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ORIGINAL RESEARCH COMMUNICATION

AU2 ▶ AU1 ▶

Redox- and Ligand Binding-Dependent Conformational Ensembles in the Human Apoptosis-Inducing Factor Regulate Its Pro-Life and Cell Death Functions

AU3 ▶ Raquel Villanueva,¹ Silvia Romero-Tamayo,¹ Ruben Laplaza,^{1,2} Juan Martínez-Olivan,¹ Adrián Velázquez-Campoy,^{1,3-5} Javier Sancho,^{1,4} Patricia Ferreira,¹ and Milagros Medina¹

Abstract

Aims: The human apoptosis-inducing factor (hAIF) supports OXPHOS biogenesis and programmed cell death, with missense mutations producing neurodegenerative phenotypes. hAIF senses the redox environment of cellular compartments, stabilizing a charge transfer complex (CTC) dimer that modulates the protein interaction network. In this context, we aimed to evaluate the subcellular pH, CTC formation, and pathogenic mutations effects on hAIF stability, and a thermal denaturation high-throughput screening (HTS) assay to discover AIF binders.

AU4 ▶ **Results:** Apoptotic hAIF_{Δ1-101} is not stable at intermembrane mitochondrial space (IMS) pH, but the 77–101 residues confer stability to the mitochondrial isoform. hAIF and its CTC populate different conformational ensembles with redox switch to the CTC producing a less stable and compact protein. The pathogenic G308E, ΔR201, and E493V mutations modulate hAIF stability; particularly, ΔR201 causes a population shift to a less stable conformation that remodels active site structure and dynamics. We have identified new molecules that modulate the hAIF NADH/NAD⁺ association/dissociation equilibrium and regulate its catalytic efficiency.

Innovation: Biophysical methods allow evaluating the regulation of hAIF functional ensembles and to develop an HTS assay to discover small molecules that might modulate hAIF stability and activities.

Conclusions: The mitochondrial soluble 54–77 portion stabilizes hAIF at the IMS pH. NADH–redox-linked conformation changes course with strong NAD⁺ binding and protein dimerization, but they produce a negative impact in overall hAIF stability. Loss of functionality in the R201 deletion is due to distortion of the active site architecture. We report molecules that may serve as leads in the development of hAIF bioactive compounds. *Antioxid. Redox Signal.* 00, 000–000.

Keywords: apoptosis-inducing factor, ligand/redox-linked allosteric conformation, thermal stability, pathogenic mutations, high-throughput screening

Introduction

AU5 ▶ **T**HE APOPTOSIS-INDUCING FACTOR (AIF) is a flavo-reductase that contributes to growth, development, and maintenance of mitochondrial function, being also a caspase-

independent programmed cell death (PCD) effector (29, 42, 43, 45, 64). Recent research is compiling evidence of the critical roles of the AIF redox state and of the redox-dependent monomer–dimer equilibrium in both functions. The human AIF (hAIF) precursor is synthesized in the

¹Departamento de Bioquímica y Biología Molecular y Celular, Facultad de Ciencias, Instituto de Biocomputación y Física de Sistemas Complejos (GBsC-CSIC and BIFI-IQFR Joint Units), Universidad de Zaragoza, Zaragoza, Spain.

²Departamento de Química Física, Universidad de Santiago de Compostela, Santiago de Compostela, Spain.

³Fundación ARAID, Diputación General de Aragón, Zaragoza, Spain.

⁴Aragon Institute for Health Research (IIS Aragon), Zaragoza, Spain.

⁵Biomedical Research Networking Centre for Liver and Digestive Diseases (CIBERehd), Madrid, Spain.

Innovation

The apoptosis-inducing factor (AIF) supports OXPHOS biogenesis and programmed cell death, sensing the redox environment, and undergoing redox-linked allosteric switches that modulate its interaction network with other biomolecules. Our study provides a comprehensive understanding of hAIF stability under several situations that mimic physiological conditions: subcellular pH, NADH-redox-linked conformational switches, and germline loss-of-function mutations. We also pave the way to identify hAIF binders that can modulate its stability or functionality in the treatment of hAIF X-linked rare diseases.

cytosol and imported into the intermembrane mitochondrial space (IMS), where the N-terminal region is enzymatically removed producing the hAIF $_{\Delta 1-54}$ mitochondrial mature form. The N-terminal transmembrane helix (residues ~60–78) tethers hAIF $_{\Delta 1-54}$ to the inner mitochondrial membrane, whereas the soluble protein moiety folds toward the IMS and incorporates the redox FAD (oxidized form of flavin adenine dinucleotide) cofactor (48, 64, 72). hAIF assists biogenesis of respiratory chain complexes through its NADH-oxidoreductase activity, by contributing to the IMS import of the coiled-coil-helix-coiled-coil-helix domain containing-4 protein (CHCHD4) (25, 41, 58, 66). Certain apoptotic stimuli induce a second N-terminal cleavage, yielding a soluble protein, hAIF $_{\Delta 1-101}$, that translocates to the cytoplasm, where it can interact with heat-shock protein 70 (Hsp70) (53), thioredoxin 1 (58), or endonuclease cyclophilin A (CypA) (10, 73). Finally, hAIF $_{\Delta 1-101}$ can translocate to the nucleus where it forms, upon recruiting CypA and histone H2AX, a DNA-degradosome that induces chromatinolysis (2, 10, 64).

Binding of NADH to hAIF elicits its reduction and the stabilization of a long-lived oxygen-stable FADH $^-$ NAD $^+$ charge transfer complex (CTC) (17, 18, 55, 56). This process triggers protein dimerization, uncouples a 50-residue region at the protein C-terminal apoptotic domain that becomes disordered in the CTC crystal structure (regulatory C-loop, 509–560), and creates a second noncatalytic and allosteric NADH binding site. In the oxidized enzyme, hAIF $_{ox}$, the C-loop 517–533 residues fold in two short α -helices, diminishing the isoalloxazine solvent accessibility (17, 18, 55, 56). The NADH-driven allosteric conformational changes propagate from the CTC active site through very specific molecular pathways that are claimed to control hAIF association with protein partners and DNA, and, as a consequence, its participation in mitochondrial homeostasis and PCD (8, 25, 57). Thus, hAIF is envisaged as a redox and/or NADH sensor that might modulate certain signal-transduction pathways (55, 62). We are also witnessing the discovery of an increasing number of hAIF allelic variants linked to a range of human rare diseases, mainly neurodegenerative mitochondrial disorders and inherited peripheral neuropathies (1, 4, 15, 22, 26, 30, 44, 50, 52, 74). Most of them (as $\Delta R201$, F210S, V243L, G262S, G308E, G338E, and Q479R) produce severe mitochondriopathies associated with substantial decrease of the CTC lifetime, reduced expression of respiratory chain complexes, and OXPHOS failure. In contrast, the E493V

mutation increases cell death *via* apoptosis, causing the Cowchock syndrome without affecting OXPHOS. Mutations are distributed through the hAIF structure, but the most deleterious ones tend to compromise structural elements involved in the binding of the adenine moieties of the NAD(H) coenzymes or the FAD cofactor (18, 57, 62). *In vitro* studies with some of these variants envisage defective folding, decreased stability, or higher propensity to FAD release (22, 57).

The NADH-driven allosteric behavior and the disease causing mutations indicate that the hAIF redox-dependent structural stability and conformation control its physiological functions. Thus, knowing how subcellular localization, ligand binding, redox state, or allelic mutations can regulate these parameters will help to understand the hAIF *in vivo* behavior. In this context, we evaluate here the hAIF stability under conditions that potentially mimic different conformational traits influenced by intracellular environment (subcellular compartments pH, NADH-redox-dependent state) and by missense pathogenic mutations. In addition, we develop and validate an FAD fluorescence-based thermal denaturation high-throughput screening (HTS) assay to search for compounds that might alter hAIF stability. To date, only one small molecule, aurintricarboxylic acid (ATA), has been identified to bind hAIF by using a photonic crystal biosensor screening (11). In this study, we propose some molecules that may serve as a starting point in the development of hAIF with low cellular toxicity.

Results

The pH of subcellular compartments modulates hAIF folding and conformation

hAIF functions in different subcellular compartments with particular pH environments that might influence its molecular properties. To evaluate this effect, we have selected pH 6.2 as typical of the IMS (49), 7.0 as representative for apoptotic cytosol (39), 7.4 for cytosol and nucleus (36, 54), and 8.0 for the mitochondrial matrix (36, 47), and physiological ionic strength. Within the pH 7.0–8.0 range, hAIF $_{\Delta 1-101ox}$ UV-visible spectra indicate that the FAD cofactor is in the oxidized state, properly incorporated, and in a similar electronic environment, whereas incubation with NADH brings about the FAD cofactor reduction with the appearance of a CTC band (Supplementary Fig. S1A; Supplementary Data ◀SF1 are available online at www.liebertpub.com/ars). At pH 6.2, the A_{280}/A_{450} and A_{380}/A_{450} ratios increase for hAIF $_{\Delta 1-101ox}$, a drift dominates the CTC spectrum, and turbidity is observed, facts indicative of low stability and some aggregation level. Since this IMS pH is relevant in our study, we have produced the hAIF $_{\Delta 1-77}$ isoform, lacking the membrane-anchoring helix but containing the IMS soluble portion. Spectra of hAIF $_{\Delta 1-77ox}$ confirm it is folded (Supplementary Fig. S1B). Henceforth, studies at pH 6.2 will take hAIF $_{\Delta 1-77}$ as IMS representative, whereas hAIF $_{\Delta 1-101}$ will be the model for other subcellular compartments. NADH presteady-state hAIF reduction kinetics corroborate CTC formation (hAIF $_{\Delta 1-101rd}$:NAD $^+$ and hAIF $_{\Delta 1-77rd}$:NAD $^+$ species (18)) with reduction rates within a factor of 1.5 (Supplementary Fig. S2 and Table 1). ◀SF2

hAIF $_{\Delta 1-101ox}$ and hAIF $_{\Delta 1-77ox}$ far-UV CD spectra are dominated by a broad negative band with minima at ~222 and

◀TI

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TABLE 1. EFFECT OF pH ON THE PRESTEADY KINETIC RATES FOR THE AEROBIC REDUCTION OF hAIF_{ox} BY NADH

Protein	pH	k_{obs} (s^{-1})		
		1 mM	[NADH] 2.5 mM	5 mM
hAIF _{Δ1-101ox}	8.0	0.53 \pm 0.01	0.88 \pm 0.02	1.17 \pm 0.01
	7.4	0.37 \pm 0.01	0.62 \pm 0.01	0.82 \pm 0.01
	7.0	0.35 \pm 0.03	0.63 \pm 0.01	0.84 \pm 0.01
hAIF _{Δ1-77ox}	6.2	0.60 \pm 0.05	0.85 \pm 0.07	1.10 \pm 0.05

Assays were performed in 50 mM phosphate buffer at a final ionic strength of 150 mM and 25°C. Final concentrations were $\sim 10 \pm 2 \mu M$ protein and 1, 2.5, or 5 mM for NADH. ($n=3$, $\pm SE$).

hAIF, human apoptosis-inducing factor; NADH, reduced nicotinamide adenine dinucleotide.

F1 \blacktriangleright ~ 208 nm (Fig. 1A), in agreement with its high α -helix content (18). Formation of the CTC slightly decreases the 208 nm minimum relative intensity (Fig. 1B), probably related to the C-loop two short α -helices unfolding (18). The hAIF _{Δ 1-101ox} near-UV/vis CD shows sharp (wider in hAIF _{Δ 1-77ox}) and broad maxima at ~ 300 and ~ 365 nm, respectively, as well as

minima at ~ 453 and ~ 477 nm, with minor pH effects (Fig. 1C). CTC formation results in lack of near-UV signal at 300 nm and of 350–500 nm features of the oxidized flavin, but shows bands at ~ 405 and ~ 600 nm consistent with CTC stabilization (Fig. 1D). Far-UV and near-UV signals are lost at denaturing temperatures (90°C), indicating that protein becomes unfolded.

Then, we have evaluated the pH effect on fluorescence emission (Supplementary Fig. S1C). WT hAIF_{ox} samples display emission maxima at ~ 337 nm with higher quantum yield at pH 8.0. CTC spectra are dominated by a ~ 460 nm band related to intrinsic NADH fluorescence, whereas the aromatic band intensity decreases due to energy transfer because of the tryptophan–coenzyme proximity. When exciting at flavin band I, fluorescence is particularly low, indicative of a strong quenching of flavin emission upon its binding to the protein. Denaturing temperatures increase fluorescent emission, consistent with flavin release.

Thermal stability of hAIF is highly modulated by formation of the CTC

Then, we have recorded thermal unfolding curves to evaluate the stability of WT hAIF _{Δ 1-101}, oxidized, and when

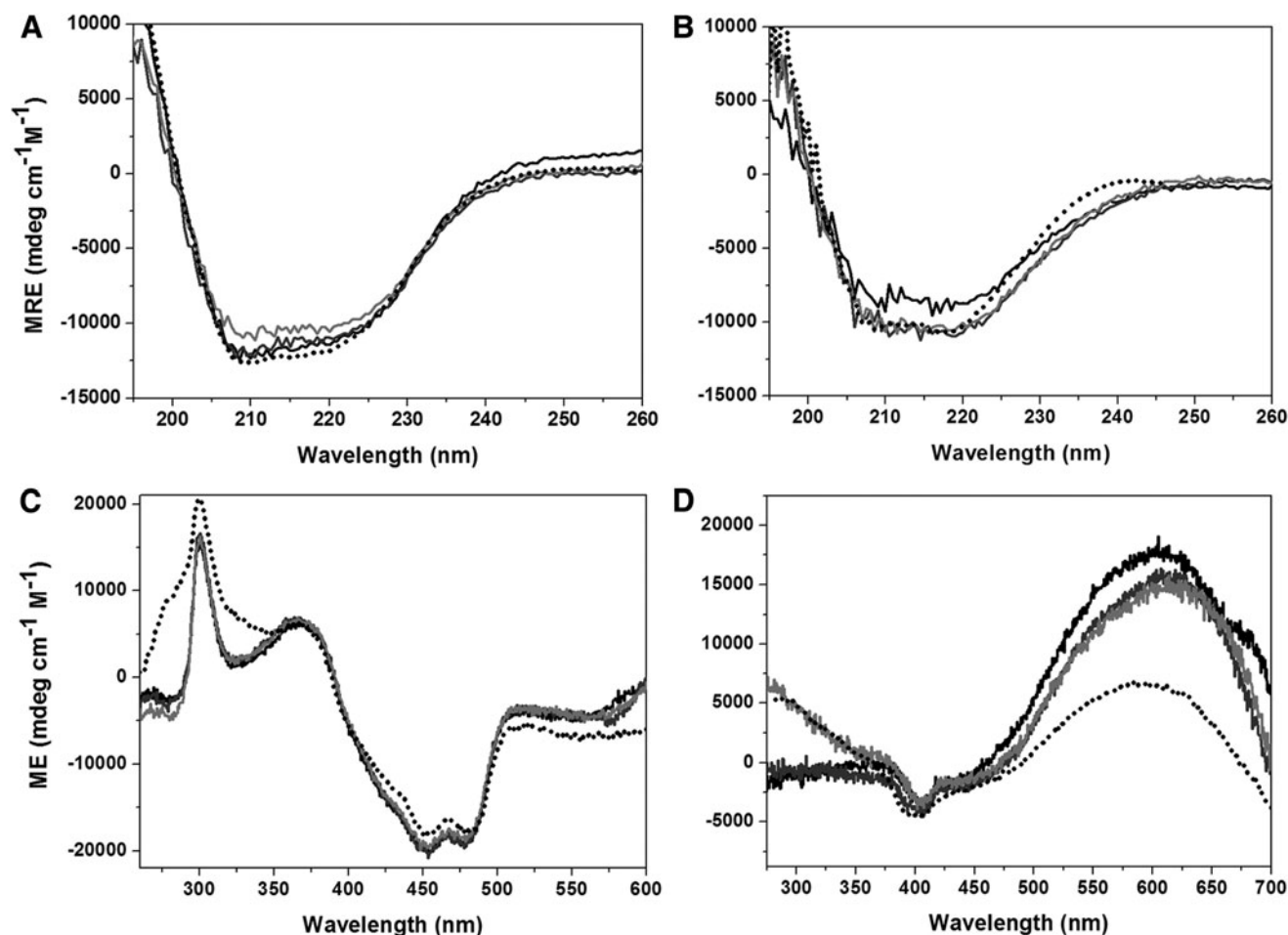


FIG. 1. Circular dichroism spectra of WT hAIF. Far-UV CD spectra of (A) 1 μM hAIF _{Δ 1-101ox} or hAIF _{Δ 1-77ox} (dashed line) and (B) a mixture of 1 μM hAIF _{Δ 1-101ox} or hAIF _{Δ 1-77ox} (dashed line) with 100 μM NADH. Near-UV/vis CD spectra of (C) 20 μM hAIF _{Δ 1-101ox} or hAIF _{Δ 1-77ox} (dashed line) and (D) a mixture of 20 μM hAIF _{Δ 1-101ox} or hAIF _{Δ 1-77ox} (dashed line) with 2 mM NADH. Spectra were recorded at 25°C in 50 mM potassium phosphate, pH 8.0 (black line), 7.4 (dark gray line), 7.0 (gray line), and pH 6.2 (black dotted line) at a final ionic strength of 150 mM. hAIF, human apoptosis-inducing factor.

forming the CTC, in the pH 7.0–8.0 range, as well as those of hAIF_{Δ1–77} at pH 6.2. Far-UV CD, near-UV/vis CD, and flavin fluorescence unfolding curves are shown in Figure 2A and B and Supplementary Fig. S3. hAIF_{Δ1–101ox} curves indicate a three-state unfolding mechanism (N ↔ I ↔ U), allowing determination of temperatures of midpoint denaturation (T_{m1} and T_{m2}) and unfolding enthalpy changes (ΔH_1 and ΔH_2) (Table 2). hAIF_{Δ1–101ox} unfolds *via* an intermediate (I) with T_{m1} revealed by the near-UV transition, and the flavin fluorescence change taking place essentially at T_{m2} . Therefore, the N ↔ I transition is caused by loss of tertiary interactions and the I state retains the isoalloxazine folded environment. Noticeably, a modestly lower population of I state is detected at pH 8.0 (mitochondrial matrix) when compared with cytosolic and nuclear pH, whereas T_{m1} is 3°C lower and the unfolding enthalpy change, ΔH_1 , is ~40% larger (Table 2). hAIF_{Δ1–77ox} unfolds *via* a four-state unfolding mechanism (N ↔ I₀ ↔ I ↔ U) at pH 6.2 (Supplementary Fig. S3C). Near-UV data suggest loss of overall tertiary structure at lower temperatures and stabilization of an additional intermediate (I₀, T_{m0}) that retains secondary structure and FAD binding, whereas T_{m1} (now I₀ → I transition) decreases compared with the apoptotic hAIF_{Δ1–101ox} values (Table 2). Finally, both pH 6.2 and the 77–101 tail have a thermal stabilizing effect for flavin being retained (T_{m2}), despite unfolding of secondary and tertiary elements starts at lower temperatures.

Formation of the dimeric hAIF_{Δ1–101rd}:NAD⁺ and hAIF_{Δ1–77rd}:NAD⁺ CTCs exerts an important impact on the overall protein unfolding mechanism and thermal stability. A two-state process is observed for hAIF_{Δ1–101rd}:NAD⁺, with T_m for CTC unfolding (T_{mCTC}) being ~7°C lower than T_{m1} and up to 13°C lower than T_{m2} (Table 2). However, a three-state process describes hAIF_{Δ1–77rd}:NAD⁺ CTC unfolding, with loss of tertiary interactions while retaining secondary structure and flavin binding. Thus, collectively, coenzyme binding and flavin reduction have a deleterious effect on the stability of N and, particularly, I states, favoring FAD release and unfolded conformers formation. Noticeably, the CTC unfolding enthalpy (ΔH_{CTC}) of hAIF_{Δ1–101ox} approximates the sum of ΔH_1 and ΔH_2 at pH 7.0 and 7.4, whereas at pH 8.0

ΔH_{CTC} is lower. To gain insight into the lower CTC stability relative to hAIF_{Δ1–101ox}, we built structural models containing the missing C-loop residues in the X-ray structure (18). Models were minimized and relaxed by short molecular dynamics (MD) simulations (Fig. 2 and Supplementary Figs. S4 and S5). Overlapping of hAIF_{Δ1–101ox} and CTC models along MD relaxation on starting models showed similar RMSD ($C\alpha$) values (Fig. 2G), with the slightly higher CTC value contributed by C-loop disorder and larger flexibility (Fig. 2H). Evolution of RMSDs for the core protein suggests that, with C-loop exception, both protein forms maintain the fold at 298 K as well as the active site architectures (Supplementary Fig. S4). The CTC structure shows larger radius of gyration, decrease in the H-bonds network, and slight increases in molecular and solvent accessible surface areas (Fig. 2I and Supplementary Fig. S5). This agrees with R201–E531 and W196–R529 interactions clapping the C-loop to the 196–201 β -hairpin in hAIF_{Δ1–101ox}, and contributing to 529–533 residues folding into an α -helix. Contrarily, in the CTC this region contributes to the C-loop disorder, making this ensemble less compact than the oxidized form and pointing out to this fact as the main reason for its lower stability.

Some allelic pathogenic mutations alter hAIF_{Δ1–101} stability

We have also evaluated the impact of missense $\Delta R201$, G308E, and E493V mutations on protein–FAD interaction integrity and hAIF_{Δ1–101} stability (Fig. 2E and Supplementary Fig. S6A–C). G308E and E493V mutations produce modest spectral perturbations (Supplementary Figs. S6 and S7), as expected from low impact in their crystal structures (50, 57, 61, 62). Nonetheless, small changes in shape and position of 380–400 nm features in their CTC vis-CD as well as low flavin fluorescence quenching in the G308E CTC are noticed (Supplementary Figs. S6H and S7D). In contrast, the R201 deletion clearly alters the protein spectroscopic properties: it modifies the flavin band I (Supplementary Fig. S6D), increases the fluorescence quantum yield of at least one tryptophan (Supplementary Fig. S6E), attenuates FAD

FIG. 2. Conformation of hAIF_{Δ1–101ox} and its CTC. Thermal unfolding curves of (A) hAIF_{Δ1–101ox} and (B) its CTC (1:100 protein hAIF_{Δ1–101ox}:NADH ratio) at pH 8.0. Thermal denaturation was monitored by far-UV CD (210 nm, *white circles*) and near-UV CD (300 nm, *black squares*) for hAIF_{Δ1–101ox}, or by far-UV CD (210 nm, *white circles*) and vis CD in the CTC (*in B*, 410 nm due to the lack of the 300 nm peak, *black squares*). Both samples were also monitored by flavin fluorescence emission (*black triangles*). The curves are shown roughly normalized from 0 to 1, and their global fits to two-transition (for A) and one-transition (for B) unfolding models are represented by the continuous lines. Curves were recorded in 50 mM potassium phosphate at a final ionic strength of 150 mM. Final snapshots of representative structural models for (C) WT hAIF_{Δ1–101ox} and (D) its CTC after modeling the missing C-loop residues (546–558 and 518–559, respectively), minimization routine, and 15 ns of MD relaxation at 300 K. Protein chain is represented as cartoon with the oxidoreductase domains in *violet* and its 191–202 β -hairpin in *hot pink* (R201 and W196 side chains in sticks), whereas the apoptotic domain is drawn in *light blue* with the C-loop (509–560 segment) in *orange*. FAD and NAD⁺ are shown in *sticks* with carbons in *orange* and *green*, respectively. Details of the active site environments in (E) hAIF_{Δ1–101ox} and (F) its CTC. Relevant residues are shown in *sticks*. Relevant interactions in the hAIF_{Δ1–101ox} model are shown in *blue dashed lines*. *Gray surfaces* show water accessible cavities in the active site environment. (G) RMSD ($C\alpha$) of hAIF_{Δ1–101ox} (*gray line*) and of its CTCs models along representative MD relaxations when overlapped on their corresponding starting models (after modeling of the missing C-loop residues and before minimization). *Red and blue lines*, respectively, represent the RMSD when considering only overlapping of the 128–508 core residues (388 $C\alpha$ atoms, *red line*). (H) RMSF of backbone atoms along a representative MD of the WT CTC model. Higher values for the *black line* in (G) are due mainly to the higher fluctuation in the position of the C-loop in the CTC. (I) Radius of gyration of hAIF_{Δ1–101ox} and of its CTC models along the MD production, colors as in (G). CTC, charge transfer complex; FAD, oxidized form of flavin adenine dinucleotide. To see this illustration in color, the reader is referred to the web version of this article at www.liebertpub.com/ars

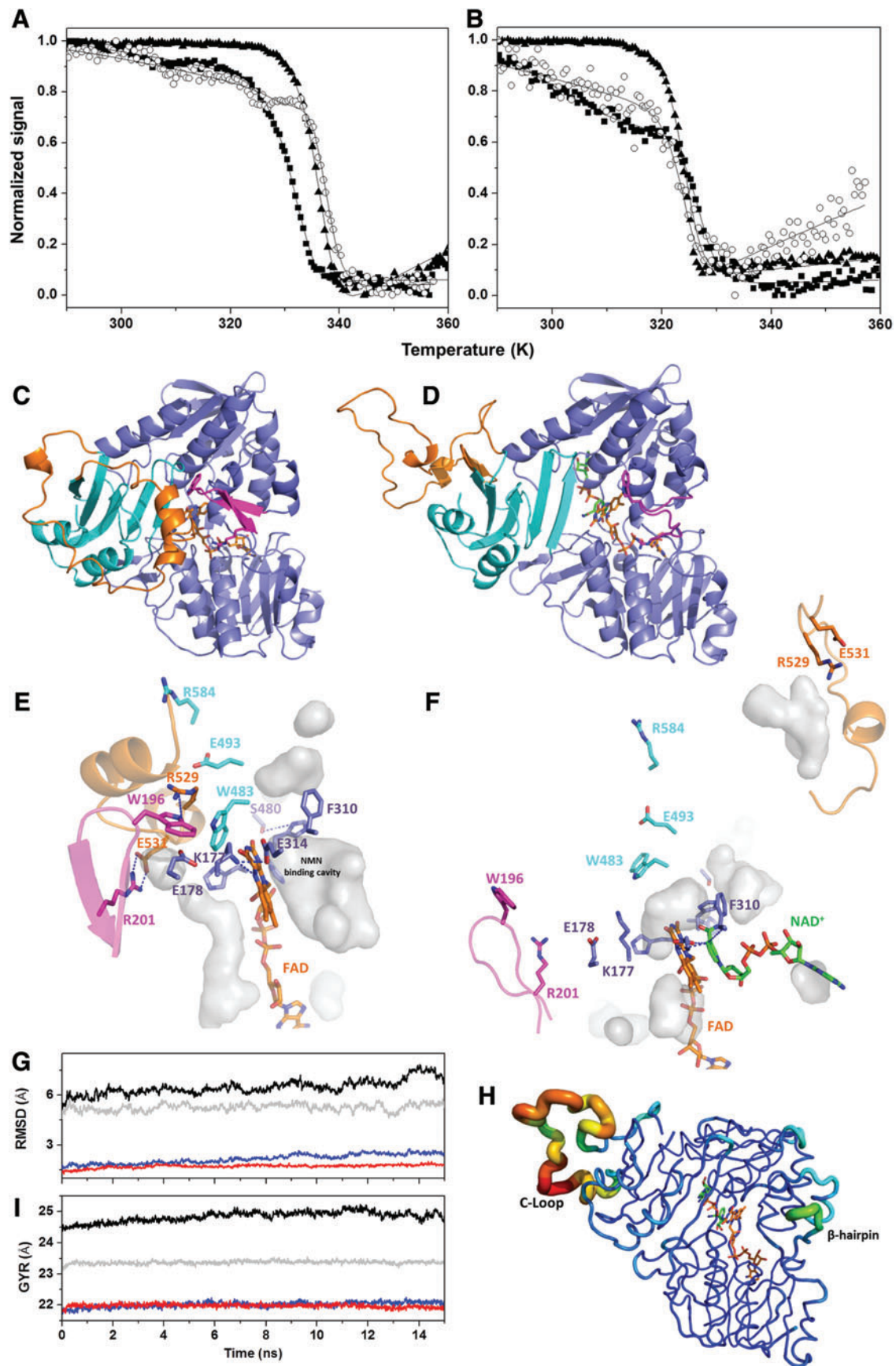


TABLE 2. EFFECT OF pH ON THE THERMAL STABILITY OF WILD TYPE hAIF $_{\Delta 1-101ox}$, hAIF $_{\Delta 1-77ox}$, AND OF THEIR CHARGE TRANSFER COMPLEXES

pH	hAIF $_{\Delta 1-101ox}$				CTC(hAIF $_{\Delta 1-101rd}$:NAD $^{+}$)			
	T_{m1} (K)	T_{m2} (K)	ΔH_1 (kcal/mol)	ΔH_2 (kcal/mol)	T_{mCTC} (K)	ΔH_{CTC} (kcal/mol)	$T_{mCTC}-T_{m1}$	$T_{mCTC}-T_{m2}$
8.0	332 \pm 1	337 \pm 1	87 \pm 6	140 \pm 10	324 \pm 1	120 \pm 10	-8 \pm 2	-13 \pm 2
7.4	334 \pm 2	336 \pm 1	52 \pm 8	110 \pm 10	327 \pm 1	180 \pm 10	-7 \pm 2	-10 \pm 2
7.0	337 \pm 1	338 \pm 1	46 \pm 3	110 \pm 10	327 \pm 1	180 \pm 10	-7 \pm 2	-11 \pm 2

pH	hAIF $_{\Delta 1-77ox}$						CTC (hAIF $_{\Delta 1-77rd}$:NAD $^{+}$)					
	T_{m0} (K)	T_{m1} (K)	T_{m2} (K)	ΔH_0 (kcal/mol)	ΔH_1 (kcal/mol)	ΔH_2 (kcal/mol)	T_{m0CTC} (K)	T_{mCTC} (K)	ΔH_{0CTC} (kcal/mol)	ΔH_{1CTC} (kcal/mol)	$T_{mCTC}-T_{m1}$	$T_{mCTC}-T_{m2}$
6.2	318 \pm 1	328 \pm 3	343 \pm 4	69 \pm 4	41 \pm 3	150 \pm 20	322 \pm 2	329 \pm 2	160 \pm 20	150 \pm 30	1 \pm 5	-13 \pm 4

Values were obtained by global fitting of the near-CD, far-UV CD, and fluorescence thermal denaturation curves to two, three, or four species unfolding mechanisms. Data were obtained in 50 mM potassium phosphate at a final ionic strength of 150 mM, from 283.15 to 363.15 K. Protein concentrations were $\sim 1 \mu M$, $\sim 2 \mu M$, and $\sim 20 \mu M$ for far-UV CD, fluorescence, and near-UV CD, respectively. ($n=3$, $\pm SD$).

fluorescence quenching (Supplementary Fig. S6F), and modifies shape and intensity of the near-UV and vis CD signals (Supplementary Fig. S7C). This is consistent with the mutation perturbing overall protein folding and FAD incorporation (22, 50). Noticeably, $\Delta R201$ CTC increases FAD fluorescence quenching when compared with $\Delta R201$ hAIF $_{\Delta 1-101ox}$ (Supplementary Fig. S6H).

The mutations have a negative impact on thermal stability when compared with WT (Fig. 3 and Supplementary Fig. S8, Table 3), particularly evident in $\Delta R201$ ($\Delta T_{m1} = -27.7^\circ C$ and $\Delta T_{m2} = -24.8^\circ C$). R201 ablation also modifies the unfolding events, so that the FAD cofactor is released in the first transition (N \rightarrow I) (Fig 3A and Supplementary Fig. S8A). Therefore, T_m for flavin release (T_{mFAD}) decreases by $\sim 33^\circ C$ in $\Delta R201$ hAIF $_{\Delta 1-101ox}$ compared with WT, indicating destabilization of the redox active center. To test the structural impact of this mutation, we have produced a $\Delta R201$ hAIF $_{\Delta 1-101ox}$ homology model. Our equilibration routine courses with an overall structure relaxation that was not observed in the WT case, whereas the organization of this structure is maintained in the MD runs to optimize the system (Fig. 4, Supplementary Fig. S4 and S5). Thus, this $\Delta R201$ model predicts important structural and dynamic rearrangements at the oxidoreductase domain, in agreement with R201 defining the β -hairpin length and interactions. These rearrangements alter the network that links the FAD cofactor, the active site residues, the central β -strand, the apoptotic domain, the C-loop, and the β -hairpin itself (Fig. 4, Supplementary Fig. S4 and S5). Increases in radius of gyration, size and number of protein internal cavities, and molecular and

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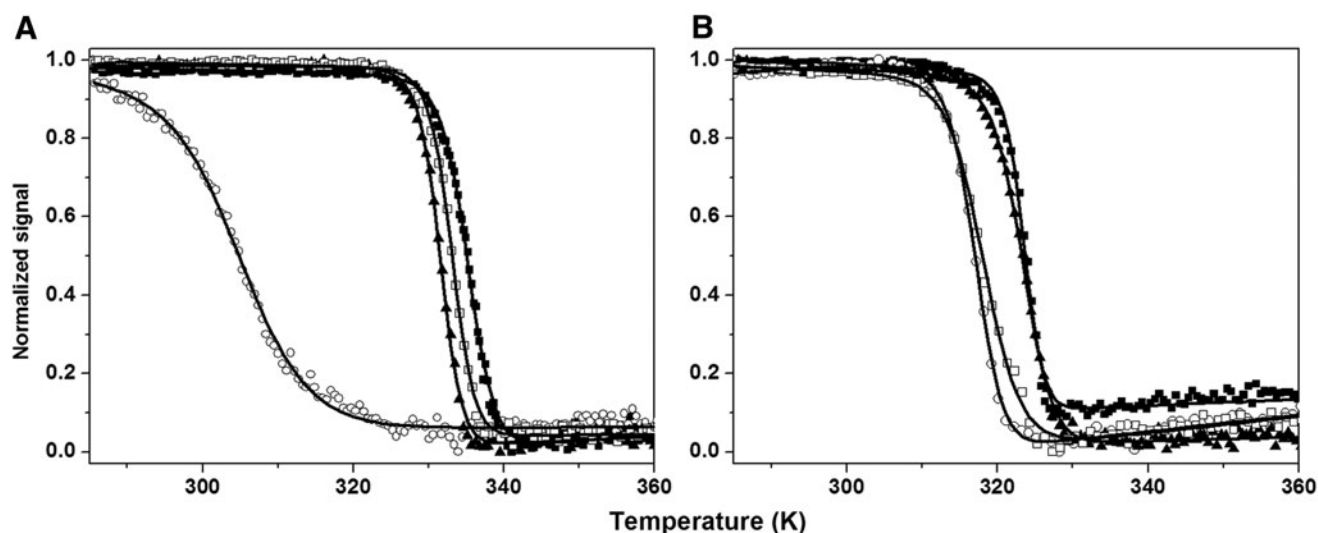


FIG. 3. Effect of $\Delta R201A$, G308E, and E493V hAIF $_{\Delta 1-101}$ pathogenic mutations on the temperature for FAD release. Curves for FAD thermal release in (A) hAIF $_{\Delta 1-101ox}$ and (B) its CTC (1:100 hAIF $_{\Delta 1-101ox}$:NADH ratio) for WT (black squares), $\Delta R201$ (open circles), G308E (black triangles), and E493V (open squares), as monitored by flavin fluorescence emission. The curves are shown roughly normalized from 0 to 1, and the fit to a two-state unfolding model is represented by the continuous lines. Data were obtained in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM. Protein concentration was $\sim 2 \mu M$. WT, wild type.

TABLE 3. EFFECT OF $\Delta R201$, G308E, AND E493V hAIF $_{\Delta 1-101}$ MUTATIONS ON THE THERMAL STABILITY OF hAIF $_{\Delta 1-101ox}$ AND OF ITS CHARGE TRANSFER COMPLEX

	hAIF $_{\Delta 1-101ox}$				CTC (hAIF $_{\Delta 1-101rd}$:NAD $^{+}$)				$T_{m1CTC}-T_{m1}$	$T_{m1CTC}-T_{m2}$
	$T_{m1}(K)$	$T_{m2}(K)$	ΔH_1 (kcal/mol)	ΔH_2 (kcal/mol)	T_{m1CTC} (K)	T_{m2CTC} (K)	ΔH_{1CTC} (kcal/mol)	ΔH_{2CTC} (kcal/mol)		
WT	332 \pm 1	337 \pm 1	87 \pm 6	140 \pm 10	324 \pm 1		120 \pm 10		-8 \pm 2	-13 \pm 2
$\Delta R201$	304 \pm 1 ^a	312 \pm 1	64 \pm 9 ^a	98 \pm 9	316 \pm 1	338 \pm 1	93 \pm 8	37 \pm 6	12 \pm 2	4 \pm 2
G308E	328 \pm 1	332 \pm 1 ^a	130 \pm 10	160 \pm 20 ^a	323 \pm 1		140 \pm 10		-5 \pm 2	-9 \pm 2
E493V	329 \pm 1	334 \pm 1	150 \pm 10	180 \pm 10	318 \pm 1 ^b		85 \pm 4		-11 \pm 2	-16 \pm 2

Data were obtained by global fitting of the near-CD, far-UV CD, and fluorescence thermal denaturation curves to two or three species unfolding mechanisms. Data were obtained in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM, from 283.15 to 363.15 K. Protein concentrations were $\sim 1 \mu M$, $\sim 2 \mu M$, and $\sim 20 \mu M$ for far-UV CD, fluorescence, and near-UV CD, respectively. ($n=3$, $\pm SD$).

^aData revealed by the FAD fluorescence denaturation curve.

^bData revealed by the far-UV and FAD fluorescence denaturation curves. Denaturation curve in the vis at 420 nm shows a T_m of 329 \pm 1 and ΔH of 150 \pm 10.

solvent accessible surface areas are also predicted (Fig. 4 and Supplementary Fig. S4F), as well as a considerable reduction in the number of internal H-bonds (Supplementary Fig. S5E). Such facts agree with increased susceptibility to proteolytic cleavage and partial FAD loss (22). Therefore, lack of R201 is expected to prevent the β -hairpin-C-loop interaction, to destabilize the 529–533 helix and to release R529 from its stacking to W196. As a consequence, FAD binding debilitates (Fig. 4G), and the active site and C-terminal loop of the oxidoreductase domain lose compactness. Such hypothesis agrees with $\Delta R201$ unfolding data relative to the WT (earlier flavin release, lower conformational stability and unfolding enthalpies, and less structured active site). The effects of G308E and E493V mutations on denaturation profiles and on T_m values are milder (3–5°C decreases) (Table 3 and Supplementary Fig. S8B, C). Nevertheless, an increasing contribution of the secondary structure of G308E to the N \rightarrow I transition suggests that in this variant the isoalloxazine environment might be the region that unfolds the last (Supplementary Fig. S8B).

All, but particularly $\Delta R201$ and E493V CTCs, are less stable than the WT CTC (Table 3 and Supplementary Fig. S8D–F). Formation of G308E and E493V CTCs decreases protein stability, and E493V vis-CD data also suggest aggregation upon denaturation. On the contrary, although the $\Delta R201$ CTC is still considerably less stable than the WT CTC, thermal stabilization is observed when compared with its oxidized state, with some secondary and tertiary interactions remaining after flavin release (Supplementary Fig. S8B). Thus, NADH binding and flavin reduction favor the $\Delta R201$ -folded state. This suggests that NAD $^{+}$ makes the active site more compact, stabilizing the overall structure and, particularly, the FAD site. In agreement, our predicted structural $\Delta R201$ CTC model was considerably more compact than the oxidized model (Fig. 4, Supplementary Figs. S4 and S5). Thus, the CTC shifts the $\Delta R201$ conformational equilibrium to a more structured ensemble than that of the oxidized state.

Finally, the $\Delta R201$ variant also decreases unfolding enthalpy relative to the WT, consistent with the mutant being less structured. On the contrary, the higher unfolding enthalpies of G308E and E493V hAIF $_{\Delta 1-101ox}$, and G308E CTC might relate to them being more prone to aggregate on un-

folding. The decreased E493V CTC enthalpy suggests that this form might be less structured, in agreement with its modestly lower T_{mCTC} .

HTS allowed the identification of molecules potentially binding hAIF $_{\Delta 1-101}$

Since fluorescence changes along thermal denaturation allow evaluating hAIF $_{\Delta 1-101}$ stability, displacements in T_{mFAD} induced by different molecules appear a feasible approach to identify ligands in HTS plate assays (13). We have used NADH (the hAIF coenzyme), the NAD $^{+}$ product, and ATA [described to interact with hAIF $_{ox}$ (11)] as controls to optimize the method. In agreement with previous sections, incubation of hAIF $_{\Delta 1-101ox}$ in a well plate with NADH excess diminishes T_{mFAD} by 14°C \pm 0.5°C (ΔT_{mFAD}) (Fig. 5A). No dose-dependent response curve is observed, possibly because of formation of (i) the CTC and (ii) the monomer-dimer transition. Incubation with NAD $^{+}$ has no effect on T_{mFAD} , consistent with its low affinity for hAIF $_{ox}$ (18). ATA decreases T_{mFAD} similarly to NADH, induces quenching of flavin fluorescence, and produces a dose-response curve (Supplementary Fig. S9) that allows estimating a K_d^{ATA} of 10 \pm 4 μM (in the range of that reported by isothermal titration calorimetry (ITC), 19 \pm 5 μM (11)). Therefore, NADH and ATA produce a ligand-induced perturbation consistent with binding and hAIF $_{\Delta 1-101}$ destabilization (Fig. 5A).

Based on these results, we have screened for compounds potentially binding hAIF $_{\Delta 1-101ox}$ using the HitFinder Collection (Maybridge) and the Prestwick Chemical Library (Prestwick Chemical). Our assessment method identifies as hits those compounds that shift the T_{mFAD} more than 3°C in the same direction by both “midpoint of unfolding” and “inflection point location” methods (13, 68). Most of the assayed compounds do not alter T_{mFAD} , a few compounds increase it by one evaluation method, and a few more compounds shift it to lower values (Supplementary Fig. S10). We have not found any compound increasing T_{mFAD} more than 3°C by both evaluation methods. However, we have identified 11 compounds (C1 to C11) shifting T_{mFAD} between -3°C and -23°C by both methods (Fig. 5B and Supplementary Table S1), which might be postulated as hAIF binders. However, other phenomena, such as protein chemical denaturation or aggregation, might be implicated in T_{mFAD}

◀F5

◀SF9

◀SF10

◀ST1

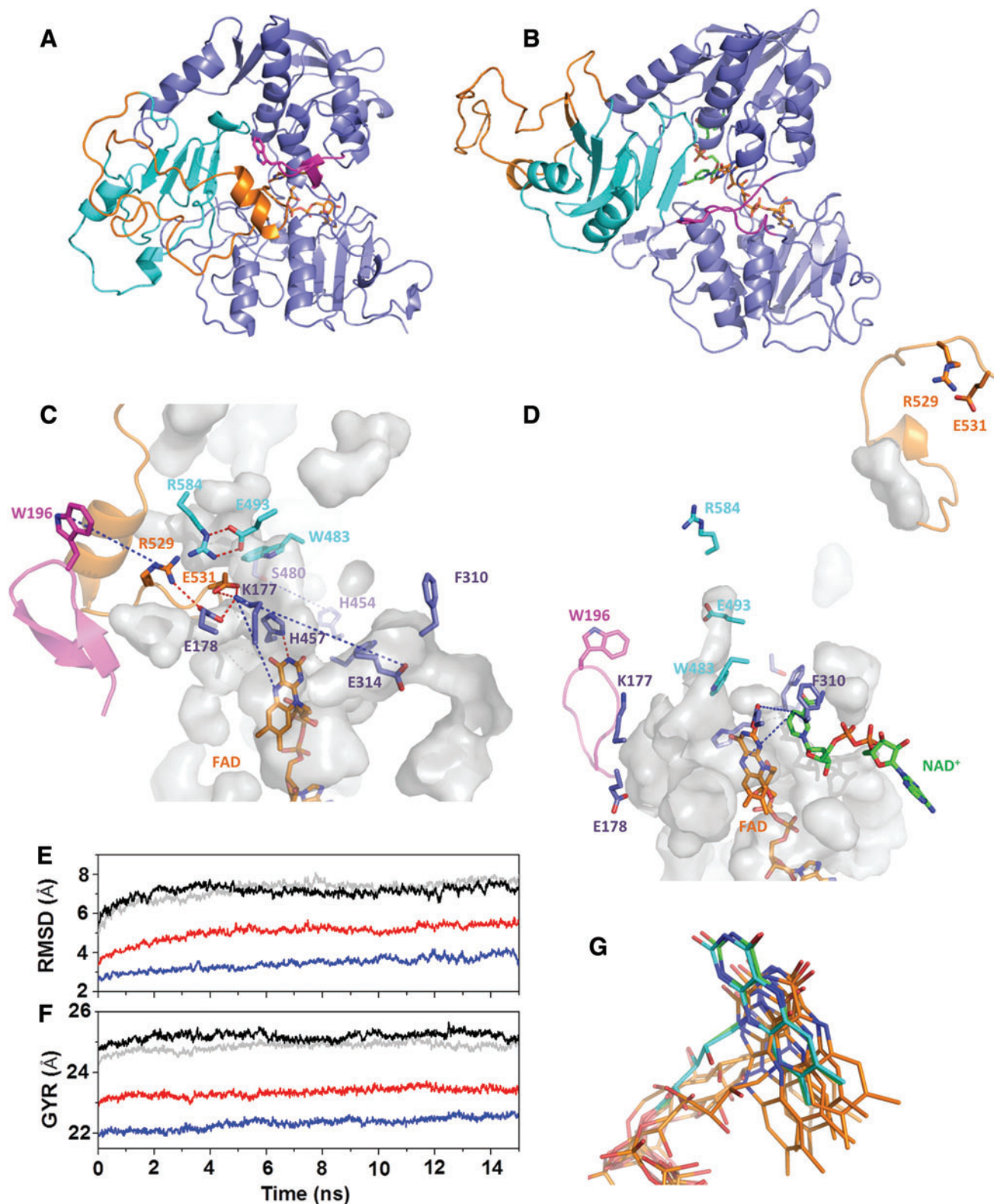


FIG. 4. Predicted hAIF $_{\Delta 1-101ox}$ conformational changes by R201 deletion. Overall representative structural models for (A) $\Delta R201$ hAIF $_{\Delta 1-101ox}$ and (B) its CTC after modeling of the missing C-loop residues and deletion of R201, minimization, and 15 ns of MD relaxation at 300 K. Protein chain is represented as cartoon with the oxidoreductase domains in violet and its 191–202 β -hairpin in hot pink (R201 and W196 side chains shown in sticks), and the apoptotic domain in light blue with the C-loop (509–560 segment) in orange. FAD is shown in sticks with orange carbons. Active site environment in (C) $\Delta R201$ hAIF $_{\Delta 1-101ox}$ and (D) its CTC. Relevant residues are shown in sticks. Relevant interactions in the WT model are shown in blue dashed lines, whereas the $\Delta R201$ structures also shown in blue the increase of the corresponding WT distances and in red dashed lines new interactions. Gray surfaces show water accessible cavities in the active site environment. (E) RMSD (C α) of $\Delta R201$ hAIF $_{\Delta 1-101ox}$ (gray line) and its CTCs (black line) models along MD when overlapped on their corresponding modeled structures before minimization. (F) Radius of gyration of $\Delta R201$ hAIF $_{\Delta 1-101ox}$ (gray line) and its CTCs along the MD. (G) Position of FAD (C in orange) in five replicas after applying the minimization routine to the $\Delta R201$ hAIF $_{\Delta 1-101ox}$ homology model, relative to the FAD position in the initial homology modeling (FAD with C in green) and in the template (4bv6, C in blue). To see this illustration in color, the reader is referred to the web version of this article at www.liebertpub.com/ars

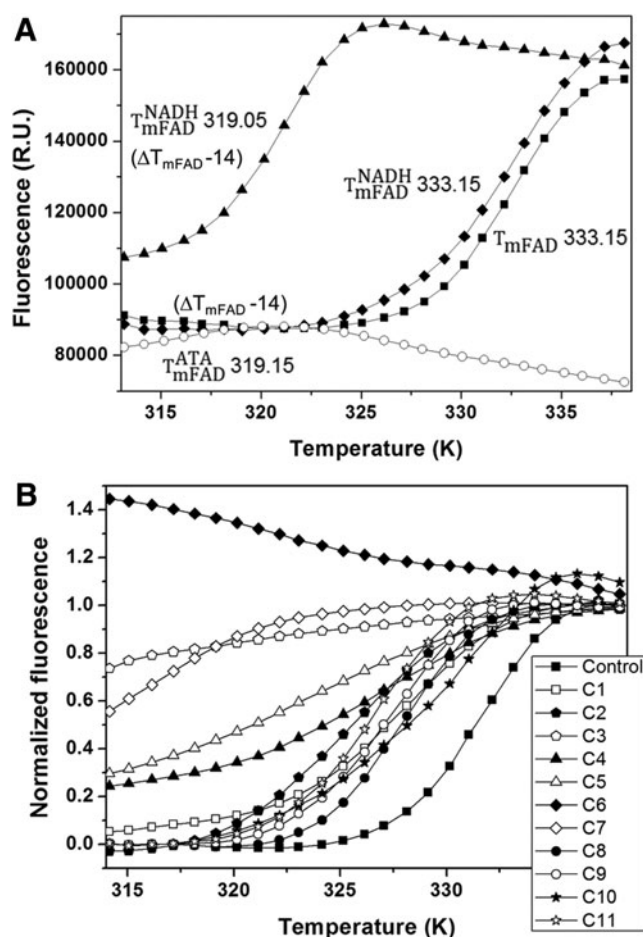


FIG. 5. Effect of binders in the thermal stability of hAIF Δ_{1-101} . (A) FAD fluorescence unfolding curves in samples containing hAIF $\Delta_{1-101OX}$ in the absence of ligands (■), and in the presence of 100 μM NADH (▲), 100 μM NAD $^+$ (◆), and 500 μM ATA (○). (B) FAD fluorescence unfolding curves in samples containing each of the 11 compounds selected by HTS as potential hAIF Δ_{1-101} binders. hAIF $\Delta_{1-101OX}$ samples incubated with 500 μM of C1, C2, C3, C4, or C5, or with 100 μM of C6, C7, C8, C9, C10, or C11 respectively, showed shifts in T_{mFAD} of -3.0 ± 0.6 , -5.0 ± 0.1 , -20 ± 1 , -3.0 ± 0.6 , -5.0 ± 0.1 , -23 ± 3 , -14 ± 1 , -5.0 ± 0.1 , -5.0 ± 0.6 , -6.0 ± 0.3 , and -6.0 ± 0.1 °C respectively. In all cases, 1 μM hAIF Δ_{1-101} was placed on each well of 96-well plates together with the ligands, and the changes in fluorescence emission at 530 nm (excitation at 450 nm) were followed during an increasing temperature ramp. Figure shows roughly normalized flavin fluorescence. The experiments were carried out in 50 mM potassium phosphate, pH 8.0.

shifts. Six compounds (C1, C2, C5, C8, C9, and C11) out of the 11 produce a dose-response ΔT_{mFAD} indicative of binding, postulating an affinity ranking $C8 \approx C11 \approx C5 > C9 > C1 > C2$ (K_d values in the 3–200 μM range).

Effect of HTS hits on cellular viability

Human HeLa cells have been exposed to a range of concentrations of these HTS hits. C1 (up to 1 mM) has no effect on cell viability. For the rest, we observe sigmoidal decreases

in cell viability with increasing concentrations, which allow determining the compound concentration for 50% cell viability (EC_{50}) (Supplementary Fig. S11). The toxicity ranges for our HTS hits vary, C9 and C8 being the more toxic (EC_{50} , 1.3 ± 0.1 and 15 ± 3 μM respectively), C5 producing an intermediate effect (112 ± 14 μM), and C11 and C2 being the less toxic compounds (279 ± 10 and 684 ± 45 μM , respectively). It is worthy to mention that C9 contains some pan assay interference compound groups and C8 produces protein precipitation.

To evaluate whether binding of HTS hits to hAIF might have a protective effect against AIF-dependent PCD, we have also incubated HeLa cells with C1, C2, C5, C8, C9, and C11 for 24 hours, and then added N-methyl-N'-nitrosoguanidine (MNNG), an alkylating agent that provokes AIF-dependent PCD (9). After 24 hours of MNNG treatment, loss of cell viability reaches $\sim 100\%$, independently on the HTS hits absence or presence. Hence, hits fail to prevent AIF-induced PCD.

Effects of selected HTS hits on the molecular and redox properties of hAIF Δ_{1-101}

Based on the ΔT_{mFAD} and cell viability effects, we have selected C2 and C11, and included in addition ATA [only producing a slight effect on macrophage cell viability after 24 hours (31, 65)], to comparatively evaluate their influence on the molecular and redox hAIF Δ_{1-101} properties. C2 and C11 do not affect the hAIF $\Delta_{1-101OX}$ visible spectrum, whereas hAIF $\Delta_{1-101OX}$ mixing with ATA shows a spectrum combining features of both (Supplementary Fig. S12A). hAIF $\Delta_{1-101OX}$ far-UV CD spectra suggest no major ATA effects on secondary structure (Supplementary Fig. S12B), whereas the spectra with C2 and C11 cannot be recorded due to their low solubility in the absence of dimethyl sulfoxide (DMSO). C2 and particularly ATA induce large changes in the hAIF $\Delta_{1-101OX}$ near-UV/vis CD (Supplementary Fig. S12C): both preventing detection of the ~ 300 nm peak [as in the CTC, Fig. 1D)] and ATA also decreasing the flavin signal intensity. ATA also produces noticeable effects in hAIF $\Delta_{1-101OX}$ fluorescence, dominating the spectrum (broad peak centered at ~ 440 nm that increases fluorescence yield in the presence of protein) and eliminating tryptophan emission (Supplementary Fig. S12D). Such observations indicate quenching of tryptophan fluorescence by energy transfer to ATA. C2 also quenches tryptophan fluorescence, whereas the effect of C11 is milder. Therefore, binding of ATA or C2 affects the environment of one or several tryptophan residues. Finally, ATA slightly increases flavin fluorescence, suggesting an effect on the hAIF $\Delta_{1-101OX}$ isoalloxazine environment (Supplementary Fig. S12E).

We have intended to evaluate affinity of hAIF $\Delta_{1-101OX}$ for each of these compounds by ITC and by difference spectroscopy, but the high amounts of ligands required to detect interaction make impossible such determination and indicate low interaction enthalpy. ATA also decreases hAIF $\Delta_{1-101OX}$ stability, while C2 and C11 change, in addition the unfolding mechanism to a two-species mechanism and C2 apparently stabilizes some structure after flavin release (Fig. 6A–C and Supplementary Table S2). We have also evaluated potential binding modes of these compounds by protein–ligand docking simulations, since attempts to cocrystallize failed. Our

◀SF11

◀AU6

◀SF12

◀F6

◀ST2

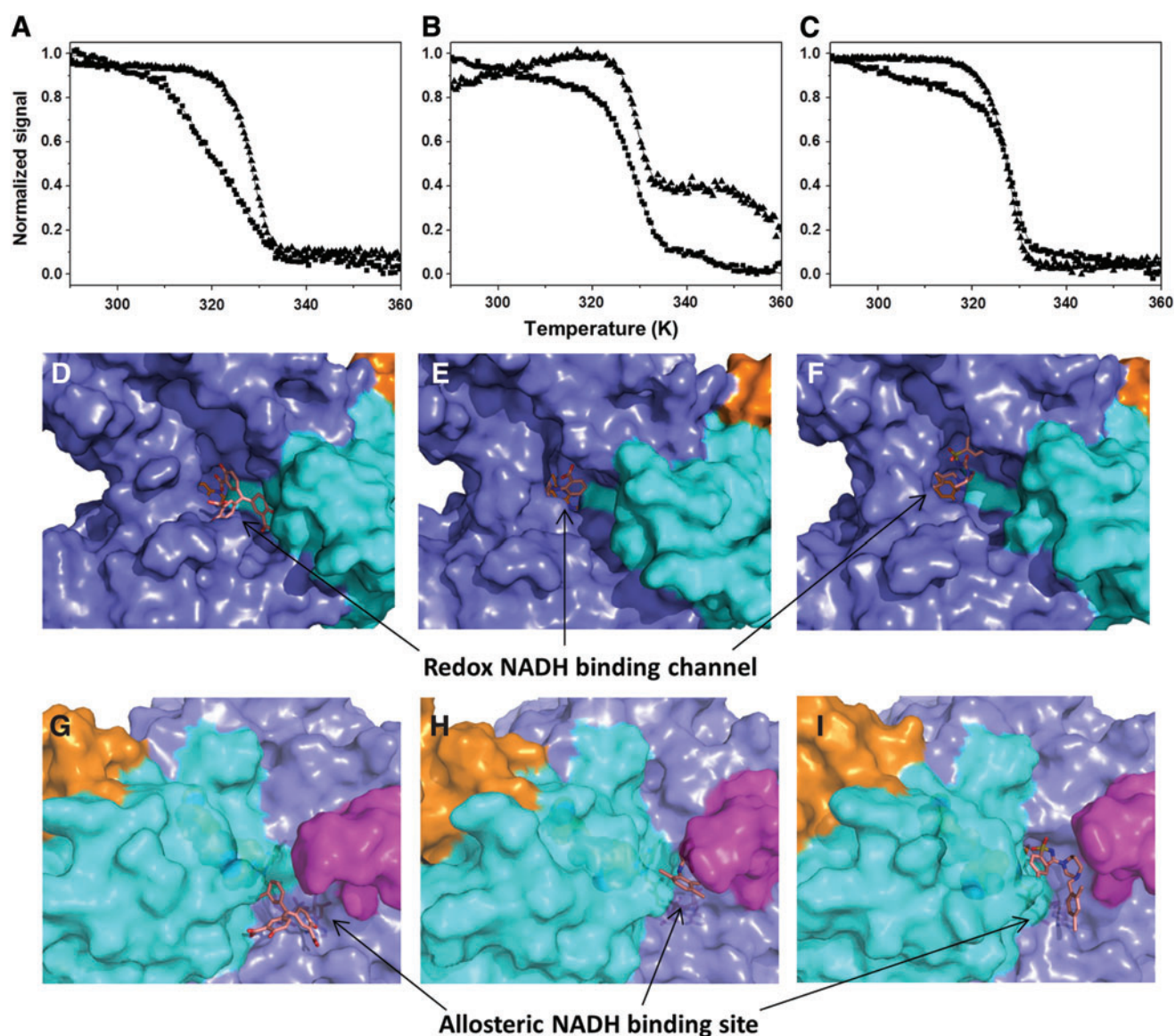


FIG. 6. Effect of selected HTS hits on hAIF $\Delta_{1-101ox}$ thermal unfolding and predicted binding modes. Thermal denaturation of hAIF $\Delta_{1-101ox}$ in the presence of (A) ATA, (B) C2, and (C) C11 was monitored by near-UV CD (300 nm, *black squares*), and flavin fluorescence emission (*black triangles*). The curves are shown roughly normalized from 0 to 1, and their global fits to a two-transition (ATA) or one-transition (C2, C11) unfolding model are represented by the *continuous lines*. Curves were recorded in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM in the presence of 500 μ M of each compound. Models for the best docking pose of (D) ATA, (E) C2, and (F) C11 to the hAIF $\Delta_{1-101ox}$ receptor and of (G) ATA, (H) C2, and (I) C11 to the hAIF $\Delta_{1-101rd}$:NAD⁺ receptor. Models were optimized by minimization and short MD equilibration routines. ATA, C2, and C11 are shown in *pink CPK sticks*. Other color codes are as in Figure 2. ATA, aurintricarboxylic acid. To see this illustration in color, the reader is referred to the web version of this article at www.liebertpub.com/ars

TABLE 4. EFFECT OF SELECTED COMPOUNDS ON THE STEADY STATE KINETIC PARAMETERS OF THE HAIF Δ_{1-101} DCPIP DIAPHORASE ACTIVITY

HIT	[HIT] (μ M)	k_{cat} (s^{-1})	K_m (μ M)	k_{cat}/K_m ($s^{-1} mM^{-1}$)
—		2.4 ± 0.1	430 ± 20	5.6 ± 0.2
ATA	100	1.0 ± 0.1	99 ± 6	10 ± 1
C2	500	2.1 ± 0.1	310 ± 10	6.8 ± 0.2
C11	100	2.1 ± 0.1	600 ± 20	3.5 ± 0.1

Assays were performed in 50 mM potassium phosphate, pH 8.0, 1% DMSO at 25°C. ($n=3$, \pm SE).

highest scoring docking clusters point to our three chemicals preferentially locating at the hAIF $\Delta_{1-101ox}$ redox NADH substrate binding channel (Fig. 6D–E). On the contrary, the NADH allosteric site is the preferred site in the CTC structure when the NAD⁺ product occupies the redox binding channel (Fig. 6H, I).

Finally, Table 4 summarizes the effect of ATA, C2, and C11 on the hAIF $\Delta_{1-101ox}$ kinetic parameters. ATA increases by nearly twice the hAIF $\Delta_{1-101ox}$ efficiency to oxidize NADH, apparently by favoring its association. C2 and C11 modify enzyme efficiency positively and negatively, respectively, by 25–30% as a consequence of changes in K_m^{NADH} . Therefore, C2, C11, and particularly ATA modify

the hAIF $_{\Delta 1-101}$ catalytic efficiency by modulating the coenzyme association/dissociation equilibrium.

Discussion

The data here presented prove that the proteolytic processing of hAIF drives its subcellular localization and provides structural plasticity for its adaptation to specific subcellular pH conditions. The apoptotic hAIF $_{\Delta 1-101}$ isoform is stable *in vitro*, able to oxidize NADH, and similarly folded at mitochondrial matrix, cytosol, apoptotic cytosol, and nuclei pH, but is unstable at the IMS pH. Nonetheless, the IMS isoform is a longer protein, hAIF $_{\Delta 1-54}$, tethered to the inner mitochondrial membrane. Evaluation of its truncated soluble isoform, hAIF $_{\Delta 1-77}$, confirms that the 77–101 traits stabilize the mitochondrial isoform. Although the pH impact on hAIF spectroscopic properties is in general mild, changes in the near-UV CD peak are noticeable under different conditions (Fig. 1C, D). In the hAIF $_{\Delta 1-77ox}$ and $\Delta R201_{ox}$ forms, this peak differs in shape and intensity, whereas it disappears when the CTC is formed, or when ATA or C2 bind (Fig. 1D, Supplementary Figs. S7C and S12C). Since ATA and C2 are predicted to bind at the WT hAIF $_{\Delta 1-101ox}$ redox NADH binding site (Fig. 6D, E), we expect induced redox-independent changes in solvent accessibility of at least one tryptophan when ligands block the NADH channel or when the C-loop is released. The effect of pH on the hAIF $_{\Delta 1-101ox}$ stability is modest, although the nonphysiological mitochondrial matrix pH slightly shifts T_{m1} and ΔH_{m1} (Table 2). hAIF $_{\Delta 1-77ox}$ unfolding curves indicate that the 77–101 region stabilizes the active site at pH 6.2, but they also point to other protein regions being more dynamic (Supplementary Fig. S3C, D). The lower stability and lack of I state along hAIF $_{\Delta 1-101rd}$:NAD⁺ CTC unfolding may seem surprising (Fig. 2B), since protein interaction with small ligands generally increases stability due to the coupling of preferential binding to the N state with unfolding equilibrium (6, 60). Nonetheless, differences between these two structures go beyond the simple coenzyme binding. Contrary to hAIF $_{\Delta 1-101ox}$, the CTC structure is a dimer with high flexibility in the regulatory C-loop (18) (with properties of an intrinsically disordered region (70)), lack of contacts between apoptotic and FAD-binding domains, and facilitated access to the flavin (Fig. 2). In this context, the ATA, C2, and C11 destabilizing effect (Fig. 6) indicates that their binding might also induce C-loop release. This suggests that the T_m decrease induced by these compounds might relate to their preferential binding to partially, or totally, unfolded conformations that become progressively populated along unfolding.

Relationships between the effects of mutations on proteins molecular functional properties and phenotypes manifestations are in general complex, but understanding them is key to improve diseases treatments (59). Destabilizing mutations may impair intracellular stability, function, regulation, and protein–protein interactions simultaneously and to different extents (21, 40). Considering the hAIF capacity to interact with several partners that modulate its cellular localization and activities, mutations causing biochemical alterations might differently compromise their *in vivo* functions. Structural bases of such intracellular interactions are in most cases hardly known, but those involving hAIF C-loop recognition are expected to be redox controlled. $\Delta R201$ was the first loss-

of-function mutation described in the *aifm1* gene causing progressive mitochondrial encephalomyopathy (22). Our unfolding data and structural models indicate very drastic impacts on hAIF $_{\Delta 1-101ox}$ stability and conformation (Fig. 4 and Supplementary Fig. S8). Break of the R201–E531 salt bridge perturbs various functional sites, confirming the β -hairpin–C-loop connection as a key element to retain a compact active site (Fig. 4 and Supplementary Fig. S8). The loss of active site compactness affects active site dynamics as well as the AIF allosteric architecture switching (8). This agrees with drastic alterations on the mutant redox properties and with OXPHOS failure playing a major role in the lethal $\Delta R201$ phenotypes (22, 50). Since the OXPHOS productive AIF–CHCHD4 interaction relies on the entire hAIF structure and is favored in the protein reduced state (25), the conformational stability decrease in the $\Delta R201$ redox active center would impact CHCHD4 recognition. Interestingly, formation of the CTC stabilizes the $\Delta R201$ variant (Figs. 3 and 4).

Differences in the switch to the CTC between WT and $\Delta R201$ variants will have major effects on hAIF *in vivo* redox conformational dynamics, as well as on the interaction networks with other physiological partners. Our models predict conformational changes in the $\Delta R201_{ox}$ variant at the region proposed for Hsp70 recognition, residues 150–268 (24, 37), which might modulate hAIF retention in the cytoplasm. Regarding the degradosome: (i) the CypA site in hAIF $_{ox}$ is mapped as containing the 370–397 3-stranded β -sheet (NADH domain) (16, 73) that apparently is only slightly altered in our $\Delta R201_{ox}$ model, but not in its CTC model. Since CypA affinity improves in the WT CTC state, a similar effect might be occurring in $\Delta R201$ CTC; (ii) the nuclear H2AX binds to the 544–554 Pro-rich C-loop motif, flexible in WT and $\Delta R201$ isoforms (3, 69); and (iii) finally, DNA putatively nonspecifically wraps around the FAD and C-terminal domains (71), with the $\Delta R201$ mutation apparently increasing its *in vitro* affinity (22). Nevertheless, considering that H2AX and DNA binding is contributed by the hAIF intrinsically disordered region, with disorder modulating the binding, the impact of $\Delta R201$ mutation might be low in the *in vivo* overall degradosome assembly, in agreement with PCD being hardly affected by this missense mutation. Noticeably, among the studied AIF variants, $\Delta R201$ is the only one stabilized by coenzyme binding (Fig. 3) and one of the few in which patients supplemented with the FAD riboflavin precursor improved clinical conditions by partially correcting OXPHOS failure (22). Thus, the FAD and NADH natural ligands might synergically contribute to *in vivo* partially overcome the $\Delta R201$ unfolding negative effect by enhancing protein stability. Such observation has important implications in the search of potential ligands correcting stability and function.

G308E and E493V variants had little effects on hAIF $_{\Delta 1-101}$ stability. G308E substitution (Supplementary Fig. S6) slows down CTC formation and abolishes the enzyme ability to discriminate between NADH and NADPH (57, 61, 62). This leads to prenatal OXPHOS-related encephalopathy, muscle atrophy, and early death. G308E oxidized structures (Supplementary Fig. S6) show that Glu308 stabilizes a different Glu336 conformer that modulates the 334–336, 365–370, and 380–388 interaction networks, which agree with its modest differences in stability (Table 3) (57, 61, 62). In its CTC form, those changes alter the interaction network of the coenzyme adenine moiety and produce its reorientation (62), envisaging

the possibility of distinct CTC conformers. The G308E mutation causes a severe complex I+IV deficiency and reduced CHCHD4 binding (4). Altogether, the loss of functionality in this variant appears related to the lower accumulation of its CTCs, rather than to an overall conformational stability lost. The E493V mutation enhances hAIF apoptotic activity, but it alters neither the interaction with CHCHD4 nor the respiratory mitochondrial activity (50). Mild changes in stability agree with the mutation having no effect on protein folding or cellular expression. Nevertheless, *in vitro* this variant has increased propensity to NADH oxidation and CTC reoxidation (50, 57), in line with its reduced unfolding enthalpy and aggregation propensity (Table 3). Thus, the E493V mutation would favor the hAIF monomeric form and its nuclear translocation. E493 is not solvent accessible, but assists folding of the regulatory C-loop and contributes to binding of the allosteric NADH. Thus, although the E493V substitution has apparently low impact on the reductase domain, it reduces the CTC dimer stability (18) and makes it less structured than in the WT. This might also have an impact on the allosteric NADH binding and/or C-loop conformation. Since disordered regions and quaternary organizations are involved in signaling, recognition, regulation, or nuclear translocation (70), we can consider that these facts affect the E493V variant interplay with its partners and, in turn, favor its nuclear translocation and DNA degradation.

Considering this, the dynamics of the domain–domain and FAD–protein interactions appear to be key elements in hAIF $_{\Delta 1-101ox}$ stabilization. hAIF $_{\Delta 1-101ox}$ and its CTC must be considered very different conformational ensembles from the stability and structural points of view, with the architecture switch toward the CTC producing a less stable and compact protein that contains an intrinsically disordered region. Such mechanisms are contributors to control hAIF physiological actions by either cellular NADH levels or interaction with other biomolecules (8, 70). Altogether, the available information gives a comprehensive picture of stability and local dynamics in aspects of the hAIF function such as the redox-derived allosteric properties and ability to form quaternary assemblies, which will influence the interactions network and the *in vivo* phenotypes consequence of missense mutations. In this context, the flavin redox state and the coenzyme binding are key elements to modulate hAIF dynamics and C-loop ensembles. Some of these hAIF properties might be also modulated by small ligands, setting a molecular framework to discover molecules able to bind hAIF. These molecules might have a rescuing function of pathogenic variants by shielding deleterious effects on protein stability and ligand recognition, but they might also be modulators of the structural protein plasticity. To date, only ATA has been described as able to bind to AIF, but its promiscuity makes of it a bad drug candidate (31, 65). The hits here identified provide a starting point to redesign and open a molecular framework to search for new pharmacological ligands targeting hAIF. Thus, this characterization provides deeper insights into the hAIF conformation as a mechanism to regulate its cellular functions, as well as some of the molecular and cellular pathogenesis associated with its defects.

Materials and Methods

Proteins production

hAIF $_{\Delta 1-101}$ samples were expressed in and purified from *Escherichia coli* as N-terminal His₆-tag proteins as previ-

ously described (18), introducing an additional chromatography step using a Superdex™ 200 10/300 GL (GE Health Care, Chicago, IL) column in 50 mM potassium phosphate (Panreac Aplichem, Barcelona, Spain), pH 8.0, 150 mM sodium chloride (Panreac Aplichem). The purified protein was dialyzed and concentrated (up to $\sim 500 \mu M$), and then diluted to the desired concentration in 50 mM potassium phosphate at the corresponding pH, 8.0, 7.4, 7.0, or 6.2, at a final ionic strength of 150 mM (described as physiological (12, 63)). pET28a-hAIF $_{\Delta 1-101}$ plasmids containing the $\Delta R201$, G308E, and E493V mutations, as well as pET28a-hAIF $_{\Delta 1-77}$, were obtained from Mutagenex (Suwanee, GA). Several independent purifications of each hAIF variant were used to reproduce experiments, confirming that results are batch independent.

Before the HTS experiments, the His₆-tag was removed by adding 0.2 U of thrombin to 1 mL of hAIF $_{\Delta 1-101}$ and loading the mixture into a 1 mL HiTrap Chelating HP column (GE Health Care) activated with NiSO₄ (Probus S.A, Badalona, Spain). The column containing the mixture was incubated for 4 hours at room temperature and then overnight at 4°C. hAIF $_{\Delta 1-101}$ was eluted with 50 mM potassium phosphate, pH 8.0. Analytical grade chemicals and buffers were used in the study: NADH/NAD⁺ (Sigma-Aldrich, 10107735001/N0632; San Luis, MO), ATA (Sigma-Aldrich; A1895), and DMSO (Sigma-Aldrich; B5879).

Spectroscopic characterization

UV–visible spectra were recorded at 25°C in a Cary 100 Bio (Agilent Technologies, Santa Clara, CA) spectrophotometer. Protein concentrations were determined using the molar absorbance extinction coefficient of the protein ($\epsilon_{450} = 13.7 \text{ mM}^{-1} \text{ cm}^{-1}$) (18). UV–vis difference spectroscopy measurements were carried out with compartmentalized cuvettes in 50 mM potassium phosphate, pH 8.0, by titrating $\sim 20 \mu M$ of hAIF $_{\Delta 1-101}$ with up to 2.9 mM of ATA. Fluorescence spectra were acquired in a thermostated Cary Eclipse Fluorescence spectrophotometer (Agilent Technologies) using 2 μM hAIF $_{\Delta 1-101}$. Protein fluorescence emission spectra were recorded upon excitation at 280 nm, whereas flavin emission was collected after excitation at 450 nm. Circular dichroism (CD) spectra were recorded in a thermostated Chirascan (Applied Photophysics Ltd., Leatherhead, United Kingdom). Far-UV CD spectra were acquired in a 0.1 cm path length cuvette using 1 μM hAIF $_{\Delta 1-101}$, whereas near-UV/vis CD spectra were recorded in a 1 cm path length cuvette using 20 μM hAIF $_{\Delta 1-101}$. Fluorescence and CD spectra were acquired at 10°C (folded state) and 90°C (thermally denatured state), both in the absence and presence of a 100-fold excess of NADH.

Kinetic measurements

Steady-state kinetic analyses were carried out in a Cary 100 Bio spectrophotometer (Varian) and measured in air-saturated 50 mM potassium phosphate, pH 8.0, using NADH (Sigma-Aldrich; 1266615-56-8) and 95 μM dichlorophenolindophenol (Sigma-Aldrich) (18). Reactions were followed at 620 nm ($\Delta\epsilon_{620\text{nm}} = 21 \text{ mM}^{-1} \text{ cm}^{-1}$) and parameters determined by data fitting to the Michaelis–Menten equation. Stopped flow measurements were carried out under aerobic conditions in an SX17.MV spectrometer (Applied

Photophysics Ltd.) using a photodiode array detector and the Xscan software (18). The observed rates for the hydride transfer (HT) event (k_{obs}) were calculated by global analysis and numerical integration methods (using all spectral data in the 400–800 nm region along time evolution) by fitting to either a single-step ($A \rightarrow B$) or a two-step ($A \rightarrow B \rightarrow C$) model. When a saturation profile on NADH concentration was observed, k_{obs} values were fitted to the $k_{\text{obs}} = k_{\text{HT}} \cdot [S]/(K_d + [S])$ equation describing formation of an enzyme–substrate complex before the HT process, where k_{HT} is the limiting rate constant for HT from the pyridine nucleotide coenzyme to the FAD cofactor of hAIF, and K_d is the hAIF coenzyme dissociation constant.

Thermal denaturation

Thermal denaturation curves were followed by fluorescence emission of the aromatic residues and the flavin cofactor, far-UV CD, and near-UV/vis CD. Denaturation curves were recorded from 10°C to 90°C with scan rates of 1°C/min and 1.5°C/min, respectively, for CD and fluorescence assays, both in the absence and presence of a 100-fold excess of NADH (Sigma-Aldrich). hAIF $_{\Delta 1-101}$ concentrations, wavelengths, and path lengths were 1 μM , 210, and 1 mm in the far-UV CD (220 nm when NADH was present); 20 μM , 300 nm, and 1 cm in the near-UV CD (or 420 and 600 nm when NADH was present); and 2 μM , 530 nm (excitation at 450 nm), and 1 cm for flavin fluorescence. Unless otherwise stated, the individual experimental data sets were globally analyzed as one-transition process (*i.e.*, two-step process, native \leftrightarrow unfolded, N \leftrightarrow U) or two-transition process (*i.e.*, three-state process, native \leftrightarrow intermediate \leftrightarrow unfolded, N \leftrightarrow I \leftrightarrow U) by applying the following equations (51):

$$S_{\text{obs}} = \frac{S_N + m_N T + (S_U + m_U T)e^{-(\Delta G/RT)}}{1 + e^{-(\Delta G/RT)}}, \quad [1]$$

$$S_{\text{obs}} = \frac{S_N + m_N T + (S_I + m_I T)e^{-(\Delta G_1/RT)} + (S_U + m_U T)e^{-((\Delta G_1 + \Delta G_2)/RT)}}{1 + e^{-(\Delta G_1/RT)} + e^{-((\Delta G_1 + \Delta G_2)/RT)}}, \quad [2]$$

where S_{obs} is the measured protein signal at a given temperature (T), S_N , S_I , and S_U are the signals (origin intercept) of native, intermediate, and unfolded protein conformations at 0 K, respectively, and m_N , m_I , and m_U are the slopes of the linear temperature dependence of those signals, respectively. In contrast, the free energy difference in Eq.1 or Eq.2 is as follows: $\Delta G_i = \Delta H_i \left(1 - \frac{1}{T_{\text{mi}}}\right) + \Delta C_{p_i} \left(T - T_{\text{mi}} - T \ln \frac{T}{T_{\text{mi}}}\right)$, where ΔH_i is the van't Hoff enthalpy for each unfolding transition, T_{mi} is the midtransition temperature for each unfolding transition, ΔC_{p_i} is the heat capacity change for each unfolding transition, and R is the ideal gas constant.

FAD fluorescence-based HTS

The thermal stability of WT hAIF $_{\Delta 1-101}$ was monitored on 96-well plates by following FAD fluorescence, which increases upon cofactor dissociation coupled to thermal un-

folding (13, 19), using a FluoDia T70 thermostated plate reader fluorimeter (Photon Technology International, Edison, NJ). The FAD fluorescence emission at 530 nm (excitation at 450 nm) was followed to determine T_m for flavin release (T_{mFAD}) of free WT hAIF $_{\Delta 1-101}$ and in the presence of known ligands and compounds from two different chemical libraries. A library of $\sim 10,000$ small molecules ($M_w < 500$) with high chemical diversity, mostly satisfying Lipinski's rules (35) and dissolved at 4 mM in DMSO, was selected from the HitFinder Collection (Maybridge Chemical Company, Ayrtrincham, UK). The Prestwick Chemical Library (Prestwick Chemical, Graffenstaden, France), composed of 1120 compounds dissolved at 10 mM in DMSO and with known bio-availability and safety in humans, was also screened. The screening of the Maybridge library was performed in two phases (13). In the first phase, three compounds were added per well in 96-well plates, at a final concentration of 100 μM each, to 1 μM of WT hAIF $_{\Delta 1-101}$ in 50 mM potassium phosphate buffer, pH 8.0 in a 100 μL final volume. The compounds in the positive wells were screened again in a second round by testing each compound individually. The screening of the Prestwick library was carried out in a single step at a final compound concentration of 500 μM and 1 μM of WT hAIF $_{\Delta 1-101}$ in 50 mM potassium phosphate, pH 8.0 in a 100 μL final volume. Internal reference control wells (WT hAIF $_{\Delta 1-101ox}$ without compounds) were distributed on columns 1 and 12 in the plates, with the same concentration of protein and DMSO as the wells containing compounds. The unfolding curves obtained for each well were analyzed using homemade software that estimates the T_{mFAD} of each well, using two methods (termed “midpoint of unfolding” and “inflection point location”) designed to complement each other and to avoid false positives (13, 68). All the HTS hits were subsequently acquired from Sigma Aldrich or Maybridge and dissolved in 100% DMSO to prepare stock solutions at 50 and 10 mM. The purity of all compounds was

>95%, as indicated by the manufacturer by high-performance liquid chromatography, thin layer chromatography, nuclear magnetic resonance, infrared, or basic titration.

Cell viability assays

Cytotoxicity of the selected compounds was analyzed using HeLa cells cultured in Dulbecco's modified Eagle medium (Pan-Biotech GmbH, Aidenbach, Germany) supplemented with 50 U/mL penicillin (Pan-Biotech GmbH), 50 mg/mL streptomycin sulfate (Pan Biotech), and 10% fetal calf serum (Pan-Biotech GmbH) at 37°C in a 5% CO₂ atmosphere. Cells were cultured in 25 mL culture flasks and subcultured every 3 days. To test the toxicity of the compounds, HeLa cells were harvested and diluted to 7000 cells per well (100 μL) in 96-well plates. After incubation for 24 hours, the compounds were added at different concentrations ranging from 5 nM to 1 mM

(except C2, 50 nM to 10 mM range). Cell viability after 24 hours of incubation was measured using CellTiter 96[®] AQueous One Solution Cell Proliferation Assay (Promega, Madison, WI) following modified manufacturer recommendations, and absorbance was recorded at 490 nm to quantify the amount of formazan produced by the cell metabolism in a Synergy HT (BioTek, Winooski, VT) plate reader. The values obtained for control wells lacking added compounds but treated with DMSO were considered to reflect 100% viability. EC₅₀ was calculated for each compound by fitting the viability at the different concentrations of compounds to a sigmoidal dose–response function (38). To test the possible protective effect of the compounds against MNNG-induced cell death, 7000 cells per well were incubated in 96-well plates for 24 hours at 37°C and 5% CO₂. Next day, 0.5 mM MNNG (Abcr GmbH AB 138809, Karlsruhe, Germany) was added on each well and the plate was incubated for 20 minutes. Then, the medium was replaced by fresh medium and the cells were incubated for another 24 hours either in the absence or presence of compounds.

Production of structural models

Models containing the missing C-loop residues (546–558 and 518–559, respectively, for WT hAIF_{Δ1–101ox} and hAIF_{Δ1–101rd}:NAD⁺), as well as deletion of R201 in the ΔR201 variant, were built using as templates the coordinates of WT hAIF_{Δ1–101ox} (PDB 4bv6) and of hAIF_{Δ1–101rd}:NAD⁺ (PDB 4bur) and the Swiss-Model server (5, 18). Cofactors or coenzymes were then reintroduced, and 50 ps MD simulations, with Generalized-Born, Molecular Volume solvation (34) and CHARMM c39b1, were carried out to remove potential clashes. Protonation states were assessed using PROPKA 3.0 (46); H131, H457, and H478 were ϵ -protonated, whereas H454 and H455 were δ -protonated. MD simulations were performed using CHARMM c39b1 and the charmm36 force field (7). Parameters for FAD and FADH[−] and other ligands were generated using the CgenFF server (67) and density functional theory (DFT). DFT calculations were performed with the Gaussian09 rev.D01 package (20) at the B3LYP/def2-SVP+GD3BJ level with a water-like polarizable continuum model (20, 23). A Monte Carlo scheme was used for neutralization, adding ions to 150 mM. A time step of 1 fs, NVT conditions, and a TIP3P water model were used. A standard workflow routine including solvation, neutralization, minimization plus heating, and equilibration for 200 ps was performed to the models, with at least three replicas. Short 15 ns MD runs were then performed at 298.15 K to ensure thermodynamic and structural stability. The CHARMM software analysis tools were used to evaluate the resulting conformational ensembles. The GOLD 5.5 software (28) and the produced WT hAIF MD structures were used to obtain interaction models with ATA, C2, and C11. Binding sites were corroborated with the FPocket suite (33). Final docked structures were obtained using default parameters and 100 Genetic Algorithm runs with a consensus docking approach between the ChemPLP and the GoldScore functions. The top ranked pose of the best cluster was then used as the starting point for short MD simulations to refine protein–ligand interactions (Fig. 6D–H). Visual MD (27) and PyMol (14) were used to analyze and to visualize structural data.

Data analysis and statistics

Data were analyzed and shown in figures using Origin (OriginLab Corp., Northampton, MA) and Pro-K (Applied Photophysics Ltd.). Results are expressed as the mean \pm the standard deviation (SD) or as the mean \pm the standard error (SE). Chemical structures of the compounds were drawn using ChemDraw Professional 16 (PerkinElmer Informatics, Cambridge, MA). The FAF-Drug3 web server was used to identify putative pan assay interference compounds (32). The Molinspiration web server (www.molinspiration.com) was used to calculate theoretical LogP values of Prestwick chemical library compounds and ATA. Corresponding values for compounds from the Maybridge chemical library were obtained from the collection database.

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Authors' Contributions

P.F. and M.M. designed the research; R.V., S.R.-T., R.L., and A.V.-C. performed research and acquired the data; R.V., R.L., J.M.-O., A.V.-C., J.S., P.F., and M.M. analyzed data; and R.V., P.F., and M.M. wrote the article.

Author Disclosure Statement

No competing financial interests exist.

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Address correspondence to: ◀AU9

Dr. Patricia Ferreira
Departamento de Bioquímica y
Biología Molecular y Celular
Facultad de Ciencias
Universidad de Zaragoza
Pedro Cerbuna 12
Zaragoza 50009
Spain

E-mail: ferreira@unizar.es

Dr. Milagros Medina
Departamento de Bioquímica y
Biología Molecular y Celular
Facultad de Ciencias
Universidad de Zaragoza
Pedro Cerbuna 12
Zaragoza 50009
Spain

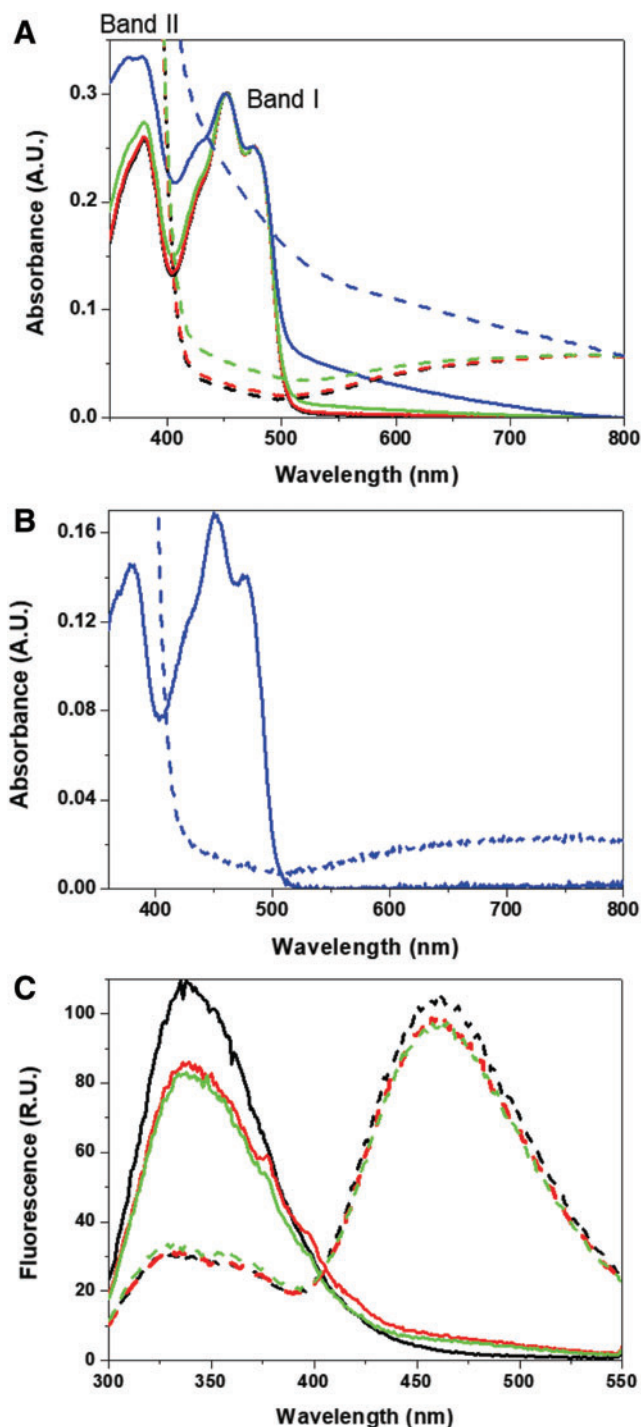
E-mail: mmedina@unizar.es

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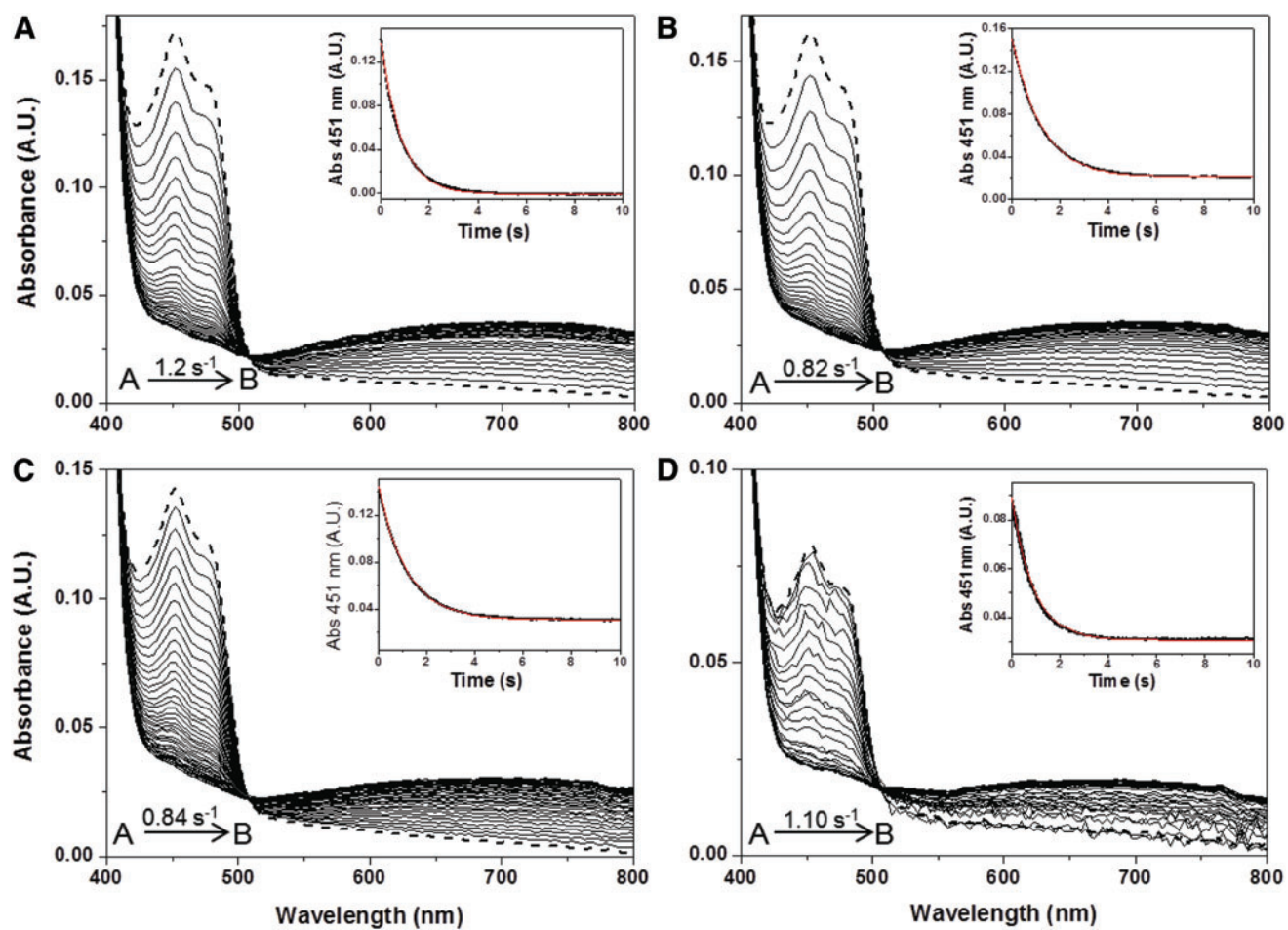
Abbreviations Used

ΔT_m	= T_m variation
AIF	= apoptosis-inducing factor
ATA	= aurintricarboxylic acid
CD	= circular dichroism
CHCHD4	= coiled-coil-helix-coiled-coil-helix domain containing 4 protein
CTC	= charge transfer complex
CypA	= cyclophilin A
DFT	= density functional theory
DMSO	= dimethyl sulfoxide
EC ₅₀	= half maximal effective concentration
FAD	= oxidized form of flavin adenine dinucleotide
Hsp70	= heat-shock protein 70
HT	= hydride transfer
HTS	= high-throughput screening
IMS	= intermembrane space
MNNG	= N-methyl-N'-nitrosoguanidine
NAD ⁺	= oxidized nicotinamide adenine dinucleotide
NADH	= reduced nicotinamide adenine dinucleotide
PCD	= programmed cell death
T_m and T_{mCTC}	= midpoint unfolding temperature for hAIF _{ox} and its CT
T_{mFAD}	= T_m for flavin isoalloxazine release
UV	= ultraviolet
WT	= wild type

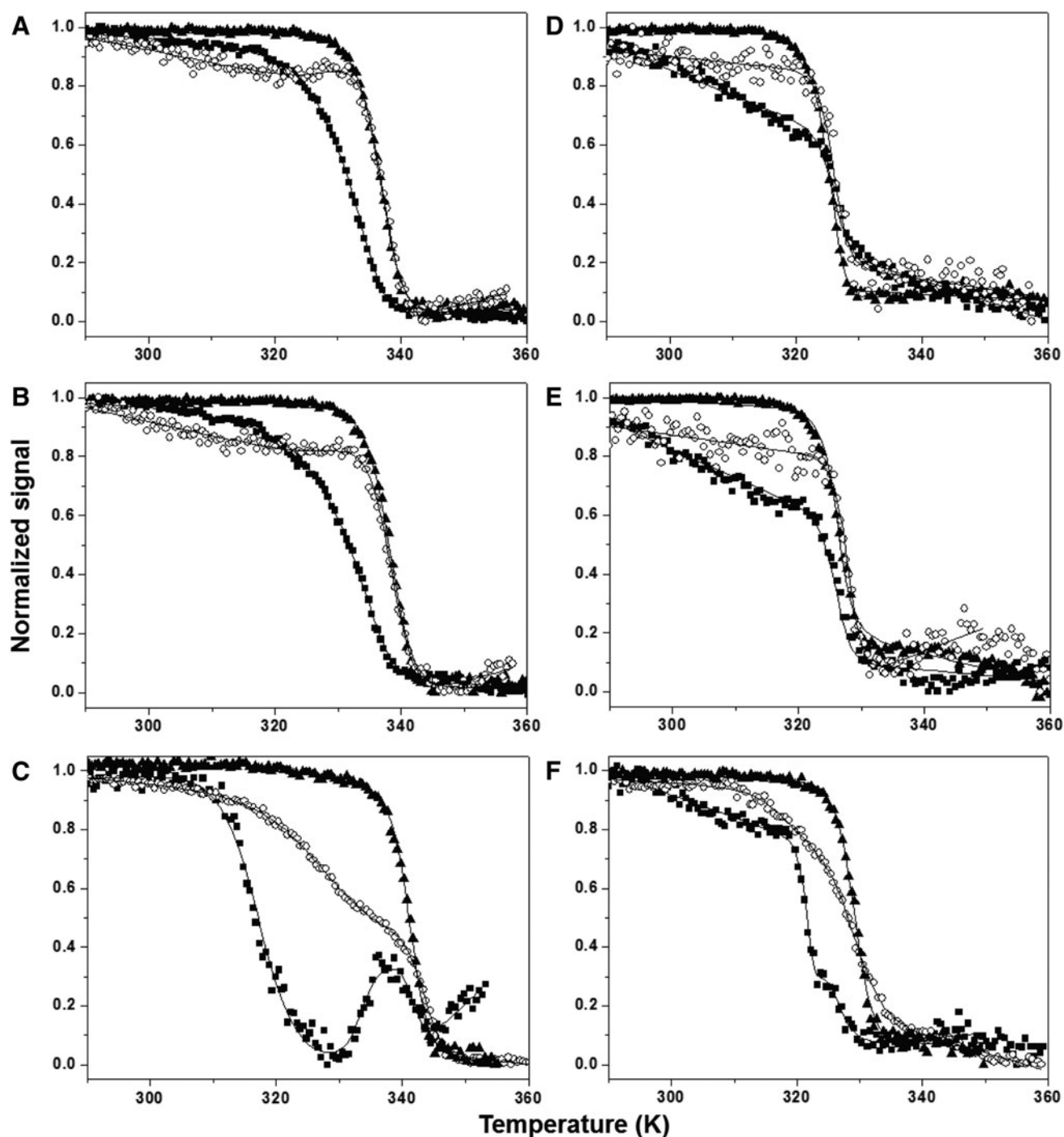
Supplementary Data



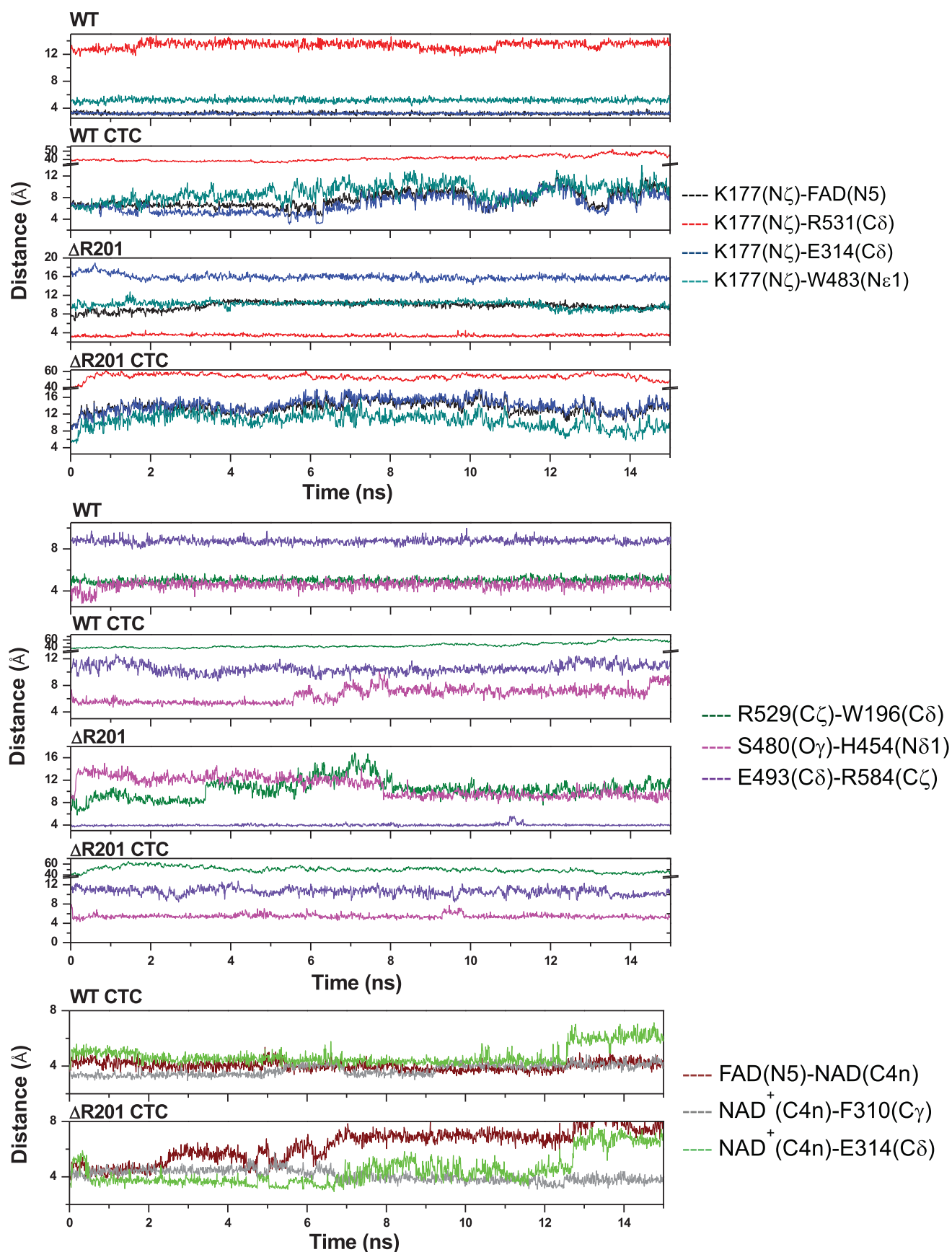
SUPPLEMENTARY FIG. S1. Spectroscopic properties of WT hAIF as a function of pH. Absorption spectra of (A) hAIF $_{\Delta 1-101ox}$ ($\sim 22 \mu M$) and (B) hAIF $_{\Delta 1-77ox}$ ($\sim 15 \mu M$) at 25°C. Position of band I and band II and A_{280}/A_{450} and A_{380}/A_{450} ratios provide information about similar or dissimilar electronic environments for the FAD cofactor. (C) Fluorescence emission spectra of hAIF $_{\Delta 1-101ox}$ ($2 \mu M$) in the aromatic region at 10°C with excitation wavelength 280 nm. Spectra were recorded in 50 mM potassium phosphate, pH 8.0 (black lines), 7.4 (red lines), 7.0 (green lines), or 6.2 (blue lines), at a final ionic strength of 150 mM. Samples containing hAIF $_{ox}$ are shown as continuous lines, and those containing hAIF $_{ox}$ mixtures with NADH in a 1:100 protein:NADH ratio are shown as dashed lines. FAD, oxidized form of flavin adenine dinucleotide; hAIF, human apoptosis-inducing factor; WT, wild type.



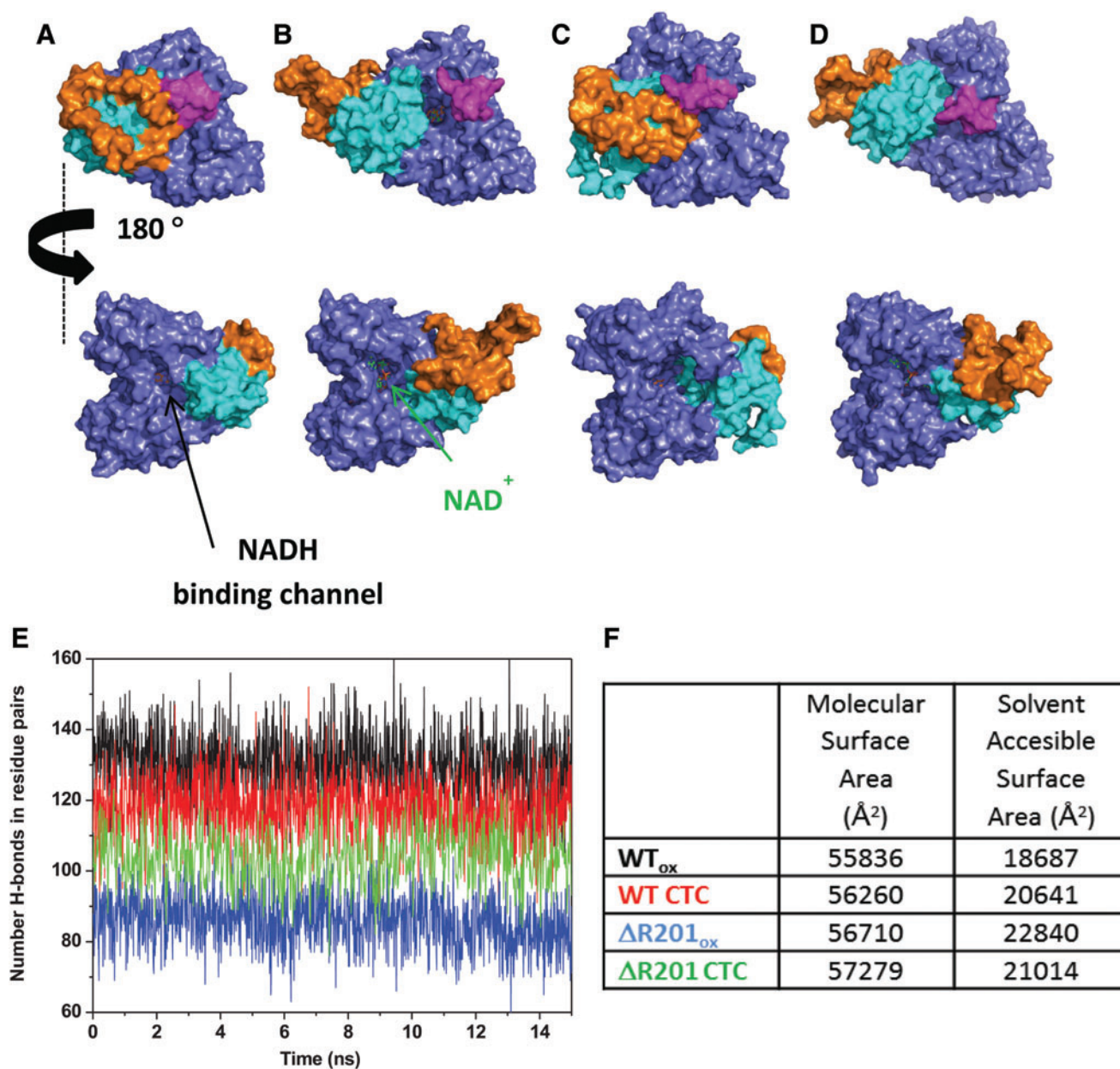
SUPPLEMENTARY FIG. S2. Spectral evolution upon reduction of hAIF $\Delta_{1-101ox}$ /hAIF Δ_{1-77ox} ($10 \pm 2 \mu M$) by NADH ($5 mM$) at (A) pH 8.0, (B) pH 7.4, (C) pH 7.0, and (D) pH 6.2. Measurements were carried out at $25^\circ C$ in $50 mM$ potassium phosphate at a final ionic strength of $150 mM$. The insets show the absorption evolution at 451 nm with the fit to a two species process (red line). Dashed lines correspond to the spectra of oxidized protein before mixing.



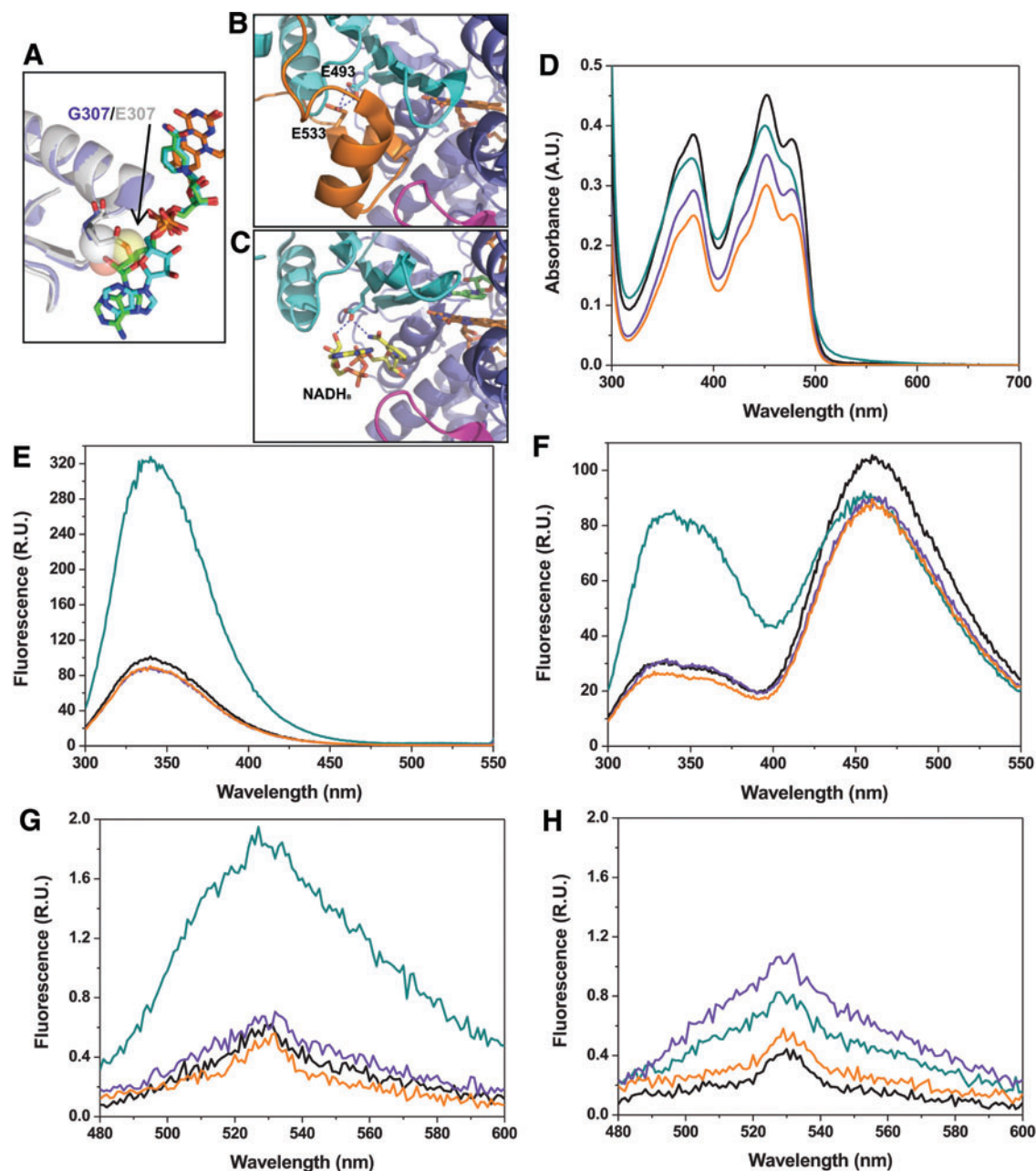
SUPPLEMENTARY FIG. S3. Thermal unfolding curves for hAIF_{Δ1-101ox} at (A) pH 7.4 and (B) pH 7.0, for hAIF_{Δ1-77ox} at (C) pH 6.2, and for their respective CTCs (1:100 protein hAIF:NADH ratio) at (D) pH 7.4, (E) pH 7.0, and (F) pH 6.2. Thermal denaturation was monitored by far-UV CD (210 nm, *white circles*), near-UV CD (300 nm, *black squares*; or 410 nm in the CTC), and flavin fluorescence emission (*black triangles*, 530 nm). The curves are roughly normalized from 0 to 1, and their global fits to three-state (for A, B, and F), four-state (C), and two-state (for D and E) unfolding models are represented by the continuous lines. Curves were recorded in 50 mM potassium phosphate at a final ionic strength of 150 mM. CD, circular dichroism; CTC, charge transfer complex; UV, ultraviolet.



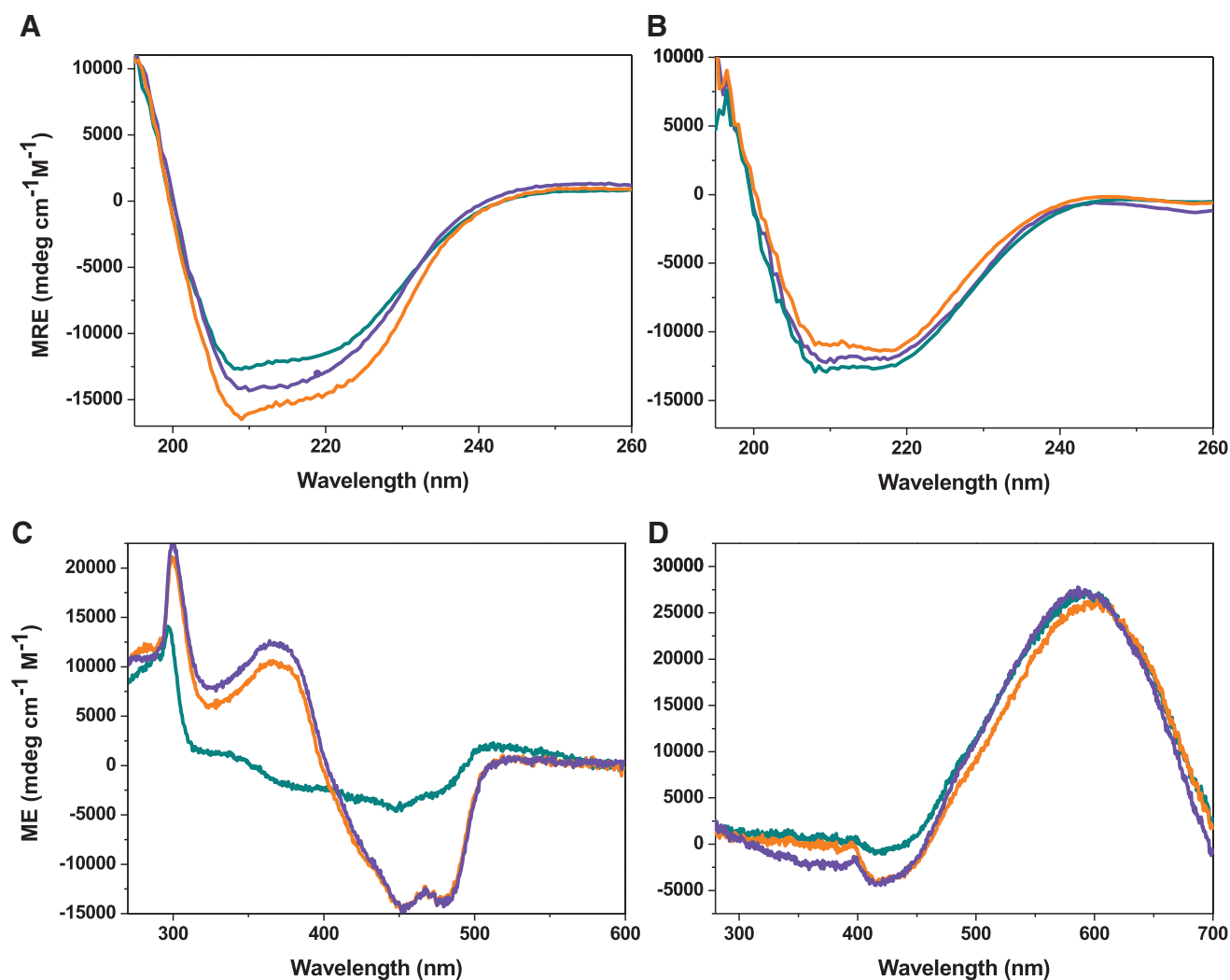
SUPPLEMENTARY FIG. S4. Dynamics of relaxation of models of hAIF Δ ₁₋₁₀₁ variants. Time evolution of selected distances along representative relaxation MD simulations of models for WT hAIF Δ _{1-101ox}, the Δ R201 variant and their corresponding CTC complexes.



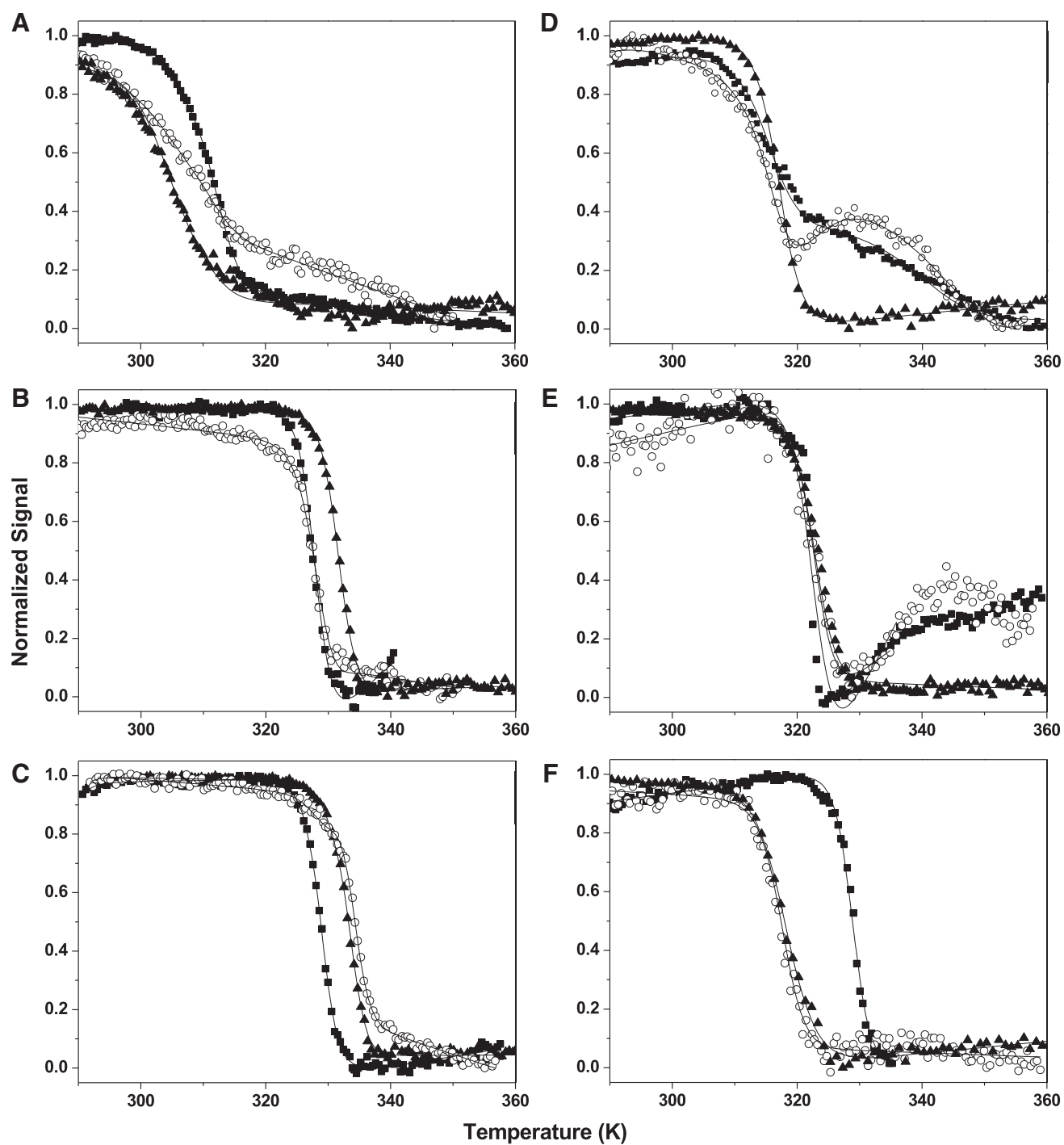
SUPPLEMENTARY FIG. S5. Conformational changes in hAIF_{Δ1-101ox} induced by CTC formation and R201 deletion. Surface representation of the representative model structures of (A) WT hAIF_{Δ1-101ox}, (B) WT CTC, (C) ΔR201 hAIF_{Δ1-101ox}, and (D) ΔR201 CTC after modeling of the missing C-loop residues (546–558), minimization, equilibration at 300 K, and short MD relaxation (15 ns) at 300 K. The oxidoreductase domain surface is in *violet* with its β -hairpin in *pink*, whereas the apoptotic domain surface is in *light blue* with the C-loop segment in *orange*. The FAD and NAD⁺ are shown in *sticks* with carbons in *orange* and *green*, respectively. (E) Number of H-bonds along the MD simulation production for the different AIF forms. Color code as in (F). (F) Summary of molecular and solvent surface area at the end of representative MD relaxations for each hAIF form.



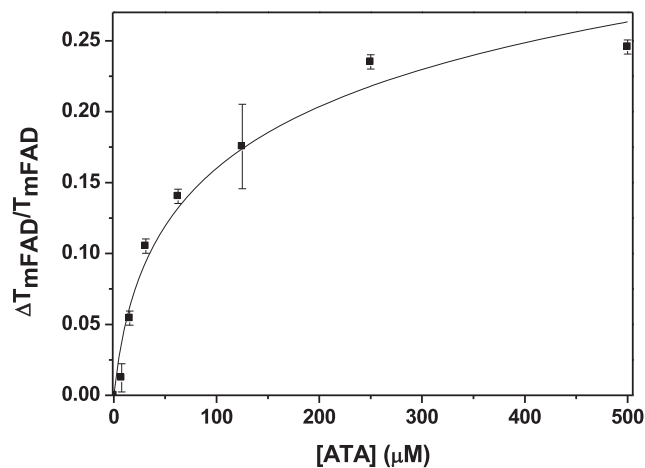
SUPPLEMENTARY FIG. S6. The Δ R201, G308E, and E493V hAIF Δ_{1-101} variants. (A) The G308E substitution alters the NAD⁺ ribose conformation in the CTC. The figure shows the equivalent position environment, G307, in the murine WT CTC structure (PDB 3gd4) in *violet* with the NAD⁺ in *green sticks*, and in the murine G307E CTC mutant structure in *white* and with the coenzyme in *blue* (PDB 5miv). *Spheres* show how the introduced Glu side chain uses the place of the coenzyme ribose. (B) E493 in WT hAIF Δ_{1-101} ox (PDB 4bv6) H-bonds E533 stabilizing the position of the C-loop short helices. (C) In the WT CTC structure (PDB 4bur), E493 contributes to stabilize the coenzyme molecule that replaces the C-loop short helices. Color code is as in Figure 3. (D) UV-visible spectra of WT (33 μ M), Δ R201 (33 μ M), G308E (26 μ M), and E493V (22 μ M) hAIF Δ_{1-101} ox variants. (E) Fluorescence emission spectra in the aromatic residues region of WT, Δ R201, G308E, and E493V hAIF Δ_{1-101} ox variants (2 μ M) recorded at 10°C with excitation wavelength of 280 nm. (F) Fluorescence emission spectra in the aromatic residues region of hAIF Δ_{1-101} variants (2 μ M) in the presence of 200 μ M NADH recorded at 10°C with excitation wavelength of 280 nm. (G) Fluorescence emission spectra in the flavin region of hAIF Δ_{1-101} ox variants (2 μ M) recorded at 10°C with excitation wavelength of 450 nm. (H) Fluorescence emission spectra in the flavin region of hAIF Δ_{1-101} (2 μ M) in the presence of 200 μ M NADH recorded at 10°C with excitation wavelength of 450 nm. All measurements were done in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM. Color code: WT (*black line*), Δ R201 (*dark cyan line*), G308E (*violet line*), and E493V (*orange line*).



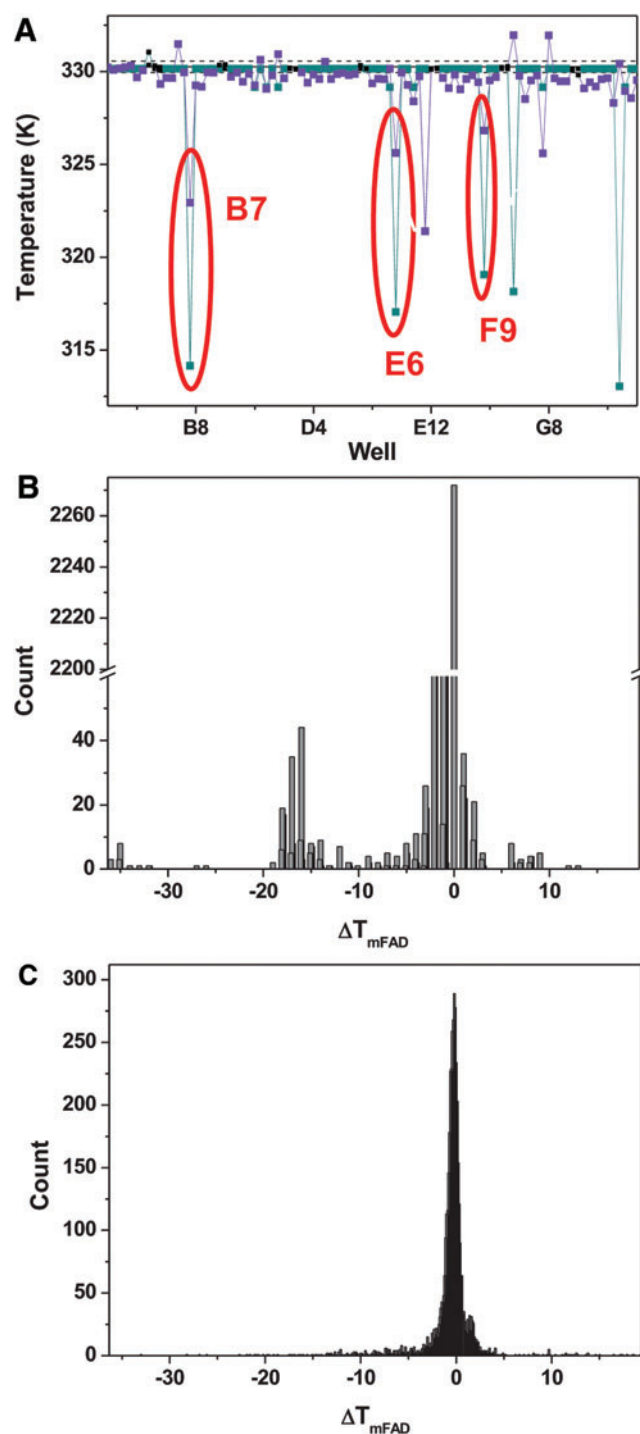
SUPPLEMENTARY FIG. S7. Circular dichroism spectra of $\Delta R201$, G308E, and E493V hAIF $_{\Delta 1-101ox}$ variants. Far-UV CD spectra of samples containing (A) 1 μM hAIF $_{\Delta 1-101ox}$ and (B) a mixture of 1 μM hAIF $_{\Delta 1-101ox}$ with 100 μM NADH for $\Delta R201$ (dark cyan line), G308E (violet line), and E493V (orange line) variants. Near-UV/vis CD spectra of samples containing (C) 20 μM hAIF $_{\Delta 1-101ox}$ and (D) a mixture of 20 μM hAIF $_{\Delta 1-101ox}$ with 2 mM NADH for $\Delta R201$ (dark cyan line), G308E (violet line), and E493V (orange line) variants. Spectra were recorded at 25°C in 50 mM potassium phosphate pH 8.0 at a final ionic strength of 150 mM.



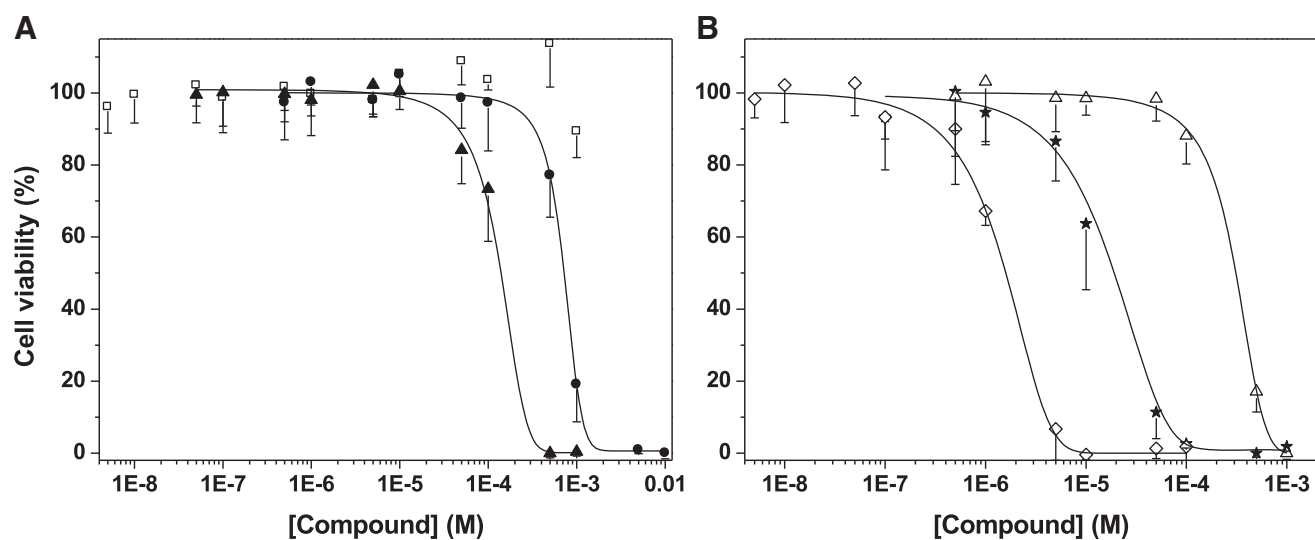
SUPPLEMENTARY FIG. S8. Unfolding properties of hAIF Δ_{1-101} variants. Thermal unfolding curves for (A) $\Delta R201$, (B) G308E, and (C) E493V hAIF $\Delta_{1-101ox}$ and for (D–F) their corresponding CTC (1:100 hAIF:NADH ratio). Thermal denaturation was monitored by far-UV CD (210 nm, *white circles*), near-UV CD (300 nm in (A–C), *black squares*) and vis CD (410 nm in (D–F), *black squares*), and flavin fluorescence emission (*black triangles*). The curves are shown roughly normalized from 0 to 1, and their global fits to three-state unfolding models are represented by *continuous lines*. Curves were recorded in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM.



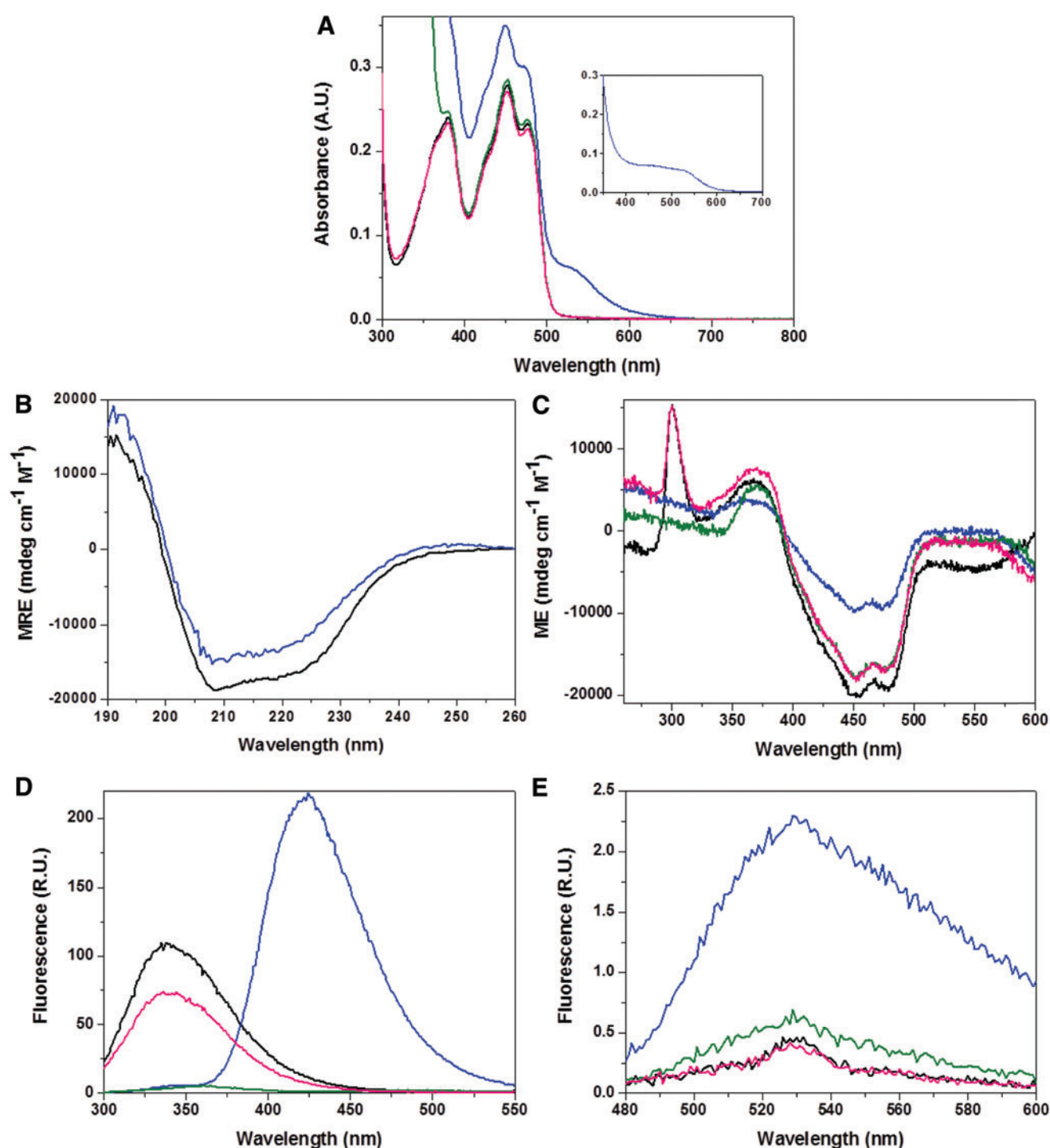
SUPPLEMENTARY FIG. S9. Interaction of hAIF_{Δ1-101} with ATA. Dose–response curve for the dependence of the WT AIF_{Δ1-101ox} T_{mFAD} on the ATA concentration. A total of 1 μM hAIF_{Δ1-101ox} was mixed with increasing concentrations of ATA in 96-well plates. Experiments were done in triplicate. The ATA dissociation constant, K_d^{ATA} , was determined by fitting the data to the equation (S1) $\frac{\Delta T_{mFAD}}{T_{mFAD}} = \frac{nRT_{mFAD}^0}{\Delta H_0} \ln\left(1 + \frac{[ATA]}{K_d^{ATA}}\right)$, where ΔT_{mFAD} estimates the extent of the ATA-induced protein destabilization ($\Delta T_{mFAD} = T_{mFAD}^0 - T_{mFAD}$); with T_{mFAD}^0 and T_{mFAD} being the midpoint denaturation temperatures in the absence and presence of ligand, respectively, ΔH_0 is the unfolding enthalpy of the protein in the absence of ATA, and n is the number of ligand molecules bound per hAIF molecule.



SUPPLEMENTARY FIG. S10. ThermoFAD HTS screening for compounds binding hAIF $\Delta_{1-101ox}$. (A) Compound-induced decrease in T_{mFAD} for hAIF $\Delta_{1-101ox}$ thermal FAD release upon unfolding as detected in a typical assay in a single 96-well plate. *Black squares* represent T_{mFAD} values for wells at columns 1 and 12 on each plate, which were used for controls in the absence of compound. T_{mFAD} values for controls are indicated as a *continuous black line*, whereas corresponding SD values are shown as *dashed black lines*. For samples containing compounds, *violet* and *cyan squares* represent T_{mFAD} values calculated using, respectively, the midpoint method and the inflection point analysis. Wells identified as containing potential hits by both methods are marked with a *red circle*. Experiments were performed in 50 mM potassium phosphate, pH 8.0, with 1 μM hAIF $\Delta_{1-101ox}$, 100, or 500 μM of the compound and 2.5% to 12.5% DMSO. (B) and (C) show the ΔT_{mFAD} histograms of all 11,424 compounds calculated using the inflection point analysis method and the midpoint analysis method, respectively. FAD, oxidized form of flavin adenine dinucleotide.

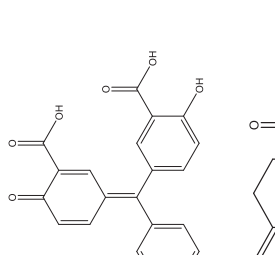
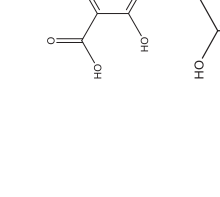
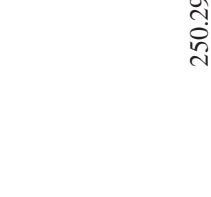

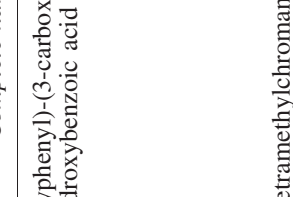



SUPPLEMENTARY FIG. S11. Viability of HeLa cells in the presence of hAIF Δ 1-101ox HTS hits. (A) C1 (□), C2 (●), and C5 (▲). (B) C8 (★), C9 (◇), and C11 (△). HeLa cells were grown for 24 hours, then treated with different concentrations of each compound and incubated for another 24 hours. Means \pm SD of at least three independent experiments are shown for each compound at each of the assayed concentrations. Dose-response dependence data were fitted to sigmoidal curves to determine EC₅₀ values.



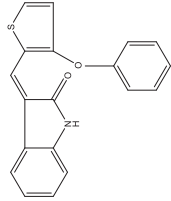
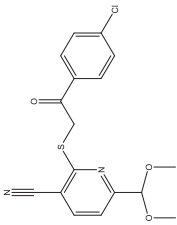
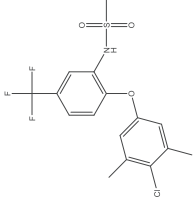
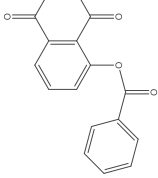
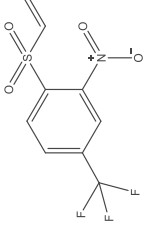
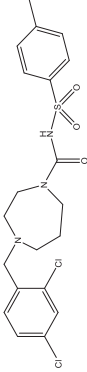
SUPPLEMENTARY FIG. S12. Effect of binders on the spectroscopic properties of hAIF $_{\Delta 1-101}$. (A) Visible absorption spectra of hAIF $_{\Delta 1-101ox}$ ($\sim 20 \mu M$) and in mixtures containing ATA ($100 \mu M$), C2 ($500 \mu M$), or C11 ($100 \mu M$). The inset shows the absorbance spectra of ATA ($500 \mu M$). (B) Far-UV CD spectra of hAIF $_{\Delta 1-101ox}$ ($1 \mu M$), and when containing ATA ($500 \mu M$). (C) Near-UV CD spectra of hAIF $_{\Delta 1-101ox}$ ($20 \mu M$) and in the presence of ATA, C2, or C11 (all at $500 \mu M$). (D) Fluorescence emission spectra in the aromatic residues region of hAIF $_{\Delta 1-101ox}$ ($2 \mu M$) in the presence of ATA, C2, and C11 (all at $500 \mu M$). (E) Fluorescence emission spectra in the flavin region of hAIF $_{\Delta 1-101}$ ($2 \mu M$) and in the presence of ATA, C2, or C11 (all at $500 \mu M$). Spectra were recorded in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM. CD spectra were recorded at 25°C. Fluorescence spectra were recorded at 10°C with excitation wavelength at 280 nm for the aromatics region and at 450 nm for the flavin region. Free hAIF $_{\Delta 1-101ox}$ is shown as a black line whereas mixtures with ATA, C2, and C11 are shown as blue, green, and pink lines, respectively.

SUPPLEMENTARY TABLE S1. PROPERTIES OF HAIF_{Δ1-101} HIGH-THROUGHPUT SCREENING HITS

C	Complete name	M _w	Structure	LogP
ATA	5-[(3-carboxy-4-hydroxyphenyl)-(3-carboxy-4-oxocyclohexa-2,5-dien-1-ylidene)methyl]-2-hydroxybenzoic acid (aurintricarboxylic acid, ATA)	422.34		3.84
C1	(±)-6-Hydroxy-2,5,7,8-tetramethylchromane-2-carboxylic acid (Trolox)	250.29		3.07
C2	2-[(2,6-Dichloro-3-methylphenyl)amino]benzoic acid sodium salt (meclofenamic acid)	318.13		5.63
C3	2-(3-Chloro-2-methylanilino)benzoic acid, 2-[(3-chloro-2-methylphenyl)amino]benzoic acid (tolfenamic acid)	261.70		5.00
C4	3,3',5-Triiodothyroacetic acid 4-(4-Hydroxy-3-iodophenoxy)-3,5-diiodophenylacetic acid (tiratricol)	621.93		5.93
C5	3-(3,5-Dibromo-4-hydroxybenzoyl)-2-ethylbenzofuran (benzbromarone)	424.08		5.70

(continued)

SUPPLEMENTARY TABLE S1. (CONTINUED)

<i>C</i>	<i>Complete name</i>	<i>M_w</i>	<i>Structure</i>	<i>LogP</i>
C6	3-[3-Phenoxy-2-thienyl)methylene]-1,3-dihydro-2H-indol-2-one	319.37		3.21
C7	2-[[2-(4-Chlorophenyl)-2-oxoethyl]thio]-6-(dimethoxymethyl)nicotinonitrile	362.83		3.28
C8	N-[2-(4-chloro-3,5-dimethylphenoxy)-5-(trifluoromethyl)phenyl]methanesulfonamide	393.80		4.57
C9	5,8-Dioxo-5,8-dihydronaphthalen-1-yl-benzoate	278.25		2.45
C10	2-Nitro-4-(trifluoromethyl) phenylvinyl sulfone	281.20		2.62
C11	N1-[[4-(2,4-dichlorobenzyl)-1,4-diazepan-1-yl]carbonyl]-4-methylbenzene-1-sulfonamide	456.38		3.92

ATA and compounds C1 to C5 were purchased from Sigma (Sigma-Aldrich, San Luis, MO), whereas compounds C6 to C11 were directly obtained from Maybridge Chemical Company, Altrincham, United Kingdom).

SUPPLEMENTARY TABLE S2. EFFECT OF
AURINTRICARBOXYLIC ACID, C2, AND C11
ON THE THERMAL STABILITY OF HAIF_{Δ1-101ox}

<i>Compound</i>	<i>T_{m1} (K)</i>	<i>T_{m2} (K)</i>
—	332 ± 1	337 ± 1
ATA	317 ± 1	328 ± 1
C2	329 ± 1	
C11	327 ± 1	

Values in the table were obtained by the global fitting of the near-CD and fluorescence thermal denaturation curves. Data were obtained in 50 mM potassium phosphate, pH 8.0, at a final ionic strength of 150 mM, from 283.15 to 363.15 K. Protein concentrations were ~2 μM and ~20 μM for fluorescence and near-UV CD, respectively.

Supplementary Reference

- S1. Cooper A and McAuley-Hecht KE. 1993. Microcalorimetry and the molecular recognition of peptides and proteins. *Phil Trans R Soc Lond A* 345, 23–35, 1993.

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- AU6: Please rewrite "After 24 hours of MNNG treatment, loss of cell viability reaches ~100%, independently on the HTS hits absence or presence" for clarity.
- AU7: Please rewrite "Thus, the FAD and NADH natural ligands might synergically contribute to *in vivo* partially overcome the $\Delta R201$ unfolding negative effect by enhancing protein stability" for clarity.
- AU8: In Ref. 17, please mention the name and location of the publisher.
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- AU10: Please rewrite "Relevant interactions in the WT model are shown in *blue dashed lines*, whereas the $\Delta R201$ structures also shown in *blue* the increase of the corresponding WT distances and in *red dashed lines* new interactions" for clarity.