

## 27216 - Fundamentals of Chemical Engineering

### Información del Plan Docente

Academic Year	2018/19
Subject	27216 - Fundamentals of Chemical Engineering
Faculty / School	100 - Facultad de Ciencias
Degree	452 - Degree in Chemistry
ECTS	6.0
Year	3
Semester	First semester
Subject Type	Compulsory
Module	---

### **1.General information**

#### **1.1.Aims of the course**

#### **1.2.Context and importance of this course in the degree**

#### **1.3.Recommendations to take this course**

### **2.Learning goals**

#### **2.1.Competences**

#### **2.2.Learning goals**

#### **2.3.Importance of learning goals**

### **3.Assessment (1st and 2nd call)**

#### **3.1.Assessment tasks (description of tasks, marking system and assessment criteria)**

### **4.Methodology, learning tasks, syllabus and resources**

#### **4.1.Methodological overview**

The course consists of 60 hours of which 48 hours will be devoted to lectures and problems and the remaining 12 hours will be used to carry out laboratory practices. The activities of laboratory practices will be carried out in Laboratory 1 of Building D of the Faculty according to the calendar that it was previously published. These practices will be carried out in teams of 2-3 students, according to the number of students enrolled. The questionnaires of the laboratory practices with the questions related to the development of the same and the data obtained will be delivered at the end of the corresponding practice. The score obtained is the same for the whole team. The two written exams during the continuous or global evaluation will be focused on the contents reviewed during the course and those exams will be announced in the classroom, in the bulletin board of the Department of Chemical Engineering and Environmental Technologies and in moodle with the less two weeks in advance. The location and timetable for office hours (advisory meetings) will be established by each teacher and will be made public at the beginning of the course in the classroom, on the bulletin board

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of the Department of Chemical Engineering and Environmental Technologies and in moodle. The dates for the global evaluation test in first and second calls will be in accordance with the academic calendar of the Faculty of Sciences and will be available on its website : <http://ciencias.unizar.es/web/horarios.do>. This link will be updated at the beginning of each academic year.

### 4.2.Learning tasks

The course includes the following learning tasks:

- Section I. Introduction. Balances of Matter and Energy in Steady State 9 h Lectures (theoretical classes and problem solving and cases) 10 hours Classes of problem solving of Balances of Matter and Energy The 9 hours of lectures will be devoted:
  - o Introduction to Chemical Engineering.
  - o Basic nomenclature and calculation methods. Unit systems. Dimensional analysis. Conversion of units.
  - o Macroscopic balances of matter and energy in processes of the chemical industry.
  - o Systematic procedure for the realization of balances of matter in steady state with and without chemical reaction.
  - o Balances of steady state energy with and without chemical reaction
  - o Simultaneous balances of matter and energy in steady state. 2.
- Section II. Transport Phenomena, Application to Equipment Design and Introduction to Separation Operations and Reactor Design 21 hours Lectures (theoretical classes and problem solving and cases) 8 hours Class of problem solving 12 h Laboratory practices in small groups (teams of 2-3 students) The 21 h of lectures that will be devoted to:
  - o Introduction to transport phenomena.
  - o Transport mechanisms. Transport equations within a fluid in molecular regime. Boundary layer theory.
  - o Transportation between phases. Individual and global transport coefficients.
  - o Application to the design of heat exchangers.
  - o Basics of separation operations. Equipment for contact between phases.
  - o Introduction to the design of absorption columns.
  - o Introduction to reactor design. Applied Chemical Kinetics.
  - o Isothermal and Adiabatic Discontinuous stirred tank Reactor.
  - o Isothermal and Adiabatic Continuous stirred tank Reactor.
  - o Continuous Isothermal and Adiabatic plug flow Reactor.

The 12 h of laboratory practices will be distributed as follows: Each laboratory team consisting of 2-3 students will perform three of the five proposed practices according to the planning carried out by the teachers responsible for the course. The laboratory sessions will be 4 hours long (including the time to solve the questionnaire) and will be done towards the end of the semester. The practices corresponding to this Section are:

- Practice 1: Transfer of matter between phases: Absorption-Desorption G-L. Determination of individual coefficients of matter transfer.
- Practice 2. Transfer of matter between phases: Absorption with chemical reaction.
- Practice 3. Transfer of matter between phases: Extraction S-L. Contact mode, number of stages and separation factor.
- Practice 4. Transfer of matter between phases: Discontinuous distillation.
- Practice 5: Ideal Piston Flow Reactor. Influence of the operating conditions on the conversion. and Ideal Perfect Mixing Reactor. Association in Series of Ideal Reactors.

Thus, the total 60 hours taught in the course include 30 hours of lectures in the classroom with the entire group, 18 hours of problem solving and 12 hours of laboratory practices. The work of the student is distributed among the hours of study, problem solving, preparation of the practices, and written evaluation tests.

### 4.3.Syllabus

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The course is divided in two sections. The following syllabus is intended to help the student the consecution of his/her formative training throughout the following activities:

### Section 1: Introduction. Mass and energy balances in steady state.

	Hours present	Activity
<b>1.9 ECTS</b>	9 h	Master class (theoretical classes and exercises)
	10 h	Exercises

The 9 h of master classes will include:

- Chemical Engineering Introduction
- Nomenclature and unit systems; dimensional analysis; units conversion.
- Mass and energy balances in steady state in chemical processes.
- Mass balances in steady state with and without chemical reaction.
- Simultaneous resolution of mass and energy balances in steady state.

### Section 2: Transport phenomena. Unit Operations and Processes. Reactor Design.

	Hours present	Activity
<b>4.1 ECTS</b>	21 h	Lectures
	8 h	Exercises
	12 h	Lab practices (2 people groups)

The 21 h of master classes will include:

- Introduction to Transport Phenomena
- Transport mechanisms. Transport equations in laminar flow regime. The boundary layer.
- Individual and global transport coefficients.
- Heat exchanger design
- Fundamentals of separation processes. Distillation
- Design of absorption and stripping towers.
- Reactor design. Chemical reaction kinetics.
- Discontinuous reactors
- The continuous plug flow reactor model
- The continuous flow stirred-tank reactor

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12 h of laboratory practices will be distributed as follows:

Each couple will carry out 3 laboratory practices, two related to Section 1 (laboratory practices 1a to 4a) and one related to section 2 (laboratory practices 5).

Each laboratory practice will last 2,5h. Each couple will carry out 2 practices from the 4 included in Section 1:

- Practice 1a: Gas/liquid absorption/desorption. Determination of individual mass transport coefficients.
- Practice 2a: Ion exchange. Determination of the breakthrough curve.
- Practice 3a: Extraction solid/liquid. Analysis of the contact mode, temperature and number of stages.
- Practice 4a: Discontinuous distillation.
- Practice 5a: continuous plug flow reactor model. Influence of the reaction conditions on the conversion. The continuous flow stirred-tank reactor. Reactors in series.

### 4.4.Course planning and calendar

Further information concerning the timetable, classroom, office hours, assessment dates and other details regarding this course, will be provided on the first day of class or please refer to the Facultad de Ciencias web (<https://ciencias.unizar.es/grado-en-quimica-0>).

Specific dates of the different activities will be announced during the classes, bulletin boards or by ADD (Moodle2 Platform).

### 4.5.Bibliography and recommended resources