bonds (aromatic ligands). Transition metal complexes with aromatic rings. Sandwich and semisandwich complexes. Complexes with three or four membered aromatic rings. Cyclopentadienyl derivatives. Binary cyclopentadienyl complexes. Metal complexes with benzene or its derivatives as ligands. Bis(arene) metal complexes. Semisandwich arene metal carbonyls. Complexes with seven or eight membered aromatic rings. Synthesis, properties and applications.

**Chapter 10.- Medical therapies with metal complexes.** Essential elements. Chelation therapy. Diagnostic agents. Radiopharmaceuticals. Therapeutic agents (anticancer, antibiotic, anti-arthritis).

## 5.4. Planning and scheduling

The information about schedules, calendars and exams is available at the websites of the Sciences Faculty, <a href="https://ciencias.unizar.es/calendario-y-horarios">https://ciencias.unizar.es/calendario-y-horarios</a>, and the Master, <a href="http://masterqmch.unizar.es">http://masterqmch.unizar.es</a>. The presentation of papers will be held according to the schedule that will be announced well in advance.

The students will be provided with different scholar material either at reprography or through the University's web tool: <a href="https://moodle2.unizar.es/add">https://moodle2.unizar.es/add</a>.

## 5.5.Bibliography and recomended resources

### **Basic Literature:**

- Activation of small molecules: organometallic and bioinorganic perspectives / edited by William B. Tolman . Weinheim : Wiley-VCH, cop. 2006
- Multiple bonds between metal atoms / edited by F. Albert Cotton, Carlos A. Murillo and Richard A. Walton . 3rd ed. New York : Springer Science and Business Media, 2005
- Organometallics: A Concise Introduction (3ª edición). Ch. Elschenbroich, A. Salzer. Ed. VCH. 2005.
- Molecular Clusters. A Bridge to Solid State Chemistry. T. P. Fehlner, J. F. Malet, J. -Y. Saillard. Ed. Cambridge University Press. 2007.
- Nanoparticles: from theory to application / edited by Günther Schmid Weinheim: Wiley-VCH, cop. 2004
- The organometallic chemistry of the transition metals / Robert H. Crabtree . 5a ed. Hoboken, N.J. : Wiley, 2009.
- Organotransition metal chemistry / Anthony F. Hill Cambridge: Royal Society of Chemistry, cop. 2009
- Uses of inorganic chemistry in medicine / edited by Nicholas P. Farrell . [1st ed.] Cambridge : Royal Society of Chemistry, 1999.
- Metallotherapeutic drugs and metal-based diagnostic agents: the use of metals in medicine / Marcel Gielen, Edward R. T. Tiekink Chichester: John Wiley & Sons, cop. 2005.
- Optoelectronic properties of inorganic compounds / edited by D. Max Roundhill and John P. Fackler, Jr. New York [etc]: Plenum Press, cop. 1999



#### **Further Literature:**

- Metal-ligand multiple bonds: the chemistry of transition metal complexes containing oxo, nitrido, imido, alkylidene, or alkylidyne ligands / William A. Nugent, James M. Mayer. [1st ed.] New York [etc.]: John Wiley & Sons, cop. 1998
- Cluster chemistry: introduction to the chemistry of transition metal and main group element molecular clusters / Guillermo González-Moraga. - [1a ed.] Berlin [etc.]: Springer-Verlag, cop. 1993
- Molecular clusters: a bridge to solid-state chemistry / Thomas P. Fehlner, Jean-François Halet, Jean-Yves Saillard Cambridge: Cambridge University Press, 2007
- Curso de iniciación a la química organometálica / Gabino A. Carriedo Ule, Daniel Miguel San José . [1ª ed.]
  Oviedo : Universidad de Oviedo, Servicio de Publicaciones, D.L.1995
- Organometallic chemistry / Gary O. Spessard, Gary L. Miessler New Jersey: Prentice-Hall, cop. 1997
- Organometallic chemistry of the transition elements / Florian P.Pruchnik; translated from polish by Stan A.Duraj . [1st ed.] New York [etc.]: Plenum Press, cop. 1990
- Organometallics. 1, Complexes with transition metal- carbon [s]-bonds / Manfred Bochmann . [1st ed. repr.] Oxford [etc.] : Oxford University Press, 2000
- Organometallics. 2, Complexes with transition metal- carbon [p]-bonds / Manfred Bochmann . 1st ed. repr. Oxford [etc.] : Oxford University Press, 2000
- Química organometálica con ejercicios corregidos / Didier Astruc ; traducción, Carmen Claver, Beatriz Alonso Barcelona : Reverté, D.L. 2003
- Metal compounds in cancer therapy / edited by Simon P. Fricker . 1st ed. London : Chapman & Hall, 1994
- Cisplatin: chemistry and biochemistry of a leading anticancer drud / Bernhard Lippert, ed. Weinheim [etc.]: Wiley-VCH, cop. 1999
- Metals in medicine / James C. Dabrowiak Hoboken: Wiley, cop. 2009
- Highly efficient OLEDs with phosphorescent materials / edited by Hartmut Yersin Weinheim: Wiley-VCH, cop. 2008
- Transition metals in the synthesis of complex organic molecules / Louis S. Hegedus . 3rd ed. Sausalito, California : University Science Books, cop. 2010

Recommended web-sites: http://www.ilpi.com/organomet