

1 On the effect of using collision/reaction cell (CRC) technology in 2 3 4 2 single-particle ICP-mass spectrometry (SP-ICP-MS) 5 6

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37 14 38 39 40 15 **ABSTRACT** 41

42
43 16 In this work, the effects of using collision/reaction cell (CRC) technology in quadrupole-based ICP-MS
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45 17 (ICP-QMS) instrumentation operated in single-particle (SP) mode have been assessed. The influence
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47 18 of (i) various CRC gases, (ii) gas flow rates, (iii) nanoparticle (NP) sizes and (iv) NP types was evaluated
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49 19 using Ag, Au and Pt NPs with both a traditional ICP-QMS instrument and a tandem ICP-mass
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51 20 spectrometer. It has been shown that using CRC technology brings about a significant increase in the
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53 21 NP signal peak width (from 0.5 up to 6 ms). This effect is more prominent for a heavier gas (*e.g.*, NH₃)
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55 22 than for a lighter one (*e.g.*, H₂ or He). At a higher gas flow rate and/or for larger particle sizes (>100
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57 23 nm), the NP signal duration was prolonged to a larger extent. This effect of using CRC technology has
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1 been further demonstrated by characterizing custom-made 50 and 200 nm Fe₃O₄ NPs (originally
2 strongly affected by the occurrence of spectral overlap) using different CRC approaches (H₂ on-mass
3 and NH₃ mass-shift). The use of NH₃ (monitoring of Fe as the Fe(NH₃)₂⁺ reaction product ion at m/z =
4 90 amu) induces a significant peak broadening compared to that observed when using H₂ (6.10 ± 1.60
5 vs. 0.94 ± 0.49 ms). This extension of transit time can most likely be attributed to the
6 collisions/interactions of the ion cloud generated by a single NP event with the CRC gas and it even
7 precludes 50 nm Fe₃O₄ NPs to be detected when using the NH₃ mass-shift approach. Based on these
8 results, the influence of a longer peak width on the accuracy of SP-ICP-MS measurement data (NP
9 size, particle number density and mass concentration) must be taken into account when using CRC
10 technology as a means to overcome spectral overlap. To mitigate the potential detrimental effect of
11 using CRC technology in the characterization of NPs *via* SP-ICP-MS(/MS), the use of light gases and
12 low gas flow rates is recommended.

13
14 **Keywords:** Nanoparticles, single-particle ICP-MS(/MS), collision/reaction cell, spectral overlap,
15 chemical resolution, peak width

1. INTRODUCTION

Downscaling of materials to the size range of nanometers (≤ 100 nm in at least one dimension) can change their properties dramatically and lead to interesting new characteristics, which opens up new fields of application. Nowadays, such engineered nanoparticles (ENPs) are widely used in various fields, ranging from material manufacturing over environmental and life sciences to food processing.[1, 2] However, this massive use of ENPs is raising concern about their potential effects on the environment and human health,[3, 4] thus requiring the development of powerful and versatile analytical tools aiming at unraveling their behavior in, e.g., environmental and biological processes.[5]

Inductively coupled plasma – mass spectrometry (ICP-MS) plays an important role in this context, as it can be deployed as a single-particle analyzer; this approach is called single-particle (SP) ICP-MS (SP-ICP-MS).[6-10] The theory of SP-ICP-MS for the characterization of colloids in aqueous suspensions was first described by Degueldre *et. al.*[11] Since then, the number of papers reporting on the use of SP-ICP-MS has grown exponentially.[12, 13] SP-ICP-MS is based on the one-by-one introduction of NPs into the ICP, in which they are individually vaporized, atomized and the atoms thus obtained ionized. The ion cloud thus formed is then introduced into the mass analyzer and the transmitted ions are finally detected. The signals thus produced are of extremely short duration (~ 0.5 ms) for conventional ICP-MS.[14-17] The number of signal “spikes” detected and their integrated intensity can be used to calculate the size distribution, particle number density and mass concentration.

Despite its potential, ICP-MS, and hence SP-ICP-MS, also suffers from drawbacks, the occurrence of spectral interferences being a major one.[18-20] Over the years, different strategies have been used to overcome spectral overlap, caused by atomic and/or polyatomic ions occurring at the same nominal mass-to-charge (m/z) ratio as that of the target nuclide. The use of high mass resolution in sector field ICP-MS (HR-SF-ICP-MS) instrumentation is an elegant approach to get rid of spectral interferences,[21, 22] but so far, this type of instrumentation has not been widely / routinely used

1 for single-particle analysis.[23] The development of (i) new HR-SF-ICP-MS instrumentation capable of
2 performing ultra-fast measurements (at dwell times down to 10 μ s),[24, 25] and of (ii) SP data
3 treatment software (e.g., Nanocount) and/or calculation tools (RIKILT Wageningen UR) might
4 promote the use of HR-SF-ICP-MS in the context of NP characterization.[23, 26, 27] Alternatively,
5 also the use of collision/reaction cell (CRC) technology in quadrupole-based ICP-MS (ICP-CRC-QMS)
6 instrumentation is a versatile approach to overcome spectral interference and it has become even
7 more attractive with the introduction of tandem ICP-mass spectrometry (ICP-MS/MS) in 2012.[28-34]
8 ICP-MS/MS benefits from a better control over the ion-molecule processes occurring within the CRC,
9 as a result of the double mass selection provided by the additional quadrupole located before the
10 CRC. In addition, present-day ICP-QMS instruments are equipped with user-friendly software for SP
11 data treatment, making this technique suited for straightforward NP characterization in routine
12 applications.

13 However, the use of the CRC technology for overcoming spectral overlap also brings about specific
14 challenges when aiming at operating ICP-CRC-QMS instrumentation in SP mode. So far, this topic has
15 been scarcely studied. Kálmista *et al.*[19] have recently demonstrated that pressurizing a CRC with
16 an inert gas (for getting rid of spectral overlap relying on kinetic energy discrimination – KED)
17 influences the signal duration of a single NP event. However, to the best of the authors' knowledge,
18 no work to date has studied the potential effect of introducing reaction gases to avoid spectral
19 overlap *via* chemical resolution, although this approach is widely used within the ICP-MS(/MS)
20 community to achieve interference-free conditions for elemental analysis. Peak broadening and peak
21 shape have been extensively studied in the case of hyphenation of, e.g., chromatographic techniques
22 or laser ablation (LA) as a sample introduction system to ICP-MS. Detection capabilities can be
23 strongly degraded when working with wide chromatographic peaks,[35] while in the context of LA-
24 ICP-MS, recent efforts have been devoted to minimize aerosol dispersion aiming to achieve shorter
25 washout times, thus providing a better imaging resolution.[36] However, peak shape and duration
26 have not been systematically addressed in the case of SP-ICP-MS yet, especially when using CRC

1 technology as a means to avoid spectral overlap. This can most likely be attributed to the fact that
2 the shortest dwell times attainable with previous-generation ICP-MS instrumentations did not allow
3 well-defined peaks to be obtained (unless the detection system was adapted in-house, e.g., by using
4 a digital oscilloscope [15]), as these dwell times were usually longer than the duration of a single NP
5 event. In other words, only recent ICP-MS instrumentation provides sufficient temporal resolution to
6 accurately define the peak generated by a single NP event with multiple data points, instead of
7 reporting a single data point per NP.[17, 37-41] Furthermore, when using shorter dwell times (≤ 0.1
8 ms) in SP-ICP-MS, a suitable data treatment software is required for the integration of the individual
9 signal values composing the transient signal for every NP.

10 In this work, we aim at assessing the effects of pressurizing the CRC of two different ICP-QMS
11 instruments (a single quadrupole and a tandem ICP-MS unit) operated in SP mode on the
12 characterization of different NP types using different CRC gases. The conclusions derived from the
13 use of Au, Ag and Pt model NPs were further tested by analyzing Fe_3O_4 NPs, which strongly suffer
14 from the occurrence of spectral interference. From all results obtained, critical considerations on and
15 practical recommendations for the use of CRC technology in ICP-MS operated in SP mode are
16 provided.

18 2. EXPERIMENTAL

19 2.1. Standards, samples and reagents

20 Ultra-pure water was obtained from a Direct-Q Milli-Q system (Millipore, France). Ag, Au, Fe and Pt
21 solutions prepared from commercially available 1 g L^{-1} single-element standards (Instrument
22 Solutions, The Netherlands) were used for method development and calibration purposes. Various
23 suspensions of spherical (nano)particles (nanoComposix Inc, Czech Republic) with different sizes and
24 chemical compositions were used in this work: 50 (51 ± 6), 80 (79 ± 7), 100 (98 ± 13) and 200 ($202 \pm$
25 29.7) nm Ag, 50 (49.9 ± 2.2 , Ultra UniformTM), 70 (71 ± 9) and 100 (97 ± 10) nm Au, 50 (46.2 ± 5.1) and

1 70 (72.2 ± 4.2) nm Pt, and custom-made 50 (54 ± 5) and 200 (196 ± 64) nm Fe₃O₄. It needs to be
2 noted that manufacturing monodisperse 200 nm Fe₃O₄ NPs proved difficult, resulting in a wider,
3 bimodal size distribution (see section 3.2.). Size distributions, particle number densities and mass
4 concentrations for these (nano)particle suspensions were provided by the manufacturer (determined
5 by TEM and gravimetric analysis). Also, 75 nm AgNPs (74.6 ± 3.8 , NIST SRM 8017) and 30 nm AuNPs
6 (27.6 ± 2.3 , NIST SRM 8012) from the National Institute of Standards Technology (NIST, USA) were
7 used as NP reference materials for the determination of the transport efficiency (TE). Additionally,
8 we also studied some of the NPs using TEM (Tecnai T20, ThermoFisher Scientific, USA) to validate the
9 SP-ICP-MS measurements and/or to provide additional information for data interpretation. The NP
10 suspensions were gently stirred and sonicated before appropriate dilution. For all the experiments
11 intended to measure the average signal duration of individual NP events, no more than 3 000 NP
12 events per minute were detected (generally, this value was lower than 1 000 NP events per minute).
13 For the determination of the size distribution of FeNPs (see section 3.2.), dilution factors leading to
14 less than 1 000 NPs per minute were used in all cases. It needs to be noted that all measurements
15 were carried out at ultra-fast data acquisition speed (dwell time of 100 μ s). The dilution factors were
16 calculated based on Poisson statistics to minimize the occurrence of double events during SP-ICP-
17 MS/(MS) measurement (see section 3.3.).

18 *2.2. Instrumentation*

19 All measurements were carried out using either of two quadrupole-based ICP-MS instruments: a
20 NexION 300X (Perkin Elmer, USA) ICP-QMS instrument and an Agilent 8900 (Agilent Technologies,
21 Japan) tandem ICP-MS/MS instrument. The NexION 300 is equipped with a quadrupole-based CRC
22 (also known as dynamic reaction cell – DRC) and with a quadrupole ion deflector to bend the ion
23 beam over a 90 degrees angle before its introduction into the CRC. Both the hardware and software
24 of this instrument were upgraded to enable faster data acquisition (down to 1 value every 10 μ s).
25 The sample introduction system comprised a 0.5 mL min⁻¹ concentric quartz nebulizer and a quartz
26 cyclonic spray chamber. The Agilent 8900 is equipped with two quadrupole mass filters (Q1 and Q2)

1 and an octopole CRC cell mounted in-between. Use of the MS/MS mode facilitates the study of the
2 reactions occurring within the CRC as a result of the different scanning tools available, such as
3 precursor and product ion scans.[34]

4 In this work, the CRC of both instruments was pressurized with different gases, He, H₂ and NH₃/He
5 (10% NH₃ in He), and the results thus obtained were compared with the standard “no gas” or
6 “vented” mode. In addition, different on-mass and mass-shift approaches typically used for
7 overcoming spectral overlap in ICP-CRC-QMS instrumentation were evaluated.

8 *2.3. Data treatment*

9 In addition to the SP software available in the MassHunter 4.4 Workstation for the Agilent 8900 ICP-
10 MS/MS unit (Agilent Technologies, Inc. 2017) and the Nano Application module of the Syngistix
11 software, version 2.1. for the NexION 300 X (Perkin Elmer Inc., 2018), an in-house developed Hyper
12 Dimensional Image Processing (HDIP) software was used to accurately calculate the NP signal
13 duration. This software was developed for post-processing of spectral data and images in the context
14 of laser ablation (LA) ICP-MS and is meanwhile commercially available from Teledyne Cetac
15 Technologies for this purpose, but was adapted to identify the individual NP events appearing
16 randomly during SP-ICP-MS measurements. By using the HDIP software, the average signal duration
17 (at different %-ages of the peak height) and the average peak shape of NP events measured under
18 different conditions could be calculated easily (**Figure 1**). To prevent data calculation artefacts, such
19 as a contribution from the background signal to the peak width, the selection criteria can be
20 modified by adjusting the software settings. HDIP uses a peak-finding approach based on a
21 parameter ‘delta’. A data point is considered a “local maximum” when its intensity is higher than
22 both that of the preceding and that of the subsequent data point by a value lower than delta times
23 that maximum signal intensity. By using the delta parameter, the sensitivity of the peak detection
24 algorithm can be adapted. This list of established maxima is further filtered by comparison of the
25 corresponding intensities to a thresholding criterion: maxima below a cut-off value (a percentage of

1 the total signal height range) are rejected. Based on a sufficient number of peaks, peak statistics can
2 be obtained and, when the algorithm is complete, the cumulative distribution function and average
3 peak profile are shown. Off-line, outliers resulting from instrument-derived signal spikes or other
4 artifacts resulting from the software data treatment were removed. For this purpose, any data point
5 more than 1.5 interquartile ranges (IQRs) below the first quartile or above the third quartile was
6 considered as an outlier.

8 **3. RESULTS AND DISCUSSION**

9 *3.1. Effect of the CRC conditions on the signal duration of single NP events*

10 *3.1.1. Effect of CRC mode*

11 To assess the effect of the CRC mode on the signal duration of single NP events, different methods
12 were developed by pressurizing the CRC system of an ICP-MS/MS instrument with He, H₂ or NH₃ and
13 by comparing the results thus obtained with those obtained in “no gas” or “vented” mode. These
14 different methods were selected aiming to assess the potential influence of different strategies that
15 are widely applied in ICP-QMS to overcome spectral overlap. An inert gas, such as He, can be used in
16 the CRC to slow down polyatomic ions more than the atomic target ions, such that the former can be
17 selectively removed *via* kinetic energy discrimination (KED mode), while in the case of H₂ and NH₃, a
18 combination of collisions and reactions is relied on to overcome spectral overlap.[28, 29, 32] For this
19 experiment, different methods for the detection of AuNPs were developed using an ICP-MS/MS
20 instrument equipped with an octopole CRC, as this can be seen as the more suitable technique to
21 study the in-cell chemistry in ICP-QMS.[34] In the case of no gas, He (5.0 mL min⁻¹) and H₂ (7.0 mL
22 min⁻¹), Au⁺ was monitored at its original mass-to-charge ($m/z = 197$ amu) ratio (on-mass), while with
23 NH₃ (2.3 mL min⁻¹ of a mixture of 10% NH₃ in He and an additional flow of 1 mL min⁻¹ of He), both Au⁺
24 and the best suited product ion of the reaction between Au⁺ and NH₃ (mass-shift approach) were
25 monitored. For mass-shift method development, the reactivity between Au⁺ and NH₃ was assessed

1 using product ion scanning (PIS) at different NH₃ gas flow rates, and Au(NH₃)⁺ (m/z = 214 amu) was
2 identified as best suited reaction product ion. Instrument settings, including the axial acceleration
3 (AA) function available for the Agilent 8900 ICP-MS/MS instrument (see section 3.2.1.), were
4 optimized to achieve the highest signal-to-background ratio. It needs to be noted that the use of
5 collision and/or reaction gases is (typically) not required for interference-free determination of
6 Au,[42] and that the purpose of developing these methods was to evaluate the influence of various
7 CRC modes on the characterization of NPs *via* SP-ICP-MS. It can be reasonably hypothesized that the
8 most important conclusions drawn by studying AuNPs as “model NPs” can also be extrapolated to
9 other NP types.

10 **Table 1** and **Figure 2** show the effect of the different CRC modes on the signal duration and peak
11 shape of individual NP events. For this experiment, highly monodisperse 50 nm AuNPs were
12 measured using all modes, except for the NH₃ mass-shift approach, for which 100 nm AuNPs were
13 selected due to the relatively low sensitivity of this method. To assess the potential influence of NP
14 size on the signal duration, 100 nm AuNPs were also measured in “no gas” or “vented” mode, as a
15 correlation between transit time of the ion cloud and particle diameter has earlier been reported on
16 in literature.[43] This behaviour has been related with the lower kinetic energy of larger particles,
17 thus traveling slower through the plasma and, thus, experiencing more diffusion and widening of the
18 ion cloud.[44] However, under the experimental conditions of this study, no significant difference
19 was found between 50 and 100 nm AuNPs (average peak width at 1% of the maximum height of 0.60
20 ± 0.10 and 0.61 ± 0.13 ms, respectively), suggesting that within this NP size range and in the case of
21 “no gas” mode, the influence of NP size can be considered negligible compared to the effect caused
22 by using the CRC technology. As can be seen, also no significant difference was found between the
23 “no gas” and H₂ modes (0.63 ± 0.10 ms), while a slight increase in signal duration (0.74 ± 0.13 ms)
24 was already noticed when pressurizing the CRC with He. However, this effect turned out to be
25 drastically different when using NH₃, both for the on-mass and the mass-shift NH₃ methods, with a
26 significant peak tailing leading to average peak widths at 1% of the maximum height of 2.59 ± 0.91

1 and 1.84 ± 0.86 ms for on-mass and mass-shift, respectively. This peak tailing of the ion burst and the
2 specific peak shape of an individual NP event have been observed in previous works and have been
3 related with the expansion of the ion cloud during the extraction through the sampler cone.[16, 45]
4 However, based on the results obtained in this work, this effect is significantly increased by the use
5 of CRC technology. In addition, **Table 1** and **Figure 2** also show an important peak broadening at 10
6 and 50% of the maximum height for both NH₃ methods, thus indicating that the effect of using NH₃ in
7 the CRC is not merely limited to a longer peak tail.

8 These results demonstrated that the use of NH₃ gas into the CRC of an ICP-QMS instrument can
9 strongly modify the occurrence of NP events during SP-ICP-MS analysis. The significant differences
10 between the use of NH₃ and the other CRC gases can most likely be attributed to the larger size of
11 NH₃ molecules compared to that of H₂ molecules and/or He atoms *i.e.*, the larger collisional cross-
12 section of the CRC gas increases the probability of collision/interaction with the ion plume created
13 for each NP event, thus significantly disturbing the kinetic energy distribution of the ions generated
14 from a single NP. Consequently, some of those ions might lose kinetic energy to a different extent,
15 and thus, they will be detected at slightly different moments in time (**Figure 3** illustrates the effect of
16 pressurizing a CRC system during the measurement of the fast transient signal for an individual NP).
17 This effect explains the longer peak width and the relatively long tail observed for the AuNPs
18 measured using both NH₃ modes. However, this explanation and the mechanism illustrated in Figure
19 3 is based on the hypothesis that the most important factor contributing to the peak broadening is
20 the collision between the ion cloud and the CRC gas, while the time required for the reaction taking
21 place in the CRC and the subsequent acceleration of the newly created molecular ions and their
22 transport out of the cell may also play an important role in the case of the mass-shift approach.
23 When taking the combination of these two effects (collision and reaction) into account, a longer
24 signal duration would be expected for the latter approach, while the experimental data rather show
25 a longer signal duration in the case of on-mass monitoring (at 1 % of the peak height). It should
26 however be taken into account that this observation might be the result of the lower sensitivity with

1 the mass-shift approach. The relatively low signal intensity observed in the NH₃ mass-shift approach
2 makes it more difficult to distinguish the peak tail from the background signal. As a result, it cannot
3 be excluded that the peak width at 1% of the peak height was underestimated. The larger standard
4 deviations on the signal duration calculated for the NH₃ methods also indicate a higher variability
5 between NP signals measured using NH₃, further supporting the different interactions between the
6 ionic cloud and the reaction gas. The occurrence of this effect at an even larger magnitude might
7 even split the population of a single NP event into various populations giving rise to partially
8 overlapping peaks, thus having a strong influence on the accurate characterization of NPs *via* SP-ICP-
9 MS (see section 3.3.).

11 3.1.2. Effect of gas flow rate

12 3.1.2.1. Effect of NP size

13 In addition to the effect of different CRC modes, the gas pressure in the CRC or the gas flow rate may
14 also have an influence on the signal duration of single NP events. For evaluating this effect, Ag
15 particles of different sizes (50, 75, 80, 100 and 200 nm) were measured using a single-quadrupole
16 ICP-MS instrument equipped with a quadrupole CRC pressurized with He at different gas flow rates
17 (0, 1, 2, 3, 4 and 5 mL min⁻¹). **Figure 4A** shows the signal duration as a function of the He gas flow rate
18 for all Ag particle sizes studied. As can be seen, no significant differences in signal duration were
19 observed in the absence of He (“no gas” mode – signal durations ranging from 0.76 to 0.81 ms,
20 expressed as peak width at 1% of the maximum height), except in the case of 50 nm AgNPs for which
21 a slightly shorter duration (0.62 ms) was found compared to all the other sizes. It needs to be noted,
22 though, that the intensities of the NP spikes generated by 50 nm AgNPs were relatively low (7.2
23 counts at the maximum of the peak on average), so that the proximity to the LOD_{size} may
24 compromise the accuracy of the signal duration measurements (see section 2.3.). **Figure 4A** indicates
25 that the addition of He into the CRC increases the average signal duration of individual NP events

1 proportionally to the gas flow rate for all the particle sizes, thus supporting the hypothesis of
2 collisions between the ion cloud and the CRC gas as reason for the anomalously wide peak profiles
3 (see **Figure 3**). In addition, **Figure 4B** shows the increase in particle event signal durations at 5 mL
4 min^{-1} of He compared to 0 mL min^{-1} of He for the AgNPs of several sizes. The signal durations were
5 established to be higher by a factor of 1.31, 1.44, 1.65, 2.25 and 2.95 at 5 mL min^{-1} He than in no gas
6 mode for particles of 50, 75, 80, 100 and 200 nm, respectively. Interestingly, these results clearly
7 demonstrate that also the particle size has an important influence on the extent of the effect exerted
8 by using CRC technology on the signal duration of NP events. This different behavior for particles as a
9 function of their size can most likely be explained by the denser ion cloud formed for the larger
10 particles, thus increasing the probability of interactions with the cell gas. Thus, in addition to the CRC
11 mode, type of gas and gas flow rate, attention also needs to be paid to the size of the particles prior
12 to SP-ICP-MS analysis using CRC technology to overcome spectral overlap (see Section 3.3.).

14 3.1.2.2. Effect of NP type

15 To assess the potential effect of the chemical composition of the NPs, the signal durations for 50 and
16 100 nm AuNPs and for 50 and 70 nm PtNPs were measured under the same conditions as described
17 in the previous section, *i.e.*, He gas flow rates ranging from 0 to 5 mL min^{-1} , and the results thus
18 obtained were compared with those of AgNPs. In this way, we aimed at assessing the potential
19 differences in peak width for individual NPs with a different chemical composition, but with
20 approximately the same diameter (Ag, Au and Pt NPs of 50 nm, Ag and Pt NPs of 70 and 75 nm,
21 respectively, and Ag and Au NPs of 100 nm). **Figures 5A** (Ag and Au NPs) and **5B** (Ag and Pt NPs) show
22 the effect of increasing the He flow rate on the signal duration for the different NP types. It is
23 interesting to note that the results obtained in this experiment for 50 nm AuNPs using a single-
24 quadrupole ICP-MS instrument equipped with a quadrupole CRC (0.62 ± 0.19 and 0.79 ± 0.25 for 0
25 and 5 mL min^{-1} He, respectively) are not significantly different from those reported in section 3.1.1.

1 for the same AuNPs using a tandem ICP-MS instrument equipped with an octopole CRC (0.60 ± 0.10
2 and 0.74 ± 0.13 for 0 and 5 mL min⁻¹ He, respectively). Self-evidently, the influence of different types
3 of mass spectrometers and/or of CRC technologies on the transit time of the ion cloud generated by
4 individual NPs merits further study, but these results seem to indicate that the effect of introducing
5 an additional quadrupole and/or of using a different type of CRC (quadrupole vs octopole) might be
6 negligible compared to the effect of adding a CRC gas.

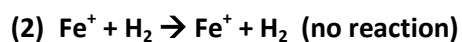
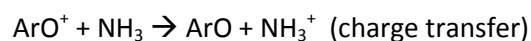
7 In the comparison of different NP types, on the one hand, very similar results were obtained for Ag
8 and Au NPs, both in terms of average signal duration and effect of the He flow rate. On the other
9 hand, significantly shorter peak widths were found for PtNPs compared to AgNPs and AuNPs of the
10 same size, while the effect of increasing the He flow also seemed less pronounced. In a previous
11 work, Kéri *et al.*[43] reported significantly longer signal durations for AuNPs than for AgNPs and tried
12 to attribute this difference to physical processes that can influence the transit time of the ion cloud
13 (*e.g.*, temperature and enthalpy of melting and boiling, and/or ionization energy). However, all
14 physical processes studied suggested that AgNPs should produce longer transient times than AuNPs,
15 and it was concluded that, most likely, some effects occurring in the plasma (*e.g.*, charge-transfer
16 reactions) can also contribute. In this work, the significant differences observed for PtNPs do not
17 seem to be related with, *e.g.*, ionization energy (Au > Pt > Ag). In addition to the physicochemical
18 properties of the constituting element, also the morphology of the NPs may influence the signal
19 duration of single NP events. In order to shed light onto these differences, TEM images of Ag, Au and
20 Pt NPs were obtained, as shown in **Figure 6**. Interestingly, it was found that – in contrast to Ag and
21 Au NPs – both 50 and 70 nm PtNPs were agglomerates of NPs of approximately 1 – 2 nm. This
22 different morphology can be the cause of the significantly shorter signal duration established for
23 PtNPs, although the influence of a different chemical composition and thus, different
24 physicochemical properties cannot be ruled out. Ag and Au NPs showed a similar behavior, but both
25 are spherical and metallic NPs from the same group within the periodic table. Thus, they are
26 expected to have rather similar chemical and physical properties. In the case of PtNPs, we cannot

1 assess the influence of the difference in chemical composition as the aberrant behavior can most
2 likely be attributed to the different morphology. Based on these results, we cannot provide a full
3 description of how NP type (composition) affects the signal duration of a single NP event, but our
4 results suggest that the use of the temporally resolved information generated by SP-ICP-MS could
5 also provide some nano-morphological information.

3.2. Characterization of Fe_3O_4 NPs via SP-ICP-MS/MS operated in different CRC modes

3.2.1. Method development for interference-free determination of Fe using CRC technology and ICP-MS/MS

As a result of their unique physical and chemical properties, iron nanoparticles (FeNPs) are of the utmost interest for environmental and biological applications.^[46-48] However, the characterization of FeNPs by SP-ICP-MS is hampered by the occurrence of interfering ions, such as the ubiquitous ArO^+ polyatomic ions with the same nominal m/z ratio as the most abundant Fe isotope (56, relative isotopic abundance of 91.75%), thus requiring the development of powerful ICP-MS methods able to address this problem of spectral overlap. Two strategies were developed for this purpose, (1) an ICP-MS/MS mass-shift approach based on the formation of an NH_3 -based reaction product ion ($Fe(NH_3)_x^+$) and (2) an ICP-MS/MS on-mass approach using H_2 as CRC gas (measurement of Fe^+ ions at their original m/z ratio). In both cases, effective removal of the ArO^+ polyatomic interference can be achieved, according to the following reactions:



1 In the first case, product ion scans were performed at different gas flow rates (between 1 and 5 mL
2 min⁻¹, see **Figure 7A**), and Fe(NH₃)₂⁺ at m/z = 90 amu was identified as the best suited reaction
3 product ion. Maximum sensitivity was obtained at 3 mL min⁻¹ of NH₃ (mixture of 10% NH₃ in He). For
4 the H₂ on-mass approach, a gas flow rate of 6.5 mL min⁻¹ of H₂ was selected after optimization of the
5 signal-to-background ratio (see **Figure 7B**). It needs to be noted that, after optimization, the
6 sensitivity of the two methods developed was approximately 110 000 cps per 1 μg L⁻¹ of an ionic
7 standard solution of Fe.

8 In addition, also the effect of using the axial acceleration (AA) option available for the Agilent 8900
9 ICP-MS/MS instrument was evaluated in the case of the NH₃ mass-shift approach. The AA function
10 allows a voltage gradient to be applied to accelerate ions along the ion guide axis of the octopole,
11 providing an enhanced sensitivity when operating an ICP-MS/MS instrument in reaction cell mode.
12 This counteracts the relatively low kinetic energy of newly created reaction product ions, reducing
13 ion transmission efficiency due to collisional scattering or space charge repulsion. The axial
14 acceleration voltage increases the kinetic energy of those reaction product ions, thus enhancing their
15 transport out of the cell and into the second quadrupole.[49] We studied the influence of AA on the
16 signal intensities of the reaction product ion selected at different gas flow rates by varying the AA
17 voltage between -2 and +2 V (a summary of these results is visualized in **Figure 8**). It was observed
18 that the optimum AA voltage increases with the increase of the gas flow rate (0.0, 0.1, 0.2, 0.5 and 1
19 V at 1, 2, 3, 4 and 5 mL min⁻¹ NH₃, respectively). Based on these results, it was demonstrated that the
20 use of higher gas flow rates in the CRC reduces the ion transmission efficiency of the reaction
21 product ions, thus necessitating a higher AA voltage to compensate for this effect. After optimization
22 of the AA, the highest signal intensity for Fe(NH₃)₂⁺ was found at 3 mL min⁻¹ of NH₃ and an AA setting
23 of 0.2 V (120 000 cps per 1 μg L⁻¹ Fe). In this specific case, the operation of the ICP-MS/MS instrument
24 with the AA off resulted in a reduction of the sensitivity (ion transmission efficiency) by a factor of
25 ~2.

3.2.2. Effect of using CRC technology on the SP-ICP-MS analysis of 50 and 200 nm Fe₃O₄ NPs

The two methods developed (based on H₂ on-mass monitoring and an NH₃ mass-shift approach, respectively) were subsequently used to study the effect of using the CRC technology on the characterization of 50 and 200 nm Fe₃O₄ NPs by means of SP-ICP-MS/MS. Both methods provide approximately the same sensitivity for the interference-free determination of ⁵⁶Fe, and thus, they are *a priori* also expected to provide approximately the same NP detection capabilities when operating the ICP-MS/MS instrument in SP mode. However, while in the case of the H₂ on-mass approach, the signal spikes corresponding to individual NP events were found to be clearly separated from the background (ionic) signal for both 50 and 200 nm Fe₃O₄ NPs, the use of NH₃ mass-shift did not allow to distinguish the 50 nm Fe₃O₄ NP spikes from the background signal. As can be seen from **Figure 9** (depicting the signal for individual events for Fe₃O₄NPs of 200 nm), this can be attributed to the effect of using NH₃ on the signal shape and height. Since the sensitivities of the H₂ on-mass and NH₃ mass-shift methods were approximately the same (see section 3.3.1.), the lower maximum peak height obtained for the same NPs using the NH₃-based method (160 counts) compared to that of H₂ (1500 counts) can only be explained by the more pronounced peak broadening, as already shown in section 3.1.1. In contrast to the significant differences in peak height, the peak areas calculated using the HDIP software corresponded to 3900 and 3700 counts for H₂ and NH₃, respectively. In addition, it needs to be pointed out that the use of AA did not improve the situation, as similar peak widths were obtained when operating the ICP-MS/MS instrument with AA on/off.

In this case, a very pronounced peak broadening was observed as a result of the combination of (i) the use of NH₃ gas (see section 3.1.1) and (ii) the measurement of relatively large particles (200 nm, see section 3.1.2.1.). As a result of these two effects, an average signal duration of 6.10 ± 1.60 ms was calculated for 200 nm Fe₃O₄ NPs using the NH₃ mass-shift approach, while for the same NPs, the average signal duration was only 0.94 ± 0.49 ms when using the H₂ on-mass method (see **Figure 9**).

Figure 10 shows the size distribution for 50 and 200 nm Fe₃O₄ NPs. It is clear that the very pronounced increase in signal duration induced by the introduction of NH₃ gas into the CRC of an ICP-

1 MS/MS instrument has a detrimental effect on the NP detection capabilities and it even precludes
2 measuring 50 nm Fe₃O₄ NPs using this approach, despite of the lower effect expected for 50 nm
3 compared to 200 nm NPs. In this context, it needs to be noted that the NH₃ flow rate was optimized
4 to achieve the highest signal-to-background ratio for ionic Fe, as this can be seen as the most
5 commonly used tuning approach. However, this selection of NH₃ gas flow rate clearly compromises
6 the NP detection capabilities due to the significant peak broadening induced by this relatively high
7 gas flow rate. In other words, it can be assumed that these circumstances do not always provide the
8 best NP detection capabilities. In order to obtain further insight, the effect of using lower NH₃ gas
9 flow rates (1.0, 1.5, 2.0, 2.5 and 3.0 mL min⁻¹) on the detectability of 50 nm Fe₃O₄ NPs was evaluated.
10 Unfortunately, none of the conditions selected allowed us to detect 50 nm Fe₃O₄ NPs, as in the case
11 of this specific reaction, a reduction in the NH₃ gas flow rate also strongly decreases the reaction
12 efficiency and thus, the sensitivity.

13 For 50 nm (H₂) and 200 nm (H₂ and NH₃), a very good agreement between the size distribution
14 obtained using SP-ICP-MS/MS and the TEM results provided by the manufacturer was obtained.
15 However, comparison of the size distribution for the 200 nm Fe₃O₄ NPs using both approaches
16 showed a slightly wider distribution for the NH₃-based method compared to the use of H₂, while also
17 NPs with sizes exceeding 300 nm appear when using the NH₃ mass-shift approach. This can also be
18 attributed to the larger signal duration and the higher probability of overlap of peaks corresponding
19 to individual NPs, as will be discussed in the next section. In a previous work, we saw that the use of
20 CH₃F as a reaction gas for the interference-free determination of SiO₂ NPs also resulted in wider size
21 distributions for the larger particles (300 and 400 nm) compared to the use of H₂.**[50]** In that work,
22 an Agilent 8800 ICP-MS/MS instrument was used and as the minimum dwell time with this
23 instrument is limited to 3 ms, we could not further study the differences between both approaches,
24 but the longer signal duration of individual NP events using CH₃F is most likely the explanation for
25 those differences.

3.3. Influence of the signal duration on the accuracy of the SP-ICP-MS measurements

The increase of the transit time of the NP ion cloud as a result of the use of CRC technology in ICP-QMS instrumentation exerts a detrimental effect on the SP-ICP-MS analysis, thus necessitating the application of some additional measures to minimize the consequences in terms of accuracy of the NP size, particle number density and mass concentration results. The occurrence of double events might be seen as one of the consequences of an anomalously high NP signal duration. In SP-ICP-MS, the NPs are randomly distributed in the diluted aqueous solutions, and this solution should be sufficiently diluted to avoid two or more particles being detected at the same time. Poisson statistics can be used to estimate the probability of one or more than one particle reaching the plasma in a fixed time interval: [9, 15, 17, 18]

$$P_x = \frac{\lambda^x}{x!} e^{-\lambda}$$

where x is the number of particles and λ is the number of particles entering the plasma in a fixed period of time (= nanoparticle flux, Q_{NP}). λ can be calculated using the following equation:

$$\lambda = Q_{NP} \times t = C_{NP} \times Q_{sam} \times \eta_{neb} \times t$$

C_{NP} is the particle number density, Q_{sam} is the sample flow rate, η_{neb} is the nebulization efficiency and t is the fixed period of time. This time has been considered as the dwell time (t_{dwell}) when working in the millisecond range, and as the peak width of the transient signal (w_{peak}) when working in the microsecond range (generally established at 0.5 ms). However, the assumption of a signal duration of ~0.5 ms is not valid when using CRC technology to avoid spectral overlap, as demonstrated in the previous sections. The peak widths obtained using this approach can range from 0.5 to 6 ms depending on the size of the NP and the type of CRC gas and the flow rate at which it is introduced into the CRC. In order to further illustrate the effect of this longer signal durations on the probability of more than one event occurring within the same time interval (particle coincidence), the probabilities were calculated at different NP fluxes (5, 20 and 50 s^{-1}) and peak widths (0.5, 1, 2.5 and 5 ms) based on Poisson statistics and the corresponding results are shown in **Table 2**. It can be seen

1 that for conventional NP fluxes, for instance 20 Hz, the probability of particle coincidence increases
2 approximately 2 orders of magnitude for NP events with a peak width of 5 ms compared to 0.5 ms.

3 This increase in the probability of particle coincidence will result in an underestimation of the particle
4 number density and an overestimation of the NP size. Depending on the type of software used for NP
5 data treatment, two individual NP events can be resolved, even if they partially overlap (see **Figure**
6 **11A**). In this specific case, accurate results can still be achieved for the particle number density, but
7 incomplete peak areas will lead to inaccurate NP sizes and mass concentrations. A further refinement
8 of this approach would lie in deconvolution of overlapping peaks, although the asymmetry of the NP
9 peaks poses additional challenges.[51] Therefore, an estimation of the signal duration of individual
10 NP events prior to SP-ICP-MS using CRC technology is recommended in order to select an appropriate
11 dilution factor. However, this approach requires an increase of the measurement time to avoid
12 inaccurate results because of poor counting statistics.

13 In addition to the occurrence of particle coincidence, the SP detection capabilities (LOD_{size}) can also
14 be affected by pronounced peak broadening (see **Figure 11B**). The overlap of the signal from the NP
15 with that of the background can potentially hamper the detection of the smallest NPs. As a result,
16 the average NP size determined could be biased high, while an inaccurate particle number density
17 will be obtained as a result of the counting of the larger NPs only.

18 In addition to the direct effect on the accuracy of the NP characterization, it needs to be noted that
19 the use of some approaches reported on in the literature to calibrate the NP size based on the use of
20 the NP signal durations can also be affected when using CRC technology.[43] If such approaches are
21 intended, it has to be considered that the duration depends on the size and the
22 collisions/interactions between the ion cloud and the CRC gas, and thus, higher variations are
23 expected.

24 Finally, also a positive side effect of the increase in signal duration for NP events using CRC
25 technology can be noted. This strategy can be used to monitor multiple (≥ 2) nuclides in SP mode.

1 Montañó *et. al.* have reported the use of microsecond dwell times for the detection of two elements
2 in the same nanoparticle, although the relatively short signal duration of individual NP events and
3 the necessity of working with a sufficiently high settling time to account for ion flight time through
4 the quadrupole and the detector (~100 μ s) leads to a poor definition of the peak shape (generally
5 only 1 data point to define the maximum of the peak).[38] This temporal resolution can clearly be
6 improved by using CRC technology, as is clearly illustrated in **Figure 9**, for which the maximum of the
7 peak using the H₂ on-mass approach is defined by 1 data point (100 μ s) only, in contrast to the NH₃
8 mass-shift approach, for which the maximum of the peak is defined by at least 5 data points (500 μ s).
9 Obviously, there is a trade-off in terms of size detectability, but there are situations where this longer
10 NP peak width can be beneficial and used for multi-nuclide monitoring using a sequential mass
11 spectrometer in the context of (i) cross-validation of the results using two isotopes from the same
12 element, (ii) measurement of bimetallic NPs (with a core of one metal and a shell of another) and (iii)
13 use of isotope dilution as a calibration strategy. Self-evidently, the use of the quasi-simultaneous
14 detection capabilities provided by time-of-flight ICP-MS (ICP-TOFMS) instrumentation provides far
15 greater capabilities for multi-isotope monitoring in this context.[52-54]

17 4. CONCLUSION

18 The use of CRC technology in SP-ICP-MS with quadrupole-based instrumentation is mandatory for
19 those NPs strongly affected by the occurrence of spectral overlap (*e.g.*, Fe₃O₄ NPs). However, this
20 work has demonstrated (using two different instrument types) that the introduction of a gas into the
21 CRC has an important effect on the signal duration of a single NP event. The use of different CRC
22 modes revealed that gases with a larger collisional cross-section (*e.g.*, NH₃) induce a more
23 pronounced peak broadening than gases with a smaller one do (*e.g.*, He and H₂). In addition, the gas
24 pressure in the CRC (determined by the gas flow rate) and even the type and morphology of the NPs
25 might influence the transit time of the NP ion cloud through the mass spectrometer. These

1 conclusions have been further illustrated by the characterization of Fe₃O₄ NPs, which are strongly
2 affected by the occurrence of spectral interference. The detrimental effect of using an NH₃ mass-shift
3 approach did not enable 50 nm Fe₃O₄ NPs to be distinguished from the background signal, while this
4 was not found to be the case when using a H₂ on-mass approach. The effect was found to be most
5 pronounced for larger particles, and thus, the characterization of NPs of smaller dimensions are less
6 affected by the use of CRC technology. To minimize the effect for larger particles, light gases and low
7 gas flow rates are recommended. In addition, the effect of a longer NP signal width on the accuracy
8 of the SP-ICP-MS measurements, such as higher probability of particle coincidence, higher LOD_{sizes},
9 and inaccurate size distribution, particle number density and mass concentration results has been
10 described into detail. This effect, however, can also be exploited for multi-isotope monitoring using
11 sequential ICP-MS instrumentation operated in SP mode.

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1 **Figure captions:**

2 **Fig. 1** SP-ICP-MS data treatment using the in-house developed HDIP software. **Figures 1A** and **1B**
3 show the raw data (intensity vs time) from one measurement replicate of a AuNP standard. **Figure 1C**
4 illustrates the identification of the signal spikes corresponding to single NP events; the selection
5 criterion was based on different parameters, such as a selected delta value, lower relative maximum
6 threshold and higher relative minimum threshold. **Figure 1D** shows the average peak profile and the
7 results obtained for the complete data set (peak profile, signal duration, number of peaks found,
8 area average, *etc*). The red triangles highlight the peak maxima, while the integration intervals are
9 visualized by green triangles.

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11 **Fig. 2** Effect of the different CRC modes on the signal duration and peak shape of individual NPs. In all
12 cases, 50 nm AuNPs were measured, except for NH₃ mass-shift, for which 100 nm AuNPs were
13 selected. The average normalized peak profile of approximately 1 000 NP events per CRC mode was
14 calculated using the HDIP software. In the case of both NH₃ methods, a mixture of 10% NH₃ in He was
15 used in combination with an extra flow of 1 mL min⁻¹ He, as recommended by the instrument
16 manufacturer.

17
18 **Fig. 3** Graphical illustration of the effect of introducing a gas in the CRC of an ICP-CRC-QMS
19 instrument on the NP signal duration of individual NP events. The graphic shows the disturbances in
20 the ion cloud generated by individual NPs as a result of interactions with a gas in the CRC system and
21 the thus caused effect on the NP signal distribution.

22
23 **Fig. 4 A:** Increase of the signal duration as a function of the He flow rate for AgNPs of various sizes
24 ranging from 50 to 200 nm. **B:** Increase of the signal duration ratio (NP signal duration at 5 mL min⁻¹
25 relative to that at 0 mL min⁻¹) for AgNPs with different sizes.

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Fig. 5 Comparison of the signal duration of Ag and Au NPs (**Figure 5A**) and Ag and Pt NPs (**Figure 5B**) as a function of increasing He flow rates.

Fig. 6 TEM images obtained for 50 nm Ag, Au and Pt NPs. It can be observed that the PtNPs are agglomerates of NPs of approximately 1 – 2 nm.

Fig. 7 A: Search for the best suited $^{56}\text{Fe}(\text{NH}_3)_x^+$ reaction product ion (using a $1 \mu\text{g L}^{-1}$ Fe solution) at different gas flow rates using product ion scanning. **B:** Optimization of the H_2 gas flow rate for the on-mass measurement of $^{56}\text{Fe}^+$ using the ramp cell gas option: signal intensity for $1 \mu\text{g L}^{-1}$ Fe and background signal (left y-axis, logarithmic scale) and signal-to-background ratio (right y-axis).

Fig. 8 Effect of axial acceleration (AA) on the intensity of the reaction product ion selected ($^{56}\text{Fe}(\text{NH}_3)_2^+$).

Fig. 9 Effect of the use of H_2 on-mass and NH_3 mass-shift approaches on the signal duration and peak shape of 200 nm Fe_3O_4 NPs. The average peak profile of approximately 1 000 NP events per CRC mode was calculated using the HDIP software.

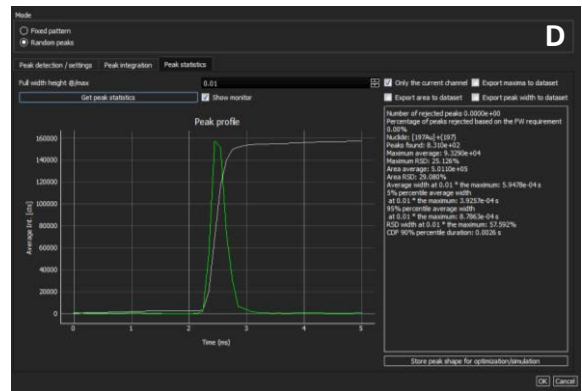
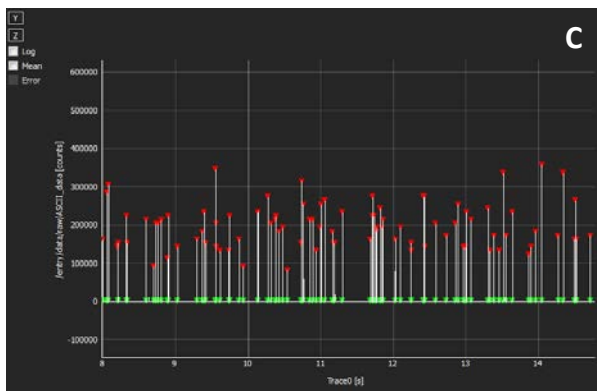
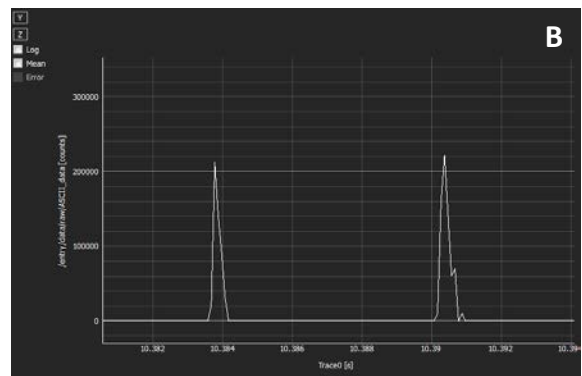
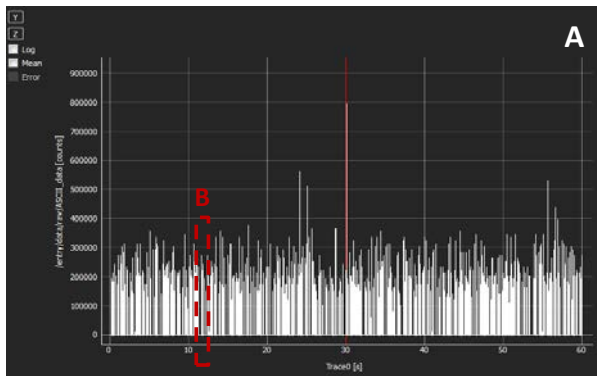
Fig. 10 Size distribution of 50 (H_2 – A) and 200 (H_2 – B and NH_3 – C) nm Fe_3O_4 NPs using SP-ICP-MS/MS. The TEM size distributions were provided by the manufacturer.

1 **Fig. 11** (A) Probability of occurrence of double events (particle coincidence) as a function of the signal
2 duration of individual NP events and (B) effect of signal broadening on the accuracy of NP
3 characterization and their LOD_{size}.

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1 Fig. 1

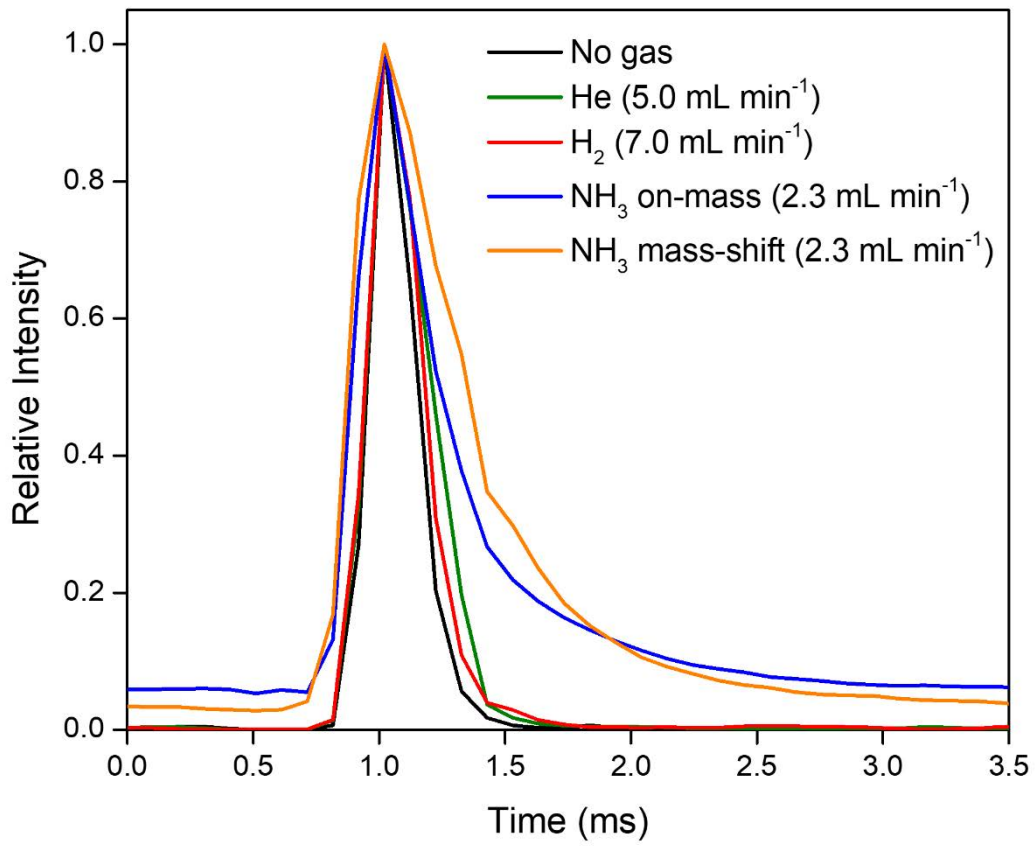
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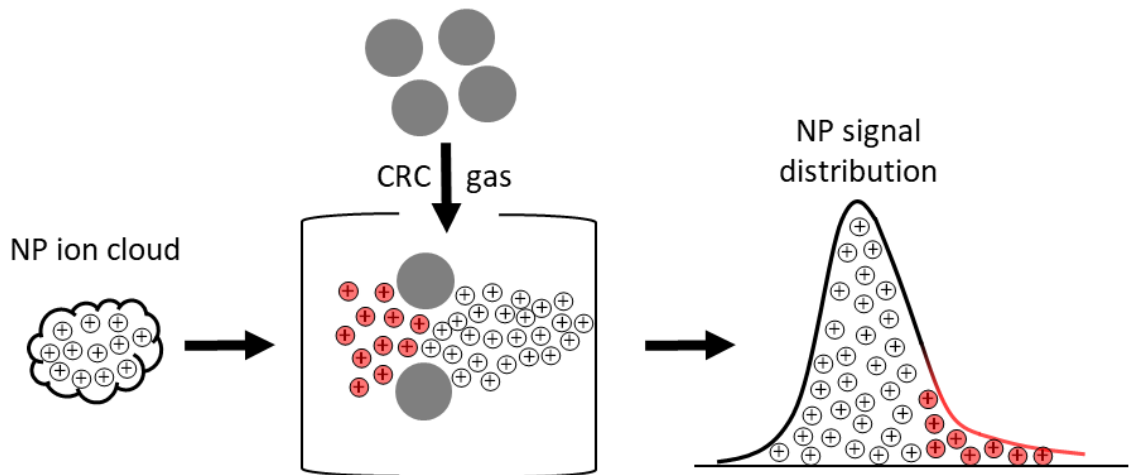


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1 **Fig. 3**

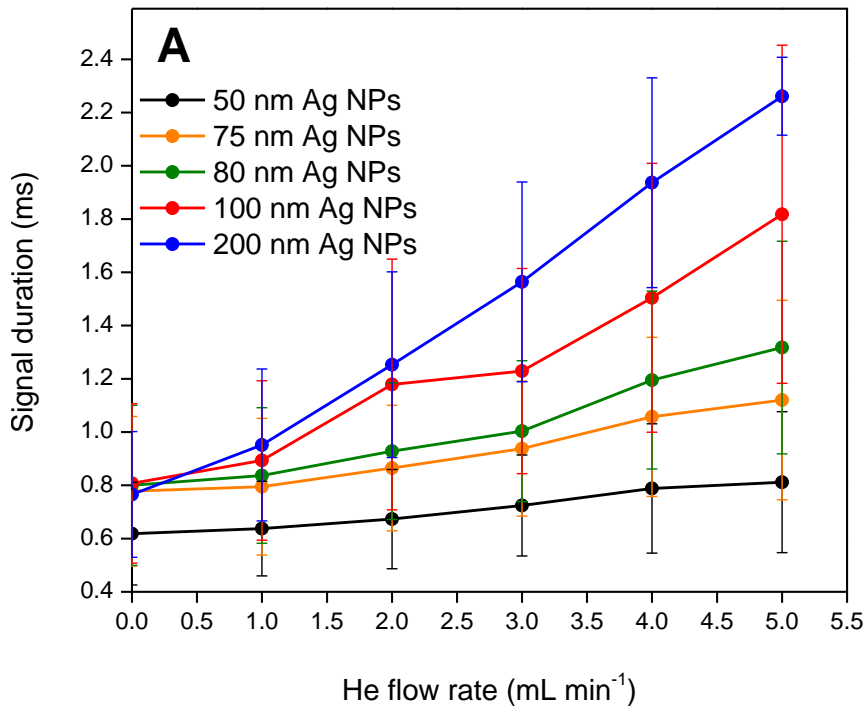


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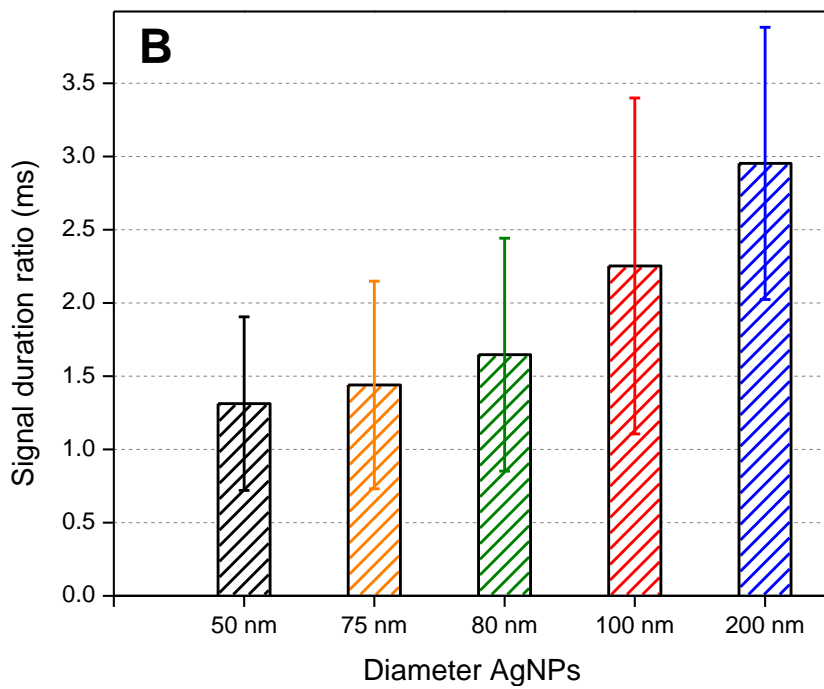
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1 **Fig. 4**



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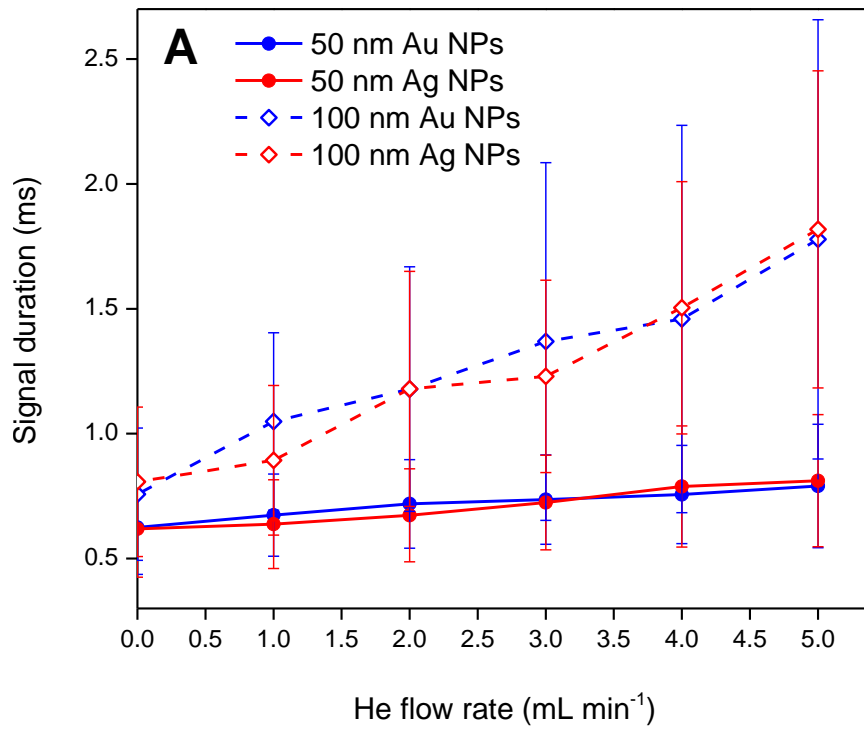
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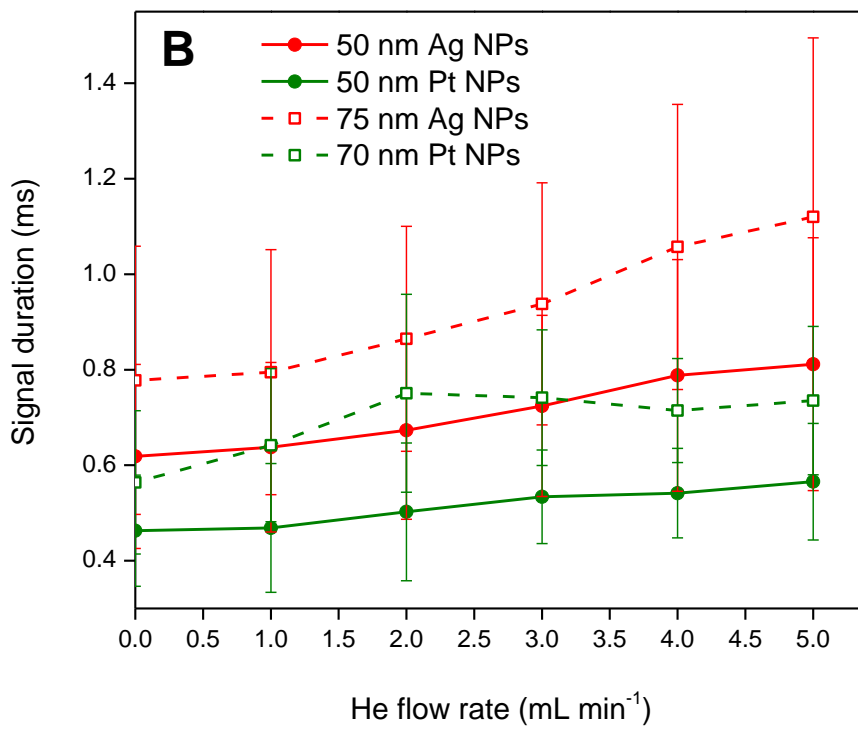
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1 Fig. 5



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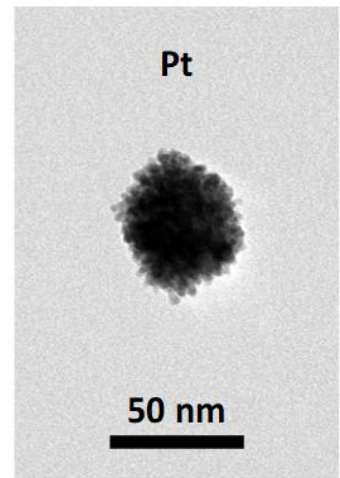
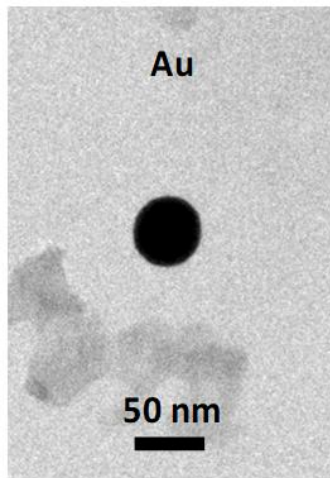
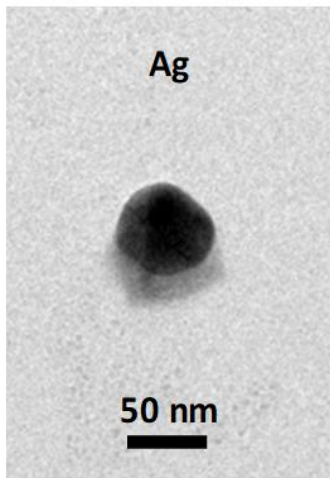


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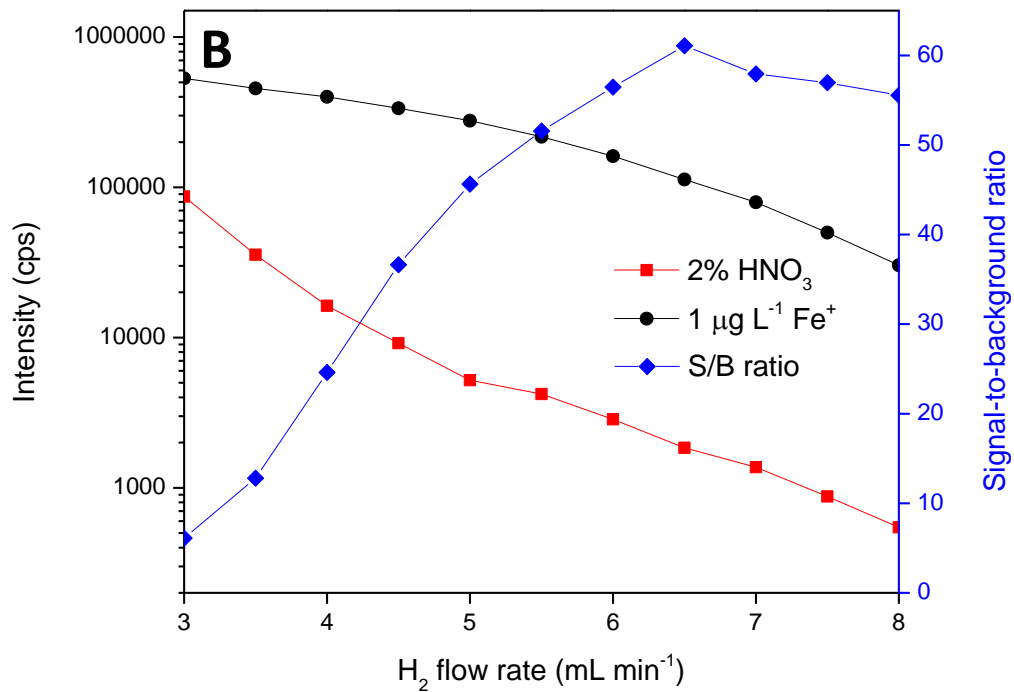
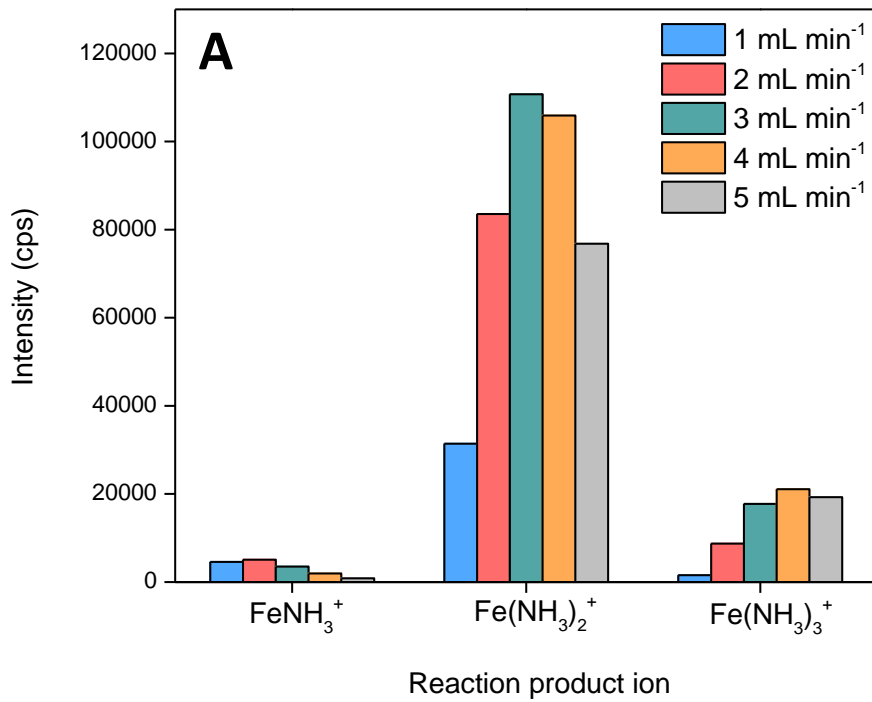
1 **Fig. 6**



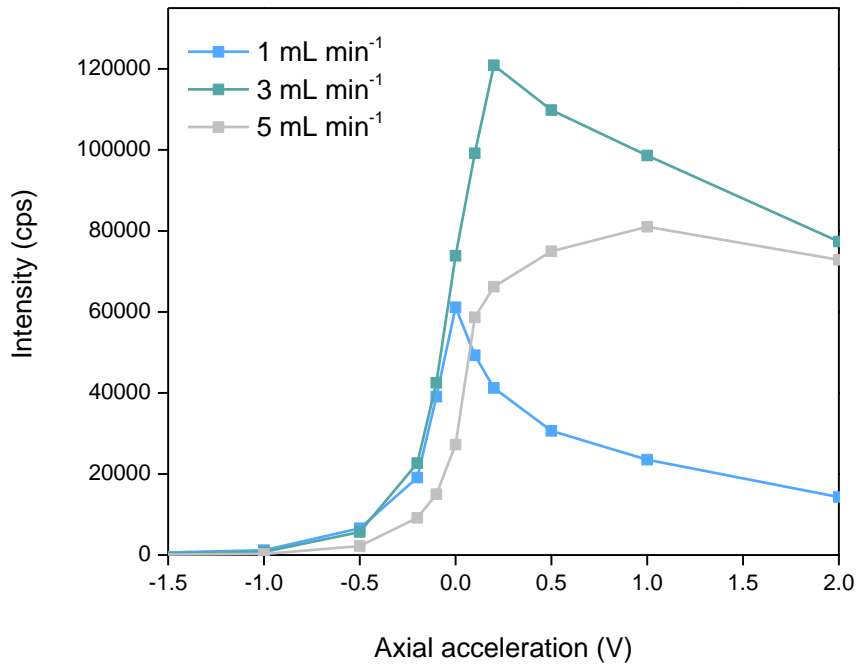
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1 **Fig. 7**



1 **Fig. 8**

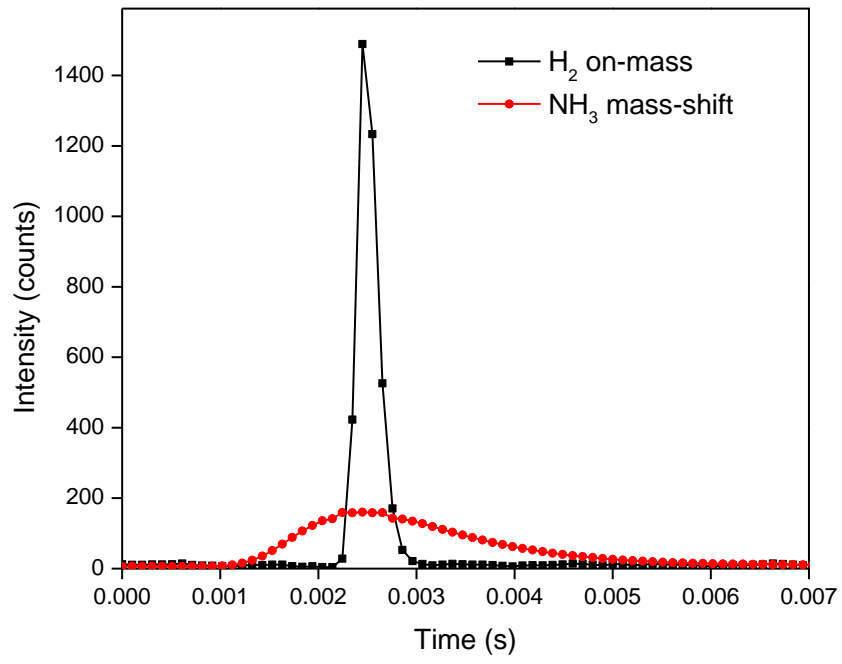


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1 Fig. 9



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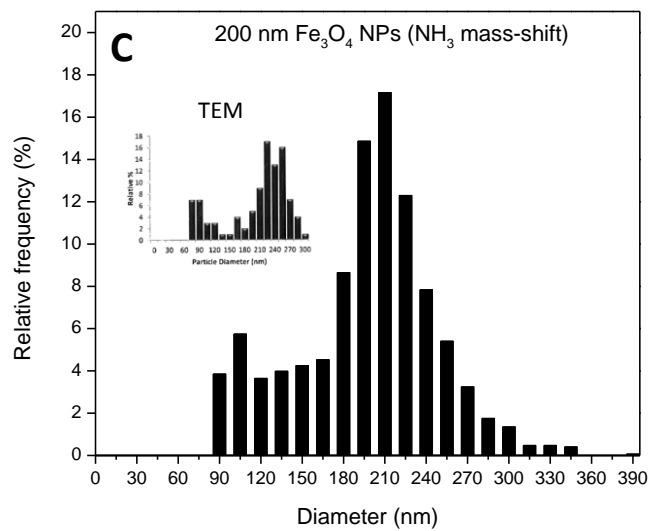
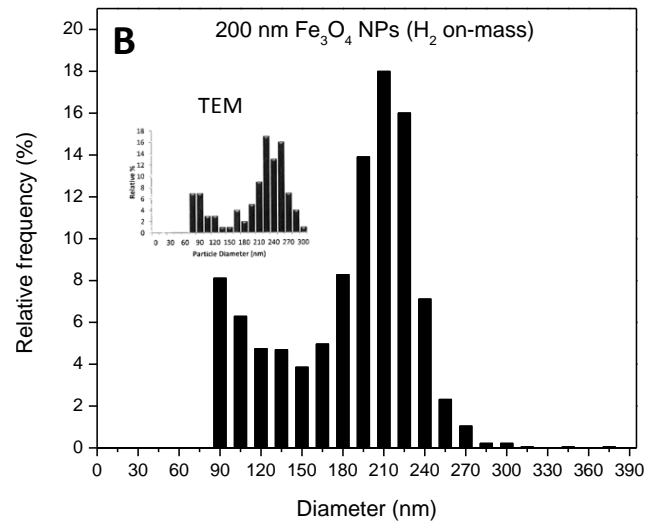
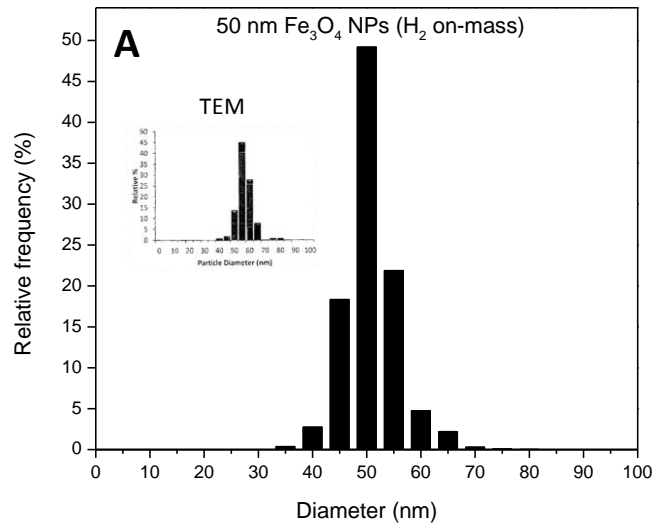
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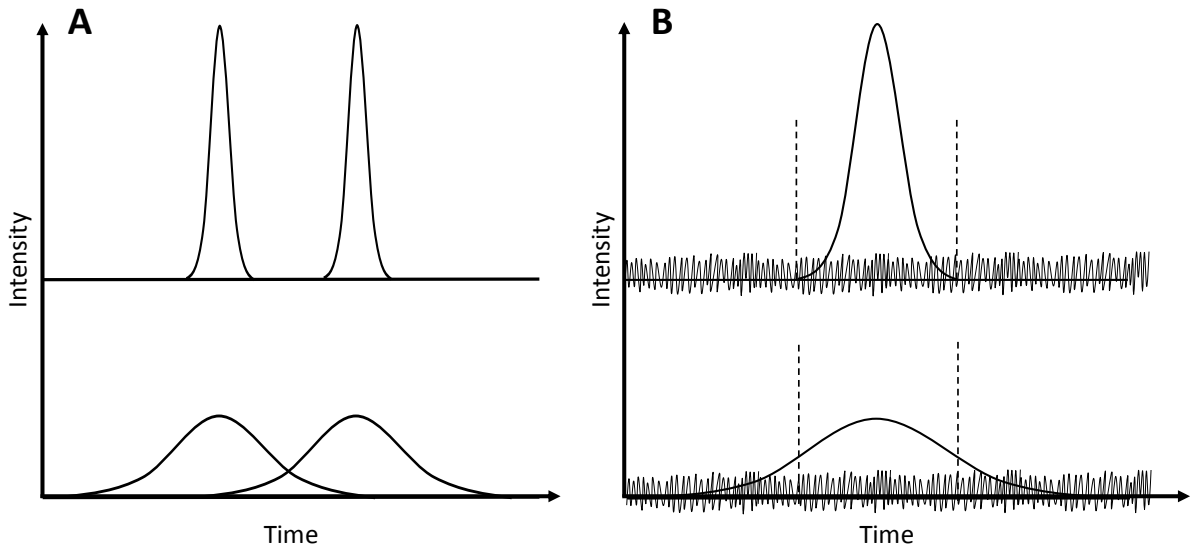
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1 **Fig. 10**



1 **Fig. 11**



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Table 1. Average peak width (signal duration) of individual NP events (signal spikes) at 1, 10 and 50% of the maximum height for the different CRC modes evaluated in this work.

| CRC mode | Gas flow rate (mL min ⁻¹) | 1% peak width | 10% peak width | 50% peak width |
|-----------------------------|--|---------------|----------------|----------------|
| | | Average ± SD | Average ± SD | Average ± SD |
| No gas | --- | 0.60 ± 0.10 | 0.40 ± 0.06 | 0.22 ± 0.03 |
| No gas* | --- | 0.62 ± 0.13 | 0.41 ± 0.08 | 0.23 ± 0.04 |
| H ₂ | 7.0 | 0.63 ± 0.10 | 0.46 ± 0.09 | 0.24 ± 0.05 |
| He | 5.0 | 0.74 ± 0.13 | 0.46 ± 0.08 | 0.24 ± 0.04 |
| NH ₃ on-mass | 2.3** | 2.59 ± 0.91 | 1.33 ± 0.79 | 0.49 ± 0.32 |
| NH ₃ mass-shift* | 2.3** | 1.84 ± 0.86 | 1.22 ± 0.61 | 0.56 ± 0.13 |

*100 nm AuNPs were measured instead of 50 nm AuNPs.

**In the case of "NH₃", a mixture of 10% NH₃ in He was used. Additionally, 1 mL min⁻¹ He was introduced automatically by the MassHunter software when the CRC is pressurized with NH₃/He.

Table 2. Probability of particle coincidence (P>1 – %) at different NP fluxes and peak widths calculated based on Poisson statistics.

| Peak width (ms) | NP flux (Hz) | | |
|-----------------|--------------|--------|-------|
| | 5 | 20 | 50 |
| 0.5 | 0.00031 | 0.0050 | 0.031 |
| 1.0 | 0.0012 | 0.020 | 0.12 |
| 2.5 | 0.0077 | 0.12 | 0.72 |
| 5.0 | 0.031 | 0.47 | 2.6 |